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Proceedings of the 2nd International Conference on Applied Biotechnology (ICAB 2014)-Volume II



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Tong-Cun Zhang · Motowo Nakajima Editors

Advances in Applied Biotechnology

Proceedings of the 2nd International Conference on Applied Biotechnology (ICAB 2014)-Volume II



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Preface

The 2014 International Conference on Applied Biotechnology (ICAB 2014), organized by Chinese Society of Biotechnology and Tianjin University of Science, was held from November 28 to 30, 2014 in Tianjin, China.

The conference served as a forum for exchange and dissemination of ideas and the latest findings in aspects of applied biotechnology. The conference was complemented by talks given by more than 30 professors and researchers.

The conference papers were submitted by more than 100 authors from different universities, institutes, and companies. Numerous fields were covered, ranging from fermentation engineering, cell engineering, genetic engineering, enzyme engineering, to protein engineering.

Special thanks are given to the Secretary Staff of the conference for the commitment to the conference organization. We would also like to thank all the authors who contributed with their papers to the success of the conference.

This book gathers a selection of the papers presented at the conference. It contains contributions from both academic and industrial researchers focusing on the research and development of applied biotechnology from all over the world. The scientific value of the papers also helps researchers in this field to get more valuable results.

Tianjin, China Tokyo, Japan Tong-Cun Zhang Motowo Nakajima

Contents

Part I Microbial Genetics and Breeding

1	Inheritance Analysis for Exserted Stigma Rate in Japonica Rice	3
2	Cloning and Expression of β-Glucosidase from Cassava in <i>Pichia pastoris</i> GS115	11
3	Construction of L-tert-Leucine Producing Strain by Expressing Heterologous Leucine Dehydrogenase and Formate Dehydrogenase in Escherichia coli	21
4	The Distribution Characteristics of Microcystis novacekii Based on 16S rDNA Sequence	31
5	The Distribution and Molecular Identification of the <i>Microcystis aeruginosa</i> in Yuqiao Reservoir Zhiyi Qiao, Da Huo, Jun Zhang, Jing Sun, Yunsi Nie and Yanan Wang	43
6	The Improved Stress Tolerance of Escherichia coli by Directed Evolution of IrrE	53

viii Contents

7	Cloning and Characterization of the <i>TMEPAI</i> Gene Promoter	67
8	Cloning and Expression of a Novel Xylanase Xyn11-1 from Alkaline Soil	75
9	To Establish the Regeneration System of Sweet Sorghum Immature Embryos	83
10	Cloning and Sequence Analysis of a Novel ACC Oxidase Gene from Peanut (Arachis hypogaea L.)	93
11	Genetic Diversity Among the Microorganisms in Daqu Used for Beidacang Liquor as Revealed by RAPD Analyses Shi-Wei Wang, Qing-Hui Wang, Li-Ping Zhai, Jun Liu, Zhi-Dan Yu, Mao-Mao Zheng and Fang Wang	99
12	Stenotrophomonas maltophilia Having Decolorization Capability of Azo Dye Isolated from Anaerobic Sludge	109
13	In Silico Cloning and Sequence Analysis of F3H Gene in <i>Raphanus sativus</i> L	117
14	The Comparative Study of Fermentation Capability of Wheat Beer Yeast	125
15	Screening of Dual Defects Strain and Effects on L-Isoleucine Production in Escherichia coli NML	135
16	Effect of MIG1 Gene Deletion on Lactose Utilization in Lac ⁺ Saccharomyces cerevisiae Engineering Strains Jing Zou, Xuewu Guo, Jian Dong, Cuiying Zhang and Dongguang Xiao	143

Contents ix

17	Enzyme Activity Analysis of Protease Produced by Marine Bacteria	153
18	Expression of Gene <i>uvrA</i> from <i>Acetobacter pasteurianus</i> and Its Tolerance to Acetic Acid in <i>Escherichia coli</i> Yu Zheng, Xingjing Chen, Jing Wang, Haisong Yin, Liqing Wang and Min Wang	163
19	Construction of Eschericha coli-Staphylococcus Shuttle Vector for EGFP Expression and Potential Secretion via Tat Pathway	171
20	SNP Affects the Mobility of Breast Cancer Cells and the Expression of Metastasis-Related Genes	181
21	Molecular Cloning and Characterization of Glycerol Dehydrogenase from Klebsiella pneumoniae Yanhua Liu, Li Zhao, Jianguo Zhang and Yu Zheng	189
22	Expression of Glucose-6-Phosphate Dehydrogenase and 6-Phosphogluconate Dehydrogenase Improve L-Citrulline Biosynthesis in argG-Deleted Corynebacterium glutamicum	197
Par	t II Optimization and Control of Biological Process	
23	Production of Amino Acids by Mixed Bacterial Strains-Mediated Solid State Fermentation of Feathers and Dynamic Changes to the Fermentation System	207
24	Effects of Culture Medium on PUFAs Production by Mortierella isabellinas	219

x Contents

25	A Simple Mechanochemical Cycle Model for Dynein	229
26	The Research on Biotransformation Pathway of Digitoxin by Aspergillus ochraceus and the Analysis of Products Activity	237
27	Astragalus membranaceus Polysaccharide-Enhanced Lymphocytes Proliferation of Yellow Drum Nibea albiflora In Vitro	249
28	Aspergillus niger Pellets Absorbed Bacillus sp. Isolated from Soybean Wastewater Sludge	255
29	Expression, Purification and Characterization of Maltase from "Quick" Baker's Yeast	265
30	Evaluation of an Ethanol-Tolerant Acetobacter pasteurianus Mutant Generated by a New Atmospheric and Room Temperature Plasma (ARTP)	277
31	A Thermotolerant Acetobacter pasteurianus T24 Achieving Acetic Acid Fermentation at High Temperature in Self-Adaption Experiment	287
32	Optimization of Cultural Conditions for Extracellular Polymeric Substances (EPS) Production by Burkholderia Using Response Surface Methodology	295
33	Effects of pH, Temperature, Storage Time, and Protective Agents on Nisin Antibacterial Stability	305

Contents xi

34	Enhanced Solvent-Stable Alpha Glycosidase Production by Bacillus licheniformis JXC-1 by Optimization of Feeding Strategies	313
35	The Effect of Different Activated Carbon and Bleaching Temperature on Kojic Acid Bleaching Wu Meng, Cuiying Zhang and Dongguang Xiao	325
36	Effect of Iodine on the Growth and Quality of Nostoc flagelliforme	335
37	The Replacement of Phe28 by Ser Enhances the Stability of the GLP-1 Analog During Fermentation	343
38	Optimization of Fermentation Condition of Man-Made Bee-Bread by Response Surface Methodology	353
39	Prediction of Lysine Acetylation Sites in Porcine Pancreas Lipase Modified by the Ionic Liquids Using Molecular Dynamics Simulations	365
40	Effect of Oxygen on Fermentation Characteristics of Three Non-Saccharomyces from Hengshui Laobaigan	381
41	Microbial Transformation of Antitumor Isatin Derivatives by Fungi	391
42	Effect of Ultrasound on Lysine Muriate Crystallization	397

xii Contents

43	Isolation and Identification of a Bacillus amyloliquefaciens Strain Against Grape Downy Mildew and Optimization for Its Liquid Fermentation Medium	409
44	The Adsorption Properties of Macroporous Resin for Fusel Oil of Luzhou—Flavor Liquor	419
45	Low Labeling ¹³ C Metabolic Flux Analysis of Saccharomyces cerevisiae Using Gas Chromatography-Combustion-Isotope Ratio Mass Spectrometry	427
46	A Comparative Study on the Antioxidant Activity of Two Polysaccharides from <i>Ganoderma lucidum</i> Ruyu Tao, Limin Hao, Shiru Jia, Xin Zheng, Jianyong Yu and Qingwu Jiang	441
47	Biofortification Using Bacteria Containing an Atrazine-Degrading Gene and Its Effects on Reactor Operating Efficiency	451
48	A Novel One-Pot Five-Component Synthesis of Tetrahydro-pyrrolo[3,4-b]pyridine-5-one via Ugi/Aza-Diels-Alder Tandem Reaction Yan Liu, Tianyi Shang, Chuanming Xu, Hui Yang, Peng Yu and Kui Lu	461
49	Study on Preparation and Application of High Esterification Green-Liquor Daqu	467
50	Metabolomics Analysis Between Wild-Type and Industrial Strains of Streptomyces avermitilis Based on Gas Chromatography—Mass Spectrometry Strategy	477

Contents xiii

51	Application of Orthogonal Design to Optimize Fermentation Conditions of Bacillus amyloliquefaciens BI ₂	487
52	Optimization of Fermentation Medium for Citric Acid Production by Aspergillus niger	497
53	Study on the Soxhlet's Extraction of Star Anise Oil and Preliminary Investigation of Its Antibacterial Activity Lin Tian and Ping Li	509
Par	t III Biological Separation and Biological Purification	
54	Production of Diosgenin from Dioscorea zingiberensis by Mixed Culture of Three Filamentous Fungi Hua Xiao, Linlin Huang, Jinxia Xie, Songtao Bie and Yu Li	521
55	Characterization of Volatile Constituents of Chinese Hawthorn (Crataegus spp.) Fruit Juices	533
56	Optimization of Extraction Conditions for Crude Antibacterial Proteins/Peptides from Clarias gariepinus By-products Yan Wang, Yunxia Xu, Junyan Mei, Chengxun Chen and Xiaomei Wang	547
57	Screening, Isolation, and Identification of Bacillus coagulans C2 in Pu'er Tea	557
58	Production and Identification of Antifungal Compounds Produced by Bacillus subtilis B579	563

xiv Contents

59	Application of Molecularly Imprinted Polymers in Purification and Separation for Epothilones	571
60	Identification of Peanut Intersectional Hybrids with SSR Markers	581
61	Partial Purification and Chemical Characterization of a Bioemulsifier and Its Application in MEOR	587
62	Optimization of a Whey Containing Medium for β-Galactosidase Production by Lactobacillus reuteri	599
63	Enzymatic Bioconversion for γ-Aminobutyric Acid by Lactobacillus brevis CGMCC No. 3414 Resting Cells Xiu-feng Shi, Bo Zheng, Chuan-you Chang, Peng Cao, Hong-jin Yang and Qiang Gao	609
Par	rt IV Progress of Biotechnology	
64	[FeFe]-Hydrogenase: Catalytic Center and Modification by Genetic Engineering	621
65	Characteristics of Staphylococcus aureus Isolates from Raw Milk	629
66	EST-SSR Marker-Based Assay for Purity Identification of Melon "Green Angle"	637

Contents xv

67	Exposure to Static Magnetic Fields Affects Insulin Secretion	
	in INS Cells	643
	Libin Mao, Zhixia Guo, Huiqin Wang, Qiongyao Wu,	
	Nan Wang and Tong-Cun Zhang	
68	The New Strategy of Breeding Cytidine Excessive	
	Biosynthesis Mutants by pyr Operon Rearrangement	
	of Bacillus amyloliquefaciens	649
	Qing Wu, Huiyan Liu, Haitian Fang, Jianguo He,	
	Xiaoguang He and Linan Yu	

Part I Microbial Genetics and Breeding

Chapter 1 Inheritance Analysis for Exserted Stigma Rate in *Japonica* Rice

Ruhua Wang, Na Yan, Xueyan Gao, Jie Yu, Fang Wang and Zetian Hua

Abstract Rice is the main crop in China. Since the discovery and exploitation of sterile strain line, three lines of *Indica* hybrid rice were bred successfully; the *Indica* hybrid rice has made great progress in production. Compared to *Indica* rice. Japonica hybrid rice is hard to study because it lacks the restorer gene. The key to promoting development of hybrid Japonica rice is to improve the yields of seed production. However, the low stigma exsertion rate of Japonica CMS Line is the main cause that restricts the yield of hybrid seed production of *Japonica* rice. As the inheritance mechanism of the stigma exsertion is complicated and difficult to study, the related study is far from enough. In this study, Jinke DS, a sterile line with higher stigma exsertion rate, and C9083, a restorer line with lower stigma exsertion rate, were used as parents for hybrid, P_1 , P_2 , F_1 and F_2 were chosen as the study materials and major gene plus polygene mixed inheritance model was chosen to analyze the genetic effects. The results showed that glume-opening single exserted stigma rate was in consistence with the E-2 model, glume-closed single exserted stigma rate was in consistence with the E-0 model, glume-opening dual exserted stigma rate, glume-opening exserted stigma rate, glume-closed dual exerted stigma rate and glume-closed exserted stigma rate were in consistence with the E-1 model.

Keywords Rice · Exserted stigma rate · Inheritance analysis

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4 R. Wang et al.

1.1 Introduction

The key to promoting development of hybrid *Japonica* rice is to improve the yields of seed production. But the low stigma exsertion rate of *Japonica* CMS Line is the main cause that restricts the yield of hybrid seed production of *Japonica* rice.

The exsertion of the stigma can increase the chances of pollination. Especially for the closed glumes, it is a good factor to overcome the barrier of pollination when the parents' flowering periods are different [1]. The average seed set of stigma exserted spikelet is 3–4 times higher than that of stigma non-exserted spikelet, and occupies 70–80 % of the outcrossing rate [2, 3]. As a matter of fact, it is significant to improve the stigma exsertion rate for hybrid rice seed production. Although the stigma exsertion rate is related with the ambient conditions, such as cultivation techniques, exogenous hormones, and so on, it is determined by the genetic factor, the variety itself [4, 5]. As the inheritance mechanism of the stigma exsertion is complicated and difficult to study, the related study is far from enough. In this study, we used mixed genetic model between major gene and multiple minor genes and segregation analysis to study the inheritance mechanism of stigma exsertion to explore the effects and relative importance of the major gene and multiple genes. We hope that this study can provide some theoretical basis for improving this trait.

1.2 Materials and Methods

1.2.1 Experimental Materials

 F_1 generation was obtained by crossing Jinke DS with C9083, which is a male sterile line with high percentage of exserted sigma and a restorer line with low percentage of exserted sigma, respectively. F_2 generation was obtained from F_1 generation plants selfing. Jinke DS was photo-thermo sensitive male sterile line japonica rice cultivated by Tianjin University of Science and Technology, with 87 % of exserted sigma rate. C9083 was restorer line rice cultivated by China National Japonica Rice R&D Center, with 3 % of exserted sigma rate.

1.2.2 Field Experiment and Character Investigation

 P_1 , P_2 , F_1 and F_2 groups were cultivated in experiment fields of Tianjin University of science and technology. There were 30 plants in each P_1 , P_2 and F_1 groups and 900 plants in F_2 group. The row and individual spacing was 30 cm \times 20 cm. The field management was the same as field production. 10 plants in P_1 , P_2 and F_1 groups, as well as 204 plants in F_2 group were randomly marked, with 3 spikes marked each. Spikelet quantity, single exserted stigma quantity, and dual exserted

stigma quantity were investigated by the early floret opening period. Glume-closed single exserted stigma quantity, dual exserted stigma, and spikelet quantity were investigated by the ending floret opening period. The percentage of exserted sigma was calculated with the formula:

$$\begin{aligned} \text{single exserted stigma rate} &= \frac{\text{single exserted stigma quantity}}{\text{spikelet quantity}} \times 100\,\% \\ \\ \text{dual exserted stigma rate} &= \frac{\text{dual exserted stigma quantity}}{\text{spikelet quantity}} \times 100\,\% \end{aligned}$$

exserted stigma rate = single exserted stigma rate + dual exserted stigma rate.

1.2.3 Genetic Analysis

The mixed genetic model between major genes and multiple minor genes [6] was applied to analyze the percentage of exserted stigma of P_1 , P_2 , F_1 and F_2 groups. Iterative algorithm ECM (Iterate expectation and conditional maximization) was applied to evaluate distribution parameters of each generation. Then AIC (Akaike's information criterion) was applied to screen the relatively optimum model. Fit tests, including homogeneity test $(U_1^2,\,U_2^2,\,U_3^2)$, Smirnov test (nW^2) , and Kolmogorov test (D_n) were then operated to select the optimum model. QTL analysis system of four generation method of major gene plus polygene and the basic symbol and parameters based on presupposition of five categories (A, B, C, D and E), totally 22 genetic models were referred to Gai and Wang [7] and Zhang [8]. Analysis software of Major gene plus polygene mixed genetic model were provided by Prof. Yuanming Zhang of Nanjing Agricultural University.

1.3 Results and Discussions

1.3.1 Character Distribution of Parents and Each Generation

C9083 was characterized as low percentage of exserted stigma (3.5 %). Jinke DS was characterized as high percentage of exserted stigma (87 %). Percentage of exserted stigma between 3.15 and 92.3 % in 204 F_2 individual plants showed a continuous distribution (Table 1.1). As shown in Table 1.1, there was a distinct peak between 30 and 45 % of exserted stigma, which showed a character of quantitative trait.

Table 1.1 Frequency distribution of trail in parents and population

Celleration	Percentage	e of glume-c	age of glume-closed exserted stigma	ed stigma					\sum_{f}	ļ	25
	0-10	10-20	10-20 20-30 30-40		40–50	90-09	40–50 50–60 60–70 70–100	70–100	Ì	۲.	
P_1^a	10								10	6.51	1.007
$\mathbf{P}_2^{\mathrm{b}}$							2	<u>«</u>	10	74.7	15.029
F_1			7	3					10	24.37	10.927
F ₂	18	41	69	34	21	12	7	2	204	204 22.31	156.89

^a P₁: C9083 ^b P₂: Jinke DS

1.3.2 Genetic Model

Four generation method of major gene plus polygene mixed inheritance model was applied to analyze the percentage of exserted stigma in four generations of crossing C9083 and Jinke DS. Maximum likelihood function values of a pair of major gene (A), two pairs of major gene (B), polygene (C), a pair of major gene plus polygene (D), and two pairs of major gene plus polygene (E); totally five categories of 22 kinds of genetic models were obtained. The AIC values of total percentage of exserted stigma are shown in Table 1.2. According to genetic model selection principle, minimum AIC value principle, five models with the minimum AIC values were taking the fit tests. Models with least number of statistics with significant meanings were chosen as the optimum model. Statistics $\left(U_1^2,\,U_2^2,\,U_3^2,\,nW^2\right)>0.05$ was considered as not significant. The results showed that the optimum model for glume-closed exserted stigma rate was mixed genetic model E-1 between two additive-dominance major genes and additive-dominance multiple minor genes (Table 1.3). Optimum models were selected with this method.

Table 1.2 The max-likelihood value and Akaike's information criterion values under various genetic models

Model	Max-likelihood-value	AIC
A-1	-927.14	1880.29
A-2	-963.36	1936.73
A-3	-944.46	1898.92
A-4	-948.65	1967.31
B-1	-879.61	1781.23
B-2	-894.32	1804.64
B-3	-948.15	1906.30
B-4	-944.15	1904.30
B-5	-919.01	1848.02
B-6	-922.27	1852.53
C-0	-878.55	1769.11
C-1	-913.19	1846.37
D-0	-849.06	1754.12
D-1	-899.76	1813.53
D-2	-913.19	1848.48
D-3	-913.19	1848.32
D-4	-913.18	1848.37
E-0	-859.34	1745.38
E-1	-859.80	1751.63
E-2	-891.56	1793.12
E-3	-913.20	1832.44
E-4	-913.20	1823.43

Table 1.3 Tests for fitness about some traits of exserted stigma

*;···E	Model	2019	211	211	112	L1172	٥
Itali	Iapolvi	Generation	U <u>ī</u>	Už	U ₃	II w	μn
Glume-opening single exserted stigma rate	E-2	P_1	0.073 (0.7868)	0.089 (0.7660)	5.008 (0.0252)	0.1633	0.2626
		\mathbf{P}_2	0.349 (0.5545)	0.064 (0.7999)	1.627 (0.2021)	0.1586	0.2828
		F_1	0.162 (0.6876)	1.402 (0.2364)	10.100 (0.0015)	0.2096	0.2644
		F_2	0.017 (0.8971)	0.008 (0.9271)	0.018 (0.8926)	0.0334	0.0469
Glume-opening dual exserted stigma rate	E-1	P_1	0.269 (0.6037)	0.128 (0.7207)	11.836 (0.0006)	0.8558	0.5474
		\mathbf{P}_2	0.963 (0.3265)	0.242 (0.6230)	3.363 (0.0667)	0.2871	0.3675
		F_1	0.246 (0.6200)	0.342 (0.5587)	18.144 (0.0000)	0.3448	0.3305
		F_2	0.799 (0.3714)	1.249 (0.2638)	1.016 (0.3134)	0.9200	0.1721
Glume-opening exserted stigma rate	E-1	P_1	0.013 (0.9092)	0.191 (0.6617)	1.712 (0.1907)	0.0622	0.1659
		\mathbf{P}_2	0.009 (0.9257)	0.005 (0.9451)	0.007 (0.9315)	0.0487	0.2016
		F_1	0.002 (0.9669)	0.025 (0.8742)	0.631 (0.4270)	0.0306	0.1653
		F_2	0.001 (0.9749)	0.000 (0.9885)	0.004 (0.9489)	0.0192	0.0284
Glume-closed single exserted stigma rate	E-0	P_1	0.000 (0.9895)	0.506 (0.4768)	8.392 (0.0038)	0.3033	0.3517
		\mathbf{P}_2	0.078 (0.7794)	0.250 (0.6168)	0.841 (0.3592)	0.0493	0.1470
		F_1	0.209 (0.6479)	0.007 (0.9349)	2.080 (0.1493)	0.1691	0.3084
		F2	0.000 (0.9849)	0.000 (0.9862)	0.000 (0.9966)	9800.0	0.0211
Glume-closed dual exerted stigma rate	E-1	P_1	0.952 (0.3292)	3.043 (0.0811)	10.233 (0.0014)	0.9127	0.5891
		\mathbf{P}_2	0.886 (0.3466)	0.927 (0.3356)	0.043 (0.8366)	0.1190	0.2627
		\mathbf{F}_1	0.006 (0.9379)	0.991 (0.3195)	13.542 (0.0002)	0.2471	0.2861
		F2	0.186 (0.6665)	3.592 (0.0580)	34.959 (0.0000)	4.9490	0.3355
Glume-closed exserted stigma rate	E-1	P_1	0.015 (0.9010)	0.346 (0.5565)	8.031 (0.0046)	0.2885	0.3356
		\mathbf{P}_2	0.158 (0.6912)	0.461 (0.4972)	1.386 (0.2390)	0.0958	0.2058
		\mathbf{F}_1	0.059 (0.8088)	0.169 (0.6810)	0.500 (0.4793)	0.0548	0.1739
		F_2	0.015 (0.9018)	0.010 (0.9215)	0.007 (0.9335)	0.0166	0.0303

1.4 Conclusion

In this study, we adopted the major gene plus poly gene mixed genetic model to analyze the characters of exserted stigma. Separately, two pairs of major genes were detected in six characters of exserted stigma, performing higher frequencies of heritability. In case of the target characters of loss and recession from generation to generation, selection should be done in the early generations to obtain the materials with stable higher percentage of exserted stigma. Related genes of high percentage of exserted stigma should be induced into varieties with good comprehensive characters to get more new varieties with higher yields, quality, and resistance in practice.

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Chapter 2 Cloning and Expression of β-Glucosidase from Cassava in *Pichia pastoris* GS115

Dongheng Guo, Hongming Tian, Yanshan Xu and Suiping Zheng

Abstract β-Glucosidases have been widely applied in the synthesis of alkyl polyglucosides. Studies have shown that cassava β-glucosidase has great synthesis ability. This study has realized the secretion of cassava β-glucosidase gene (mebgl) in Pichia pastoris GS115. The recombinant yeast expression vector pPICZaA-MEBGL has been successfully constructed. The constructed plasmid was linearized and integrated into P. pastoris GS115 strain by electroporation. Positive clones were selected on YPDZ plates and then cultured in shake flask. The supernatant was collected and used for alkyl polyglucoside synthesis. The synthesis reaction was conducted under conditions: 40 °C, pH 5.0, 10 % water content, while the conditions have not been optimized. The results of these studies have indicated that 5'AOX promoter was the best one for expression; in the shake flask fermentation, the strain had the maximum hydrolysis activity of 60 U/L, while the optimal temperature was 35 °C, the optimal pH was 6.0, pNP-Glc was used as substrate; the molecular mass of the recombinant monomer protein was estimated to be 70 kDa; the recombinant protein showed high ability to transfer glucose from pNP-Glc to n-hexyl alcohol with high yields of 60 %, and the yields would substantially increase after optimized. This recombinant cassava β-glucosidase is expected to promote the industrialization process of alkyl polyglucoside enzymatic synthesis.

Keywords Cassava β-glucosidase • *Pichia pastoris* GS115 • Secretory expression • Enzymology properties

2.1 Introduction

β-Glucosidases (E.C. 3.2.1.21) catalyze the hydrolysis of β-D-glucosidic linkages between β-D-glucose and aglycone or sugar. β-Glucosidases exhibit similar specificity for β-glucoside substrates, have a similar molecular weight (about 55–65 kDa)

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D. Guo et al.

and pH optima (between pH 5 and 6) [1], and are evolutionarily related. However, they may have different substrate specificities, suggesting their diverse biological functions. According to the classification of glycosyl hydrolases based on amino acid sequence similarities [2], β -glucosidases belong to GH1, GH3, GH5, GH9, GH30, GH116 families. A large proportion of β -glucosidases belong to glycoside hydrolase families 1 and 3 (GH1 and GH3), and hydrolyze their substrates via double-displacement mechanism [3]. The crystal structures of β -glucosidases belonging to GH1 family have been reported. But to GH3 family, only four crystal structures have been reported. β -Glucosidases that belong to GH1 and GH3 family can catalyze reverse hydrolysis and transglycosylation reactions, leading to synthesis of oligosaccharides and alkyl polyglucosides.

Cassava \(\beta\)-glucosidase belongs to GH1 family, it can transfer a nonreducing glucosyl group from a glycoside or carbohydrate to water (hydrolysis) or another alcohol (transglycosylation) via retaining mechanism. Cassava β-glucosidase possess $(\beta/\alpha)_8$ barrel structure. It has two carboxylic acid residues on β -strands 4 and 7, which act as the general acid-base catalyst and the nucleophile, respectively [4]. Site-directed mutagenesis is carried out to study the function of active-site amino acid residues based on a homology model, while the MODELLER program is used [5]. The enzyme activity is destroyed when Glu-413 is changed to Gly, which is consistent with that it being the catalytic nucleophile. The Gln-339/Glu mutation also destroys activity, confirming a function in positioning the catalytic diad. The Phe-269 contributes to the cyanogenic specificity of the cassava β-glucosidase. The recombinant cassava β-glucosidase is successfully expressed in Saccharomyces cerevisiae. Its Michaelis constants for the natural substrate linamarin ($K_{\rm m} = 1.06$ mM) and the synthetic *p*-nitrophenyl β -D-glucopyranoside (pNP-Glc; $K_{\rm m} = 0.36$ mM) are very similar to the plant enzyme. Linamarase gene is intracellular expressed in *Pichia pastoris* GS115 [6]. SDS/PAGE analysis shows that the molecular weight of recombinant protein is around 71 kDa. The optimal temperature of recombinant protein is about 37 °C, and the optimal pH is about 5. The $K_{\rm m}$ is 1.70 mmol/L, and $V_{\rm max}$ is 8.36 μ mol/(min mg), while pNP-Glc is used as the substrate.

The cassava β -glucosidase has very low hydrolysis activity, but shows high synthesis ability [7]. The alkyl polyglucosidase synthesis ability of β -glucosidases from cassava, Thai rosewood, and almond has been compared. Cassava β -glucosidase shows greater ability to transfer glucose from pNP-Glc to secondary alcohol acceptors than other β -glucosidases, and it is the only one to synthesize tertiary alkyl β -glucosides with high yields. However, it could not catalyze the synthesis of alkyl polyglucosides through reverse hydrolysis reaction. Cassava β -glucosidase requires active glycosyl donors and could not use mono- or disaccharides as sugar donors in alkyl polyglucoside synthesis.

The *P. pastoris* expression system is widely used as eukaryotic expression system [8, 9]. The system has many advantages, such as relatively simple operation, fast growth, high expression, low miscellaneous proteins, postprocessing, protein

folding, and posttranslational modification. And the AOX1 promoter can ensure a high expression of exogenous genes with methanol as the sole carbon and energy source.

In this study, we constructed the recombinant expression vector pPICZ α A-MEBGL. The constructed plasmid was integrated into *P. pastoris* GS115 strain. After screened by nutrient deficiencies plate and shake flask fermentation, we got the positive recombinant strains. And we carried out preliminary enzymatic properties study on the recombinant cassava β -glucosidase, such as hydrolysis and synthesis ability.

2.2 Materials and Methods

2.2.1 Materials, Strains, and Media

Escherichia coli Top10FTM was used for the amplification of all plasmids. The *P. pastoris* GS115 strains were used as host strains for β-glucosidase expression. The cloning and expression vector pPICZαA was purchased from Invitrogen.

LB culture medium: 1 % tryprone, 0.5 % yeast extract, 1 % NaCl. Zeocin was added to the medium when the medium was used for screening. In the medium, the concentration of zeocin was 25 μ g/mL. YPD, BMGY, BMMY reference Invitrogen company operating manuals. YPDZ medium, which the concentration of zeocin was 100 μ g/mL, was used for screening.

2.2.2 PCR of Cassava β-Glucosidase Gene

The Genebank ID of *mebgl* gene was gi:249261. The gene was synthesized after codon optimization. The oligonucleotide primers contain 5' *Eco*RI and 3' *Kpn*I restriction endonuclease sites. Upstream primer contains *Eco*RI restriction endonuclease site (outlined by underline), "His" label and protected bases: 5'CAGGAATTCCATCATCACCATCACCATACAGATGACGATGACGACAA 3'. Downstream primer contains *Kpn*I restriction endonuclease site and protected bases: 5'CGGGGTACCCTACATAACGTAAAACTTTC3'. The PCR amplification of *mebgl* gene was conducted under conditions: 2 μL 100 ng/μL template, 25 μL 2× KOD FX Buffer, 10 μL 2 mmol dNTPs, 2.5 μL 10 μmol/L each primer, 1 μL KOD FX DNA polymerase, 9.5 μL ddH₂O; initial denaturation at 94 °C for 5 min, 30 cycles (10 s at 94 °C, 30 s at 55 °C, and 96 s at 68 °C) were followed by a final incubation at 68 °C for 10 min.

D. Guo et al.

2.2.3 Cloning and Construction of the Yeast Expression Vector

The PCR products were digested with EcoRI and KpnI, gel purified, and subsequently inserted into the EcoRI–KpnI sites in pPICZ α A, giving pPICZ α A-MEBGL. The recombinant plasmid was transformed into $E.\ coli$ Top10FTM via CaCl2 transformation method. Single colonies of $E.\ coli$ Top10FTM that contained pPICZ α A-MEBGL were grown in LB resistance screening medium, at 37 °C, 200 rpm for 14–16 h. The recombinant plasmid was extracted, confirmed by restriction digestion, and DNA sequencing.

2.2.4 Yeast Transformation and Screening

Electrotransformation was carried out with the Bio-Rad Gene Pulser system (Bio-Rad, Richmond, CA, U.S.A.) according to the manufacturer's instructions, 0.1 μg linearized plasmid was used. After that, the GS115 cells were coated in YPDZ plates. The plates were cultured at 30 °C for 3 days. Single colonies of *P. pastoris* GS115 were grown in YPDZ medium, and confirmed by colony PCR.

2.2.5 Expression of Cassava β-Glucosidase in Pichia pastoris GS115

The strains were cultured at BMGY, at 30 °C, 200 rpm for 20–24 h. Then the culture was transformed into BMMY which contained 2 % (v/v) methanol. The initial OD $_{600}$ was controlled at one. After that, the strains were cultured at BMMY for 144 h, at 30 °C, 200 rpm. 2 % (v/v) fermented liquid was extracted and 2 % (v/v) methanol was added every 24 h for the induction of the recombinant proteins.

2.2.6 Purification of Recombinant Cassava β-Glucosidase

The fermentation liquor was collected after the strains were cultured for 144 h. The supernatant was used for purification after filtered by 0.22 μm filter membrane. Then the supernatant was desalted through desalting crude (HiTrapTM). After that, the supernatant was used for Ni-chelating affinity chromatography (HisTrapTMFF). The recombinant protein was eluted by 10, 20, 50, 100, 200, 300 mM imidazole. Then the eluent was concentrated through 10 kDa filter membrane bag. The recombinant protein was used for further analysis.

2.2.7 SDS-PAGE Analysis

SDS-PAGE electrophoresis was performed on 6 % spacer polyacrylamide gel and 12 % (w/v) separation polyacrylamide gel. After electrophoresis, proteins were stained in 0.05 % (w/v) Coomassie Brilliant Blue R250 (Sigma-Aldrich) solution.

2.2.8 The Analysis of Hydrolysis and Transglycosylation Activities

The hydrolysis activity of cassava β -glucosidase was assayed with colorimetric method while pNP-Glc was used as substrate. The reaction was conducted under conditions: 250 μ L cassava β -glucosidase liquid and 250 μ L 5 mM pNP-Glc were preheated at 40 °C for 5 min, respectively, then they were mixed for reaction at 40 °C for 30 min, 0.2 M sodium acetate (pH 5.0) buffer solution was used. The reaction was stopped by adding 500 μ L 1 M sodium carbonate, and the p-nitrophenol released was measured by its absorbance at 405 nm. One unit of activity is defined as the amount of enzyme used to release 1 μ mol of p-nitrophenol in 1 min.

The temperature optimum was determined by measuring p-nitrophenol released from pNP-Glc at temperatures ranging from 10 to 70 °C at 10 °C increments, and the assay was completed as described above. The pH optimum was determined by hydrolysis of pNP-Glc at 40 °C for 30 min in the following 0.2 M buffers: glycine–HCl, pH 2–3; sodium citrate, pH 3–4; sodium acetate, pH 4–5; MES, pH 5–6; sodium phosphate, pH 6–8; Tris–HCl, pH 8–9; and CAPS, pH 9–10.

The preliminary experiments of transglycosylation reaction were conducted under conditions: 30 mM pNP-Glc; 10 % (v/v) 0.2 M pH 5.0 sodium acetate (containing recombinant cassava β -glucosidase); 90 % (v/v) n-hexyl alcohol, at 40 °C for various times (from 1 to 6 h) before analysis by HPLC.

Waters 2695, Waters 2424 ELS Detector and C18 chromatographic column were used for HPLC analysis. Samples were filtered by 0.22 μ m filter membrane before detected. The mobile phase was methol–water mixture (80:20, v/v). The detection was conducted under conditions: 1.0 mL/min flow rate, 30 °C for C18 column, 60 °C for drift tube, 30 psi for nitrogen pressure.

2.3 Results and Discussions

2.3.1 Construction of Recombinant Expression Vector pPICZaA-MEBGL

The recombinant plasmid pPICZαA-MEBGL (Fig. 2.1) was constructed by the products of PCR based on the method as described by Sect. 2.2.3. Three different recombinant plasmids were constructed, which contain GAP, modified AOX and

D. Guo et al.

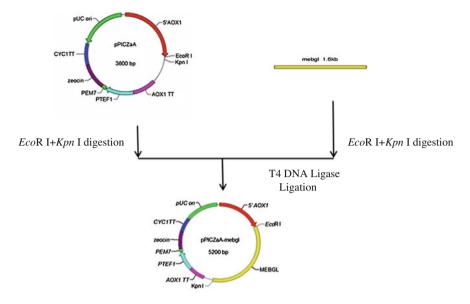


Fig. 2.1 Sketch of construction of recombinant expression plasmid pPICZαA-MEBGL

5'AOX promoter, respectively. The recombinant plasmids were confirmed by restriction digestions (Fig. 2.2a) and DNA sequencing. The targeted band was between 1,500 and 2,250 bp.

2.3.2 Identification of Recombinant Pichia pastoris GS115/pPICZαA-MEBGL

Single colonies of *P. pastoris* GS115 were grown in YPDZ medium, and then confirmed by colony PCR (Fig. 2.2b). The results showed that most single colonies were positive clone. The positive clones were used for further screening.

2.3.3 The Screening of Recombinant Pichia pastoris GS115

The recombinant and control strains were cultured at the conditions as described by Sect. 2.2.5. There were no difference between their growth situations, but the recombinant strains show better hydrolytic activities, 60 U/L after been induced for 144 h (Fig. 2.3). The hydrolysis activities of recombinant strains were increased along with the induction time, but the control strains were decreased. And among GAP, modified AOX and 5'AOX, 5'AOX promoter was the best one for expression.

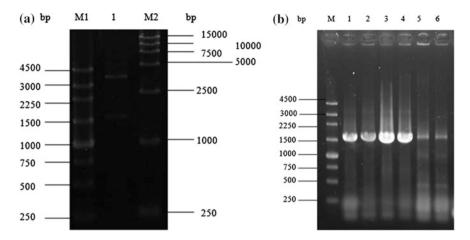


Fig. 2.2 a Identification of recombinant plasmid pPICZαA-MEBGL. *M1* 250 bp DNA marker; *I* Double restriction of pPICZαA-MEBGL by *Eco*RI and *Kpn*I; *M2* DL15000 DNA marker. **b** Identification of recombinant *Pichia pastoris* GS115/pPICZαA-MEBGL. *M1* 250 bp DNA marker; *1*, 2, 3, 4, 5, 6 recombinant *P. pastoris* GS115

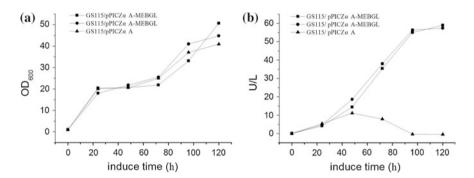


Fig. 2.3 The culture of recombinant *Pichia pastoris* GS12. **a** OD₆₀₀ of recombinant *P. pastoris* GS115/pPICZαA-MEBGL (GS115/pPICZαA as control); **b** supernatant hydrolytic activities of recombinant *P. pastoris* GS115/pPICZαA-MEBGL (GS115/pPICZαA as control)

2.3.4 Analysis of SDS-PAGE

The supernatant of the recombinant and control strains were applied for SDS-PAGE analysis. The molecular mass of the recombinant monomer protein was estimated to be 70 kDa (Fig. 2.4). This is consistent with literature reports that the recombinant protein was 71 kDa [6].

D. Guo et al.

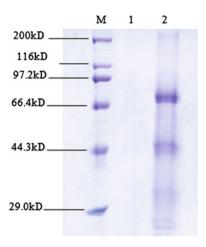


Fig. 2.4 SDS-PAGE analysis of the recombinant cassava β-glucosidase expressed in *Pichia pastoris* GS115 (GS115/pPICZαA as control); *M* protein marker; *I* supernatant of GS115/pPICZαA; 2 supernatant of *P. pastoris* GS115/pPICZαA-MEBGL

2.3.5 Optimization of Temperature and pH for Hydrolytic Activity of Recombinant Cassava β-Glucosidase

The recombinant cassava β -glucosidase was sensitive to temperature and pH values. As shown in Fig. 2.5a, when the temperature increased from 10 to 35 °C, the activity increased gradually. The maximum activity was achieved at 35 °C. If the temperature continued to rise, the activity decreased seriously. At 60 °C, the activity only retained 14.0 % of maximal activity.

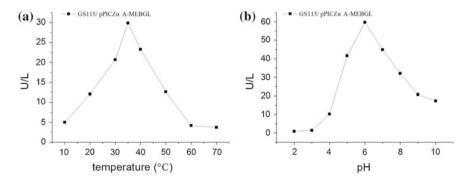


Fig. 2.5 Optimization of temperature and pH for hydrolytic activity of recombinant cassava β -glucosidase. **a** Optimization of temperature for hydrolytic activity of recombinant cassava β -glucosidase. **b** Optimization of pH for hydrolytic activity of recombinant cassava β -glucosidase

As shown in Fig. 2.5b, the recombinant β -glucosidase had an optimal pH of 6. At pH 5.0 and pH 7.0 the activity kept about 60–70 % of the maximum activity. There were only 17.1 and 29.0 % of maximal activity, when the pH values were 4.0 and 10.0.

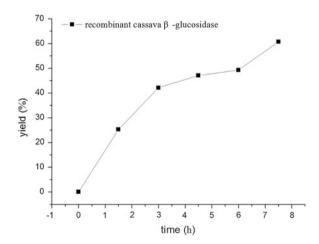
2.3.6 Purification of Recombinant Cassava β-Glucosidase

The fermentation liquor was desalted through desalting crude, then the supernatant was used for Ni-chelating affinity chromatography. The results indicated that most of the proteins could not bind to nickel column, a few proteins were eluted by 10 mM imidazole, and no proteins were eluted by other concentrations of imidazole. The supernatants that containing unbound proteins were concentrated through 10 kDa filter membrane bag, so were the eluents. Both of the concentrated solutions were used for SDS-PAGE analysis and transglycosylation reactions. The results of SDS-PAGE indicated that most of recombinant cassava β -glucosidase could not bind to nickel column. The concentrated cassava β -glucosidase was used for the synthesis of alkyl polyglycoside.

2.3.7 The Preliminary Experiment: Transglycosylation of Recombinant Cassava β-Glucosidase

A certain amount of cassava β -glucosidase, the hydrolysis activity was 0.1 U, was used for transglycosylation reaction. The preliminary experiment of transglycosylation reaction of cassava β -glucosidase showed great alkyl polyglycoside synthesis ability (Fig. 2.6). This was consistent with literature reports that cassava β -glucosidase showed better alkyl polyglycoside synthesis ability than Thai rosewood and

Fig. 2.6 The synthesis of n-hexyl β-D-glucopyranoside through transglycosylation, while pNP-Glc was used as the substrate



D. Guo et al.

almond β -glucosidases [7]. The cassava β -glucosidase could synthesize tertiary alkyl polyglycosides, while the other two β -glucosidases could not. Cassava β -glucosidase was important to the enzymatic synthesis of alkyl polyglycoside for its high synthesis ability.

2.4 Conclusion

This study realized the expression of β -glucosidase gene (mebgl) from cassava in *P. pastoris* GS115. SDS/PAGE analysis showed that the molecular mass of the recombinant monomer protein was around 70 kDa. The optimal temperature of this enzyme was about 35 °C, and the optimal pH was about 6.0, the maximum supernatant hydrolytic activity was 60 U/L, while pNP-Glc was used as the substrate. Study had shown that cassava β -glucosidase shows greater ability to transfer glucose from pNP-Glc to primary and secondary alcohol acceptors than Thai rosewood and almond β -glucosidases. And it is the only one β -glucosidase that is able to synthesize tertiary alkyl β -glucosides. This β -glucosidase is expected to promote the industrialization process of alkyl glucoside enzymatic synthesis.

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Chapter 3 Construction of L-tert-Leucine Producing Strain by Expressing Heterologous Leucine Dehydrogenase and Formate Dehydrogenase in Escherichia coli

Junzhen Bai, Yajian Song, Xuegang Luo, Haixu Yang, Wen Du and Tongcun Zhang

Abstract L-tert-Leucine is an unnatural amino acid that is a key intermediate for the synthesis of several important drugs. The L-tert-Leucine synthesis can be performed continuously by the collaboration of leucine dehydrogenase and formate dehydrogenase. In this study, recombinant strains of Escherichia coli expressing leucine dehydrogenase (LeuDH) and formate dehydrogenase (FDH), respectively, and the strain co-expressing the two enzymes were constructed. The activity for the two enzymes of the cell extraction from different recombinant strains was determined. L-tert-Leucine was successfully synthesized by the recombinant strains, and the yield in different conditions was compared. The production of L-tert-Leucine was the highest when cell extraction of strains containing pLeuDH and pFDH, 1 mL cells extract could produce 4.5 mg L-tert-Leucine, while 1 mL whole cells could only produce 1.05 mg L-tert-Leucine. The yield of L-tert-Leucine was 3.375 mg/mL cell extraction of the strain containing pLeuDHFDH when NAD was added, while the yield fell to 1.635 mg/mL when the whole cell was used.

Keywords L-tert-Leucine • Leucine dehydrogenase • Formate dehydrogenase

3.1 Introduction

Amino acids are extensively used as chiral starting materials, auxiliaries, and catalysts in modern organic synthesis [1]. Incorporation of unnatural amino acids into peptide has become an important strategy in the synthesis of biologically active

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J. Bai et al.

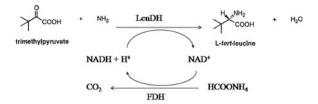


Fig. 3.1 Synthesis of L-tert-Leucine by coupling of enzyme reactions

compounds owing to the resistance to enzymatic degradation [2]. One of the most interesting amino acids in this regard is L-tert-Leucine, which has a sterically demanding side-chain. L-tert-Leucine can be used for templates in asymmetric synthesis and as building blocks for pharmaceutically active compounds. L-tert-Leucine itself is a component of several pharmaceutical development projects as tumor fighting agents or HIV protease inhibitors. A whole range of ligands for asymmetric catalysts has also been developed with L-tert-Leucine, mostly based on the oxazolidine moiety [3].

A method for enzymatic synthesis L-tert-Leucine was performed by reductive amination of trimethylpyruvate with leucine dehydrogenase (LeuDH) [4–6]. However, the cost of coenzyme hampered the application of this method to industrial production. A multi-enzyme reaction system for simultaneous coenzyme regeneration has been proposed to overcome this problem. As shown in Fig. 3.1, L-tert-Leucine was produced from trimethylpyruvate with LeuDH and the necessary cofactor NADH is regenerated by a second enzyme, formate dehydrogenase (FDH).

In this study, we constructed recombinant strains of *Escherichia coli* expressing LeuDH and FDH, respectively, and co-expressing the two enzymes together to produce L-*tert*-Leucine. The crude enzyme activity and production of L-*tert*-Leucine were determined and compared.

3.2 Materials and Methods

3.2.1 Bacterial Strains, Plasmids, and Other Materials

Bacillus cereus (TCCC 11114) used in this study was obtained from the preservation management center of College of Biotechnology, Tianjin University of Science and Technology. Candida boidinii (2.1645) used in this study was obtained from China General Microbiological Culture Collection Center (CGMCC). Vector plasmids pET-28a used for gene cloning and expression were purchased from YouBio (Changsha, Hunan, China) and host strain E. coli BL21 (DE3) was purchased from Invitrogen (Beijing, China). Restriction enzymes and other DNA-modifying enzymes were purchased from Takara (Dalian, China) and used according to the manufacturers' instructions. NAD and NADH were obtained from

Gener	Primer	Template
LeuDH	Forward 5'-GGCCCATGGCATTAGAAATCTTCGAATAC-3' ^a Reverse 5'-GATGGATCCGCGACGGCTAATA ATGTC-3' ^b	Genome of Bacillus cereus
FDH	Forward 5'-TCGGGATCCATGAAAAAAGTCGTCACG GTTTGCC-3' ^b Reverse 5'-CCGCTCGAGTTACGCCAGTGCCG CTTCG-3' ^c	Genome of Candida boidinii

Table 3.1 Primers and template DNAs used for construction of pLeuDH and pFDH

Solarbio (Beijing, China) and L-Leucine was obtained from Sigma (St. Louis, United States). All other chemicals were commercially available and of analytical grade.

3.2.2 Construction of pLeuDH and pFDH

DNA synthesis by PCR was performed with a DNA Thermal Cycler (Perkin-Elmer) in 25 μ L of a mixture containing deoxynucleoside triphosphate at a concentration of 1 μ L, 1 μ L primers, 1 μ L template DNA, 2.5 μ L reaction buffer (Takara), and 2.5 μ L PfuDNA polymerase (Takara). The PCR program consisted of predenaturation at 95 °C for 8 min, denaturation at 95 °C for 30 s, annealing at 57 °C for 1 min, and extension at 72 °C for 10 min for a total of 32 cycles. The genome of *B. cereus* was carried out with a DNA extraction kit from CWBIO (Beijing, China). The genome of *C. boidinii* was obtained as described by Manuel et al. [7]. All the primers used are shown in Table 3.1.

3.2.3 Construction of pLeuDHFDH

Plasmid for the simultaneous expression of LeuDH and FDH (pLeuDHFDH) was constructed as shown in Fig. 3.2. Plasmids pLeuDH and pFDH encoding the genes for LeuDH [8] and FDH [9], respectively, were prepared as described previously and used as template DNAs for PCR (Table 3.2). The LeuDH gene was cloned by PCR by using pLeuDH as template and was introduced into pET-28a at the *Bam*HI and *Sac*I sites, and then the FDH gene was cloned by PCR by using pFDH as template and was introduced into the plasmid at the *Sal*I and *Xho*I sites to produce plasmid pLeuDHFDH (Fig. 3.2).

^a The underlined site is the NcoI restriction site

^b The underlined site is the *BamHI* restriction site

^c The underlined sites are the *XhoI* restriction sites

J. Bai et al.

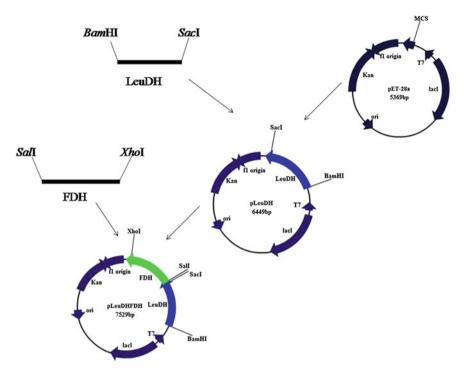


Fig. 3.2 Construction of the plasmid pLeuDHFDH used for production of L-tert-Leucine by expression in Escherichia coli cells

Table 3.2 Primers and template DNAs used for construction of pLeuDHFDH

Gener	Primer	Template
LeuDH	Forward 5'-GCGGATCCATGGCATTAGAAATCTTCGAATA-3' ^a Reverse 5'-GGCGAGCTCTTAGCGACGGCTAATAATGTCA TG-3' ^b	pLeuDH
FHD	Forward 5'-GCGTCGACAAGGAGATATACCATGAAGATCGTTTT AGTCTTATACGA-3' ^c Reverse 5'-CCGCTCGAGTTATTTCTTATCGTGCTT ACCATAA-3' ^d	pFDH

^a The underlined site is the *BamHI* restriction site

3.2.4 Cultivation of Recombinant E. coli Strains and Enzyme Assays

Plasmid pLeuDH, pFDH, or pLeuDHFDH was introduced into *E. coli* BL21 (DE3) and the resulting transformant was cultivated in 5 mL Luria-Bertani medium [10]

^b The underlined sites are the *XhoI* restriction sites

^c The underlined sites are the SacI restriction sites

^d The underlined sites are the SalI restriction sites

supplemented with 50 μ g/mL kanamycin, 34 μ g/mL chloramphenicol, and 1 mM isopropyl- β -D-thiogalactopyranoside in a test tube of 12 mL volume, shaking at 37 °C with 220 rpm. The volume of 5 mL cells were collected at the beginning of the stationary phase of the culture when the cells OD₆₀₀ reached 2.5, and then suspended in 5 mL lysis buffer (100 mM Tris/HCl, pH 7.5). Cell extracts were prepared by sonication. The enzyme activity of FDH and LeuDH were assayed with a MAPADA UV-3200 spectrophotometer by determining the NADH at 340 nm using the methods described previously [11–14]. One unit of the enzyme was defined as the amount of enzyme that catalyzes the production (or consumption) of 1 μ mol of NADH per min under the standard assay conditions.

3.2.5 Production of L-tert-Leucine

The clone cells were cultured and collected as described above and suspended in lysis buffer (100 mM Tris/HCl, pH 7.5). Cell extracts were prepared by sonication. The standard reaction mixture for L-tert-Leucine synthesis contained 200 mM trimethylpyruvate, 200 mM ammonium formate, 0.4 g/L NAD supplemented with water in a final volume of 30 mL and initiated by the addition of 2 mL washed cells or cell extracts treated as described above shaking at 30 °C for 24 h. The reaction mixture without NAD was also used for leucine synthesis and the other components were as the same as the standard reaction mixture.

3.2.6 Identification of Reaction Products

The product was derivatived with OPA/BOCCys and analyzed by HPLC using an LC-18DB column (5 μ m, 4.6 \times 250 mm, Agilent). The gradient elution profile, at 1 mL/min, was as follows: 95 % A and 5 % B at 0 min, 5 % A and 95 % B at 30 min (A: 20 % 0.02 M sodium acetate, and 1 % acetic acid, pH 7.2; B: acetonitrile/MeOH (1:1 by vol). Ultraviolet detector was used to determine the product at 340 nm. LC-MASS was used to identification molecular weight of the production.

3.2.7 GenBank Accession Numbers at NCBI

The obtained sequences of LeuDH and FDH were performed at the website of NCBI. The GenBank accession number of *leudh* is KM454878, and the number of *fdh* is KM454879.

J. Bai et al.

3.3 Result

3.3.1 Construction of LeuDH and FDH Expression Strains

Strain containing pLeuDH used for expressing LeuDH was constructed. The specific activity for LeuDH of the cell extract was 0.183 U/mL. Strains containing FDH used for expressing FDH was constructed and the activity for FDH of the cell extract was 0.084 U/mL. Recombinant strain containing pLeuDHFDH used for

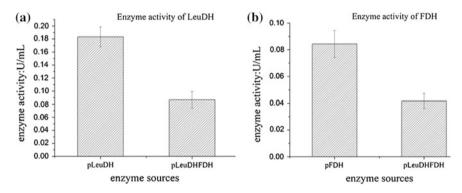


Fig. 3.3 The enzyme activity of LeuDH (a) and FDH (b) of recombinant strains of *E. coli* BL21 (DE3)

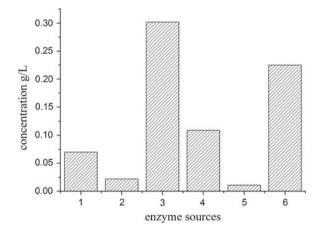


Fig. 3.4 Concentration of the L-*tert*-Leucine in the reaction system using different recombinant strains and conditions. *I* Whole cells of the stains containing pLeuDH and pFDH with NAD; 2 whole cells of the strains containing pLeuDH and pFDH without NAD; 3 cell extraction of the strains containing pLeuDH and pFDH with NAD; 4 whole cells of the strains containing pLeuDHFDH with NAD; 5 whole cells of the strain containing pLeuDHFDH without NAD; 6 cell extraction of the strains containing pLeuDHFDH with NAD

simultaneous expression of LeuDH and FDH was also constructed. LeuDH and FDH were tandem expressed using the T7 promoter of pET28a. The activity for LeuDH of the cell extract of the strain was 0.087 U/mL, which reached 47.5 % of the strain containing pLeuDH. The activity for FDH of the cell extract of the strain was 0.042 U/mL, which reached 50 % of the strain containing pFDH (Fig. 3.3).

3.3.2 Production of L-tert-Leucine

L-tert-Leucine synthesized by the constructed strains in different conditions was compared. Figure 3.4 showed the L-tert-Leucine concentration of the 30 mL reaction systems after 24 h reaction using 2 mL whole cell or cell extraction as enzyme source. The production of L-tert-Leucine was the highest when cell extraction of strains containing pLeuDH and pFDH was used together as the enzyme source and NAD was added into the system. At this condition, 1 mL cells extract could produce 4.5 mg L-tert-Leucine (Figs. 3.3 and 3.4), while 1 mL whole cells could only produce 1.05 mg L-tert-Leucine. The yield of L-tert-Leucine was 3.375 mg/mL cell extraction of the strain containing pLeuDHFDH when NAD was added, while the yield fell to 1.635 mg/mL when the whole cell was used. What is more, the result concluded that the yield of L-tert-Leucine was dramatically low when the reaction system without NAD was used.

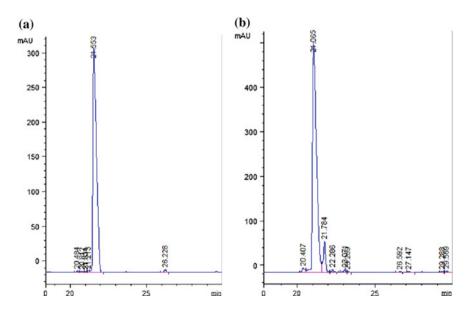


Fig. 3.5 HPLC result of the standard L-tert-Leucine (a) and product of the reaction system (b)

J. Bai et al.

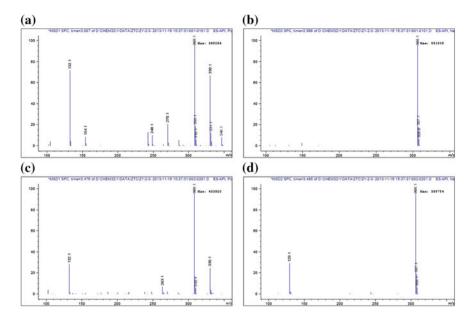


Fig. 3.6 LC-MASS of the L-*tert*-Leucine and product of the reaction system: **a** standard substance (+); **b** standard substance (-); **c** product of reaction system (+); **d** product of reaction system (-)

3.3.3 Identification of L-tert-Leucine

The reaction liquid was detected by HPLC (Fig. 3.5) and LC-MASS (Fig. 3.6). The results show that the appearance time of product was the same as the standard substance at 21 min (Fig. 3.5). The molecular weight of product was 131.1 (+), 130.1 (-) and was identical to the standard substance.

3.4 Discussion

In this study, we successfully constructed L-tert-Leucine producing strains by introducing leudh from B. cereus and fdh from C. boidinii into E. coli. Strains expressing LeuDH and FDH, respectively, were constructed first. As substrate, trimethylpyruvate can be transformed into L-tert-Leucine using the cell extract of the two strains together but cannot be transformed using the whole cells [4, 5]. It suggested that the trimethylpyruvate could not overcome the barrier of the cell membrane and get into the cells freely.

Meanwhile, the strain co-expressing LeuDH and FDH was constructed, the LeuDH and FDH activity of the crude enzyme can reach about half the strains expressing the two enzymes, respectively, and the yield of L-*tert*-Leucine could reach about 75 % of it [15]. It concluded that we have successfully constructed

L-tert-Leucine producing strain by simultaneously expressing two heterologous genes in *E. coli*, however, the product yield was lower than using two strains expressing LeuDH and FDH separately. In further study, we will attempt to optimize the codon of genes and change different inducible promoters to enhance the expression of the two enzymes and optimize their ratio in order to improve the yield of L-tert-Leucine.

The addition of NAD could remarkably enhance the product yield, which suggested that the coenzyme produced by cells itself could not satisfy the reaction system and NAD/NADH could successfully achieve mutual conversion in this system. In addition, this reaction system is also suitable for the synthesis of other natural or unnatural amino acids which contain different side chains and can be transformed by LeuDH.

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J. Bai et al.

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Chapter 4 The Distribution Characteristics of *Microcystis novacekii* Based on 16S rDNA Sequence

Da Huo, Yang Luo, Yunsi Nie, Jing Sun, Yanan Wang and Zhiyi Qiao

Abstract *Microcystis* has been regarded as the most common cause of water bloom. The spatial and temporal distribution of *Microcystis* provides a method of bloom warning. Microcvstis novacekii was the dominant cvanobacteria species in Yuqiao reservoir in recent years. Seven sampling stations were set and the phytoplankton were collected in both July and October. Comparing morphological observation with the molecular biological method, which is based on 16S rDNA sequencing, the recognizing research of M. novacekii was conducted. The results showed that the average biomass of M. novacekii was 0.92 ± 0.23 mg/L; cell density was $1833.48 \pm 457.70 \times 10^4/L$ affected by light, temperature, nutrients, and other environmental factors. In July, the maximum biomass was 2.21 mg/L at Center station and the minimum was 0.83 mg/L at the same site; in October, the maximum was 0.18 mg/L at intake site, the minimum was 0.09 mg/L at Center West. The significant seasonal fluctuation meant that M. novacekii favored high temperature, which is the same habit with universal *Microcystis* algae. 30 clones of M. novacekii were collected and amplified the 16S rDNA fragments. Results proved that the molecular identification of *Microcystis*, which is based on the conservative fragment of 16S rDNA, was consistent with the traditional morphology judgement, we confirmed that environmental factors have a huge impact on the distribution of Microcystis and the molecular biology techniques based on 16S rDNA sequencing were reliable for Microcystis identification.

Keywords *Microcystis novacekii* • Distribution • 16S rDNA • Molecular identification

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D. Huo et al.

4.1 Introduction

China has suffered from the most serious and widely distributed cyanobacteria bloom. In recent years, cyanobacteria bloom frequent caused local water quality deterioration, ecological disasters happened frequently [1], after the outbreak of algal blooms, the accumulated death cells would released a lot of cyanobacterial toxins, odor, toxic and harmful substances threats to the water security [3]. The main toxic substance released by *Microcystis* was microcystin (MC), which partly cyanobacteria produced, and the molecular producing mechanism has been figured out clearly. It was produced through non ribosome ways by Giant enzyme complex include peptide synthetase, polyketide synthases (PKSs) and other modification enzyme [8, 18].

Hence the researches of the toxigenic identification and temporal, and spatial distribution of the *Microcystis* was imminent and important.

Cyanobacteria species have traditionally been identified on the basis of their microscopic morphology, physiology, and staining characteristics, which are inadequate and inaccurate, leading to misidentification due to their similarities [22]. *Microcystis* strains usually form scum [13], and thus *Microcystis* spp. to *Aphanocapsa* spp. is often misidentified in the same way. *C. raciborskii* has similar morphology with other cyanobacteria, such as *Anabaenopsis* and *Raphidiopsis*, making their quantification more difficult [5].

The lack of a simple and rapid method to detect the composition of algae in water bodies was a barrier of *Microcystis* identification. DNA barcode technology can directly detect the distribution of algal species and analyze the composition of algae, making the algae research toward a higher level. Using barcode technology to identify algae species of algae can make people exempt tedious microscopy analysis steps to determine if the species occurred.

Yuqiao reservoir—the drinking water source of Tianjin has usually appeared the water bloom in small scale, which caused the potential risk on the safety of drinking water [9, 16]. According to the reports, *Microcystis novacekii* is one of the dominant species of cyanobacteria in recent years. The laws of spatial and temporal distribution of *M. novacekii* may allow us to understand the algae growth causation better. These consequences combined with the rapid identification using molecular technology may provide a basis for early warning and ecological control in the future

The purpose of this study was to analyze the regular water bloom in the Yuqiao reservoir. We investigated the distribution patterns of *Microcystis* algae in Yuqiao reservoir and established a set of standard operation process to distinguish it, in the hope of giving a prewarning of *Microcystis* bloom.

4.2 Materials and Methods

4.2.1 Sampling

Yuqiao reservoir located in Ji county in Tianjin, covers a flow area of $2,060 \text{ km}^2$ with a total storage capacity of $1.559 \times 106 \text{ m}^3$. It is an important adjusting reservoir and functions as the water resource for the drinking water, industry, and agriculture of Tianjin. The dam is located in the west of the reservoir; there is a water intake site in the south, this water drains into the culvert in Tianjin (Fig. 4.1).

The sites for sampling are shown in Fig. 4.1 and the GPS coordinates in Table 4.1. There are three important rivers—Sha River, Lin River, and Guo River in the east. The dam is located in the west of the reservoir (site 1, outlet of Yuqiao). There is a water intake site in the south (site 7), and the water drains into the culvert in Tianjin.

Seven sampling sites were studied at reservoir in July and October 2012 at 9:30–10:00 a.m. A 1-L sample of surface (50 cm) water was collected with a stainless steel Kitahara's sampling apparatus for the quantity of phytoplankton at each site. Another 1-L water sample was taken with a sampling tube and directly transported

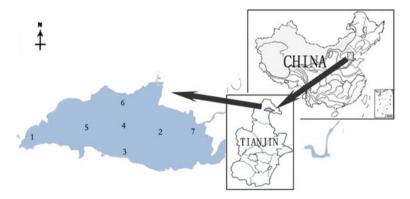


Fig. 4.1 Map showing the sampling sites of Yuqiao reservoir

Table 4.1 The coordinate of sampling sites

	Sites	Latitude (N)	Longitude (E)
1	Intake	40°01.624′	117°26.593′
2	Center east	40°01.967′	117°33.178′
3	Center south	40°01.583′	117°34.967′
4	Center	40°02.533′	117°31.255′
5	Center west	40°02.172′	117°29.493′
6	Center north	40°03.167′	117°31.017′
7	Feng hill	40°01.647′	117°31.255′

D. Huo et al.

to the laboratory for molecular identification. The other samples were collected with phytoplankton net for qualitative research and fixed with Lugol's iodine solution.

4.2.2 The Qualitative and Quantitative Analyses

The phytoplankton samples were preserved with 1.5 % Lugol's iodine solution and were concentrated to 30 mL after sedimentation for 48 h. After mixing, 0.1 mL concentrated samples were estimated by using a Sedgwick–Rafter counting chamber under an Olympus optical microscope at $800\times$. Classification and identification of phytoplankton species and genera were based on Hu et al. [6]. The phytoplankton biomass was estimated from the approximate geometric volume of each taxon, assuming that the biomass per μ m³ equal 1–6 μ g fresh weight. The geometric dimensions were measured on 10–30 individuals of each phytoplankton species in each sample. The cell numbers of individual species were quantified. Phytoplankton abundance was expressed as the number of cells or the total algal biomass per liter sample.

4.2.3 Molecular Identification

30 monoclonal *M. novacekii* were transformed to 0.5 mL PCR tubes after washing with pure water. DNA extracting method based on patent (CN201410177188.7). Specific primers used in this study was

27F: 5'TGGCGGCGTGCCTAACACATGCAAG-3' 1494RC: 5'AGGAATGACTTCGGGCGTGACCAG-3' [11],

which is conserved sequence of 16S rDNA 3' end and the 5' end of 23S rDNA. The 25 μ L PCR contained 1 μ L of DNA template, 4 mol/L dNTPs mix (25 mM), 1× PCR Buffer, 1 μ L MgCl₂ (50 mM), 0.5 μ L forward primer (10 mM), 0.5 μ L reverse primer (10 mM), and 5 U *EasyTaq* DNA *Polymerase* (TransGen Biotech, Beijing, P.R. China) then add molecular biology grade H₂O in a total volume of 25 μ L. PCR was performed using the following conditions: 94 °C for 10 min followed by 30 cycles of 94 °C for 1 min, 54 °C for 2 min, 72 °C for 1 min, then 1 cycle of 72 °C for 10 min; PCR was performed on Eppendorf amplification instrument.

The PCR products were checked by electrophoresis in a 1 % agarose gel. Gels were stained with ethidium bromide and visualized under UV in a transilluminator. PCR products were sent for sequencing (Genewiz, Beijing, China).

4.2.4 Alignment and Phylogenetic Analysis

Sequences were assembled by Software ContigExpress and aligned with Clustal X 1.81. All sequences were confirmed by doing a BLAST search against the GenBank nr database (http://blast.ncbi.nlm.nih.gov/Blast.cgi). Neighbor-joining tree computed using MEGA 5.05 and the Kimura 2-parameter distance nucleotide model. The tree was evaluated by 1,000 bootstrap replications.

Statistical analysis was performed with SPSS 14.0 for Windows. The results were expressed as means and standard deviations (±SD).

4.3 Results

4.3.1 Spatial and Temporal Distribution of Microcystis novacekii

The average biomass of *M. novacekii* was 1.70 ± 0.45 mg/L (average cell density $3396.85 \pm 909.88 \times 10^4$ ind/L) in July and 0.14 ± 0.06 mg/L (average cell density $278.28 \pm 114.55 \times 10^4$ ind/L) in October, Fig. 4.2 shows the maximum of biomass was 1.74 mg/L (average cell density 3470.02×10^4 ind/L) at intake site and the minimum of 0.09 mg/L (average cell density 186.03×10^4 ind/L) at Center West.

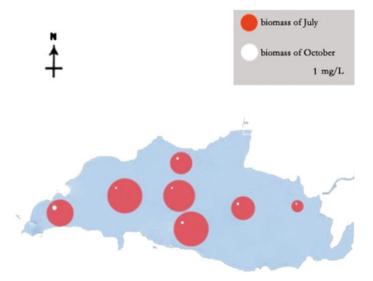


Fig. 4.2 The biomass of *Microcystis novacekii* in Yuqiao reservoir (1 mg/L)

D. Huo et al.

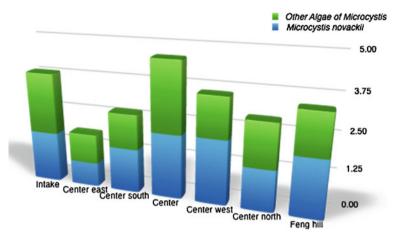


Fig. 4.3 Microcystis novacekii biomass proportion in July (1 mg/L)

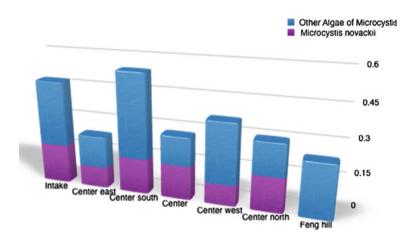


Fig. 4.4 Microcystis novacekii biomass proportion in October (1 mg/L)

The rapid growth of M. novacekii was related to the high temperature and intense illumination in summer. Two-month biomass of M. novacekii had significant difference (P < 0.01). In July, M. novacekii grown to dominant populations, the average biomass takes part of 52 % in Microcystin algae. The peak value appeared at Center point, which was 58 % (Fig.4.3). In October, it changes to another situation. M. novacekii is no longer a dominant species in Microcystis. The special circumstance appears at Feng hill site, no M. novacekii was detected at that site (Fig. 4.4).

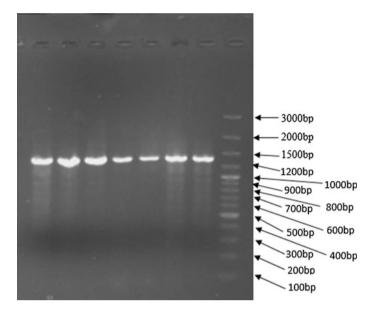


Fig. 4.5 Microcystis novacekii 16S rDNA fragments amplified result

4.3.2 Molecular Identification

The 30 PCR productions were confirmed by electrophoresis and sequencing, the size of fragments was in the range of 1,257–1,276 bp, and the homology of *M. novacekii* 16S rDNA fragment was between 95 and 99 % (Figs. 4.5 and 4.6).

The average content of A, T, G, C four bases are 25.4, 19.2, 32.2, 23.3 %. A+T: 44.6 %, G+C: 55.4 %. Bases composition was similar between individual clones. Using MEGA to aggregate difference sequence and evaluate Base transversion rate. Table 4.2 shows the base substitution of *M. novacekii* 16S rDNA.

Monoclonal 16S rDNA gene in sample 30, according to Table 4.2, the frequency of transitions and transversions were all 0, except A–G transition which was 1. We divided the samples and GenBank downloaded results into two groups; the group genetic distance of the first group was 0.00086, the second group was 0.00074; the genetic distance between the two groups was 0.00082 (Fig. 4.6).

4.4 Discussion

4.4.1 Seasonal Distribution of Microcystis novacekii

The rainy season of Yu Qiao is from June to September, and the normal river flow period appears in May [23]. The biomass distribution of *M. novacekii* in summer

D. Huo et al.

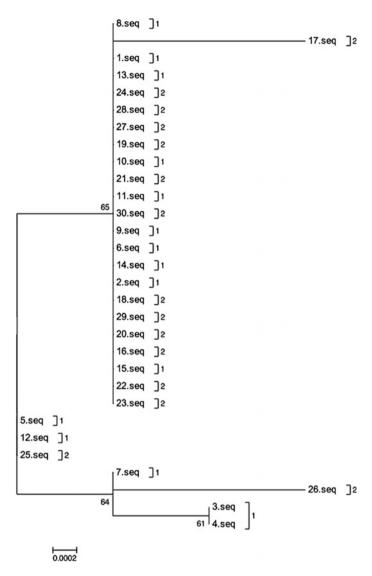


Fig. 4.6 Neighbor-joining bootstrap consensus tree. Numbers on branches represent the bootstrap support for each node in 1,000 replicates

Table 4.2 Base substitution of Microcystis novacekii 16S rDNA

	ii	si	sv	R	TT	TC	TA	TG	CC	CA	CG	AA	AG	GG	Total
Average	1,259	1	0	4.1	241	0	0	0	293	0	0	320	1	405	1260.8

was significantly higher than in autumn (P < 0.01). M. novacekii shows growth advantage in flood season. Enough water, high temperature of water, and high pH value are suitable for algae growth [21]. In recent years, the dominant bloom algae were Microcystis [10]. According to the results from analysis of water samples, compared with other researches, we concluded that the distribution of M. novacekii shows seasonal changes.

We did a comparison between each station. The biomass of M. novacekii shows a significant difference in July (P < 0.05). In this study, Intake was a special position where the highest biomass (2.21 mg/L) appears at this position. Wind and temperature have a significant impact on the distribution of Microcystis [15]. After verification, there was high temperature and precipitation in July 2013. Strong northwest wind leads to the high biomass of Microcystis at intake.

4.4.2 Monoclonal DNA Extracting Methods of Microcystis

Normal DNA extracting methods of *Microcystis* was DPS purification [2], extraction—centrifugation method [17], Bio kit method, and even execute PCR directly without DNA extraction [12, 20]. These methods cannot extract *Microcystin* monoclonal DNA, and the operation is cumbersome and prone to errors. In this study, repeated freezing and thawing methods were used to extract monoclonal DNA of *Microcystis*. The whole progress of extraction is less than 1 h, no need for purification, and the production of extraction was effective for accurate analysis of monoclonal DNA. The study provided a new high-throughput method of DNA analysis.

4.4.3 DNA Barcoding Fragment of Microcystis novacekii

In 30 samples, the base is almost no transitions and transversions, only one base transition appears between A and G, and the genetic distance was very small. This proves the interspecific conservation of *M. novacekii* 16S rDNA. 16S rDNA fragment has a higher mutation rate, the intergeneric difference is very large, but the interspecific difference is small [19, 14]. 16S rDNA gene cannot have high genetic diversity [7]. However, there are significant differences in the primary structure of interspecific *Microcystis* [4]. Classification and identification of species herein may be based on the similarity between the values of 16S rRNA nucleotide sequence, which can be used between the morphologically similar algae.

According to the NJ tree, there were several samples of this study that were not in the same branch, it may be the cryptic species that existed in samples we collected. There was system isolation between *Microcystis* in Taihu Lake [15].

D. Huo et al.

It shows that the wind may change the phenotype of *Microcystis* and lead to the same phenotype with different genotypes. Nevertheless, most of the samples belonging to the same branch proves that this area has a large conservation among species. DNA barcode technology was a powerful tool for *Microcystis* identification.

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Chapter 5 The Distribution and Molecular Identification of the Microcystis aeruginosa in Yuqiao Reservoir

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Abstract Yuqiao reservoir, the only drinking water source of Tianjin in the downstream of Luan river basin, has occurrence of water bloom, at a small scale, annually. Hence the research in *Microcystis* distribution spatially and temporally has been conducted. In this study, we focus on Microcystis aeruginosa, which is one of the most contributing causes of the bloom. Samples are collected from seven sites in July and October 2012 and molecular techniques conducted to identify the selected clone from the samples by 16S rDNA sequencing. According to the results, M. aeruginosa has grown vigorously in July with the average biomass of 1.59 ± 0.39 mg/L and the cell density of $3191.25 \pm 768.04 \times 10^4$ /L. The maximum value occurs in the site of the northern reservoir (the average biomass and the cell density were $1.37 \pm 1.07 \text{ mg/L}$ and $2723.33 \pm 2131.13 \times 10^4/\text{L}$ respectively). By analyzing the sequencing results and comparing with the morphological characters, the accuracy of molecular identification based on the 16S rDNA region is confirmed. From the BLAST alignments in NCBI, we find that the homology of 30 clones of M. aeruginosa is between 97 and 99 %; A+T (56.6 %), which is higher than C+G (43.4 %). The samples almost match the same branch of phylogenetic tree.

Keywords Microcystis aeruginosa · Distribution · Molecular identification · 16S rDNA

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⁴³

Z. Qiao et al.

5.1 Introduction

Cyanobacterial dominance in phytoplankton has become a common feature of eutrophic freshwaters worldwide [8]. Considerable concern has arisen over the temporal and spatial distribution of the water bloom in ecological research [4, 7, 9, 12].

Cyanobacterial or *Cyanophyt*—the oldest and the most primitive algae reproduces rapidly and blooms often arise in many lakes and reservoirs in China during spring through autumn [1]. The influencing factors of *Cyanophyta* bloom have been widely investigated [10]. Recently, Yuqiao reservoir—the drinking water source of Tianjin discovered the water bloom regularly in small scale, which risks potentially the safety of drinking water. In the last 9 years, six dominants (*Microcystis aeruginosa, Microcystis viridis, Microcystis novacekii, Microcystis wesenbergii, Microcystis flos-aquae* (Wittr). Kirchner, and Microcystis incerta Lemm.) occurred successively with abrupt succession every 2–3 years [11]. *M. aeruginosa* is one of the dominants that occurred the most in the last 2 years. *M. aeruginosa* cells usually aggregate to the visible mass and form the cooper green/odour/toxic in most cases [13] of algal bloom.

The distribution regular of dominant provides data for controlling and early-warning of bloom. Usually, the taxonomy of the species is time-consuming. The accuracy and consistency can be ensures by professional researchers only, which greatly impedes the expanded monitoring and research of algal bloom. In this study, the DNA of *M. aeruginosa* of 30 clones directly selected from the wild samples was extracted (the national patent number of the method of the DNA extract in this research is 201410177188.7); 16S ribosomal RNA genes (16S rDNAs) are amplified by PCR. By analyzing sequencing data, we explore the possibility of a simple-quick method to identify the taxonomic status accurately.

5.2 Materials and Methods

5.2.1 Sampling and Counting

Yuqiao Reservoir is located in Ji county of Tianjin, about 30 km west to east. The reservoir has a surface area of 86.8 km² and a mean depth of 16 m. The drainage basin is 2,060 km² with a total water circulation of 155,900,000 m³ per year, when the yearly mean rainfall value is 750 mm. Yuqiao Reservoir is the main water resource for the supply of Tianjin from 1961. There are three import rivers—Sha River, Lin River, and Guo River in the east. The dam is located in the west of the reservoir; there is a water intake site in the south (site 1, outlet of Yuqiao), then the water drains to the culvert to Tianjin (Fig. 5.1).

Water samples were taken below the surface (50 cm) on seven sites of the reservoir in July and October 2012 at 9:30–10:00 a.m (Table 5.1). The phytoplankton samples

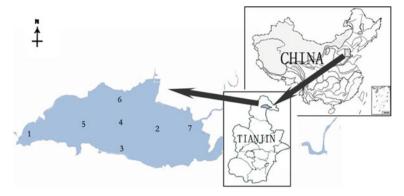


Fig. 5.1 Map showing the sampling sites of Yuqiao reservoir

were preserved with 4 % formalin solution and concentrated to 30 mL after sedimentation for 48 h. After mixing, 0.1 mL concentrated samples were counted directly using an Olympus optical microscope at 800×. Classification and identification of phytoplankton species and genera were based on Hu et al. [5].

The phytoplankton biomass was estimated from the approximate geometric volume of each taxon, assuming that the biomass per μm^3 equals 1–6 μg fresh weight. The geometric dimensions were measured on 10–30 individuals of each phytoplankton species in each sample. The cell numbers of individual species were quantified. Phytoplankton abundance was expressed as the number of cells or the total algal biomass per liter sample. Statistic analysis was performed with SPSS 14.0 for Windows. The results were expressed as means and standard deviations ($\pm SD$) (Table 5.2).

5.2.2 Molecular Identification

30 monoclonals of *M. aeruginosa* algae were transferred to 0.2 mL EP tubes by micropipette after washing with ultrapure H₂O. The DNA extracting method used national patent (CN201410177188.7). Specific primers used in this study were

Table 5.1	The coordinate of
sampling s	ites

	Sites	Latitude (N)	Longitude (E)
1	Intake	40°01.624′	117°26.593′
2	Center east	40°01.967′	117°33.178′
3	Center south	40°01.583′	117°34.967′
4	Center	40°02.533′	117°31.255′
5	Center west	40°02.172′	117°29.493′
6	Center north	40°03.167′	117°31.017′
7	Feng hill	40°01.647′	117°31.255′

Z. Qiao et al.

27F: 5'TGGCGGCGTGCCTAACACATGCAAG-3' and 1494RC: 5'AGGAATGA CTTCGGGCGTGACCAG-3' [10] which is the sequence of 16S rDNA with the length of about 1,300 bp.

The 25 μ L PCR contained 1 μ L of DNA template, 4 mol/L dNTPs mix (25 mM), 1× PCR Buffer, 1 μ L MgCl₂ (50 mM), 0.5 μ L forward primer (10 mM), 0.5 μ L reverse primer (10 mM), and 5 U EasyTaq[®] DNA Polymerase (TransGen Biotech, Beijing, P.R. China), then ultrapure H₂O was added in a total volume of 25 μ L. PCR was performed using the following conditions: 94 °C for 10 min followed by 30 cycles of 94 °C for 1 min, 54 °C for 2 min, 72 °C for 1 min, then 1 cycle of 72 °C for 10 min; PCR was performed on Eppendorf amplification instrument.

PCR product was checked by electrophoresis in 1 % agarose gel (1× TAE buffer) under the electric voltage of 90 mV/cm². Gels were stained with ethidium bromide (0.5 μ g/mL) and visualized under UV in a transilluminator. Then purification and sequencing were done (Genewiz, Beijing, China).

5.2.3 Alignment and Phylogenetic Analysis

Sequences were assembled by Contig Express and aligned with Clustal X 1.81. All sequences were confirmed by doing a BLAST search against the GenBank database (http://blast.ncbi.nlm.nih.gov/Blast.cgi). Neighbor-joining tree was computed using MEGA 5.05 and the Kimura 2-parameter distance nucleotide model. The tree was evaluated by 1,000 bootstrap replications.

Statistical analysis was performed with SPSS 14.0 for Windows. The results were expressed as means and standard deviations (±SD).

5.3 Results and Discussion

5.3.1 The Distribution of the Microcystis aeruginosa

Microcystis aeruginosa is a typical high temperature pond species whose average cell density $(3191.25 \pm 768.04 \times 10^4 \text{ ind/L})$ and biomass $(1.59 \pm 0.39 \text{ mg/L})$ in July is significantly higher than that in October $(331.31 \pm 74.99 \times 10^4 \text{ ind/L})$, $0.17 \pm 0.04 \text{ mg/L}$ (P < 0.01). Owing to the effect of the southeast wind in summer and autumn, *M. aeruginosa* spreads more rapidly in the center north (Fig. 5.2, site 6).

By comprehensive analysis from the biomass of seven sites, the northwest of the reservoir (site 5 and 6) had distributed more *M. aeruginosa* cell than other regions. The result is mainly contributed by the data of July (Table 5.2). Because the competitive inhibition of vascular plants, *M. aeruginosa* cannot grow fast in feng hill (site 7). The encouraging result is that the biomass of the outlet of Yuqiao (site 1) is relatively low.

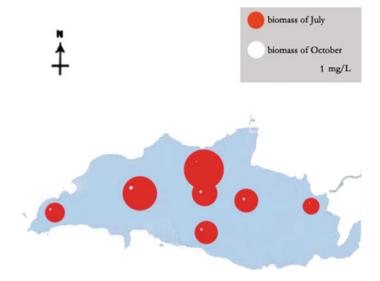


Fig. 5.2 The biomass of *Microcystis aeruginosa* in Yuqiao reservoir (mg/L)

Table 5.2 The spatial changes in *Microcystis aeruginosa* (mg/L)

Sites	July	October	Average	Standard deviations
1	1.21	0.12	0.67	±0.55
2	1.44	0.15	0.80	±0.65
3	1.39	0.14	0.77	±0.63
4	1.53	0.17	0.85	±0.68
5	2.11	0.18	1.15	±0.97
6	2.43	0.30	1.37	±1.07
7	1.02	0.11	0.57	±0.46

Similarly, the speared trend of *M. aeruginosa* for 2 months is that the cell biomass accumulated more in the western and northern regions than in other regions of the lake.

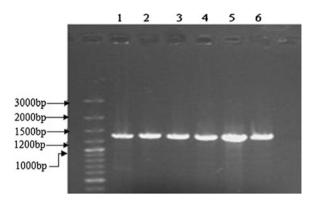
5.3.2 Molecular Identification

30 PCR products were confirmed by electrophoresis and sequencing; fragments were in the range of 1,200–1,500 bp, and the homology of *M. aeruginosa* 16S rDNA fragment was between 95 and 99 % (Fig. 5.3).

After the DNA sequence assembly, BLAST analysis was conducted by NCBI, which showed that the sequences of 30 nomoclones had high homology (about 98–99 %) with the reported sequences of *M. aeruginosa* in GENBANK, respectively.

48 Z. Qiao et al.

Fig. 5.3 The electrophoretic confirming of 16S rDNA of *M. aeruginosa*



From these results, we can identify the species of 30 nomoclones which have been selected randomly from field water, they are all *M. aeruginosa*.

After the integrating and comparing of 30 sequences used by Clustal, continued analysis was done by MEGA 4.1. Results show that the average contents of A, T, G, C were 25.0, 31.6, 19.6, 23.8 %, respectively. A+T: 44.6 %, G+C: 55.4 %. The genetic distance was between 0.000 and 0.001.

At the same time, the high homology sequences were downloading from GENEBANK for the construction of phylogenetic tree with our sequences.

5.4 Discussions

5.4.1 The Regularities of Algal Distribution Can Provide Reference to the Monitoring and Early-Warning

The bloom of *M. aeruginosa* would usually release microcystin (a toxics protein) which has raised concern for many years. At the same time, it is a typical high temperature pond species. From the results of this experiment, the management should pay extra attention during summer. According to the study, the optimum temperature of *M. aeruginosa* is 28.8-30 °C. The high light intensity also helps in the growth of the cyanobacteria. The average cell density of July was $3191.25 \pm 768.04 \times 10^4$ ind/L when it was $331.31 \pm 74.99 \times 10^4$ ind/L in October in the same year. Research has proved that the outbreak of water bloom was controlled by many external factors: lake depth, station morphometric, substratum, and hydraulic regime [2]. This partly coincides with this investigation, where the results clearly demonstrate that wind can entrain the *M. aeruginosa* downward,

wind-driven horizontal *M. aeruginosa* drift. The data indicate that the highest *M. aeruginosa* biomass values were in the north-west portions of the reservoir. The biomass in the north region has a dramatic season change, with the maximum value in July but relatively lower than other regions in October. These may be caused by the southeast monsoon in summer and the northwest monsoon in winter.

The same pattern was approved by satellite images acquired in the past 20 years, also showing that the cyanobacterial bloom appeared predominantly in the western and northern portions of the lake [3]. Thus, during the growth season, *M. aeruginosa* biomass is distributed downwind along the dominant wind direction, and wind-driven drift is a major factor involved in the downwind accumulation.

Above all, we should pay more attention to the north-west region in summer in Yuqiao reservoir, especially after a windy day.

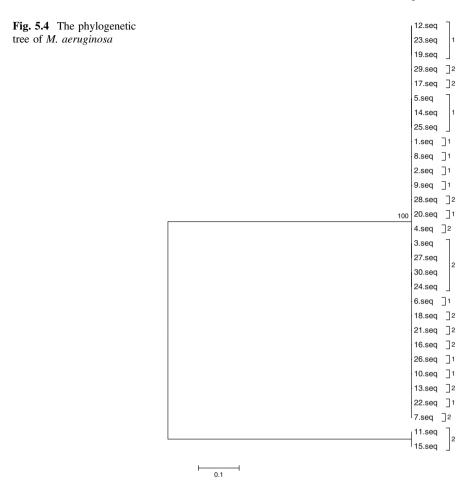
5.4.2 A Simple Fast Method to Identified Algal Is Necessary

Limited by the cell size of *Microcystis* (diameter usually at 2–6 µm), most molecular research has to expand the culture of monoclones before DNA extracting which is time-consuming, easy polluting, or extracting the total DNA from the samples directly that cannot analyze accurate sequencing. The procedure of extracts of the total DNA in this research (National invention patent CN201410177188.7) can solve this problem. The DNA of the monoclone *Microcystis* cells which are selected from the wild samples directly can be instantly guaranteed for PCR and sequencing. The procedure makes the larger scale and genetic analysis of prokaryotic algae possible and also provides the convenient identification of cyanobacteria for laypeople (Fig. 5.4).

5.4.3 The Genetic Relationship of Microcystis aeruginosa

The high (98–99 %) homology of 30 clones explains the conservative evolution of *M. aeruginosa*, and confirms the consistence of the identification from morphology and molecular biology. By the analysis of MEGA and comparison with the sequences downloaded from Gene Bank, the 30 clones belong to the same branch-*M. aeruginosa*. 16S rDNA region was a widely used DNA barcoding fragment [6, 13] with compatible length (1,100–1,600 bp) and amount of information, consisting of the coding region that is highly conserved and the moderate and high variation region. So the identical procedure based on DNA barcoding of *M. aeruginosa* is reliable.

Z. Qiao et al.



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Chapter 6 The Improved Stress Tolerance of Escherichia coli by Directed Evolution of IrrE

Jianmei Luo, Jiajia Liu, Yuanyuan Zheng and Min Wang

Abstract IrrE, a global regulator of *Deinococcus radiodurans*, has proven to be effective in enhancing microbial stress tolerance. In this paper, IrrE from D. radiodurans R1, was introduced into Escherichia coli and directed evolved by error-prone PCR. The influence of mutation of IrrE on the cell growth and tolerance to various stresses was further investigated. First, one of the mutations, designated M4 with higher ethanol tolerance was obtained by error-prone PCR using the pET-28a(+)-irrE(W) as a template. The OD₆₀₀ value of M4 reached 1.6 after 28 h cultivation under 8 % ethanol, while no obvious cell growth was observed in the recombinant strain harboring plasmid pET-28a(+)-irrE(W) and the control strain harboring plasmid pET-28a(+). The cell viability of M4 under different stress shock conditions (such as pH = 5.0, pH = 10.0, 3 mmol/L sorbitol and 15 % methanol), the cell growth rate, and the final biomass were improved obviously. The sequence comparison of irrE revealed that there were two sense substitutions (C24T and G530A). The substitution of G530A caused one amino acid change Gly177Glu. The homology modeling of IrrE was built according to the known structure of IrrE protein from D. deserti, which showed that the amino acid mutation located in HTH motif of IrrE. The work laid a foundation for further research on the relationship of IrrE structure and host cell tolerance to stress.

Keywords IrrE \cdot *Escherichia coli* \cdot *Deinoeoccus radiodurans* \cdot Stress tolerance \cdot Error-prone PCR

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54 J. Luo et al.

6.1 Introduction

Recently, global transcriptional machinery engineering (gTME) has been widely explored to improve strain phenotypes by engineering transcription factors or RNA polymerase subunits, thereby altering transcriptional profiles of target strains; various degrees of success have been achieved [1–3]. The genus *Deinococcus* is known for its ability to survive extreme stress conditions, conferred by a unique pool of genes.

Deinococcus radiodurans is one of the most radio-resistant organisms known. This bacterium can survive extreme cold and dehydration as well as exposure to vacuum and acidic conditions [4]. Genetic analysis of a DNA damage-sensitive strain of *D. radiodurans* R1 led to the discovery of a novel regulatory protein, IrrE (GeneID: 7738927) [5, 6]. IrrE, a newly identified gene switch responsible for extreme radio-resistance of *D. radiodurans*, plays a central regulatory role in multiple DNA damage repair and protection pathways in response to radiation stress [6]. Interestingly, the heterologous expression of *irrE* could enhance the tolerance to osmotic pressure in *Escherichia coli* [7], salt tolerance in *E. coli* and *Brassica napus* [8], the anti-oxidation and anti-ultraviolet irradiation in *Bacillus subtilis* [9].

Escherichia coli is one representative strain of prokaryotes and model microorganism, which has been widely used for the production of bio-fuels and other chemicals. But the strain always faces various abiotic stresses in the industrial application, such as organic solvents, extreme pH, osmotic pressure, and so on, which seriously restricts its application efficiency.

In our previous work, the wild *irrE* gene from *D. radiodurans* R1 has been heterologously expressed in *E. coli* BL21(DE3). The results showed that the growth rate and the final biomass of recombinant strain containing pET-28a(+)-*irrE*(W) under stress were significantly higher than the control strain (harboring the empty vector pET-28a(+)). In order to obtain the mutants with higher tolerance to stress, IrrE was further directed evolved by error-prone PCR using pET-28a(+)-*irrE*(W) as a template in this paper. The cell growth performance and tolerance to stress of the evolved strain were investigated. The sequence comparison between the wild and evolved IrrE, and homologous modeling analysis by SWISS MODEL were further studied. The results could better understand the relationship of IrrE structure and host cell tolerance to stress.

6.2 Materials and Methods

6.2.1 Materials

Restriction enzymes, DNA-modifying enzymes, DNA markers, and T4 Ligases were purchased from TaKaRa (Dalian, China). The kits for DNA purification, gel recovery, and plasmid mini-prep were purchased from Omega Bio-Tek (USA).

Protein Marker II was purchased from Biomed (Beijing, China). Taq DNA polymerase, dNTPs was purchased from Sangon (Shanghai, China). The kit for protein purification was purchased from ComWin (Beijing, China). Sequence analysis was performed by BGI (Beijing, China). All cloning steps were performed in $E.\ coli\ DH5\alpha$ or BL21(DE3). The plasmid pET-28a(+) was stored in Key Laboratory of Industrial Fermentation Microbiology (Tianjin University of Science & Technology).

6.2.2 Strains

Escherichia coli BL21(DE3)/pET-28a(+) and E. coli BL21(DE3)/pET-28a(+)-irrE (W) have been successfully constructed in our previous work, which were grown in Luria-Bertani (LB) medium containing 50 μg/mL kanamycin at 37 °C. When required, 2 mmol/L IPTG was added to the medium to induce the expression of IrrE.

6.2.3 Library Construction and Selection

Using the plasmid pET-28a(+)-*irrE*(W) as a template, error-prone PCR method was employed to construct the library. The mutation ratio was controlled by adjusting the concentration of manganese ions and magnesium ions.

A total volume of 100 μ L of reaction mixture contained 10 μ L 10 \times Taq DNA polymerase buffer, 0.2 mmol/L dATP and dGTP, 1 mmol/L dCTP and dTTP, 10 ng of template, 0.05–0.25 mmol/L MnCl₂, 1–7 mmol/L MgCl₂ and 5 units (U) of Taq DNA polymerase. A pair of primers used was F1 (5'-CCCATGGCTGCAGAAAT-GCCCAGTGC-3') and R1 (5'-CATGCCATGGTCTAGATGTGCAGCG-3'). PCR was carried out at 94 °C for 30 s, 63 °C for 30 s, and 72 °C for 1 min for a total of 30 cycles. The products of error-prone PCR were purified, digested, and inserted into *Nco*I and *Eco*RI sites of plasmid pET-28a(+)-*irrE*(W) replacing the wild-type *irrE* gene. The ligation products were transformed into *E. coli* BL21(DE3) competent cells. Cells were plated on LB-agar plates containing 50 mg/mL kanamycin, incubated at 37 °C overnight, and scraped off to create a liquid library. The library size was approximately 10^6 .

The liquid library was inoculated into the challenging medium. After cultivation at 37 °C on a rotary shaker (180 r/min) to the late logarithmic phase, the cultures were transferred into the fresh challenging medium to cultivate for 2–3 times under the same conditions. The unique difference was that the ethanol concentration in the challenging medium was gradually increased until the final concentration reached 8 %. Finally, the cultures were spread onto LB agar plates containing 50 mg/mL kanamycin. After incubation at 37 °C overnight, the grown colonies were

J. Luo et al.

preliminary considered as mutants with ethanol tolerance, which should be further screened.

The single colony on the LB-agar plates was, respectively, inoculated into LB liquid medium containing 50 $\mu g/mL$ kanamycin. After incubation overnight at 37 °C on a rotary shaker (180 r/min), the OD_{600} value of culture was adjusted to the same level and transferred into the fresh challenging medium (LB medium containing 50 $\mu g/mL$ kanamycin, 2 mmol/L IPTG and different concentration of ethanol (4, 8, 12 % (v/v)) using 0.5 % (v/v) of the seed culture. The cell was grown at 37 °C on a rotary shaker (180 r/min). The culture was withdrawn during growth and OD_{600} value was detected. The higher ethanol tolerance mutants were selected by comparison of cell growth rate and the final biomass.

During the above selection process, the plasmid of the mutant with higher ethanol tolerance was verified by PCR and enzyme digest methods, and then retransformed into fresh *E. coli* DH5a cells. The mutant growth in challenging media was retested to separate the effects of the mutated irrE gene from any possible spontaneous chromosomal mutations acquired during the selection process.

6.2.4 Cell Growth Assay

To test the influence of $\it irrE$ mumants on the cell growth, strains were inoculated into LB liquid medium containing 50 $\mu g/mL$ kanamycin. After cultivation overnight at 37 °C on a rotary shaker at 180 r/min, the OD_{600} value of cultures were adjusted to the same level and transferred into the fresh challenging medium (LB medium containing 50 $\mu g/mL$ kanamycin and 2 mmol/L IPTG using 0.5 % (v/v) of the seed culture. The cells were grown at 37 °C on a rotary shaker at 180 r/min. The culture was withdrawn during growth and OD_{600} value was detected to obtain the growth curve.

6.2.5 Shock Experiments and Cell Growth Under Different Stress Conditions

To test the tolerance of strain toward extremely stresses, strains were grown overnight at 37 °C on a rotary shaker at 180 r/min in LB medium containing 50 mg/mL kanamycin. The cultures were transferred into the fresh LB medium containing 50 mg/mL kanamycin using 1 % (v/v) of the seed culture and grown at 37 °C on a rotary shaker at 180 r/min. When reaching an OD_{600} value of 0.6, the cultures were added 2 mmol/L IPTG and inoculated for 6 h to induce the expression of IrrE. Then 1 mL of culture was withdrawn and the cells were harvested by centrifugation at 6,000 r/min for 10 min. The same mass cells were transferred to 50 mL challenging

narooning p21 20d(1) W2(H)								
Stresses	Sorbitol (mol/L)	pН	Methanol (%, v/v)	Ethanol (%, v/v)	H ₂ O ₂ (%, v/v)			
Condition	3.0	pH = 2; pH = 12	10	12	1			
Time (min)	30	30	30	30	30			

Table 6.1 Shock experiment conditions of the control strain (*E. coli* BL21(DE3) harboring plasmid pET-28a(+)), the recombinant strain harboring pET-28a(+)-*irrE*(W) and the mutant harboring pET-28a(+)-*irrE*(M)

medium containing $50 \mu g/mL$ kanamycin and different kinds of stresses (Table 6.1). During incubation, optical density at 600 nm was used to measure cell growth.

To test the cell growth performance in challenging medium, strains were grown overnight at 37 °C on a rotary shaker at 180 r/min in LB medium containing 50 mg/mL kanamycin, respectively. The OD_{600} values of culture were adjusted to the same level and transferred to the fresh challenging medium (LB medium containing 50 µg/mL kanamycin, 2 mmol/L IPTG and different challenging substances (4 % ethanol, 5 % methanol and 1 M sorbitol)) using 0.5 % (v/v) of the seed culture. During incubation, optical density at 600 nm was used to measure cell growth.

6.2.6 Mutational Sites Analysis of IrrE

The mutated *irrE* gene was sequenced and then translated into the amino acids by software DNAMAN. To analyze the influence of the mutant sites on the function and structure of IrrE, the homology modeling was built based on the solved structure of IrrE protein from *Deinoeoccus deserti* by SWISS-MODEL.

6.3 Results and Discussion

6.3.1 Library Construction and Selection of Ethanol-Tolerant Mutants

6.3.1.1 The Development of Error-Prone PCR Conditions

Low concentration of Mg^{2+} is required for Taq DNA polymerase while high level can promote the stability and the complementarity of base pairs. Mn^{2+} can reduce the specificity of the polymerase to the template, which is advantageous for mutation. Therefore, error-prone PCR conditions were optimized by adjusting the concentration of Mg^{2+} (1–7 mmol/L) and Mn^{2+} (0.05–0.25 mmol/L). When the concentration of Mg^{2+} was set at 3 mmol/L, all stripes were clear and bright, and no

58 J. Luo et al.

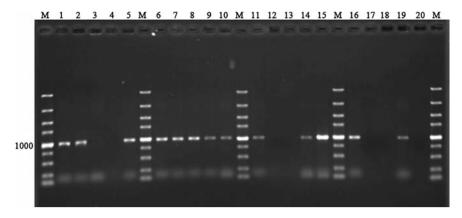


Fig. 6.1 Error-prone PCR conducted with different concentrations of $\mathrm{Mn^{2+}}$ and $\mathrm{Mg^{2+}}$. M DL5000 DNA Marker. I-5 The concentration of $\mathrm{Mn^{2+}}$ is 0.05, 0.1, 0.15, 0.2, 0.25 mmol/L, respectively, at $\mathrm{Mg^{2+}}$ is 1 mmol/L. 6-10 The concentration of $\mathrm{Mn^{2+}}$ is 0.05, 0.1, 0.15, 0.2, 0.25 mmol/L, respectively, at $\mathrm{Mg^{2+}}$ is 3 mmol/L. 11-15 The concentration of $\mathrm{Mn^{2+}}$ is 0.05, 0.1, 0.15, 0.2, 0.25 mmol/L, respectively, at $\mathrm{Mg^{2+}}$ is 5 mmol/L. 15-20 The concentration of $\mathrm{Mn^{2+}}$ is 0.05, 0.1, 0.15, 0.2, 0.25 mmol/L, respectively, at $\mathrm{Mg^{2+}}$ is 7 mmol/L

obvious dispersion was observed under different concentrations of Mn²⁺ (Fig. 6.1). So 3 mmol/L of Mg²⁺ was used for the error-prone PCR.

6.3.1.2 Library Construction

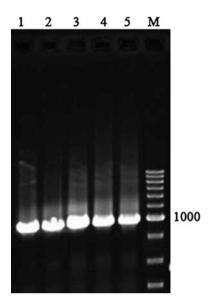
Error-prone PCR was employed to construct the library using the plasmid pET-28a (+)-irrE(W) as a template. PCR was carried out under different Mn²⁺ concentrations (0.05–0.25 mmol/L) and 3 mmol/L Mg²⁺. As presented in Fig. 6.2, the target band was clear and bright. The size of irrE was about 1,000 bp [8]. Therefore, IrrE library (about 10⁶) was constructed by using the above PCR conditions and the detailed procedures are described in Sect. 6.2.3.

6.3.1.3 Selection of Ethanol-Tolerant Mutants

The liquid library was screened according to the detailed procedures described in Sect. 6.2.3. After incubation in the challenging medium (LB medium containing 50 μ g/mL kanamycin, 2 mmol/L IPTG and 8 % ethanol), the cultures were spread onto LB agar plates containing 50 mg/mL kanamycin. After incubation at 37 °C overnight, 24 single colonies (named M1-M24) were grown and preliminary considered as mutants with higher ethanol tolerance.

24 single colonies were respectively inoculated into LB medium and the cell growth curves under different ethanol stresses were determined according to the detailed procedure shown in Sect. 6.2.3. By comparison of cell growth rate and the

Fig. 6.2 Error-prone PCR conducted with different Mn²⁺ concentrations. *M* DL500 DNA marker. *I*–5 The concentration of Mn²⁺ is 0.05, 0.1, 0.15, 0.2, 0.25 mmol/L, respectively, at Mg²⁺ is 3 mmol/L



final biomass, four mutants named M1-M4 were screened and grown well in the presence of 4 % ethanol (Fig. 6.3a). When the ethanol concentration was increased to 8 %, one mutant (M4) showed good growth performance and a maximum OD_{600} of 1.6 was reached after 28 h (Fig. 6.3b). Further increase in ethanol concentration (12 %) and no obvious cell growth was observed in all mutants. So mutant M4 was considered as the better ethanol-tolerant strain.

The cell growth performance of the control strain ($E.\ coli\ BL21(DE3)$) harboring plasmid pET-28a(+)), the recombinant strain harboring pET-28a(+)-irrE(W), and the mutant M4 harboring pET-28a(+)-irrE(M4) were compared in LB medium containing 8 % ethanol. As shown in Fig. 6.4, the cell growth of the control strain harboring plasmid pET-28a(+) and recombinant strain harboring pET-28a(+)-irrE(W)) were completely inhibited, while the irrE mutant M4 strain showed rapid growth after a 20 h lag phase. The maximum OD_{600} reached 1.6 after 28 h cultivation.

6.3.2 The Influence of Expression of IrrE on Cell Growth Under No Stress

As shown in Fig. 6.5, the lag phase of the control strain (*E. coli* BL21(DE3) harboring plasmid pET-28a(+)) and the recombinant strain harboring pET-28a(+)-*irrE*(W) under no stress condition were 0–8 h, while the lag phase of the *irrE* mutant M4 was 2 h shorter than the other two strains. After 14 h cultivation, the

J. Luo et al.

Fig. 6.3 The growth curves of *E. coli* BL21/pET-28a(+)-*irrE*(M1-M4) under various concentrations of ethanol. **a** 4 % ethanol. **b** 8 % ethanol

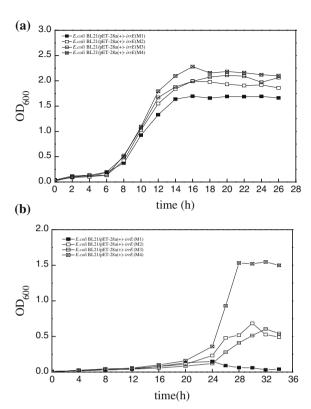
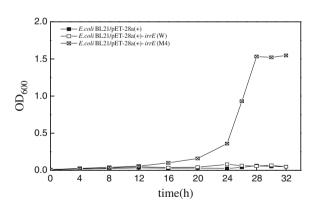


Fig. 6.4 The growth curves of *E. coli* BL21/pET-28a(+)-*irrE*(M4), *E. coli* BL21/pET-28a(+)-*irrE*(W) and the control strain under 8 % ethanol



final biomass of the recombinant strain harboring pET-28a(+)-irrE(W) and the irrE mutant M4 was silimilar and 10 % more than the control strain harboring plasmid pET-28a(+). This indicated that the expression of irrE shortened the cell lag phase and increased the final biomass.

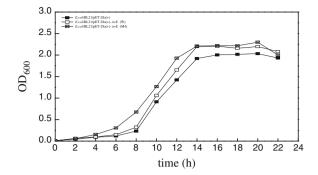


Fig. 6.5 The growth curves of *E. coli* BL21/pET-28a(+)-*irrE*(M4), *E. coli* BL21/pET-28a(+)-*irrE* (W) and the control strain under no stress condition

6.3.3 The Influence of Expression of IrrE on Cell Viability and Growth Performance Under Different Stress Conditions

6.3.3.1 The Influence of Expression of IrrE on Cell Viability Under Stress Shock

After shocked with 1 % H_2O_2 and 10 % ethanol for 30 min, the cell viability of the mutant M4 was less than 15 %, which was higher than those of the control strain and wild type strain. For other shock conditions (pH = 5, pH = 10, 15 % methanol and 3 M sorbitol), the cell viabilities of mutant M4 were 56.9, 70, 55, 51.6 %, respectively, which increased by 25.0, 30.4, 17.5, and 6.7 %, respectively, compared with the wild type strain and increased by 36.4, 45.4, 35.0, and 16.0 % compared with the control strain (shown in Fig. 6.6). These results indicate that the expression of mutant *irrE* could remarkably enhance the tolerance of *E. coli* to stresses of 3 M sorbitol, pH = 5, pH = 10, and 15 % methanol.

6.3.3.2 The Influence of Expression of IrrE on Cell Growth Performance Under Different Stress Conditions

In the presence of 4 % ethanol, the lag phase of the control strain and the wild-type stain was 10 h, while the lag phase of the *irrE* mutant M4 was shortened to 6 h and its cell growth rate and final biomass were obviously increased, indicating that the IrrE mutation enhanced the strain tolerance to 4 % ethanol (Fig. 6.7a). Under the other stress conditions (5 % methanol and 1 M sorbitol), the control strain, the wild-type stain and the *irrE* mutant M4 showed the similar growth tendency. However, the *irrE* mutant M4 showed higher growth rate and final biomass. In the presence of 5 % methanol, the *irrE* mutant M4 reached a maximum OD₆₀₀ of 2.2 after 18 h

62 J. Luo et al.

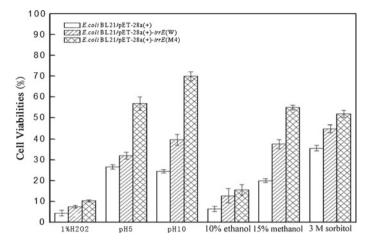


Fig. 6.6 The cell viability of *E. coli* BL21/pET-28a(+)-*irrE*(M4), *E. coli* BL21/pET-28a(+)-*irrE* (W) and the control strain under various kinds of shock conditions

cultivation, which was 18 and 25 % higher than the control strain and wild-type strain, respectively (Fig. 6.7b). In the presence of 1 M sorbitol, the irrE mutant M4 reached a maximum OD_{600} of 2.0 after 16 h cultivation, which was 17 and 42 % higher than the control strain and wild-type strain, respectively (Fig. 6.7c). These results indicate that the expression of the mutated IrrE could promote host cell tolerance to organic solvents and osmotic pressure.

6.3.4 The Mutation Site Analysis of IrrE

The *irrE* sequence of mutant M4 was analyzed. The whole length of *irrE* was 987 bp, encoding 328 amino acids. The comparison result between the wild-type and mutated *irrE* showed that there were two sense substitutions (C24T and G530A). The substitution of G530A caused one amino acid change Gly177Glu.

The IrrE protein structure of *D. deserti* has been solved, which shows 64 % homology with the IrrE protein from *D. radiodurans*. Therefore the homology structural model of IrrE from mutant M4 was built using SWISS-MODEL according to the known structure of IrrE protein from *D. deserti*. The domain boundary of IrrE from *D. radiodurans* was as follows: the N-terminal domain (residues 1–161); the middle domain (residues 162–203); and the C-terminal domain (residues 204–328) (Fig. 6.8a). The results show that the mutational amino acid was located in helix-turn-helix (HTH) motif of IrrE, which was considered to be involved in DNA binding and recognition [10] (Fig. 6.8b).

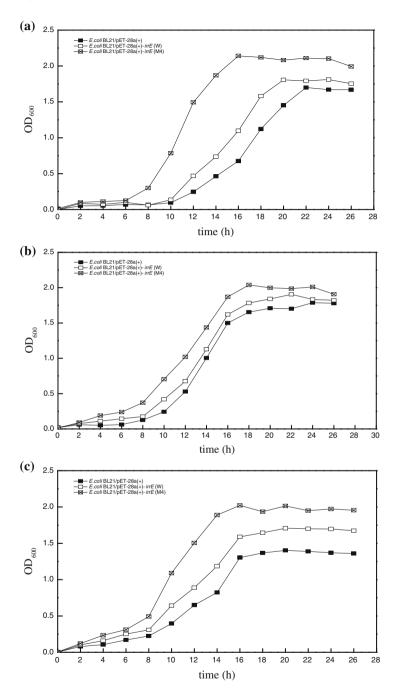


Fig. 6.7 The growth curves of *E. coli* BL21/pET-28a(+)-irrE(M4), *E. coli* BL21/pET-28a(+)-irrE(W) and the control strain under certain concentration of stress conditions. **a** 4 % ethanol. **b** 5 % methanol. **c** 1 M sorbitol

J. Luo et al.

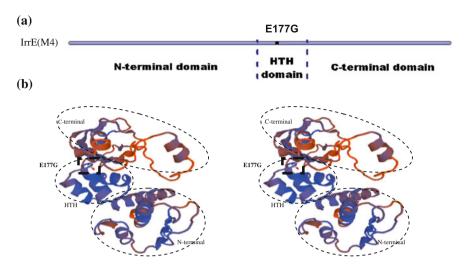


Fig. 6.8 Summary and location of mutation sites of IrrE. **a** Summary of mutation site in the amino acid sequence of the mutant M4, the full sequence is divided into three parts by two *dashed lines*. Each part represents a separate domain of *irrE* (from *left* to *right* N-terminal domain, HTH domain, C-terminal domain). **b** Location of mutation sites of the mutant M4 in a modeled structure of *irrE* from *D. radiodurans*

6.4 Conclusions

The *irrE* gene from *D. radiodurans* R1 was directed evolved by error-prone PCR method and was introduced into *E. coli*. One mutant M4 with better stress tolerance was obtained, which showed good cell growth performance in the presence of 8 % ethanol. Also, the cell viabilities of mutant M4 were promoted under extremely stress shock conditions (pH = 5, pH = 10, 15 % methanol and 3 M sorbitol). On the other hand, mutant M4 exhibited higher growth rate and final biomass in the presence of challenging medium (4 % ethanol, 5 % methanol and 1 M sorbitol) compared with the control and wild type strains. These results indicate that the expression of mutational IrrE could further improve the tolerance of *E. coli* to various stresses.

The comparison result between the wild-type and mutated *irrE* showed that there were two sense substitutions (C24T and G530A). The substitution of G530A caused one amino acid change, Gly177Glu. The homology modeling of IrrE was constructed using the known IrrE from *D. desertias* as a template, indicating the mutational amino acid was located in HTH motif of IrrE, which was thought to be involved in DNA binding and recognition [10].

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Chapter 7 Cloning and Characterization of the *TMEPAI* Gene Promoter

Ailong Guo, Ping Qin, Weiwei Shi, Yuyin Li and Aipo Diao

Abstract *TMEPAI* was originally identified as a highly androgen-induced gene by serial analysis of gene expression in androgen-treated prostate cancer cell. *TMEPAI* is highly expressed in many tumor cells. However, little is known about the transcriptional mechanism regulating *TMEPAI* gene expression, and its promoter has not yet been characterized. In this study, the 5' regulatory region of the *TMEPAI* gene was characterized, the 5' flanking sequence of *TMEPAI* gene was successfully amplified by PCR, and the 925 bp fragment of *TMEPAI* gene promoter was inserted into pGL4-Basic vector to measure its activity. Furthermore, the transcriptional regulators in *TMEPAI* promoter region were analyzed by bioinformatics. Taken together, these results will help to understand the expression regulation of *TMEPAI* and its role in tumorigenesis.

Keywords TMEPAI • Promoter • Bioinformatics

7.1 Introduction

The gene encoded transmembrane prostate androgen-induced protein (*TMEPAI*) is located on the chromosome 20q13.31-q13.33, which was discovered during systematic search for androgen-induced genes [1]. TMEPAI protein is highly expressed in many tumor including prostate, breast, and colon cancers [2–4]. TMEPAI protein is predominantly expressed in the prostate gland and is directly regulated by androgen receptor [5]. *TMEPAI* gene encodes a protein of 287 amino acids with a

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A. Guo et al.

trans-membrane Ib domain, and two PY motifs that can interact with the WW domain of the E3 ubiquitin ligase Nedd4 [6]. Structural analysis of the *TMEPAI* protein indicates similarities to the Nedd4 ubiquitin ligase binding protein coded by the N4wbp4 gene suggesting a role for *TMEPAI* in the ubiquitin-proteasomal pathway. Assessment of *TMEPAI* functions suggests specific interaction between *TMEPAI* and Nedd4 uniquitin E3 ligase [7–10]. *TMEPAI* has been reported highly expressed in many cancers, overexpression of *TMEPAI* promoted cell proliferation of prostate cell, and knockdown of *TMEPAI* expression decreased the growth of lung cancer cell Calu3 [11–14].

In this article, in order to analyze the transcriptional regulation of *TMEPAI* expression, the promoter of the *TMEPAI* gene was cloned and analyzed. This study provides the basis to understand the transcription regulation of *TMEPAI* gene, and further considers the regulatory mechanism will help to develop therapeutic reagents for human cancers using *TMEPAI* as the target.

7.2 Materials and Methods

7.2.1 Genomic DNA Isolation and Cloning of the TMEPAI Promoter

Genomic DNA was extracted from human blood using a genomic DNA extraction kit (Promega, Madison, WI). Primers were designed based on the sequence of the 5′ untranslated region of *TMEPAI* (Gen Bank ID: 56937). Primers for amplification of promoter were: forward primer 5′-CGGGGTACCAAGGCGACACACTAAATTTT TAAGAG-3′ and reverse primer 5′-CCGCTCGAGCGCTCCGAGACCGCGGT-3′. The isolated genomic DNA was used as the template to amplify the *TMEPAI* promoter region. *KpnI* and *XhoI* restriction sites were introduced into the forward and reverse primers, respectively. The PCR product was purified using a gel purification kit and cloned into the T/A cloning vector pUCm-T (Shanghai Sangon Biological Engineering Technology, Shanghai, China). *TMEPAI* promoter fragment was cloned into the pGL4 basic vector (Shanghai Sangon Biological Engineering Technology, Shanghai, China). The 925 bp 5′ blank region of *TMEPAI* gene was generated by digesting the plasmid pT/A-925 with *KpnI* and *XhoI*, and subcloned the released fragment into the *KpnI/XhoI* site of pGL4 basic vector to obtain plasmid pGL4-925. The positive clones of pT/A-925 were isolated and sequenced.

7.2.2 Cell Culture and Transfection

Human hepatic carcinoma cell line (HepG2), human alveolar epithelial lung cell line (A549), human breast cancer cell line (MDA-MB231), Hela cells, human

gastric cancer cell line (MGC-803), and human prostate cancer cell line (DU145) were cultured in Dulbecco's modified Eagle' medium (DMEM) containing 10 % fetal bovine serum (FBS), 2 mM glutamine, 1,000 U/mL penicillin, and 0.01 mg/mL streptomycin at 37 °C in a humidified incubator with 5 % carbon dioxide. Cells were transfected using Turbofect (Thermo Scientific) according to manufacturer's recommendations. 10⁵ cells were seeded in each well of 12-well plate and cultured for 24 h before transfection. The cells were transfected with 1 µg of *TMEPAI* promoter construct or pGL4 basic vector.

7.2.3 Analysis of Luciferase Activity

Luciferase activity was measured 48 h after transfection. Cells were harvested and washed twice with PBS, then lysed in cold lysis buffer (25 mmol/L glycylglycine, pH 7.8; 1 % Triton X-100; 15 mmol/L MgSO₄; 4 mmol/L EGTA, and 1 mmol/L dithothreitol). For luciferase assay, 100 μL 25 mmol/L luciferin and 5 μL assay cocktail (1 mol/L adenosine triphosphate; 15 mmol/L KH2PO4, pH 7.8; 15 mmol/L MgCl₂) were added to 45 μL cell lysate, and the luiferase activities were measured using a FLUOstar OPTIMA fluorescence reader.

7.2.4 Bioinformatics Analysis of TMEPAI Promoter Region

The putative binding sites for transcription factors with 5'-regulatory region of *TMEPAI* gene were analyzed with the following software packages: Match 1.0 public (http://www.gene-regulation.com), and TRANSFAC database (http://www.gene-regulation.com/pub/databases.html). Transcription start sites were identified by using DBTSS (http://www.biomedsearch.com/sci/Dbtss.com) and BDGP (http://www.Fruitfly.org/seq-tools/promoter.html).

7.2.5 Statistical Analysis

Experiments were repeated at least thrice, with two replicates per sample for each experiment. The data were analysis of variance using the Student's *t*-test. Values of p < 0.05 were considered to indicate statistical significance.

70 A. Guo et al.

7.3 Results

7.3.1 Identification of the Upstream Promoter Region of TMEPAI Gene

To better understand the mechanism involved in *TMEPAI* gene expression, the promoter region of *TMEPAI* was cloned and analyzed. A blast search using *TMEPAI* (Gene ID: 56937) as query was performed online (basic local alignment search tool: BLAST; http://blast.ncbi.nlm.nih.gov/Blast.cgi) and revealed the 5'-upstream sequence of the *TMEPAI* gene.

A genomic DNA fragment that spans positions -761 to +164 relative to the transcriptional initiation site about 925 bp (Fig. 7.1) was amplified by PCR. The fragment was then cloned into the T/A cloning vector pUCm-T, and sequenced.

7.3.2 Construction of TMEPAI Promoter Plasmids

The pUCm-T-925 plasmid was digested with *Kpn*I and *Xho*I, and then the promoter region of *TMEPAI* was subcloned into pGL4-Basic vector, which was digested with the same restriction enzymes. The reporter plasmid was named pGL4-925. The recombinant plasmid was identified by enzymes digestion and PCR (Fig. 7.2).

7.3.3 Analysis of TMEPAI Promoter Activities

In order to analyze *TMEPAI* gene promoter activities, pGL4-925 and pGL4-Basic were transfected into six cell lines including HepG2, A549, MDA-MB231, HeLa, MGC-803, and DU145. We observed that the highest *TMEPAI* promoter activity was found in HeLa cells (Fig. 7.3). Other cells showed lower promoter activity, and

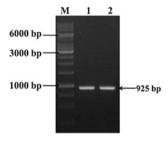


Fig. 7.1 PCR was performed to amplify *TMEPAI* gene promoter. PCR product was separated on 0.8 % agarose gel. *M* 1 kb ladder DNA marker. *I–2 TMEPAI* gene promoter

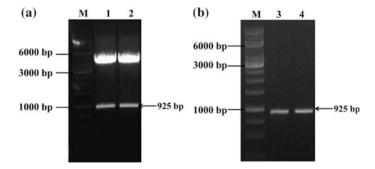


Fig. 7.2 Identification of the recombinant plasmids with *KpnI* and *XhoI* digestion (a) and PCR (b) *M* 1 kb ladder DNA marker. *I*–*4 TMEPAI* gene promoter

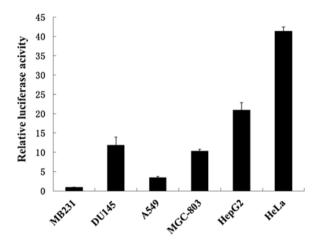


Fig. 7.3 Determination of the *TMEPAI* promoter activity in different cell *lines* by luciferase assay pGL4-925 and pGL4-Basic were transiently transfected into six different cell *lines*. After 48 h transfection, cell lysates were prepared and assayed for luciferase activity. Relative pGL4-Basic activity of representative experiment is shown

the MDA-MB-231 cells showed the lowest promoter activity. Therefore, pGL4-925 can be used to screen anticancer reagents with the *TMEPAI* promoter as the target.

7.3.4 Bioinformatic Analysis of TMEPAI Promoter Region

TFSEARCH database was used to analysis the binding sites for transcriptional regulators in *TMEPAI* promoter. The promoter does not contain a TATA box, however, a CAAT box and a GC box were identified, which are both often encountered in TATA-less promoters and they contribute to the correct RNA polymerase positioning. In addition, the sequence search in *TMEPAI* promoter

72 A. Guo et al.

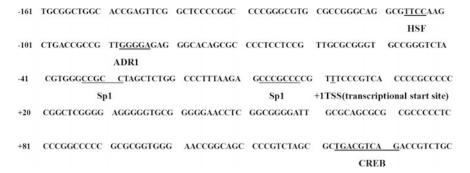


Fig. 7.4 Nucleotide sequence of the *TMEPAI* gene promoter region. The numbering of the sequence is relative to the TSS. Putative binding sites for the transcriptional factors are *underline* and *labeled* below. Transcriptional start site was predicted with DBTSS database analysis

region revealed several regulatory elements such as the binding sites for transcriptional factors Sp1, CREB, ADR1, HSF, and AP-2 (Fig. 7.4).

7.4 Discussion

TMEPAI was identified originally as a highly androgen-inducible gene with prostate-abundant expression that was restricted to prostatic epithelial cells [15]. Studies of TMEPAI have suggested its role in regulating cell growth as noted by the androgen induction of TMEPAI expression, elevated TMEPAI expression in nontumorigentic revertants of tumor cell lines, and the TMEPAI expression alterations (up- or down-regulation) in human tumors [1, 16]. Functional analysis indicates that TMEPAI is a Nedd4 E3 ligase binding protein and plays a role in down-regulation of androgen receptor through a negative feedback loop between androgen receptor and TMEPAI [15]. Studies also suggest that TMEPAI is involved in other cancers through regulation of TGF-β, PI3K and WNT pathway [17, 18].

A 925 bp fragment of *TMEPAI* promoter region was amplified from human genomic DNA by PCR (Fig. 7.1). Transcriptional regulation is the result of the balance between chromatin remodeling and RNA polymerase activity on accessible regulatory regions [19], the consensus recognition sequences for several transcription factors in *TMEPAI* promoter were analyzed by the TRANSFAC database (http://www.gene-regulation.com/pub/databases.html), and some of them are transcription factors including Sp1, CREB, ADR1, HSF, and AP-2 (Fig. 7.4). These results serve as a basis for future studies to understand the regulation of *TMEPAI* expression at transcriptional level.

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Chapter 8 Cloning and Expression of a Novel Xvlanase Xvn11-1 from Alkaline Soil

Kun Li, Zhongyuan Li, Xuegang Luo, Cuixia Feng, Cuigiong Wang, Minghui Zhang and Tongcun Zhang

Abstract A novel xylanase of family 11 (Xyn11-1) was obtained from the metagenomic DNA of alkaline soil by touchdown-PCR and thermal asymmetric interlaced (TAIL) PCR methods. Xvn11-1 is composed of 645 nucleotides, which encodes a signal peptide of 19 amino acids and a mature protein of 196 amino acids. It is a novel GH11 xylanase sharing the highest identity (79 %) with the reported GH11 xylanase (XP 008721536) in GenBank database. In order to detect its biological activity, the recombinant plasmid xyn11-1-pET28a(+) was constructed and recombinant protein was successfully expressed in heterologous hosts Escherichia coli BL21 (DE3) induced by isopropy-beta-p-thiogalactopyranoside (IPTG). Using the 3,5-dinitrosalicylic acid (DNS) method, the xylanase activity of crude intracellular protein is 11.32 U/mL. The optimal inducing IPTG concentration and inducing temperature was examined at 0.2 mM and 15 °C, respectively.

Keywords Xylanase • Alkaline soil • Gene cloning • Recombinant expression • Optimization

8.1 Introduction

Hemicelluloses are the second most abundant plant polysaccharides in nature. Due to their potential role as sustainable energy sources, hemicelluloses are increasingly becoming an important concern [1]. Xylan is the major carbohydrate of hemicelluloses, its complete degradation requires the synergistic action of a variety of

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enzymes, including xylanase, β -xylosidase, α -glucuronidase, α -arabinofuranosidase, and acetylxylan esterase [2]. Among them, xylanase plays a crucial role in the hydrolysis of the xylan backbone [3], in which xylanase cleaves the β -1,4-glycosidic bond between xylose residues to release xylooligosaccharides. For some biotechnological applications, xylanase is useful, especially in the papermaking, bioconversion, and food industry [4]. Due to the increasing demand for xylanase production, exploring new enzyme sources has been a hot topic in recent years [5]. Therefore, we try to explore some novel alkaline xylanase from alkaline soil, in this study, a novel xylanase of family 11 (*Xyn11-1*) was obtained from the metagenomic DNA of alkaline soil, and it was heterologously expressed in *Escherichia coli* BL21 (DE3).

8.2 Materials and Methods

8.2.1 Strains, Plasmids, and Culture Conditions

Escherichia coli DH5α and the vector PMD-19T (TaKaRa, Dalian, China) were used for gene cloning. *E. coli* BL21(DE3) and vector pET28a(+) obtained from Invitrogen (Carlsbad, CA, USA) were used for gene expression. *E. coli* DH5α and *E. coli* BL21(DE3) was grown in LB medium (10 g/L tryptone, 5 g/L yeast extract, and 10 g/L NaCl) supplemented with 50 mg/L ampicillin or kanamycin, respectively, at 37 °C with shaking at 220 rpm.

8.2.2 Alkaline Soil Sampling and Metagenomic DNA Extraction

Alkine soil was obtained from the coastal saline area in Tianjin (China) and stored in -20 °C. The pH was measured by pH meter (SevenEasy, Shanghai). About 1 g soil sample was first grinded by liquid nitrogen, and the metagenomic DNA was extracted by CTAB-SDS method [6]. The crude DNA was purified by DNA purification kit (Solarbio, Beijing, China), and was detected by electric.

8.2.3 Cloning and Sequence Analysis of Xylanase Gene Xyn11-1

The xylanase gene *xyn11-1* was obtained by touchdown-PCR and TAIL-PCR methods. First, the conserved sequence of *xyn11-1* was amplificated by Touchdown-PCR with the conserved primer GH11F and GH11R according to Wang et al. [7].

Primers	Primer sequence (5′–3′)
X11-F	AACTGCTACCTGKCNITNTAYGGNTGG
X11-R	CCGCACGGACCAGTAYTGNKIRAANGT
11-1sp1D	TACCATCGATCGAGGGCTGGTTGGTGC
11-1sp1U	CCTCGTCGAGTACTACGTGATTGAATCGTACG
11-1sp2D	ACGTAGATGTTGTAGGTTCCGCCGTCGG
11-1sp2U	GTCGAGTACTACGTGATTGAATCGTACGGCAC
11-1sp3D	CGTACGATTCAATCACGTAGTACTCGACGAGG
11-1sp3U	CACCAACCAGCCCTCGATCGATGGTAC
11-1-28-F	ATGCgaattcCGCCCATTCGACTTCCTCGACGATATCG
11-2-28-R	GGGctcgagaGTGGGTTTGCACGTACACTTCGGACTCG

Table 8.1 Primers used in this work

The letters in lowercase were the endonuclease site

The nucleotide fragment was sequenced by GENEWIZ, Beijing, China. The full gene sequence was amplificated and the six specific primers designed based on the conserved sequence using Tail-PCR method. The primers used in the study are shown in Table 8.1. The sequence was assembled by Vector IN7.0 (Invitrogen), and was blasted by BLASTX (http://blast.ncbi.nlm.nih.gov/Blast.cgi). The accessory number of the protein sequence of Xyn11-1 on NCBI was KM893863.

8.2.4 Heterologous Expression of Xyn11-1

The full-length gene sequence of xyn11-1 was digested with EcoRI and XhoI, ligased with pET28a(+) vector, and transformed into $E.\ coli$ DH5 α competent cells. The correct recombinant plasmid xyn11-1-pET28a(+) was transformed into $E.\ coli$ BL21(DE3) competent cells. The clone was incubated in LB medium for OD₆₀₀ reach to 0.6, and then the clone was induced at 25 °C using 1 mM IPTG for 20 h. The cells was collected by centrifugation at 7,000 g, 4 °C for 10 min, and then dissolved by sonication in phosphate buffer (8 g/L NaCl, 3.58 g/L Na₂H-PO₄·12H₂O, 0.2 g/L KCl, 0.27 g/L KH₂PO₄). Cell debris and supernatant was collected to SDS-PAGE analysis, respectively.

8.2.5 Enzyme Activity Assays

The 3,5-dinitrosalicylic acid (DNS) method was used to assay xylanase activity [8]. One unit of xylanase activity was defined as the amount of enzyme that released 1 μ mol of reducing sugar from the substrate equivalent to xylose per minute under the assay conditions.

78 K. Li et al.

8.3 Results

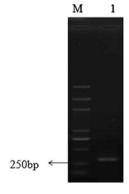
8.3.1 Cloning and Sequence Analysis of the Xylanse Gene Xyn11-1

With the degenerate primers by touchdown PCR, about 250 bp DNA fragment was amplificated as shown in Fig. 8.1. The conserved sequence of *xyn11-1* was sequenced and show 87 % identity with the reported GH11 xylanase (XP_008721536). The flanking sequences were amplicated by Tail-PCR. After sequence assembling, the ORF of xylanase *Xyn11-1* was obtained which contains 645 bp in length and 19 amino acid signal peptide and 196 aa mature protein. The protein sequence of *Xyn11-1* shared 79 % identity with endo-1,4-beta-xylanase from *Cyphellophora europaea* CBS 101466 (ETN36718).

8.3.2 Expression of Recombinant Enzyme

The structure of recombinant plasmid *xyn*11-1-pET28a(+) is shown in Fig. 8.2. *Xyn*11-1 was successfully expressed in *E. coli*, and its activities in cellular and extracellular fractions were compared. Major portions of the total activity were detected in the intracellular, because there were no signal peptides in the expression vector or in the recombinant protein sequence. The recombinant enzymes showed the apparent molecular masses of approximately 30.0 kDa for *Xyn*11-1 in SDS–PAGE gel (Fig. 8.3), which was consistent with their calculated molecular masses. Using the DNS method, the xylanase activity of crude intracellular protein is 11.32 U/mL.

Fig. 8.1 Conserved region of Xyn11-1 by touchdown PCR. *M* DM2000 marker; *I* GH11 xylanase conserved region from the metagenomic DNA of alkaline soil



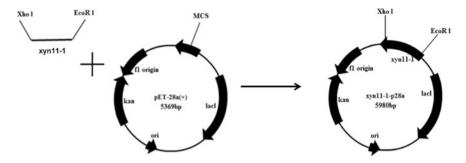
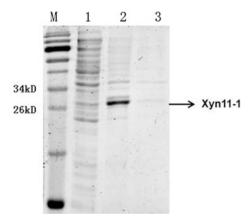


Fig. 8.2 The structure of recombinant plasmid xyn11-1-pET28a(+)

Fig. 8.3 SDS–PAGE analysis for the expression of *Xyn11-1*. *M* protein marker; *1* cell sonication debris; 2 cell sonication supernatant; *3* LB culture medium



8.3.3 Optimization of IPTG Concentration

IPTG is an expensive chemical for industrial production. Using excess IPTG will cause unnecessary waste and, meanwhile, it is poisonous to cells. So the IPTG concentration is also very crucial for the expression level of fusion protein *Xyn11-1*. 0.2, 0.4, 0.6, 0.8, 1 mM were used in this study, and the results suggest that the optimal IPTG concentration for the expression of the fusion protein was 0.8 mM (Fig. 8.4).

8.3.4 Optimization of Inducing Temperature

Escherichia coli usually grows better at temperatures of 37–39 °C, but this temperature is not suitable for the recombinant enzyme, because it might form inclusion bodies under these temperatures. In order to find the optimal temperature for the

K. Li et al.

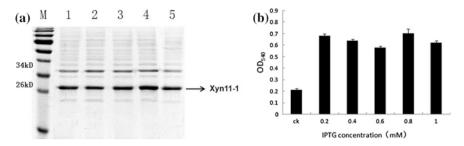


Fig. 8.4 Optimization of IPTG concentration for the expression of Xyn11-1. **a** SDS-PAGE analysis with different IPTG concentrations: (*M*) protein marker, (*I*) 0.2 mM, (2) 0.4 mM, (3) 0.6 mM, (4) 0.8 mM, (5) 1 mM. **b** Enzyme activity asays with different IPTG concentrations

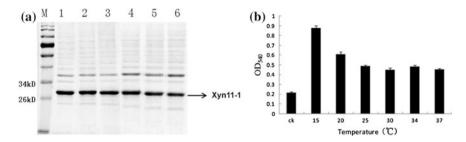


Fig. 8.5 Optimization of temperature for the expression of *Xyn11-1*. **a** SDS–PAGE analysis with different inducing temperature: (*M*) protein marker, (*I*) 15 °C, (2) 20 °C, (3) 25 °C, (4) 30 °C, (5) 34 °C, (6) 37 °C. **b** Enzyme activity assays under different inducing temperature

soluble expression protein, several different temperature grades were under experiment, including 15, 20, 25, 30, 34, 37 °C. The results under different temperature conditions are shown in Fig. 8.5 and implies that 15 °C was the most suitable temperature for *Xyn11-1*.

8.4 Discussion

Xylanase is the most critical xylan degrading glycoside hydrolase, xylan could be degraded into oligosaccharides, which could be further degraded to xylose by glucosidase and other side-chain hydrolase. In this study, a novel GH11 xylanase Xyn11-1 was obtained, and the recombinant plasmid Xyn11-1-pET28a(+) was successfully constructed and expressed in $E.\ coli.$ Meanwhile, we carried out the work to optimize the inducing conditions for the soluble expression of Xyn11-1. The results showed that the optimal IPTG concentration and inducing temperature was examined at 0.2 mM and 15 °C. The xylanase activity of crude intracellular protein is 11.32 U/mL.

As the data showed, the enzyme activity of *Xyn11-1* in the prokaryotic expression system was not as satisfactory as expected. This is mostly because the novel xylanase *Xyn11-1* was from fungal genome. And as is shown in many studies [9], GH11 xylanases are usually expressed better in *Pichia methanolica* Expression System. In view of the work we have done, now we are structuring *P. methanolica* Expression System and characterizing the recombinant protein.

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Chapter 9 To Establish the Regeneration System of Sweet Sorghum Immature Embryos

Xiaomu Chen, Oujing Li, Lili Shi, Xianhua Wu, Boxian Xia and Zhongyou Pei

Abstract Sweet sorghum is an important forage and energy crop. In this experiment, sweet sorghum immature embryos were taken to be explants, through inducting and regenerating for sweet sorghum callus of different genotypes, to obtain the best culture medium formula and create appropriate conditions for establishing efficient regeneration system. The result showed that, for callus induction, the response of immature embryos varied with the genotype, but the medium is not. Of these, Xinliang 52 showed the highest frequency of callus induction under appropriate culture conditions. As regards regenerable callus formation, the response of callus varied with the genotype and medium. 07-27 regeneration occurred at high frequencies when MSR and MBR were added in the regeneration medium. And there is nothing on NBI medium for the whole genotypes. So NBI is not suitable for callus regeneration culture.

Keywords Sweet sorghum \cdot Immature embryo \cdot Tissue culture \cdot Medium \cdot Genotype

9.1 Introduction

Sweet sorghum is a variant of ordinary sorghum. It has affluent stem juice, high sugar content, and efficient ethanol conversion frequency. It was used to make silage and process sugar and ethanol. At the same time, sweet sorghums are a kind of green energy crops, which have such attractive prospects in the field of renewable energy that it has been closely highlighted by experts and scholars

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X. Chen et al.

worldwide [21]. But sweet sorghum seed resource is deficient, traditional breeding application is limited, and as a result, the process of variety improvement and new variety breeding is slow [23]. Therefore, the biological technology assisted breeding is taken seriously, of which the most important is tissue culture. Plant tissue culture technology is not restricted by the growth season. The genetic transformation of sweet sorghum could be achieved in a relatively short period of time, so as to enrich seed resource. However, sweet sorghum is considered to a plant that has difficulty with plant tissue culture, and is yet to establish efficient regeneration system in vitro.

Sorghum immature embryos [4–6, 12, 16], stem tips [18, 22], anther [8, 14, 19], spikes [1, 17, 20], mature embryos, mature seeds and mature seeds germination buds [2, 3, 9, 11, 24] in vitro have all obtained regeneration plants. In addition, sorghum immature embryos were considered to be the best explants for tissue culture. Gamborg et al. [6] were the earliest researchers to study sorghum immature embryo tissue cultivation. They used immature sorghum embryos, which are harvested 12–18 days after pollinating and are cultured in vitro. They obtained the regeneration shoots, and finally the mature regeneration plants. Dunstan et al. [4], Elkonin [5], Ma and Liang [12], Shi et al. [16], Wang et al. [20] obtained regeneration plants and Fuguang Han obtained callus by using the sorghum immature embryos. What is more, Liu et al. [10] achieved the transgenic plants of sorghum immature embryos during the genetic transformation study.

In this experiment, we regarded immature embryos of ten sweet sorghum varieties as explants, studying different mediums and different genotypes and whether they had effect on sweet sorghum callus induction and regeneration or not, so that we can establish a highly efficient regeneration system of sweet sorghum.

9.2 Materials and Methods

9.2.1 Materials

The varieties were sweet sorghum materials with good agronomic traits. Xinliang 52, BJ-285, 07-27, BJ-299, Rome, M81-E, Cowley, Tianza2, Liaotian3, and Sanrio were provided by the Agronomy department of Tianjin Agricultural University, and were planted in the Crop Arboretum of East Campus of Tianjin Agricultural University.

Constant temperature and humidity box were bought from MMM Group; plant growth chamber were bought from BINDER company; tissue culture box were bought from PERCIVAL company.

9.2.2 Medium

9.2.2.1 Induction Medium

$$\begin{aligned} \text{MSI} &= \text{MS medium} + 0.5 \, [\text{g/L}] \text{CH} + 1.0 \, [\text{g/L}] \text{Pro} + 30 \, [\text{g/L}] \text{Sur} \\ &+ 2.0 \, [\text{mg/L}] 2, \text{4-D}. \end{aligned}$$

MBI = MS majority + MS minority + B₅ organic + 0.5 [g/L]CH + 1.0 [g/L]Pro +
$$30$$
[g/L]Sur + 2.0 [mg/L]2,4-D.

$$\begin{split} NBI = & N_6 \text{ majority} + B_5 \text{ minority} + B_5 \text{ organic} + 0.5 \text{ [g/L]CH} \\ & + 1.0 \text{ [g/L]Pro} + 30 \text{[g/L]Sur} + 2.0 \text{ [mg/L]2,4-D.} \end{split}$$

9.2.2.2 Subculture Medium

$$MSS = MS \text{ medium} + 0.5 [g/L]CH + 1.0 [g/L]Pro + 30[g/L]Sur + 2.0 [mg/L]2,4-D.$$

9.2.2.3 Regeneration Medium

$$\begin{split} \text{MSR} &= \text{MS medium} + 8.0 \text{ [g/L]D-sor} + 0.8 \text{ [g/L]CH} + 0.5 \text{ [g/L]Vc} \\ &+ 30 \text{[g/L]Sur} + 2.0 \text{ [mg/L]KT} + 0.5 \text{[mg/L]6-BA} + 0.2 \text{ [mg/L]IAA}. \end{split}$$

$$\begin{split} \text{MBR} &= \text{MS majority} + \text{MS minority} + \text{B}_5 \text{organic} + 8.0 \text{ [g/L]D-sor} \\ &+ 0.8 \text{ [g/L]CH} + 0.5 \text{ [g/L]Vc} + 30 \text{[g/L]Sur} + 2.0 \text{ [mg/L]KT} \\ &+ 0.5 \text{ [mg/L]6-BA} + 0.2 \text{ [mg/L]IAA}. \end{split}$$

$$\begin{split} NBR &= N_6 \text{ majority} + B_5 \text{minority} + B_5 \text{organic} + 8.0 \text{ [g/L]D-sor} \\ &+ 0.8 \text{ [g/L]CH} + 0.5 \text{ [g/L]Vc} + 30 \text{[g/L]Sur} + 2.0 \text{ [mg/L]KT} \\ &+ 0.5 \text{ [mg/L]6-BA} + 0.2 \text{ [mg/L]IAA}. \end{split}$$

9.2.2.4 Rooting Medium

MSRt = 1/2 MS majority + MS minority + MS organic + 30[g/L]Sur + 0.2 [mg/L]IAA.

In the above medium, the PH is adjusted to 5.8 before adding 6.8 g agar; all the mediums are autoclaved for 20 min at 113 °C.

X. Chen et al.

9.2.3 Immature Embryos Callus Induction

Panicles were covered with pollination bags before flowering to ensure self-fertilization. 20–25 days after flowering, immature seeds were collected from these plants. The seed surfaces were wiped with 70 % alcohol and then placed in the refrigerator at 4 °C for pretreatment for about 24 h. Immature seeds were immersed in 70 % ethanol for 1 min, and then immersed in 0.1 % mercuric chloride for 10 min. After this treatment, finally the seeds were rinsed six times in sterile distilled water. Then the seeds were placed on sterilized Whatman filter papers to dry. Immature embryos (IEs) were aseptically removed and placed (scutellum-side up) on induction medium in a plastic Petri dish (90 × 15 mm), and maintained for 3–4 weeks in dark at 28 °C. Then the embryos were transferred to the subculture medium and cultured for 3–4 weeks. The percentage of embryos producing callus was calculated.

9.2.4 Immature Embryo Callus Regeneration

The callus that grew well was selected from subculture medium and transferred to the regeneration medium, and kept for 3–4 weeks at 28 °C with a photoperiod of 16/8 h in light/dark. Then the number of explants forming shoots were counted and the percentage of embryos forming shoots was calculated.

9.2.5 Rooting Culture and Transplantation

The shoots of buds growing up to 3–5 cm were transferred to the rooting medium for 2 weeks. The bottle vents were opened, when there were around 3–4 roots. To adapt for 1–2 days, the medium were washed from roots, and the shoots were transplanted to pots that contained vermiculite and nutrient soil for 2–3 weeks, and then grown to maturity in a greenhouse. When transplanting shoots, the roots are protected from damage, and the nutrition soil is kept moist.

9.2.6 Data Analysis

The callus induction rate = (No. callus/No. inoculated embryos) \times 100 %

The regeneration rate = (No. regenerated callus/No. inoculated callus) \times 100 %

The data were analyzed using SPSS statistical software. Significance was determined by analysis of variance (ANOVA) and the differences between the means were compared by Duncan's new multiple range test.

9.3 Results

9.3.1 Immature Embryos Callus Induction

After culturing for 3–4 weeks, all the ten materials on different induction mediums could induce callus (Fig. 9.1a), but the morphological character and callus induction rate of these ten materials are distinct. The callus of 07-27, Cowley, Sanrio, Rome, Tianza2 and BJ299 were light yellow granular. The callus of Liaotian3, Xinliang 52, BJ-285, M81-E were white granular. 07-27, Xinliang 52, Sanrio, Rome and BJ299 secreted mucilage. Liaotian3, Cowley, Xinliang 52, M81-E, Rome and Tianza2 secreted brown substance. 07-27, BJ-285, Sanrio, Rome, Tianza2, BJ299, Liaotian3 and Cowley had close texture. Xinliang 52 and M81-E were opposite.

9.3.1.1 Effect of Genotype on the Callus Induction Rate

After culturing for about 3–4 weeks, immature embryos had come into being callus. As a result, callus induction rate of different genotypes has a significant difference. Of which Xinliang 52 showed the highest frequency, followed by Cowley, Tianza2, and BJ-299; BJ-285 induction rate is the worst (Table 9.1).

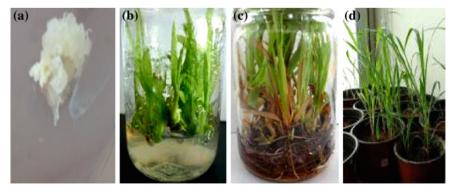


Fig. 9.1 a Induction of 07-27; b regeneration of 07-27; c rooting of 07-27; d transplantation of 07-27

X. Chen et al.

Table 9.1 Genotype as the main effect to analyze significance

Varieties	Percentage of callus formation (%) ^a
Xinliang 52	90.8 ± 4.9 a
Cowley	71.9 ± 3.5 b
Tianza2	$66.5 \pm 3.7 \text{ bc}$
BJ-299	64.1 ± 7.2 bc
Sanrio	61.2 ± 9.9 c
07-27	57.8 ± 15.3 cd
Rome	$51.4 \pm 3.6 \text{ de}$
Liaotian3	50.4 ± 7.9 de
M81-e	45.7 ± 7.5 e
BJ-285	27.4 ± 8.4 f

 $^{^{}a}$ Each value is the mean percentage of three experiments. Numbers (mean \pm standard deviation) followed by the same letter are not significantly different at the 5 % level by Duncan's new multiple range test

9.3.1.2 Effect of Medium on Callus Induction Rate

According to the data which have been counted, Callus induction rate of different mediums has no significant difference (Table 9.2). This indicates that MSI, NBI, and MBI could all be used for callus induction.

9.3.2 Immature Embryo Callus Regeneration

After regenerating culture for 3–4 weeks, callus appeared green based on different regeneration mediums. After 4 weeks' culturing, there were visible buds, and buds grow into shoots for 5 weeks (Fig. 9.1b). Then it was transferred to the rooting medium (Fig. 9.1c). In the process of regeneration, Liaotian3, Cowley, Xinliang 52, BJ-285, Sanrio, Cowley and Rome secreted brown substance. Tianza2 secreted little brown substance. M81-E and 07-27 do not secrete colored substance.

Table 9.2 Medium as main effect to analyze significance

Medium	Percentage of callus formation (%) ^a
MSI	$58.3 \pm 18.3 \text{ a}$
NBI	$60.9 \pm 17.4 \text{ a}$
MBI	58.1 ± 19.3 a

 $[^]a$ Each value is the mean percentage of three experiments. Numbers (mean \pm standard deviation) followed by the same letter are not significantly different at the 5 % level by Duncan's new multiple range test

Varieties	Medium	No. of calli No. of green		Percentage of regeneration			
		tested	dots	(%)			
Xinliang 52	MBR	105	2	1.90			
BJ-285	MSR	78	2	2.57			
07-27	MSR	77	31	40.26			
	MBR	91	37	40.66			
M81-E	MBR	61	3	4.92			
Tianza2	MSR	54	1	1.85			
	MBR	92	1	1.09			
Liaotian3	MSR	117	18	15.39			
	MBR	141	24	17.02			
Cowley	MSR	158	7	4.43			

Table 9.3 Callus regeneration results of different genotypes on different mediums

9.3.2.1 Effect of Genotype on the Regeneration Rate of Immature Embryo Callus

After regenerating for about 3–4 weeks, some callus had grown into shoots, and it is easily found that not all the genotypes can differentiate for callus regeneration. Callus regeneration rate for 07-27 is the highest, followed by Liaotian3; the regeneration rate of other genotypes is between 1 and 5 % (Table 9.3).

9.3.2.2 Effect of Medium on the Regeneration Rate of Immature Embryo Callus

The result was achieved based on statistical data. 07-27, Liaotian3, and Tianza2 can all regenerate shoots both on MSI and MBI; Xinliang 52 and M81-E only regenerate shoots on MBI, Cowley and BJ-285 only regenerate shoots on MSI. But there is nothing on NBI. So it is clear that MSI and MBI are suitable for callus regeneration (Table 9.3).

9.4 Discussions

The experiment demonstrated that genotypes not only had significant effect on callus induction culture, but also had great influence on callus regeneration culture. Thus selecting excellent genotype is very important. This agrees with Shi et al. [16]. In addition, we found that the induction rate and regeneration rate of other genotypes were low. In contrast, 07-27 showed the high frequency of callus induction and regeneration. What is more, the culture process is not brown, and finally obtained regeneration plants. Therefore, we thought that 07-27 was an ideal

90 X. Chen et al.

material for tissue culture and genetic improvement of sweet sorghum. Besides, Zhao et al. [24] thought that the M81-E was very good genotype, but the M81-E in this experiment, the callus induction rate, and regeneration rate were very low, and not suitable for tissue culture of sorghum immature embryos.

For callus induction culture, MSI, MBI, and NBI have no significant difference. But the result was the opposite for callus regeneration culture. Both MSR and MBR regenerated shoots, NBR had nothing. Therefore, in comprehensive consideration of sweet sorghum tissue culture, MS is a suitable basic medium. In the experiment, the regeneration rate is very low for a majority of genotypes, which reveals that it is difficult to develop the sorghum tissue culture and find an appropriate medium for sweet sorghum tissue culture. Hagio [7] thought that adding cytokinins (BA or Kin) into callus inducing medium will increase the regeneration rate, when using MS basic medium. What is more, adding PVP and Pro into regeneration medium can also increase the cell regeneration rate. But there is no use of adding cytomin into regeneration medium. However, Oldach and Morgenstern [13] considered that L₃ medium was more suitable for sorghum callus induction and regeneration, but also the medium supplemented with AgNO₃ can significantly increase the regeneration rate. Robert et al. [15] found that the regeneration medium mM11 supplemented with 3 mg/l BAP and 1 mg/l TDZ and removing 2,4-D with light treatments for 2 weeks can increase the regeneration rate.

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Chapter 10 Cloning and Sequence Analysis of a Novel ACC Oxidase Gene from Peanut (*Arachis hypogaea* L.)

Quanxi Sun, Xiuzhen Wang, Yueyi Tang, Qi Wu, Yunyun Wang, Qingyun Zhang, Guangying Cao, Shuo Meng and Chuantang Wang

Abstract Ethylene plays important roles in seed dormancy. As a key enzyme in ethylene biosynthesis, ACC oxidase gene (*ACCO*) converts 1-aminocyclopropane-1-carboxylic acid (ACC) into ethylene. To study the mechanism underlying peanut seed dormancy, cDNA sequence of the candidate gene was isolated from peanut using RACE technique. As in the situation in *Arabidopsis*, the *ACCO* from peanut (*AhACCO* gene) contained four exons and three introns. Homology comparison analysis showed 68–87 % sequence similarity between *AhACCO* and *ACCO* genes of other selected plants.

Keywords Peanut · ACC oxidase · RACE · Homology comparison

10.1 Introduction

Peanut (*Arachis hypogaea* L.) is one of the five most important oilseed crops worldwide. The duration of seed dormancy is of great importance to peanut. Seeds without dormancy or short dormancy duration would germinate before they are harvested, leading to reduced seed oil, poor seed quality, and high risk for aflatoxin contamination [1].

A complex network of plant hormones, including abscisic acid (ABA), gibberellin (GA), ethylene, auxin, brassinosteroids, and jasmonates, control seed dormancy, and germination [2]. In our previous study, ethylene could break the dormancy of peanut seed [3], but its mechanism is still unclear. In the past decades, ethylene signaling had been shown to play roles in mediating plant response to

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94 Q. Sun et al.

biotic and abiotic stresses [4–7]. In recent years, ethylene biosynthesis has proved to be necessary for several physiology functions. For example, in a maize ACC synthase mutant where the first step in ethylene biosynthesis was affected, drought tolerance was enhanced and drought-induced leaf senescence was inhibited [8]. ACC oxidase gene (*ACCO*) catalyzes the final step of ethylene biosynthesis, converting 1-aminocyclopropane-1-carboxylic acid (ACC) to ethylene [9]. The genes have been isolated from several organisms including white clover [10], alfalfa [11], pear [12], and tomato [13]; however, the *ACCO* gene from peanut has not been isolated yet, and its function in seed dormancy remains largely unknown.

Here we report the isolation and analysis of a full-length cDNA encoding an ACC oxidase from peanut.

10.2 Materials and Methods

10.2.1 Plant Material

Huayu 33, a peanut cultivar from Shandong Peanut Research Institute, was grown in a temperature-controlled chamber at 20 $^{\circ}$ C with a photoperiod of 16 h light and 8 h dark. After about 1 month, the leaves were collected and immediately frozen in liquid nitrogen and stored at -80 $^{\circ}$ C.

10.2.2 Nucleic Acid Manipulation and PCR-Based Cloning

Total RNAs were isolated from leaves of Huayu 33 using Trizol reagent (Invitrogen) according to the manufacturer's protocol. The first-strand cDNA was synthesized by M-MLV reverse transcriptase following SMARTer™RACE cDNA Amplication Kit Use Manual (Clontech). We performed PCR using Phusion (Thermo) according to its protocol. The PCR products were separated by electrophoresis though a 1.5 % agarose gel, and purified using a Gel Extraction Kit (TransGen). All purified PCR products were cloned into pEASY-blunt simple vector (TransGen) and sequenced.

10.2.3 DNA and Protein Sequence Analysis

The DNA sequences were sequenced in Sangon (Shanghai, China). The open reading frame and deduced amino acid sequence were analyzed using ORF Finder (http://www.ncbi.nlm.nih.gov/projects/gorf/). Homology comparison and phylogenetic analysis of *ACCO* genes were performed by DNAMAN Sequence Analysis Software (Lynnon Biosoft).

10.3 Results and Discussion

10.3.1 Bioinformatics Analysis and Degenerate Primers Design

Seven *ACCO* homolog genes from different organisms were obtained using NCBI blast. These sequences were aligned with DNAMAN Software (Fig. 10.1). Two degenerate primers (AhACCO-F1 and AhACCO-R1) (Table 10.1) were designed based on the conserved sequences underlined in Fig. 10.1.

10.3.2 Isolation of AhACCO Gene from Peanut

An about 600-bp fragment (*AhACCO* middle fragment) was obtained by conventional PCR using primers: AhACCO-F1 and AhACCO-R1 (Fig. 10.2). The fragment was purified from gel, and ligated to pEASY-blunt simple vector (TransGen). The selected clones were sequenced using M13-47 primer. The sequences showed high homology to *ACCO* genes from other sources (data not shown). Gene-specific primers (AhACCO-R2 and AhACCO-F2) were designed based on the sequence information (Table 10.1) to amplify the 3'- and 5'-regions.

To amplify the 3'-region of the cDNA, the gene-specific antisense primer Ah-ACCO-F2 was used with the universal primer UPM provided in the kit. A ~ 500 bp PCR product was cloned and sequenced. The 5'-region of the cDNA was PCR-amplified using sense primer AhACCO-R2 combined with UPM. A ~ 300 bp PCR product was obtained and sequenced (Fig. 10.2). The middle fragment was

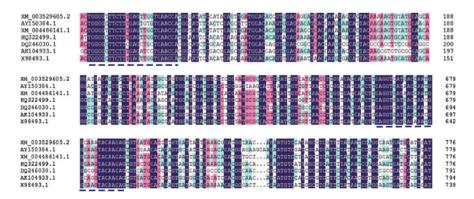


Fig. 10.1 Alignment of the *ACCO* mRNA sequences from different organisms. Note two conserved sequences (a and b) were *underlined* and used to design degenerate primers. XM_003529605.2: *Glycine max*, AY150384.1: *Arabidopsis thaliana*, XM_004486141.1: *Cicer arietinum*, HQ322499.1: *Solanum lycopersicum*, DQ246030.1: *Zea mays*, AK104933.1: *Oryza sativa*, X98493.1: *Nicotiana tabacum*

Q. Sun et al.

Table 10.1 Primers use in this study

Primer	Sequence
AhACCO-F1	TGGGGYTTCTTYGAGHKBSTGAAC
AhACCO-R1	CTCTTGTAYYKSCCRTTDGTKATSACCTC
AhACCO-F2	GAAGGTGTTTCATGGGTC
AhACCO-R2	ACTCCAAGCCTTTGCTTGCCAC
AhACCO-F3	ACTTGTAGTGAGAGACCAAAC
AhACCO-R3	GCTTTCATAATCCATATTGCTTAA

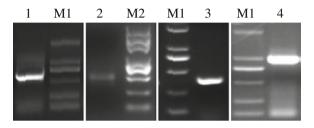


Fig. 10.2 Isolation of peanut *ACCO* cDNA. *1* PCR amplified middle fragment of *AhACCO*, 2 3′-RACE product, 3 5′-RACE product, 4 full-length cDNA of *AhACCO*. *M1* Trans2K DNA marker (TransGen), *M2* Trans2K plus II DNA Marker (TransGen)

overlapped with 5'/3'-PCR product sequence by about 150 and 250 bp, respectively, confirming that they were ultimately derived from a single gene. The full-length cDNA of *AhACCO* was isolated using primers AhACCO-F3 and AhACCO-R3 (Table 10.1) and its deduced amino acids sequence were predicted with ORF Finder.

The AhACCO genomic DNA sequence was downloaded from PeanutBase and the Peanut Genomics Initiative (http://www.peanutbase.org/). The AhACCO gene was found to contain four exons and three introns, similar to AtACCO from Arabidopsis (Table 10.2). This indicated that the AhACCO in this study was an ACCO gene. The exons of the gene from Arabidopsis and peanut were much of a size, while the introns varied widely.

10.3.3 Homology Comparison and Phylogenetic Analysis

AhACCO was predicted to encode a protein of 321 amino acids with a molecular mass of 36.3 kDa. The deduced amino acid sequence of AhACCO aligned with other ACCO proteins is shown in Fig. 10.3. To examine the relationships among different

Table 10.2 The exon and intron structures of ACCO genes from Arabidopsis and peanut

Gene	Accession number	ORF size (bp)	Exon 1 (bp)	Intron 1 (bp)	Exon 2 (bp)	Intron 2 (bp)	Exon 3 (bp)	Intron 3 (bp)	Exon 4 (bp)	Protein (aa)
AtACCO	CP002684.1	960	114	174	227	88	334	172	285	320
AhACCO	_a	963	108	85	227	156	334	508	294	321

^a No accession number is available for this gene

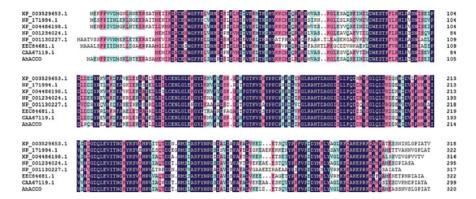


Fig. 10.3 Alignment of the predicted protein sequence of *AhACCO* with sequence predicted from other known *ACCO* genes. XP_003529653.1: *G. max*, NP_171994.1: *A. thaliana*, XP_004486198.1: *C. arietinum*, NP_001234024.1: *S. lycopersicum*, NP_001130227.1: *Z. mays*, EEC84681.1: *O. sativa*, CAA67119.1: *N. tabacum*

sources of ACCO genes, sequences from representative plants were selected to construct the phylogenetic tree by the DNAMAN software (Fig. 10.4). The amino acid sequence of AhACCO exhibited 87 % identity to ACCO from *Glycine max* (XP_003529653.1) and *Cicer arietinum* (XP_004486198.1), 81 % identity to *Solanum lycopersicum* (NP_001234024.1) and *Nicotiana tabacum* (CAA67119.1), and 72, 72 and 68 % identity, respectively, to *Arabidopsis thaliana* (NP_171994.1), *Oryza sativa* (EEC84681.1) and *Zea mays* (NP_001130227.1) (Fig. 10.3).

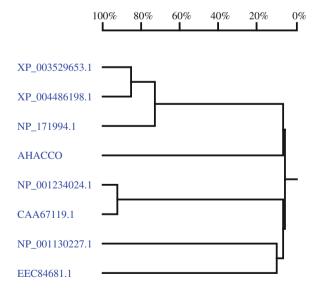


Fig. 10.4 Phylogenetic analysis of ACCO genes. The tree shown was constructed using the amino acid sequences by DNAMAN software. XP_003529653.1: *G. max*, NP_171994.1: *A. thaliana*, XP_004486198.1: *C. arietinum*, NP_001234024.1: *S. lycopersicum*, NP_001130227.1: *Z. mays*, EEC84681.1: *O. sativa*, CAA67119.1: *N. tabacum*

98 Q. Sun et al.

10.4 Conclusion

In this study, a novel cDNA, *AhACCO*, was isolated from peanut cultivar Huayu 33 using RACE technique. Homology comparison and phylogenetic analysis indicated that it encoded an ACC oxidase. Its detailed function in peanut seed dormancy, however, is yet to be studied in the future work.

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Chapter 11 **Genetic Diversity Among** the Microorganisms in Dagu Used for Beidacang Liquor as Revealed by RAPD Analyses

Shi-Wei Wang, Oing-Hui Wang, Li-Ping Zhai, Jun Liu, Zhi-Dan Yu, Mao-Mao Zheng and Fang Wang

Abstract To assess the genetic diversity among five microorganisms (namely, DSL(CS)-01, DSL(CS)-02, DSL(PDA)-C, DSL(PDA)-E, DSL(PDA)-F), RAPD (random amplified polymorphic DNA) analysis was performed. Seven primers with stable polymorphisms were screened from 64 primers. PCR was carried out with the seven primers. The data indicated that the molecular markers gave different levels of polymorphism. 94 RAPD bands were obtained and out of them 64 were polymorphic bands (68.09 %) and the best primer screened was b8. The primer b8 was considered as the most effective RAPD marker for the five microorganisms above. The dendrogram built on the basis of data from RAPD analysis represented the genetic distances among the five microorganisms. Understanding the genetic variability among the microorganisms opens up a possibility for developing a molecular genetic map that will lead to the application of marker-assisted selection tools in genetic improvement of the microorganisms.

Keywords Genetic • Diversity • Microbial • Strain • Dagu

11.1 Introduction

The development and advancement of liquor-making industry in China is dependent on scientific development and technical innovation [2]. Heilongjiang Beidacang Group Co. Ltd. was founded in 1914, when the name "Juyuan permanent pot", is one

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⁹⁹

100 S.-W. Wang et al.

of the eight distilleries. Beidacang spirit was the earliest popular liquor in Northeast China. As a saccharifying ferment, Dagu is made from "Snake Eyes" sorghum as raw material. The raw material is made into Dagu billets first through crushing and water pressed at a certain temperature and humidity. The microorganisms in Daqu mainly are mold, less yeast, and bacteria [5]. The bacteria mainly included lactic acid bacteria, acetic acid bacteria, bacillus, etc. [4]. Knowledge of the microbial diversity in Dagu provides the basis for dominant microflora analysis and understanding of the role of microbes in the Dagu production process. Dagu is a multistrain mixed starter. Bacteria such as butyric acid bacteria and caproic acid bacteria are Clostridium prazmowski, which are the functional bacteria in pit mud and create a long aftertaste in liquor [9]. The utilization of molecular biology and genomics in the production of liquor is important [3, 12, 14]. In our earlier work, 11 microbial strains were isolated from "Beidacang" Daqu using different culture media (Czapek's medium, PAD medium and LB medium) and temperatures (37 and 60 °C). By morphological preliminary evaluation, these strains consisted of Aspergillus, Rhizopus, Saccharomyces as well as several strains like Bacillus sp. But for the five strains like Bacillus (DSL(CS)-01 (37 °C), DSL(CS)-02 (55 °C), DSL(PDA)-C (37 °C), DSL (PDA)-E (55 °C) DSL(PDA)-F) (55 °C), it is difficult to analyze and differentiate them only from the morphology. In this paper, we will study the five strains using RAPD molecular marker.

As a type of PCR, the DNA segments amplified are random. Several arbitrary, short primers (10 nucleotide length) are created, then PCR proceeds. RAPD markers are able to differentiate between genetically distinct individuals. It can also analyze genetic diversity. RAPD has been used to characterize and trace the phylogeny of diverse microbial species. The analysis of genetic variation and relatedness in microbial strains of Daqu are of great value for genetic resources conservation and application. The objectives of this investigation were to determine the genetic variability among five microorganisms (namely, DSL(CS)-01, DSL (CS)-02, DSL(PDA)-C, DSL(PDA)-E, DSL(PDA)-F) at the molecular levels using RAPD markers and to use the data to construct a phylogenetic tree. The genotype-specific markers were also determined.

11.2 Materials and Methods

11.2.1 Microbial Materials

Materials Five microorganisms (namely, DSL(CS)-01, DSL(CS)-02, DSL(PDA)-C, DSL(PDA)-E DSL(PDA)-F) were separated from Daqu used in Heilongjiang Beidacang Group Co. Ltd. China.

Now they are reserved in the molecular laboratory of Qiqihar University.

Microbial culture The five microbial materials were activated and cultured in different media and at different temperatures. The two strains (DSL(CS)-01,

DSL(CS)-02) were cultured in Czapek's solid media at 37 °C, the three strains (DSL(PDA)-C, DSL(PDA)-E DSL(PDA)-F) were in PDA solid media at 37 °C. After 24 h, the different plate colonies were observed.

11.2.2 RAPD Analysis

DNA extraction The colonies in different solid plates were transferred to the corresponding liquid media (Czapek and potato dextrose). They were cultivated under 37 °C at 200 rpm for 2 days in a shaking incubator. DNA was extracted from the mycelium by the method reported [13]. Purity extracted genomic DNA was examined by UV spectrophotometer and the gel with 1 % agarose.

Primers screened and used Seven stable primers were screened from 64 primers. They were d2, b17, b18, b1, b3, b8, c9, respectively. These primers were found to be very stable and had good polymorphism in our previous work. So, the nine primers will be used for RAPD. The primer sequences are shown in Table 11.1.

PCR analysis PCR was performed in a total volume of 20 μL containing 10 ng DNA, 200 μM dNTPs, 1 μM of 6 arbitrary 10-mer primers (Sangon Biotech (Shanghai) Co. Ltd, China), 1 unit of Taq polymerase (Beijing TransGen Biotech Co., Ltd., China) and $10 \times Taq$ polymerase buffer (Beijing TransGen Biotech Co., Ltd., China). For DNA amplification thermal cycler (MG96G/Y, Hangzhou Longgene Scientific Instruments Co., Ltd., China) was programmed as follows: 94 °C for 5 min followed by 40 cycles, 94 °C for 1 min, 36 °C for 1 min, 72 °C for 1 min, and 72 °C for 7 min. The amplification products were analyzed by electrophoresis in 1 % agarose in TBE buffer, stained by ethidium bromide, and photographed under UV light.

Polymorphism and cluster analysis The RAPD gel images were scanned using AlphaImager EP gel imaging systems (USA) and analyzed with Alpha Viewer Software (USA). The bands were sized and then binary coded by "1" or "0" for their presence or absence in each genotype. The genetic diversity of strains was analyzed by DNA fingerprints. Resembled indexes were calculated by the reported method [10]. Cluster analysis was based on similarity matrices obtained with Statistical Product and Service Solutions (SPSS) V19.0 Software to estimate the dendrogram.

Table 11.1	Names and	sequences of	t RAPD	primer	used t	o asses	s the	genetic	variability	among
the five mic	roorganisms									

Primer	Sequence	Primer	Sequence	Primer	Sequence
name		name		name	
b1	GTTTCGCTCC	b17	AGGGAACGAG	d2	TGCCGAGCTG
b3	CATCCCCCTG	b18	CCACAGCAGT		
b8	GGTGACGCAG	c9	TGGACCGGTG		

102 S.-W. Wang et al.

11.3 Result and Discussion

11.3.1 Petri Dishes of Five Microorganisms

Bacillus is a genus of Gram-positive, rod-shaped bacteria and a member of the phylum *Firmicutes*. *Bacillus* sp. can be obligate aerobes, or facultative anaerobes. Under stressful environmental conditions, the bacteria can produce oval endospores. These characteristics originally defined the genus, but not all such species are closely related, and many have been moved to other genera of *Firmicutes*.

As shown in Fig. 11.1, the five microorganisms were identified preliminarily as genus Bacillus, according to their colonial and cellular morphologies as well as their spore staining in earlier work (Note: The results are not shown here).

11.3.2 Extraction of DNA from Five Microorganisms

Breaking the cells open is commonly referred to as cell lysis, to expose the DNA within. This is commonly achieved by grinding the sample. The membrane lipids were removed by adding SDS.

DNA ladder (λ-*Hin*dIII digest DNA Marker) is in lane 1; the DNA in other lanes (2, 3, 4, 5, 6) one by one is the DNA from DSL(CS)-01, DSL(CS)-02, DSL(PDA)-C, DSL(PDA)-E, DSL(PDA)-F.

As shown in Fig. 11.2, DNA of all strains were extracted and purified by analysis of agarose gel electrophoresis (1 % agarose). The results showed that the DNA purified can be used as the template for RAPD.

11.3.3 Screening of Primers and RAPD Analysis

RAPD markers are easier and quicker to use and preferred in applications where relationships between closely related breeding lines are of interest.

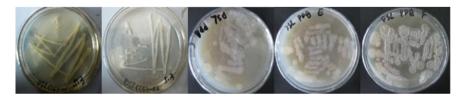
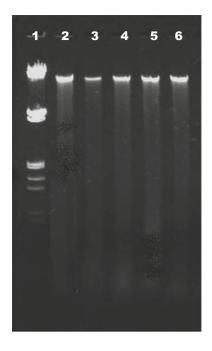


Fig. 11.1 Colonial morphology of microorganism on the petri dishes (they are DSL(CS)-01, DSL (CS)-02, DSL(PDF)-C, DSL(PDF)-E, DSL(PDF)-F, respectively, from the *left* to the *right*)

Fig. 11.2 The DNA purified from five microbe strains using agarose electrophoresis (agarose 1 %)



As shown in Fig. 11.3a, there are obvious polymorphisms in five RAPD fingerprinting obtained with DSL(CS)-01 strain DNA as a template using five primers (b1, b8, c9, b18, d2) except for primer b3. But in Fig. 11.3b, for DNA template from strain DSL(CS)-02, only four better RAPD finger printings were obtained (b18, b8, c9, b1). No RAPD finger printing occurred with the primer b3 or the primer b17. So we can find that some primers (such as b1, b8, c9, b18, d2,) are better than the primers like b3 and b17.

As shown in Fig. 11.4a–c stand for the RAPD fingerprinting from strains DSL (PDA)-C, DSL(PDA)-E, and DSL(PDA)-F, respectively. There are obvious polymorphisms in RAPD fingerprinting obtained (like Fig. 11.4b, except for primer b3).

In Fig. 11.4a, it is seen that three primers (b8, b18, and d2) obtained good polymorphisms, but the other three primers (b1, b3, c9) had almost no polymorphisms. However, the other primers did not make polymorphisms in Fig. 11.4c except for primers b8 and c9. Among all, primer b8 was the best primer chosen for RAPD experiment below. According to Figs. 11.3 and 11.4, the total bands are 94 and the genetic special DNA bands are 64. The results show that five strains have a high level of genetic diversity and the percentage of DNA polymorphic was 68.09.

Recently, the identification of microorganisms is dependent on the application of RAPD markers. DNA sequences are independent of environmental conditions, identification can be determined at any stage of microbial growth [1]. RAPD markers are easier and quicker to use and preferred in applications where relationships between closely related breeding lines are of interest.

104 S.-W. Wang et al.

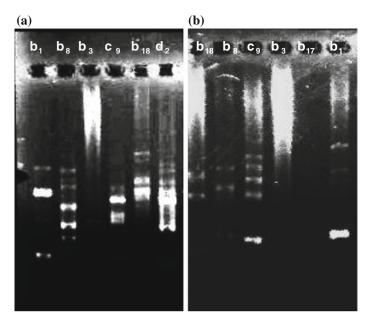


Fig. 11.3 RAPD finger printings obtained with DSL(CS)-01 strain (a) and DSL(CS)-02 strains (b) with different primers using agarose electrophoresis (agarose 1 %)

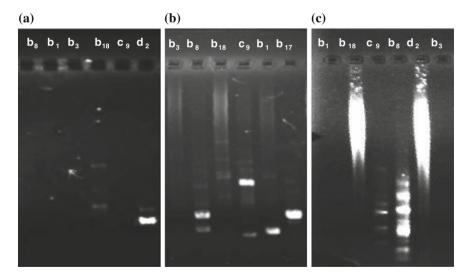
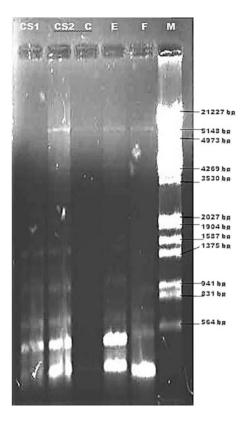


Fig. 11.4 RAPD finger printings obtained with DSL(PDA)-C strain (a), DSL(PDA)-E strain (b), and DSL(PDA)-F strain (c) using different primers (agarose 1 %)

Fig. 11.5 RAPD finger printings from five different strains using primer b8 (1.5 % agarose). *M* DNA marker (λDNA/*Hind*III+*Eco*RI) (125, 564, 831, 941, 1375, 1584, 1904, 2027, 3530, 4268, 4973, 5148, 21227 bp)



As shown in Fig. 11.5, the obvious RAPD finer printings were obtained with DNA from the five strains (DSL(CS)-01, DSL(CS)-02, DSL(PDA)-C, DSL)PDA)-E, DSL(PDA)-F) as templates, respectively, using primer 8.

Obviously, in Fig. 11.5 there is no band for DNA template from strain DSL(CS)-01 in size 5,000 bp, but other DNA templates from the other four strains can make the same band in the same size. Many bands are located in a range in size between 5,000 and 564 bp. These bands are dim. For the five strains (DSL(CS)-01, DSL (CS)-02, DSL(PDA)-C, DSL)PDA)-E, DSL(PDA)-F), there are at least 3, 8, 1, 5, and 4 dim bands, respectively, in this size range. However, at a range below 564 bp, there are very obvious different bright bands for four strains (DSL(CS)-01, DSL (CS)-02, DSL(PDA)-E, DSL(PDA)-F).

A comparison between seven random primers on PCR showed that every strain had its special DNA fingerprints (see Figs. 11.3 and 11.4).

Fingerprints of primer b8 showed that the strain DSL(CS)-01 had five bands; DSL(CS)-02 had nine bands (including two bright bands). For the strain DSL(PDF) had only two dim bands. But for DSL(PDA)-E and DSL(PDA)-F, they had at least 6–8 bands, the former have more bands than the latter.

S.-W. Wang et al.

11.3.4 Genetic Similarity Index

DNA bands of RAPD-PCR from seven primers were grouped by cluster analysis. The genetic similarity index for nine strains was from 0.9091 to 0.2324. The average genetic similarity index was 0.5282 (Table 11.2).

As shown in Table 11.2, the highest genetic similarity was found between the strain DSL-CS(01) and the strain DSL-CS(02), and the lowest genetic similarity was found between the strain DSL-CS(01) and the strain DSL(PDA)-C.

11.3.5 Genetic Relation and Clustering Analysis

The RAPD-based dendrogram group investigated microorganisms in two main clusters. The first cluster included DSL(PDA)-F and DSL(PDF)-C, the second cluster was divided into two subclusters. The first cluster contains the strain DSL (CS)-01 and DSL(CS)-02 and the second has only DSL(PDA)-E (Fig. 11.6).

As shown in Fig. 11.6, the genetic relation and clustering analysis showed that there was closest genetic relationship between the strains DSL(CS)-01 and DSL (CS)-02, they may be different specials which came from the same genus. Because of gene mutation, the latter was more thermophile than the former. The strain DSL (CS)-02 can grow well at 55 °C unlike DSL(CS)-02 (its most suitable growth

			•			
		DSL(CS)-01	DSL(CS)-02	DSL(PDA)-E	DSL(PDA)-F	DSL(PDA)-C
	DSL(CS)-01	1.0000				
	DSL(CS)-02	0.9091	1.0000			
	DSL(PDA)-E	0.8000	0.7273	1.0000		
	DSL(PDA)-F	0.6000	0.7273	0.3333	1.0000	
	DSL(PDA)-C	0.2324	0.2857	0.3333	0.3333	1.0000

Table 11.2 Matrix of genetic similar index of five strains separated from "Beidacang" Daqu

In the table there are five microorganisms compared to each other

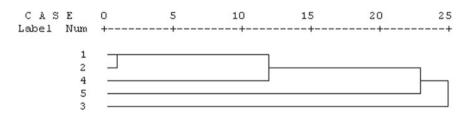


Fig. 11.6 Phylogenetic tree constructed with UPMGA of five strains of "Beidacang" Daqu. "1, 2, 3, 4, 5" stands for strain DSL(CS)-01, DSL(CS)-02, DSL(PDA)-C, DSL(PDA)-E, DSL(PDA)-F, respectively

temperature is 37 °C). The clustering analysis also showed that there was a closer genetic relationship between the strains DSL(CS)-01 and DSL(PDA)-E. The two strains clustered into one category between 10 and 15 in the rescaled distance cluster combine. However, the strain DSL(PDA)-F had farther genetic relative relationship with strains such as DSL(CS)-02, DSL(CS)-01 and DSL(PDA)-E (between 20 and 25 in the rescaled distance cluster combine). The strain DSL (PDA)-F is close to the strain DSL(PDA)-C in genetic relationship. But the strain DSL(PDA)-C had the lowest genetic similarity with the other four strains.

RAPD markers are easier and quicker to use and are preferred in applications where relationships between closely related breeding lines are of interest. It was reported that more than 200 colonies were isolated and characterized. The isolates were discriminated by phenotypic and conventional biochemical taxonomic methods [11]. The results revealed the presence of bacteria, molds, and yeasts. Bacteria consist of bacillus, Acetobacter, Lactobacillus, and Clostridium, among which Bacillus strains were found to be predominant. So, Bacillus strains are very important microbes in the process of liquor production. Maotai flavor substances were mainly from functional bacteria in high temperature Dagu. These functional microorganisms are of great importance. Some hemophilic bacteria were isolated from high temperature Dagu of Maotai flavor liquor [15]. The result showed that the strain B3 was functional bacteria of Maotai. The strains from high temperature Maotai Daqu had a close relationship with the quality, production, and flavor liquor. The brewing microorganisms were the basis of Maotai unique flavor [6]. Maotai microorganism has obvious diversities [7]. The thermophiles Bacillus is the effective strain of Maotai liquor [2]. The furan, aromatic, and pyrazine compounds were characteristic components of Maotai flavor [16]. The future development direction of liquor will be from the research of functional bacteria to the development on microbial community [2, 8].

11.4 Conclusion

The results of this study indicate that DNA markers represent efficient tools for estimating the genetic variability and the genetic relationships among the microorganisms. The markers generated are enough to distinguish between the different genotypes used. The genotype-specific molecular markers were determined and these markers can be considered as useful markers for the other microorganisms in Daqu used for Beidacang liquor. This opens up a possibility for developing a molecular genetic map that will lead to the study of other microorganisms in Daqu used for Beidacang liquor.

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108 S.-W. Wang et al.

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Chapter 12

Stenotrophomonas maltophilia Having Decolorization Capability of Azo Dye Isolated from Anaerobic Sludge

Wei Feng, Peng Song, Yang Zhang and Zixing Dong

Abstract Strain 5-8 was isolated from anaerobic sludge which was collected from a sewage treatment plant located at Shandong Province. Strain 5-8 has efficient ability to decolorize Fast Yellow2G. Fast Yellow2G was decolorized by more than 80 % in 48 h, and completely decolorized in 60 h. Strain 5-8 was observed to form yellow colored, smooth, circular, convex colonies of 5 mm diameter with entire margins on LB medium. Cells of strain 5-8 were observed to be gram-negative, aerobic, motile with a single polar flagellum and rod-shaped with dimensions of 0.3–0.4 × 0.6–1.5 μm. Growth was observed at 18–40 °C (optimum 35–37 °C), pH 6.0–9.0 (optimum pH 7.0–8.0) and with 0.5–5 % (w/v) NaCl (optimum 1 %). Neighbor-joining phylogenetic tree based on 16S rRNA gene sequences showed that strain 5-8 belongs to the genus *Stenotrophomonas*, clustering coherently with the type strain of *Stenotrophomonas maltophilia* showing sequence similarity value of 99 %.

Keyword Decolorization azo dye phylogenetic analyses *Stenotrophomonas*

12.1 Introduction

Azo dyes are synthetic organic colorants, which are characterized by the presence of one or more azo groups. The largest class of dyes in industrial use is known as azo dyes in number and amount produced [1]. Dyes are usually released into the environment as industrial effluents from textile and dye manufacturing. Release of such colored compounds into the environment is undesirable not only because of

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110 W. Feng et al.

their aesthetic appearance, which may drastically affect photosynthesis in the aquatic ecosystem, but also because many of the dyes and/or their breakdown products are toxic and mutagenic to life [2]. The presence of dyes or their degradation products in water even at very low concentrations can be toxic and sometimes carcinogenic, mutagenic, or teratogenic to various organisms, including humans [3]. Therefore, azo dyes are of significant concern as pollutants of the environment, and removal of dyes from environment using different methods has become a focus for environmental scientists.

Physical and chemical methods are not suitable to treat dye effluent due to their operational costs or secondary sludge disposal problems [4]. Biological methods represent a more proper way of textile azo dye removal. Several microorganisms such as algae, yeast, filamentous fungi, and bacteria individually or in consortium are shown to degrade azo dyes in the presence of nutrients [5]. Various woodrotting fungi were able to decolorize azo dyes using peroxidases or laccases. But fungal treatment of effluents is usually time-consuming. Under static or anaerobic conditions, bacterial decolorization generally demonstrates good color removal effects. However, aerobic treatment of azo dyes with bacteria usually achieved low efficiencies because oxygen is a more efficient electron acceptor than azo dyes [6]. Bacteria studied for biodegradation include Pseudomonas NBAR, Rhodopseudomonas palustris, etc. [7]. Use of microbial consortium for biodegradation of dyes has also been reported [8]. Therefore, the isolation of effective strains to decolorize azo dyes is a key point. During the course of our research, we have isolated a bacterial strain that highly decolorized the toxic azo dye Xylene Fast Yellow2G from anaerobic sludge samples by an enrichment culture.

12.2 Materials and Methods

12.2.1 Dyes and Chemicals

All chemicals including azo dyes (Xylene Fast Yellow2G) (Fig. 12.1) used in this study were provided by Shanghai Sangon Biological Engineering Technology & Services Co., Ltd.

12.2.2 Bacterial Strains and Culture Conditions

Anaerobic sludge was collected from a sewage treatment plant located at Shandong Province, eastern China and used as a source for the isolation of bacterial strains.

Fig. 12.1 The chemical structure of Xylene Fast Yellow2G

Strain 5-8 was isolated by the standard dilution plating technique at 30 °C on the medium including azo dye (0.5 % glucose, 0.2 % ammonium sulfate, 0.05 % polypepton, 0.05 % yeast extract, 0.03 % $\rm K_2HPO_4$, 0.02 % $\rm KH_2PO_4$, 0.01 % $\rm MgSO_4 \cdot 7H_2O$, and 0.005 % azo dye, pH 6.8) and cultivated routinely at 37 °C in LB medium (containing 0.005 % azo dye). Strain 5-8 was maintained on LB media at 4 °C for short-term preservation and as a glycerol suspension (20 %, w/v in distilled water) at -80 °C for long-term preservation.

Cellular biomass required for DNA extraction was obtained from cultures grown in LB media at 37 °C for 2 days. The decolorization experiments were also performed in LB medium (containing 0.005 % azo dye) at 37 °C.

12.2.3 Morphological, Physiological and Biochemical Characterization

Colony morphology was examined after 3 days of incubation at 35 °C on LB medium. Cell morphology was investigated by light microscopy and Gram reaction was observed by staining using a Gram staining kit according to the manufacturer's protocol. Motility was assessed under phase contrast microscopy.

Strain 5-8 was cultured over the temperature range of 5–50 °C (at intervals of 5 °C) and the pH range of 5.0–10.0 (at intervals of 1 pH unit) in LB. Growth at various NaCl concentrations was tested over the range of 0.5–10 % (w/v) NaCl [0.5 % (w/v) and 1–10 % (w/v) at intervals of 1 %] by incubating under conditions of 35 °C and pH 7.0.

Oxygen requirement, activities of catalase, urease and oxidase, hydrolysis of casein, starch, Tweens 20, 40 and 80, aesculin, nitrate reduction, Voges–Proskauer test, indole and H₂S production were performed according to the conventional methods described by Cowan and Steel [9] and Smibert and Krieg [10]. The utilization of sole carbon and nitrogen sources was determined according to the method described by Gao et al. [11].

12.2.4 16S rRNA Gene Sequencing and Phylogenetic Analyses

Extraction of genomic DNA was performed according to the method of Sambrook and Russell [12]. PCR amplification of the 16S rRNA gene was performed using the following two primers: 27F(50-AGA-GTTTGATCMTGGCTCAG-30) and 1492R(50-GGTTACCTTGTTACGACTT-30) [13].

The purified gene fragment was sequenced by Shanghai Sangon Biological Engineering Technology & Services Co., Ltd. Phylogenetic neighbors were identified using GenBank (http://www.ncbi.nlm.nih.gov/blast/), and multiple alignments

112 W. Feng et al.

were generated using CLUSTAL X [14]. A phylogenetic tree was constructed from the evolutionary distance data with the neighbor-joining [15], maximum likelihood PHYLIP (phylogeny inference package), version 3.5c. Department of Genetics, University of Washington, Seattle) and maximum parsimony [16] methods using MEGA 5.0, and bootstrap values were calculated based on 1,000 replicates.

12.3 Results

12.3.1 Decoloriation of Azo Dyes by Growing Cells

Xylene Fast Yellow2G decolorizing bacteria were isolated by the enrichment culture containing Xylene Fast Yellow2G in the medium. Among isolated bacteria, 15 strains showed the decolorization of Xylene Fast Yellow2G with different rates in the liquid medium. Of these, an isolate 5-8 was selected as having the most efficient ability to decolorize Fast Yellow2G. Fast Yellow2G were decolorized more than 80 % (Fig. 12.2) in 48 h, and almost completely decolorized in 60 h.

12.3.2 Morphological, Physiological, and Biochemical Characterization

Strain 5-8 was observed to form yellow colored, smooth, circular, convex colonies of 5 mm diameter with entire margins on LB.

Cells of strain 5-8 were observed to be gram-negative, aerobic, motile with a single polar flagellum and rod-shaped with dimensions of 0.3– 0.4×0.6 – $1.5 \mu m$. Growth was observed at 18–40 °C (optimum 35–37 °C), pH 6.0–9.0 (optimum pH 7.0–8.0) and with 0.5–5% (w/v) NaCl (optimum 1%) (Table 12.1).

Fig. 12.2 The color contrast of two culture disposed by centrifugation with one culture bacteria (*right*) and the other blank (*left*), after 48 h

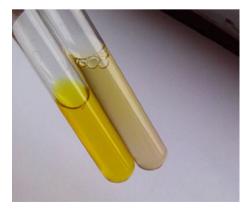


Table 12.1 Part of the physiological and biochemical characteristics of strain 5-8

Characteristic	Strain 5-8	
Cell size (µm)	0.3~0.4 × 0.6~1.5	
Temperature growth range (°C)	18–40	
Optimum temperature	35	
pH range	6.0-9.0	
NaCl range (%, w/v)	0.5–5	
Nitrate reduction to nitrite	+	
Tween 80	+	
Carbon source utilisation	+	
Lactose		
Xylose	_	
Maltose	+	
Sucrose	+	
Mannose	+	
Inulin	_	
Salicin	+	
Galactose	_	
I-Arabinose	-	
Mannitol	-	
Cellobiose	+	
D-Arabinose	_	
Raffinose	-	
Trehalose	_	
Melibiose	-	
Hydrolysis of Gelatin	+	
Hydrolysis of Starch	-	

All data are obtained from this study unless indicated otherwise. +, positive; -, negative. All strains could use Lactose, Maltose, Sucrose, Mannose, Salicin, Cellobiose as Carbon source; none of strains could use Xylose, Inulin, D-Arabinose, Raffinose, Trehalose, Melibiose; hydrolysis of Gelatin positive and Starch hydrolysis was negative

12.3.3 16S rRNA Gene Sequencing and Phylogenetic Analysis

Comparison of the almost complete 16S rRNA gene sequence (1,440 bp, Fig. 12.3) of strain 5-8 showed that it is most closely affiliated to the genus *Stenotrophomonas*. The highest 16S rRNA gene sequence similarity was found with *Stenotrophomonas maltophilia* (99 %). The neighbor-joining phylogenetic tree showed that 58 clustered with *S. maltophilia* 6B2-1 (bootstrap value of 90 %). The relationship between these two strains was maintained using maximum likelihood and maximum parsimony (Fig. 12.4).

114 W. Feng et al.

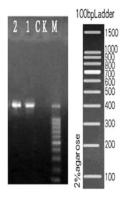


Fig. 12.3 Electrophoretogram of PCR amplified 16S rRNA gene. *Marker* 100 bp ladder; 1, 2 amplified 16S rRNA gene of strain 5-8 by primer 27F and 1492R

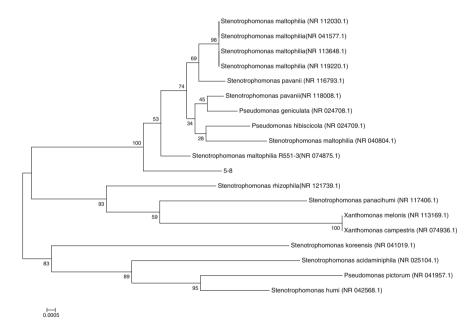


Fig. 12.4 Neighbor-joining phylogenetic tree based on 16S rRNA gene sequences showing the positions of strain 5-8 and related taxa. Bootstrap values (expressed as percentages of 1,000 replications) are shown at branching points. *Bar* 0.0005 substitutions per site

12.4 Discussion

16S rRNA gene sequence analysis indicated strain 5-8 to be most closely related to *S. maltophilia* R551-3 (99 % sequence similarity). Generally accepted criteria for delineating bacterial species state that strains showing less than 97 % of 16S rRNA gene sequence similarity or lower than 70 % of DNA–DNA relatedness in current bacteriology are considered to belong to different species [17].

Goodfellow et al. [18] reported that the DNA-DNA relatedness provides a reliable way of distinguishing between representatives of species that share high 16S rRNA gene sequence similarity.

Xylene Fast Yellow2G was used as the experimental dye in cell's decolorization rate. The culture condition for the highest decolorization rate of Xylene Fast Yellow2G by the strain was at 37 °C, pH 6–9. Most wastewaters from textile industries have a neutral to alkaline pH (around 7–11). The strain 5-8 used in the present study was able to degrade the dye efficiently in the range of pH 6–9 similar to earlier report.

Citrobacter sp. CK3 exhibited decolorization of Reactive Red 180 at pH 6.0 and 7.0, with 96 % decolorization in 48 h, while at pH 10 and 11, the decolorization of dye was completely inhibited [19] In view of this, the strain 5-8 seems to be better candidate for bioremediation with low pH.

Stenotrophomonas maltophilia sp. was able to degrade wide variety of textile dyes and consistent with earlier reports of the bacterial species [20, 21]. In this study, strain 5-8 also showed ability to decolorize the structurally different textile dyes in plain distilled water. This approach can provide more practical and cost-effective way of bioremediation.

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Chapter 13 In Silico Cloning and Sequence Analysis of F3H Gene in *Raphanus sativus* L.

Guang Ma and Jiping Guo

Abstract Flavanone 3-hydroxylase is a key enzyme in the biosynthetic pathway of plant flavonoids. The complete cDNA sequence of *Raphanus sativus* L. F3H gene was in silico cloned using *Brassica napus* F3H-2 gene cDNA sequence (DQ513329.1) as a probe. Then the hydrophilicity, secondary structure, and advanced structure of F3H protein in *R. sativus* L. were analyzed using bioinformatics methods. The results showed that the cDNA was 1,437 bp, contains an open reading frame of 1,077 bp, encoding 358 amino acids. The protein coded by *R. sativus* L. F3H gene cDNA showed 98 % similarity to *B. napus* F3H-2. Its secondary structure contains 38.55 % alpha helix, 41.9 % random coil and 4.93 % beta turn. The homology analysis of 3D structure showed that the three-dimensional structure of the protein is a compact globular structure.

Keywords Flavanone 3-hydroxylase • 3D structure • *Raphanus sativus* L. • Cloning • Bioinformatics

13.1 Introduction

Flavanone 3-hydroxylase (F3H; EC 1.14.11.9) is a key enzyme of flavonoid biosynthesis in plant. It catalyzes hydroxylation of C-ring third position hydroxyl catalytic at flavanone to form two hydrogen flavanols [1–3]. It was first found in the crude extracts of violet, and the properties studied by parsley cell culture [4]. Subsequently, F3H gene was cloned from the petunia and showed high activity by functional expression in *Escherichia coli* [5]. Later, more characteristics of F3H gene in different species were reported such as barley, begonia, alfalfa, maize, *Arabidopsis thaliana* and perilla. In different species, expression characteristics of F3H are different. In petunia and snapdragon, regulation of flavonoid biosynthesis pathway of the preliminary steps and late stage is different. In petunia F3H gene belongs to the early gene, but it belongs to the late genes in antirrhinum. In maize, the flavonoid biosynthesis pathway is through

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G. Ma (⊠) · J. Guo

118 G. Ma and J. Guo

co-regulation to produce anthocyanin in organizations, but flavonol F3H gene expression is only consistent with accumulation of flavonol in anthers [6]. In the study of alfalfa, F3H is also expressed in the roots and nodules, but its role is not clear.

Flavonoids have very high application value in medical and health care. As an important flavonoid biosynthesis gene, F3H gene of *Brassica napus* had been cloned, however, it was little reported in other *Brassica* plants. There is no report on *Raphanus sativus* L. In this study, with the *B. napus* F3H gene cDNA as probe, the F3H gene of *R. sativus* L. was in silico cloned. We obtained the F3H cDNA coding sequence. The sequence and coding protein characteristics are studied, and the three-dimensional structure was constructed homology. The study has laid a solid basis for further research on F3H function.

13.2 Materials and Methods

13.2.1 BLAST Searching of Raphanus sativus L. EST Databases

The cDNA sequence of *B. napus* flavanone 3-hydroxylase 2 (F3H-2) mRNA (GenBank: DQ513329.1) was used as a probe to search the *R. sativus* L. expressed sequence tag (EST) database (http://www.ncbi.nlm.nih.gov/dbEST/) for a homologous clone, using the BLAST program. The EST sequence of score \geq 100 and length \geq 100 bps selected from the blast result, were generated contigs (http://pbil.univ-lyon1.fr/cap3.php). The longer contig was used as the second probe. The above step was not repeated until the newly generated probe could not be elongated. This approach led to a sequence as a putative *R. sativus* L. F3H gene cDNA.

13.2.2 Bioinformatics Analysis of the Raphanus sativus L. F3H Gene

The putative *R. sativus* L. F3H gene cDNA was analyzed by bioinformatics software. The cDNA sequence was submitted to open reading frame software (http://www.ncbi.nlm.nih.gov/gorf/gorf.html), to find the start codon and termination codon positions. The protein sequence of F3H gene of *R. sativus* L. and *B. napus* were aligned by multiple sequence alignment software clustalw2.1 (http://www.ebi.ac.uk/Tools/msa/clustalw2/) to compare their protein similarity. Primary structure characteristics of *R. sativus* L. F3H protein was analyzed using online server (http://www.expasy.org/tools/protparam.html). Its secondary protein structure was analyzed in online server (http://npsa-pbil.ibcp.fr/cgi-bin/npsa_automat.pl?Page=/NPSA/npsa_sopma.html). Hydrophobicity analysis used ProtScale program (http://www.expasy.org/cgi-bin/protscale.pl). The neighbor-joining phylogenetic tree was constructed by Mega 5.05 software.

13.2.3 The Prediction of F3H Protein Three-Dimensional Structure

The three-dimensional structure of the protein was predicted using Swiss-Model online server (http://swissmodel.expasy.org/workspace/index.php?func=modelling_simple1). The three-stage structure model is shown by the CN3D software.

13.3 Results

13.3.1 In Silico Cloning of Raphanus sativus L. F3H Gene

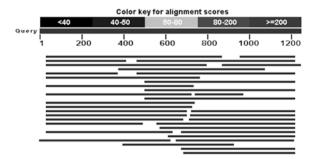
With the cds of *B. napus* F3H-2 mRNA (GenBank: DQ513329.1) as probe, 24 EST sequences (score \geq 100 and length \geq 100) were found by Blastn searching the *R. sativus* L. EST database, in NCBI (Fig. 13.1).

The selected sequences were saved in a file in FASTA format. The file was submitted to online CAP software. One contig was obtained by assembling EST sequences. It was putative *R. sativus* L. F3H cDNA (Fig. 13.2). Searching the cDNA sequence for potential coding regions by ORF finder, an entire open reading frame (ORF) of 358 amino acids was detected with a potential start codon at the 196th site and a stop codon at the 1,272th site (Fig. 13.2).

13.3.2 Gene Comparison Between Raphanus sativus L. F3H and Brassica napus F3H-2 cDNA

The alignment of *R. sativus* L. and *B. napus* F3H-2 amino acid sequences, constructed using ClustalW program, suggest that *R. sativus* L. F3H amino acid sequences are similar to *B. napus* F3H-2. There are only 7 different amino acids of a total of 358 amino acids (Fig. 13.3).

Fig. 13.1 BLAST result of F3H cDNA in the *Raphanus sativus* L. EST databank



120 G. Ma and J. Guo

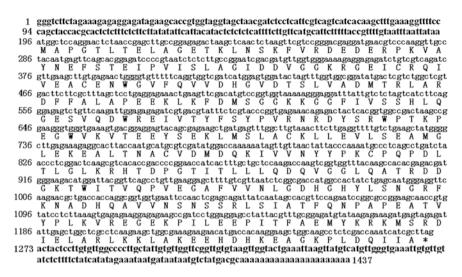


Fig. 13.2 Nucleotide sequence and deduced amino acid sequence of R. sativus L. F3H gene

```
R. sativus L. MAPGTLTELAGETKLNSKFVRDEDERPKVAYNEFSTEIPVISLAGIDDVGGKRGEICRQI 60
          MAPGTLNELAGETKLNSKFVRDEDERPKVAYNEFSTEIPVISLAGIDDVGEKRGEICRQI 60
B. napus
          R. sativus L. Veacenwgvfovvdhgvdtslvadmtrlardffalapeeklkfdmsgckkggfivsshlq 120
B. napus
          VEACENWGVFQVVDHGVDTSLVADMTRLARDFFALPPEEKLKFDMSGGKKGGFIVSSHLQ 120
           ********************
R. sativus L. gesvodwreivtyfsypvrnrdysrwptkpegwykvteeyseklmslackllevlseamg 180
B. napus
          GESVQDWREIVTYFSYPVRNRDYSRWPTKPEGWVKVTEEYSERLMGLACKLLEVLSEAMG 180
           *************************
R. sativus L. LEKEALTNACVDMDQKIVVNYYPKCPQPDLTLGLKRHTDPGTITLLLQDQVGGLQATRDD 240
B. napus
           LEKEALTNACVDMDQKIVVNYYPKCPQPDLTLGLKRHTDPGTITLLLQDQVGGLQATRDD 240
           *********************
R. sativus L. GKTWITVQPVEGAFVVNLGDHGHYLSNGRFKNADHQAVVNSNSSRLSIATFQNPAPEATV 300
B. napus
          GKTWITVQPVEGAFVVNLGDHGHYLSNGRFKNADHQAVVNSNSSRLSIATFQNPAPEATV 300
           *************************
R. sativus L. YPLKVREGEKPILEEPITFAEMYKRKMSRDIELARLKKLAKEEHDHKEACKPLDQIIA 358
B. napus
          YPLKVREGEKPILEEPITFAEMYKRKMSRDLELARLKKLAKEEHDHKEAAKPLDQIIA 358
```

Fig. 13.3 Homology of R. sativus L. and Brassica napus F3H

13.3.3 Secondary Structure of Raphanus sativus L. F3H Protein

The secondary structure of *R. sativus* L. F3H protein was predicted utilizing online service (http://npsa-pbil.ibcp.fr/cgi-bin/npsa_automat.pl?page=/NPSA/npsa_sopma. html). The results showed that *R. sativus* L. F3H protein have 38.55 % alpha helix, 15.36 % extended strand, 41.9 % random coil, and 4.19 % beta turn.

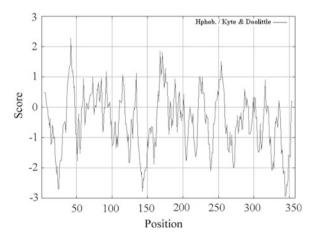
13.3.4 Hydrophilicity Prediction of Raphanus sativus L. F3H Protein

Hydrophilicity of R. sativus L. F3H protein was predicted utilizing Program of ProtScale (Kyte and Doolittle). The results showed that most sites of R. sativus L. F3H protein is in the hydrophilic region (Score: 2.278 to -2.293) (Fig. 13.4). It was concluded that the R. sativus L. F3H protein is a hydrophilic protein.

13.3.5 Analysis of Raphanus sativus L. F3H Phylogenetic Tree

F3H cDNA sequence of *R. sativus*, *Brassica rapa naringenin*, *B. napus* (F3H-1), *B. napus* (F3H-2), *B. rapa* subsp. *campestris* (F3H-1), *A. thaliana*, *Arabidopsis lyrata*, *Eutrema salsugineum*, *Parrya nudicaulis*, *M. incana*, *Capsella rubella*, *Cardamine resedifolia* were downloaded from genebank database. These sequences were stored in a FASTA file including F3H cDNA sequence of *R. sativus* L. The neighbor-joining phylogenetic tree was constructed with the FASTA file by software MEGA 5.05 (Fig. 13.5). The phylogenetic tree analysis showed *B. napus*, *B. rapa*, *R. sativus* L. clustered into one branch. *R. sativus* L. F3H cDNA is close to *B. napus* and *B. rapa*. The genetic relationship between the F3H cDNA is consistent with the phylogenetic tree.

Fig. 13.4 Hydrophilicity profile of *R. sativus* L. F3H protein



122 G. Ma and J. Guo

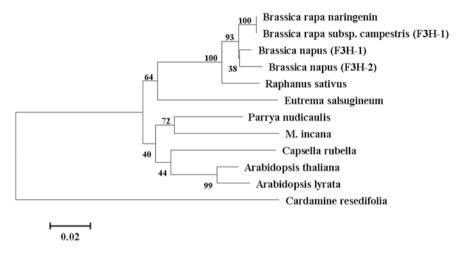


Fig. 13.5 Analysis of R. sativus L. F3H protein phylogenetic tree

13.3.6 Advanced Structure of Raphanus sativus L. F3H Protein

The structure prediction from primary to advanced structure is an important task in the field of protein research. The three-dimensional structure model of *R. sativus* L. F3H protein was predicted by the Swiss-Model server, by homology modeling, based on the available structures. The homologous sequence of pdb1gp6A and pdb1w9yA with high homology and three-dimensional structure were two models. Because the C terminal of *R. sativus* L. F3H is poor homology to *B. napus*, and less than 10 amino acids, the template of pdb1w9yA matching 2-332 to *R. sativus* L. F3H was selected for homology modeling (Fig. 13.6). The results were close to the protease real space conformation.

Fig. 13.6 The predicted three-dimensional structure of *R. sativus* L. F3H protein



13.4 Discussion

In silico cloning is a method developed in recent years for functional gene identification using genome and EST database. Compared to traditional methods, such as molecular hybridization, screening of genomic or cDNA library, it is advanced for low cost, high efficiency, easy operation, etc. With more and more EST and genome sequencing data being reported, it would become possible and feasible to isolate and identify the functional genes by in silico cloning. Many successful examples strongly support the fact that in silico cloning is an absolutely feasible tool for gene cloning and presents some advantages compared to the traditional methods.

In this study, we present in silico cloning and characterization of one flavonoid biosynthetic genes, flavanone-3 hydroxylase of *R. sativus* L. The F3H transcripts are abundant in seed of *R. sativus* L. In addition, a sequence similar to the Myb-like box A/CACC T/AAA/CC found in several genes of the flavonoid pathway. It was thought to be involved in the regulation of the flavonoid biosynthesis pathway and is present in inverse orientation between -1,024 and -1,016 upstream from the start codon of the *R. sativus* L. F3H gene. Future analysis of *R. sativus* L. plants transformed with the F3H promoter-GUS fusions may demonstrate the involvement of the Myb-like sequence in the expression of this gene. The structure and function of *R. sativus* L. F3H were analyzed and predicted using bioinformatics methods successfully. The results revealed that it is a convenient technique for cloning novel gene by searching EST database with homologous gene of model living things. To our knowledge, it was the first report on cloning of *R. sativus* L. F3H cDNA with in silico cloning. This research achievement will provide the theory and reference for flavonoid biosynthesis research in *R. sativus* L.

Acknowledgments This work was supported by the Hebei Higher School Science and Technology Research Youth Fund Project (No. QN2014328) and Hebei province Science and Technology Research and Development Program (No. 12212702).

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124 G. Ma and J. Guo

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Chapter 14 The Comparative Study of Fermentation Capability of Wheat Beer Yeast

Jinxu Sun, Yue Wang and Cuicui Wang

Abstract In this study, the wheat beer fermentation performances of three beer yeasts S-15, S-16 and S-17 were comparatively studied. Here the best strain was selected for fermentation of wheat beer. Comparison of three strains had been experimented at fermentation capability, such as sugar-reducing rate, higher ethanol concentration, fermentation degree, agglutinating, and alcohol yield. Among these three yeasts, sugar-reducing rate of S-17 was the quickest and up to 0.1452 Brix/h. The data showed that S-17 could rapidly utilize sugar to ferment and was better in the brewing process to initiate fermentation. After 96 h of fermentation, alcohol yield of S-17 was slightly higher than other yeasts in agreement with sugarreducing rate and up to 3.55 % (v/v). The real fermentation degree and the apparent fermentation degree of S-17 were also higher than the other two yeasts. Three beer yeasts belong to weak yeast coherency and the sequence of yeast coherency was S-17 > S-16 > S-15. The total content of high alcohol for S-17 was 126.39 mg/L, in which, the main ingredient is isopentanol and up to 56.48 %. The content of the high alcohol produced by S-17 during beer fermentation is in the normal range. Under the same conditions, strain S-17 showed faster fermentation speed, shorter fermentation period, higher alcohol degree, and stronger cohesion property. So S-17 was selected for wheat beer fermentation.

Keywords Wheat beer • Fermentation degree • High alcohol • Alcohol

14.1 Introduction

In recent years, world barley production had slumped because of dry climate, so barley price was higher than beforehand. A lot of beer companies were looking for some sort of magic bullet that would solve this problem. Based on this, wheat came

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J. Sun et al.

into view. Wheat can be used in place of barley and has long been known as the main raw materials for beer production. Wheat beer was also taking off rapidly in the emerging economies such as China. The first wheat beer in China was made and used in Shandong province.

Wheat beer was popular among consumers because of its wide source of raw materials, cheap price, mellow taste, and unique flavor. However, there are still many problems with respect to wheat beer fermentation technology, of which the most important is the strain selection for wheat beer fermentation. The content of ethanol and high alcohols produced by strain fermentation greatly affect the quality and flavor of wheat beer [1]. In this study, we determined the fermentation parameters of beer yeast strain S-15, S-16 and S-17, including the fermentation degree, high alcohol content, and alcohol content index, in different culture medium in order to select superior strains suitable for wheat beer fermentation.

14.2 Materials and Methods

14.2.1 Material and Equipment

Beer yeast strains, S-15, S-16 and S-17, were stored in Department of Life Sciences Laboratory of Hengshui University; P-type Pipette was purchased from Jia'an Analytical Instrument Company in Shanghai); gas chromatograph (7890A) was purchased from Agilent Technologies Instrument Co. Ltd.; isobutanol, *n*-butanol and isoamyl alcohol were obtained from Ji'nan Aohui Chemical Co. Ltd.

14.2.2 Medium

Yeast culture medium (YEPD) (w/v): natural pH, 121 °C, 0.1 MPa sterilized for 20 min.

Malt extract medium: the amount ratio of crushed malt to distilled water is 1:4, the medium was mashed according to certain route.

14.2.3 Standard Solution of High Alcohol

0.1 mL chromatography grade of N-propanol, isobutanol, isopentanol, n-butanol was accurately mixed with 60 % ethanol solution up to 10 mL to yield the 1 % mixed standard solution, and stored at 4 °C.

14.2.4 Determination of Alcohol Content by Rapid Oxidation Method

As described in the literature [2, 3], distilled alcohol may be oxidized by excess potassium dichromate to glycolic acids under acidic conditions:

$$2C_2H_5OH + 2Cr_2O_7^{2-} + 16H^+ \rightarrow 4Cr^{3+} + 3CH_3COOH + 11H_2O$$

The remaining potassium dichromate was reduced by potassium iodide:

$$Cr_2O_7{}^{2-} + 6I^- + \ 14H^+ \rightarrow 2Cr^{3+} + \ 3I_2 + \ 7H_2O$$

Precipitated iodine was titrated by standard sodium hyposulfite $I_2+2S_2O_3^{2-}\to 2I^-+S_4O_6^{2-}$.

Alcohol concentration was calculated by dose of sodium hyposulfite.

14.2.5 Determination of the Fermentation Degree

(1) Determination of the appearance fermentation degree

1 g yeast mud is added in an Erlenmeyer flask filled with 150 mL wort, and the flask was delivered to the incubator for fermentation. The temperature was set at 25 °C. The flask was shaken once every 8 h and removed after being incubated for 3–4 days. The yeasts were removed by filtration and the proportion of fermentation broth measured, then the residual extract concentration was calculated.

(2) Determination of real fermentation degree

After fermentation, the yeasts were removed from the wort by filtration and the filtrate was heated to evaporate the ethanol by a simmer until the filtrate accounted for 1/3 of the original volume. The original volume was restored by adding distilled water, then the specific gravity was determined and the residual extract concentration calculated.

14.2.6 Determination of Cohesion Property

1 g of processed yeast sample was added and suspended in a centrifuge tube supplemented with 10 mL acetate buffer. The yeast suspension was heated in a water bath at 20 °C for 20 min. The tubes were gently shaken for 5 min to make the yeasts resuspended, and then left to stand. At 20-min intervals, the volume of

J. Sun et al.

precipitated yeasts was recorded once at 1 min. The accumulated deposition amount of the yeasts up to 10 min was defined as Burns Value [4]. This experiment is known as the Burns Test [5]. The volume of yeast sedimentation >1.0 mL is defined as strong agglutination, while the volume <0.5 mL, weak agglutination.

14.2.7 Beer Fermentation Experiment

Wort concentration: 10 Brix; yeast inoculum: 10 %; fermentation temperature: 20 °C; fermentation period: 4 days; main fermentation period: the first 2 days. Timing sampling was performed to measure the fermentation performance.

14.2.8 CO₂ Weightlessness Experiment [6]

The flask supplemented was weighed with distilled water and fermented liquid each 12 h and calculated the weight loss of CO_2 .

14.2.9 Determination of Fermentation Speed

135 mL of 10 Brix wort was added into a 250 mL sterilized erlenmeyer flask. The seed liquid was inoculated at a ratio of 10 %, and fermented at 20 °C. The concentration of the fermentation liquid was determined everyday and records made for consecutive comparison.

14.3 Results and Analysis

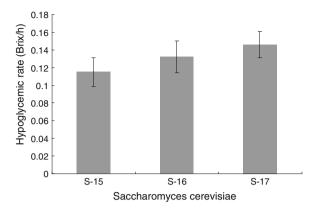
14.3.1 Fermentation Performance Comparison of Different Strains

Wheat beer fermentation yeasts S-15, S-16 and S-17 were selected for this study. Fermentation speed, alcohol content, fermentation degree, and high alcohol content of the three yeasts were compared to select the best strain for beer fermentation.

14.3.1.1 Fermentation Speed

After expanding culture, the selected strains of beer yeasts S-15, S-16, and S-17 underwent fermentation culture (10 % inoculation, fermentation temperature 20 °C,

Fig. 14.1 The hypoglycemic rate comparing of three brewer yeast for reducing sugar in 48 h



liquid volume 75 %) for 48 h. Then sugar content of the fermentation liquid was determined and hypoglycemic rate was simultaneously calculated. The results are shown in Fig. 14.1.

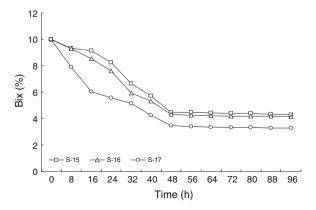
Figure 14.1 showed that of the three strains of beer yeast cultured for 48 h, strain S-17 exhibited maximum hypoglycemic rate of 0.1452 Brix/h. The results indicate that strain S-17 can quickly utilize sugar for fermentation.

Quick fermentation initiation and rapid fermentation mediated by the strain S-17 is beneficial for beer fermentation. So, compared to strains S-15 and S-16, strain S-17 is more advantageous for wheat beer fermentation.

After expanding culture, the selected strains S-15, S-16 and S-17 were inoculated in the wheat beer fermentation medium and incubated for 96 h. The sugar content was determined at each 8 h, and the result is shown in Fig. 14.2.

The results show that the three strains were at a main fermentation stage within the first 48 h. Strain S-17 exhibited the fastest hypoglycemic speed, followed by strains S-16 and S-15. After 48 h fermentation, the three strains entered the post fermentation period, during which period, all of the three strains showed slow hypoglycemic speed with little change. After 96 h, reducing sugar of strain S-17 fell to 2.68 Brix. In the

Fig. 14.2 The reducing sugar curve of three brewer yeasts



J. Sun et al.

fermentation process, all beer yeasts showed an adaptation period after being inoculated in wort, thus, strain with fast fermentation initiation and short adaptation period will soon become the dominant bacteria. No bacteria contamination is conducive to fermentation. Comparison of hypoglycemic correlation curves of the three strains revealed that strain S-17 was suitable for the fermentation production.

14.3.1.2 Alcohol Comparison

After expanding culture, the selected strains of beer yeasts S-15, S-16 and S-17 underwent fermentation culture (10 % inoculation, fermentation temperature 20 °C, liquid volume 75 %) for 96 h. Then the alcohol content was determined and the result is shown in Fig. 14.3.

The data revealed that the alcohol content produced by the three strains showed minor differences. Relatively, the alcohol content of strain S-17, was more than that of the other two strains and reached 3.55~% (v/v), which is consistent with the hypoglycemic results.

14.3.1.3 Comparison of Fermentation Degree

After fermentation for 96 h, the real fermentation degree and the apparent fermentation degree of the three strains were determined. The results are shown in Fig. 14.4. The data showed that the apparent and real fermentation degree of strain S-17 is higher than the other two strains, which indicates that strain S-17 showed better fermentation performance.

14.3.1.4 Cohesion Comparison

Cohesion phenomenon, which is one of the main physiological characteristics of yeast, has been found and applied for a long time. Different yeast strains exhibited

Fig. 14.3 The alcohol content comparing of three brewer yeast

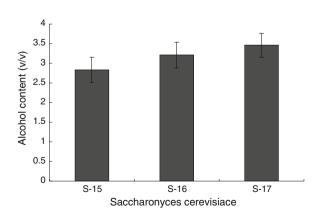
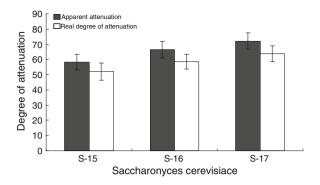


Fig. 14.4 The fermenting degree comparing three brewer yeasts

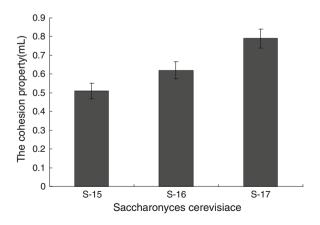


large differences in cohesion property. The sedimentation speed is quick for the yeast with strong cohesion, and this kind of yeast is easily separated from fermented liquid, and vice versa. Thus, when selecting fermentative strains, yeast with moderate cohesion performance should be recommended. The best strain could not only achieve a high fermentation degree but shows high sedimentation speed, which is easy to separate [7]. The cohesion property of the three yeasts after fermentation was determined by Burns Test as shown in Fig. 14.5. The test showed that the selected three yeast strains all exhibited weak cohesion, wherein strain S-17 showed slight stronger coherency, and strain S-15 was the weakest. Through artificial evaluation, the order of the cohesion effect of the three strains after fermentation was S-17 > S-16 > S-15.

14.3.1.5 Overall Fermentation Characteristics Comparison of the Three Yeast Strains

After fermentation for 96 h, overall fermentation index comparison of the three yeasts is shown in Table 14.1.

Fig. 14.5 The cohesion comparing of three brewer yeasts



J. Sun et al.

Strain	Brix (v/v, %)	Apparent attenuation (%)	Real degree of attenuation (%)	Cohesion (mL)
S-17	3.55	72.35	66.80	0.8
S-16	3.15	66.40	58.25	0.6
S-15	2.89	57.12	50.21	0.5

Table 14.1 The fermentability comparing of three brewer yeasts

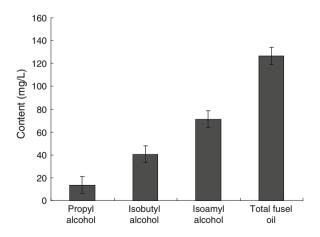
Table 14.1 shows that strain S-17 exhibited faster fermentation speed, shorter fermentation cycle, higher alcohol content, higher fermentation degree, and stronger coherence, compared to the other two strains. Therefore, strain S-17 was selected as the starting strain for subsequent experiments.

14.3.2 Determination of High Alcohol Content of Strains S-17 During Fermentation

As one of the main sources of beer aroma, high alcohols endow beer rich taste and make the wine full-bodied. But excessive high alcohols will affect the taste of beer. The content of senior alcohols was generally 50–150 mg/L in the common beer of low temperature fermentation. The content of senior alcohols exceeded 100 mg/L will make the beer smelly. Therefore, in the beer production process, higher alcohol content must be controlled at a proper level by a series of measures, including yeast screening, raw materials selection and match, and fermentation conditions control, etc., among which, the yeast strain is the main factor affecting the formation of high alcohol content. Thus, selection of beer yeast suitable for fermentation is of vital importance.

After the fermentation of strain S-17, the content of high alcohols was determined. As can be seen from Fig. 14.6, the high alcohols content of strain S-17 after

Fig. 14.6 The fusel oil content of S-17 after fermenting



fermentation was 126.39 mg/L, wherein the isopentanol content was most abundant, accounting for 56.48 % of the total amount. The high alcohol content was within the normal range of beer, but slightly higher. High senior alcohol content is disadvantageous to human health and affects the beer flavor. Thus, to make the strain suitable for industrialized application, the content of high alcohols should be reduced.

14.4 Conclusion

In this study, the fermentation performance of three strains of wheat beer fermentation yeast S-15, S-16, and S-17 were compared. Of the three strains, S-17 exhibited fast fermentation speed, short fermentation cycle, high alcohol content, high fermentation degree, and strongest cohesion under the same conditions. High alcohols content produced by the strain S-17 reached a total amount of 126.39 mg/L, wherein the content of isoamyl alcohol was most abundant. The high alcohol content was within the normal range of beer, though a little higher. High content of senior alcohols would bring about unhealthy impacts on the human body and influence the beer flavor. Therefore, in order to make industrialized application of the yeast strain, the content of high alcohols should be decreased.

Acknowledgments This work was supported by the Science and Technology Research Project for Hebei Higher Education (Z2013091).

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Chapter 15 Screening of Dual Defects Strain and Effects on L-Isoleucine Production in Escherichia coli NML

Linan Yu, Huiyan Liu, Haitian Fang and Qing Wu

Abstract Original strain *Escherichia coli* N12 with lifting the repression of L-isoleucine was mutated mutagenesis by diethylsulfate (DES), then L-isoleucine producing strain named *E. coli* NML were obtained. The effects of *E. coli* K12, N12, and NML in fermentation process on biomass, yield of L-isoleucine and glucose consumption rate were studied by carrying out fed-batch fermentation by analyzing the metabolic sysnthesis process. At 48 h, L-isoleucine production in the fermentation medium of N12 and NML was 2.25 and 4.64 g/L, respectively, and it was not detected in *E. coli* K12. Results show that mutant strains with dual genetic markers (Met⁻ + Leu⁻) can help with the synthesis of L-isoleucine, which also provide some frame of reference for the production of other amino acids.

Keyword Escherichia coli NML · Screening · L-Isoleucine · Production

15.1 Introduction

L-Isoleucine (L-Ile), one of the eight essential amino acids and one of the three branched-chain aliphatic amino acids (BCAAs) that are essential substrates for protein synthesis in all organisms, holds an important place in the amino acid market share [1, 2]. However, productive capacity of L-isoleucine cannot meet the huge market of China. Traditionally, strain improvement was achieved mainly by multiple rounds of random mutagenesis and selection, which are still very useful nowadays [3–5]. In *E. coli*, Ile is synthesised from aspartate by a 10-step reaction (Fig. 15.1) [6, 7]. The first 5 reactions convert L-aspartate to L-threonine, and the second 5 reactions convert L-threonine to L-Ile (Fig. 15.1). Several enzymes in the pathway are regulated by feedback inhibition. Aspartate kinase (AK), homoserine dehydrogenase (HD), and homoserine kinase (HK) are inhibited by L-threonine,

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136 L. Yu et al.

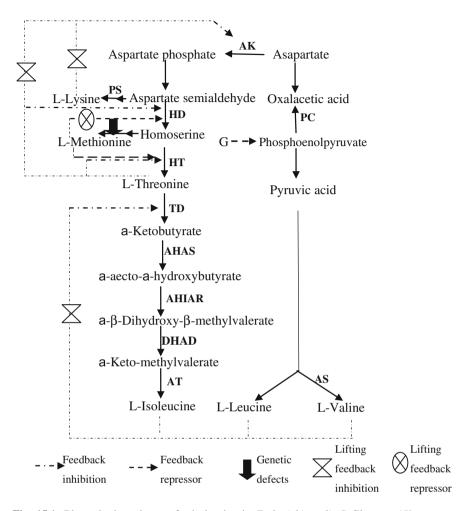


Fig. 15.1 Biosynthesis pathway of L-isoleucine in *Escherichia coli*. *G* Glucose; *AK* aspartate kinase; *ASD* aspartyl semialdehyde dehydrogenase; *HD* homoserine dehydrogenase; *HK* homoserine kinase; *TS* threonine synthase; *TD* threonine dehydratase; *AHAS* acetohydroxy acid synthase; *AHIAR* acetohydroxy acid isomeroreductase; *DHAD* dihydroxy acid dehydratase; *AT* branched chain acid aminotransferase

threonine dehydrogenase (TD) is inhibited by L-Ile, and acetohydroxy acid synthase (AHAS) is inhibited by all three branched-chain amino acids (Fig. 15.1). Because of its longer biosynthesis pathway and tighter regulation, bacterial strains for L-Ile production have not been developed as well as those for the production of amino acids which have a simpler biosynthesis pathway, such as L-lysine [8–10]. Therefore, to improve the efficiency of Ile production, rational metabolic engineering strategies should be implemented.

To accumulate more L-isoleucine, cutting off certain metabolic pathways of other amino acids is important. In this paper, we obtained NML that the synthesis route of

methionine and leucine were cutted by mutating by diethylsulfate (DES) with *E. coli* N12 with analog resistance and studied the changes of biomass, glucose consumption rate, and production of L-isoleucine in the fermentation process. We obtained *E. coli* NML by mutation screening for defects of methionine, lysine, and leucine in *E. coli* N12 with diethylsulfate (DES) by analyzing the pathway and regulation of L-isoleucine biosynthesis. The strategies and researchers for developing L-isoleucine over-producer by metabolic engineering way are proposed and reviewed [11].

15.2 Materials and Methods

Escherichia coli K12 and N12 were preserved in Bioengineering Laboratory of Ningxia University. Methionine (Met) and leucine (Leu) were from Sigma.

The complete medium (CM) contained glucose (5 g/L), tryptone (10 g/L), sodium chloride (5 g/L), and agar (25 g/L). The minimal medium (MM) contained glucose (10 g/L), (NH₄)₂SO₄ (1.5 g/L), KH₂PO₄·3H₂O (1.5 g/L), MgSO₄·7H₂O (0.5 g/L), vitamin H (50 µg/L), MnSO₄·H₂O (0.01 g/L), FeSO₄·7H₂O (0.01 g/L), and agar (20 g/L) and then mixed methionine and/or leucine (0.1 g/L) in MM to be supplemental medium (SM). The production medium contained glucose (10 g/L), (NH₄)₂SO₄ (3.5 g/L), MgSO₄·7H₂O (0.05 g/L), KH₂PO₄·3H₂O (0.1 g/L), and corn steep liquor (15 ml/L).

The Ile production experiments pick out a single point of access to the colonies corresponding basic medium and selective medium after handling 30 min by 1 % diethyl sulfate (DES) and then verify that the amino acid-deficient strain through filter paper method [12]. *E. coli* cells were precultured in seed medium at 36 °C and 200 rpm for 18–24 h and measured OD₆₀₀. The preculture was inoculated into 50 ml fermentation medium to a final optical density (OD₆₀₀) of 1.0 and shaken at 200 rpm for 48 h at 36 °C [13, 14]. Samples were taken every 4 h to determine residual glucose with a biosensor (Institute of Biology, Shandong Academy of Science, China). The biomass was calculated by measuring the OD of the fermentation broth using a spectrophotometer at 600 nm and one unit OD 600 nm was equal to 0.3 g in 1 L of dry cell weight (DCW) [15, 16]. The amino acid concentration was determined using reversed-phase high-pressure liquid chromatography [11, 17].

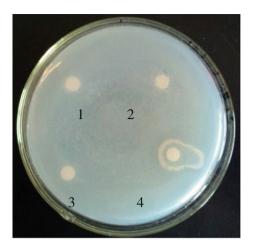
15.3 Results and Analysis

15.3.1 Screening of Genetic Defects on L-Isoleucine Producing Strain

A was determined as the deficient strain when the cells grown around the filter paper containing the amino acid A and did not grow around B [18]. Figure 15.2

L. Yu et al.

Fig. 15.2 The result of auxotrophic test. NML (Met + Leu -). *1* methionine; 2 leucine; 3 lysine; 4 mixing of three amino acid



shows that the strain grown around the sterile filter that contained methionine, leucine, and mixture of methionine, leucine, and lysine did not grow surrounding the sterile filter paper involves lysine, then it was NML (Met + Leu).

15.3.2 Comparison of Growth and Yield of L-Isoleucine in Different Mutants

To study the effect of genetic defects on L-Ile biosynthesis, N12 and NML, the strains selection by mutation were generated, analyzed, and compared to the initial *E. coli* K12.

The growth of these strains in fermentation medium revealed that the K12, N12, and NML strains exhibited similar growth rates (Fig. 15.3a). After 48 h cultivation,

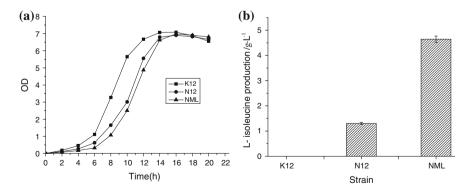


Fig. 15.3 Fermentation of *E. coli* K12, N12 and NML. **a** Growth of different mutant strains; **b** production of L-isoleucine in different mutant strains

the production of L-Ile in these *E. coli* strains was determined (Fig. 15.3b). L-Ile production in the fermentation medium of N12 and NML was 2.25 and 4.64 g/L, respectively, which was higher than the production of the initial strain K12 (Not detected), which indicated that auxotrophic mutants contributed to the synthesis of Ile and also provided a reference for the production of other amino acids.

15.3.3 Fermentation Processing Control of L-Isoleucine in Mutant Strain E. coli NML

The production of L-isoleucine is associated with growth status [14]. As shown in Fig. 15.4, 0–28 h, the rate of glucose consumption, biomass, and L-isoleucine production showed an increasing trend. 28–36 h, the rate of increase of biomass and L-isoleucine production slowed with the rate of consumption of sugar into the stationary phase. After 36 h, biomass and L-isoleucine production entered into the stationary phase with the rate of consumption of sugar declining rapidly.

15.4 Conclusion and Discussion

To improve the efficiency of Ile production, rational metabolic engineering strategies should be implemented. Breeding of auxotrophic and structural analogues resistant mutants is conductive to accumulate L-Ile by changing the biosynthesis of

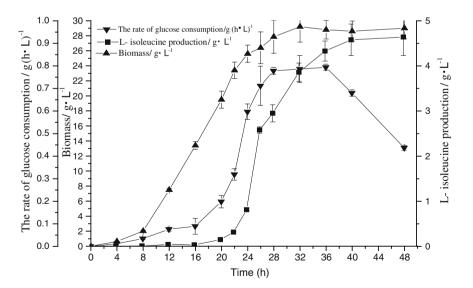


Fig. 15.4 The curve of L-isoleucine fermentation processing in E. coli NML

140 L. Yu et al.

L-Ile [19, 20]. It accumulated more L-Ile by filtering strains with auxotrophic and structural analogs resistant mutants [18, 21]. In this study, L-Ile production in the fermentation medium of NML was higher than *E. coli* N12 and *E. coli* N12 was higher than the wild type K12. At 48 h, production of L-Ile in the fermentation medium of *E. coli* N12 and NML was 2.25 and 4.64 g/L, respectively, and it was not detected in *E. coli* K12.

In *E. coli* N12, lifting of the feedback repression of L-isoleucine was conducted to accumulate more L-isoleucine than the wild type K12. In *E. coli* NML, the biosynthesis of methionine and leucine was done by mutating in *E. coli* N12, which led to increase in L-isoleucine production. These results suggest that breeding of auxotrophic strain is a useful metabolic engineering strategy to improve L-isoleucine biosynthesis and also provide some reference for other amino acids.

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Chapter 16 Effect of MIG1 Gene Deletion on Lactose Utilization in Lac⁺ Saccharomyces cerevisiae Engineering Strains

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Abstract A lactose-consuming Saccharomyces cerevisiae strain EY-510 was constructed by expressing LAC4 and LAC12 gene of Kluyveromyces marxianus in the host strain AY-5. Mig1 is a zinc finger DNA-binding protein that plays a critical role on glucose repression in S. cerevisiae. In order to study in anaerobic condition the degree of glucose repressing galactose metabolism, a deletion fragment MIGIA-KanMX-MIG1B was transformed into AY-5, resulting a Δmig1 strain DY-510. In order to study whether the presence of glucose inhibits the consumption of lactose, a $Lac^{+}\Delta mig 1$ strain RY-510 was constructed by transforming the deletion fragment into EY-510. Galactose consumption was initiated at higher glucose concentrations in the MIG1 deletion strain RY-510 and DY-510 than in the corresponding wild-type strain AY-5 and EY-510, wherein galactose was consumed until glucose was completely depleted in the mixture. On lactose medium, the duration of fermentation for RY-510 was 168 h, whereas the duration for EY-510 was 252 h. The lactose uptake rate was 0.357 g/L/h for RY-510 and that was 0.238 g/L/h for EY-510. The ethanol productivity of RY-510 was 0.127 g/L/h and that was 0.085 g/L/h for EY-510. And in cheese whey powder solution medium, RY-510 was able to produce 30.25 g/L ethanol from 76.8 g/L initial lactose in 190 h, during which EY-510 was able to consume 70.8 % of the initial lactose and produced 24.14 g/L ethanol. Therefore, relieving glucose control provides an approach for constructing lactose-consuming S. cerevisiae.

Keywords MIG1 · Lac⁺ S. cerevisiae · Glucose repression · Whey cheese

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16.1 Introduction

Fuel ethanol fermentations from different renewable resource have received high attention due to increasing petroleum shortage. Whey is a by-product of milk industry and the annual world production is 160 million tons. Whey contains 5 % lactose, which means 8 million tons of lactose needs disposal each year. There are much research on using whey to produce single cell protein or bioethanol because of its high carbohydrate content and availability [1, 2].

Only a few yeast strains are able to ferment lactose to produce ethanol, including, e.g., *Kluyveromyces lactis*, *Kluyveromyces marxianus*, and *Candida pseudotropicalis* [3]. The most common distillers yeast *Saccharomyces cerevisiae*, however, cannot ferment lactose since it lacks both lactose permease and β -galactosidase, which are responsible to transport the lactose into cytoplasm and hydrolyze lactose to monsaccharides. These characteristics prevent *S. cerevisiae* from fermenting cheese whey to ethanol. Most of *the Kluyveromyces* genus is able to ferment lactose which is present in cheese whey. However, *S. cerevisiae* is more excellent on ethanol production than that of *Kluyveromyces*. Alternatives have been sought, such as hydrolyzed lactose by β -galactosidase from another microorganism and subsequent fermentation by *S. cerevisiae*, protoplast fusion between *Saccharomyces* and *Kluyveromyces*, or expression of heterologous genes in *S. cerevisiae*, which are responsible for hydrolyzing lactose.

A Lac⁺ strain EY-510 was constructed which expressed two key genes *LAC4* and *LAC12* of *K. marxianus* on wild-type *S. cerevisiae* strain AY-5. As lactose was first hydrolyzed into glucose and galactose, the presence of glucose can inhibit the transcription of *GAL* genes which encode the enzymes needed by galactose metabolic pathway, i.e., glucose repression. *Mig1*, a zinc finger class of DNA-binding protein, plays a critical role in glucose repression on galactose metabolism. In this work, the *MIG1* gene of the Lac⁺ strain EY-510 was deleted, resulting RY-510. The ability of the two lactose-consuming strains to utilize lactose was examined using anaerobic shake flasks. In addition, the degree of glucose repression was studied in anaerobic batch cultivations on glucose–galactose mixture medium.

16.2 Materials and Methods

16.2.1 Strains, Media, and Growth Conditions

The yeast strains used in this study are listed in Table 16.1. Yeast strain AY-5 and TY-3 and *Escherichia coli* strain DH5 α used in this study were obtained from the Yeast Collection Center of Tianjin Industrial Microbiology Key Laboratory of Tianjin University of Science and Technology, P.R. China. *Escherichia coli* strain DH5 α was grown in LB broth with ampicillin for plasmid maintenance. *S. cerevisiae* and *K. marxianus* strains were incubated in YPD medium at 30 °C. The

Strains	Genotype	Source
AY-5		Guo et al. [5]
TY-5		Guo et al. [5]
EY-510	ath1(1678, 2355): PGK-LAC12 nth1(1215, 1664): PGK-LAC4	This study
RY-510	ath1(1678, 2355): PGK-LAC12 nth1 (1215, 1664): PGK-LAC4 mig1 (499, 1123): KanMX	This study
DY-510	mig1(499, 1123): KanMX	This study

Table 16.1 Yeast strains used in this study

recombination strains were screened on YPD medium supplemented with 1,000 µg/mL G418 and 250 µg/mL zeocin. Lactose fermentation media (containing 5 g/L (NH₄)₂SO₄, 1 g/L KH₂PO₄, 0.5 g/L MgSO₄·7H₂O, 10 mL/L trace element solution, 1 mL/L vitamin solution, and 60 g/L lactose) was prepared according to the method by Verduyn [4] with slight modifications. The galactose and glucose media were prepared similar to lactose fermentation media, replacing lactose with the corresponding sugar. The sugar mixture media (containing 30 g/L glucose, 30 g/L galactose, 5 g/L (NH₄)₂SO₄, 1 g/L KH₂PO₄, 0.5 g/L MgSO₄·7H₂O, 10 mL/L trace element solution, and 1 mL/L vitamin solution) was used to analyze the glucose repression exerted on galactose metabolism.

16.2.2 Plasmid and Strain Construction

The genomic DNA used as templates for DNA amplification was isolated from strains AY-5 and TY-3. The recombinant plasmid pUC-MABK used for MIG1 deletion provided by Zhang et al. [6]. The construction of integrative plasmids pPNLZ and pPALK, which were constructed to express LAC4 and LAC12 genes of K. marxianus was described by Ref. [7]. The pPNLZ and pPALK plasmid were transformed into the AY-5 a/α, respectively, selecting for zeocin⁺ and G418⁺ mutant strains on resistance plates. Then, the G418 resistance gene was deleted through transforming the plasmid pGAPza expressing the Cre recombinase. Finally, the expected phenotypes of AY-5 a/α were selected and mated to obtain the hybrid EY-510. In order to study the effect of glucose repression on galactose-lactose metabolism, a recombinant cassette MIG1A-Kan-MIG1B fragment that was amplified from plasmid pUC-MABK was transformed into AY-5 a/α and EY-510 a/α, resulting DY-510 and RY-510, respectively. The purpose of construction of DY-510 was mainly to verify whether deletion of the MIG1 gene could eliminate glucose repression on galactose metabolism. The recombination of the three plasmids was verified by PCR [8], and primers for verifying the recombination of plasmids, pPNLZ, pPALK, and pUC-MABK were listed in Table 16.2.

J. Zou et al.

Plasmid name	Primer name	Sequence
pPNLZ	N up1	AGGTATGGTGGAGCATTTTATT
	N down1	TTTCGGTTAGAGCGGATGT
	N up2	GCTGTAGGTATCTCAGTTCGGT
	N down2	AACTTGGGTAGTGCTTTCTCC
pUC-MABK	R1-U	TACAAACTCCCACCCTCG
	R1-D	TCATTGGCAACGCTACCT
	R2-U	GCTGTAGGTATCTCAGTTCGGT
	R2-D	TGAAT CCGACTGAGAAATGG
pPALK	A up1	ATCGTCCCTATTCCATTTACTA
	A down1	CAAACCGTTATTCATTCGTG
	A up2	GCTGTAGGTATCTCAGTTCGGT
	A down2	AGTTTCTAATAGCGATACCACCA

Table 16.2 Primers for verifying the recombination of plasmid pPNLZ, pPALK, and pUC-MABK

16.2.3 Standard Solutions for the Medium Used

The trace element solution and the vitamin solution had the following compositions: Trace element solution: 3 g/L EDTA; 0.09 g/L CaCl₂·2H₂O; 0.90 g/L ZnSO₄·7H₂O; 0.60 g/L FeSO₄·7H₂O; 200 mg/L H₃BO₃; 156 mg/L MgCl₂·2H₂O; 80 mg/L Na₂MoO₄·2H₂O; 60 mg/L CoCl₂·2H₂O; 60 mg/L CuSO₄·5H₂O; and 20 mg/L KI. The pH of the trace element solution was adjusted to 4.00 with NaOH, and autoclaved afterward.

Vitamin solution: 50 mg/L p-biotin; 200 mg/L para-amino-benzoic acid; 1 g/L nicotinic acid; 1 g/L Capantothenate; 1 g/L pyridoxine·HCl; 1 g/L thiamine·HCl; and 25 g/L m-inositol. The pH was adjusted to 6.5 and stored at 4 °C after sterile filtration.

16.3 Result and Discussion

16.3.1 Construction of MIG1 Deletion Mutant

The linearized pPNLZ, pPALK were transformed into AY-5, resulting a Lac⁺ strain EY-510. Chromosomal DNA of EY-510 was prepared and used as a template for the PCR. A 1.7 kb fragment was amplified when using the forward primer (N-up1) upstream of *NTH1* together with the backward primer (N-down1) in the plasmid pPNLZ, which verified the existence of the linearized pPNLZ in the *NTH1* locus (Fig. 16.1a). In the same way, a 2,600 bp fragment was amplified using the forward primer N-up2 in the pPNLZ together with a down primer N-down2 in *NTH1* downstream (Fig. 16.1c). The recombination of the pPALK plasmids was verified

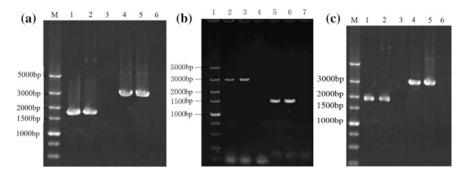


Fig. 16.1 Verification of the transformants of pPNLZ (a), pUC-MABK (b), and pPALK (c). a *M* is DL5000 marker; *1*–2 a 1,700-bp fragment amplified from the EY-510 genome with N-up1 and N-down1 as primers; *3* no fragment amplified from the AY-5 genome with N-up1 and N-down2 as primers; *4*–5 a 2,600 bp fragment amplified from the EY-510 genome with N-up2 and N-down2 as primers; *6* no fragment amplified from the AY-5 genome with N-up2 and N-down2 as primers. **b** *I* DL5000 maker; *2*–*3* a 1.5-kb fragment amplified from the mutant RY-510 with R2-U and R2-D as primers; *4* no PCR product amplified from the Lac⁺ strain EY-510 with R1-U and R1-D as primers; *7* no PCR product amplified from the Lac⁺ strain EY-510 with R1-U and R1-D as primers. **c** M is DL5000 marker; *1*–2 a 1,900-bp fragment amplified from the EY-510 genome with A-up1 and A-down1 as primers; *3* no fragment amplified from the EY-510 genome with A-up2 and A-down2 as primers; *6* no fragment amplified from the AY-5 genome with A-up2 and A-down2 as primers; *6* no fragment amplified from the AY-5 genome with A-up2 and A-down2 as primers;

through the same way and the verification fragment was shown in Fig. 16.1c. The verification of the recombinant cassette *MIG1A-Kan-MIG1B* fragment that was amplified from pUC-MABK by PCR and the result was shown in Fig. 16.1b.

16.3.2 Cultivation on Mixed Glucose-Galactose

In *S. cerevisiae*, glucose was exhausted before galactose. EY-510, RY-510, DY-510, and AY-5 were cultivated on the mixture of 30 g/L glucose and galactose to analyze the influence of glucose control on galactose metabolism in the anaerobic shake flask fermentations.

The sugar consumption can be divided into two phases for AY-5 (Fig. 16.2a). In the first phase, while glucose was present in the medium, the galactose consumption was inhibited. Until glucose in the mixture was completely consumed, the galactose just began to be utilized. The duration of cultivation for the AY-5 strain lasted about 114 h. The same phenomenon was also observed in strain EY-510, in which glucose was consumed before galactose (Fig. 16.2c). Although there was glucose repression in the two strains AY-5 and EY-510, the fermentation time was significantly different. The difference might be caused by the heterologous *LAC12* gene expressed in EY-510 which could help to transport galactose in *S. cerevisiae*

J. Zou et al.

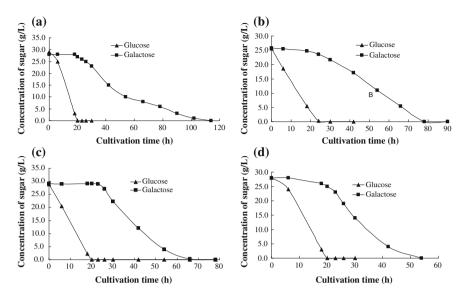


Fig. 16.2 Concentrations of glucose (*filled triangles*) and galactose (*filled squares*) during the anaerobic cultivation of strains **a** AY-5, **b** DY-510, **c** EY-510, and **d** RY-510 on a galactose–glucose mixture. After inoculation with 10 % fresh yeast cells was added into 150 mL galactose–glucose mixture medium, the residual concentration of glucose and galactose in the mixture medium was analyzed every 12 h until the total sugar was completely depleted. Glucose and galactose were detected by HPLC using an Aminex HPX-87H column (Bio-Rad) at 65 °C for separation, using 5 mM H₂SO₄ as the mobile phase at a flow rate of 0.6 mL/min, and were detected by a refractive index detector (Waters 410)

[6]. In strains DY-510, galactose metabolism was initiated despite the high glucose concentration, and the cultivation duration was 78 h, reduced by 34 % compared with the wild-type AY-5 (Fig. 16.2b). In RY-510, the same phenomenon was also observed that glucose and galactose were simultaneously utilized (Fig. 16.2d). As the exogenous *LAC12* gene expressed in RY-510, the fermentation duration of it was significantly shorter than that of DY-510 strain, reduced about 24 h. The data demonstrated that the deletion of *MIG1* gene had partly relieved glucose repression of galactose metabolism.

16.3.3 Characterization of Fermentation on Lactose

In the same culture condition, the two Lac⁺ strains EY-510, RY-510 were cultured on lactose medium. The lactose utilized ability, fermentation period, and ethanol productivity were compared (Fig. 16.3). EY-510 was able to grow on lactose medium but the growth was slow. After 252 h of cultivation on 60 g/L lactose, the residual lactose was 0.24 g/L and the ethanol was 21.38 g/L. The lactose uptake rate

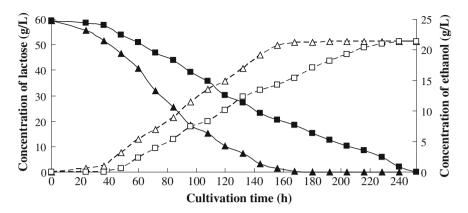


Fig. 16.3 Profiles of Lactose consumption (*solid symbols*, *solid line*) and ethanol production (*open symbols*, *dashed line*) in EY-510 (*solid* and *open squares*) and RY-510 (*solid* and *open triangles*). The duration of fermentation for the *MIG1* deletion strain was 168 h. The residual lactose was 0.15 g/L and the final concentration of ethanol was 21.34 g/L. However, the duration of fermentation for EY-510 was 252 h, the residual lactose was 0.24 g/L, and the concentration of ethanol was 21.38 g/L

of EY-510 was 0.238 g/L/h and the ethanol productivity was 0.085 g/L/h. The performance of RY-510, in which the *MIG1* gene was deleted, differed from that of EY-510 and the lactose was able to be depleted in 168 h. The residual lactose of RY-510 was 0.15 g/L and the final ethanol concentration was 21.34 g/L, which the ethanol productivity was 0.127 g/L/h and the lactose uptake rate was 0.357 g/L/h. The differing performance of RY-510 from EY-510 in utilizing lactose might be due to glucose repression. In EY-510, as glucose and galactose were liberated from lactose, glucose repressed galactose consumption, thereby affecting lactose consumption. The deletion of the *MIG1* gene in RY-510 removed the associated mechanism, accelerating the lactose consumption.

16.3.4 Characterization of Fermentation on Concentrated Cheese Whey Powder Solution (CWPS)

Two transformants were cultivated on CWPS which corresponds to 76.8 g/L initial lactose in the anaerobic shake flask fermentations. During the fermentation of RY-510, the lactose was nearly depleted after about 190 h, resulting in the production of 30.25 g/L ethanol, with a corresponding productivity of 0.159 g/L/h. The EY-510 was able to utilize lactose to produce ethanol in CWPS. During the fermentation, only 70.8 % lactose was utilized in 190 h and 24.14 g/L ethanol was produced. The ethanol productivity was 0.127 g/L/h. Therefore, RY-510 enables utilization of lactose in whey more efficiently to produce more ethanol than EY-510 (Fig. 16.4).

J. Zou et al.

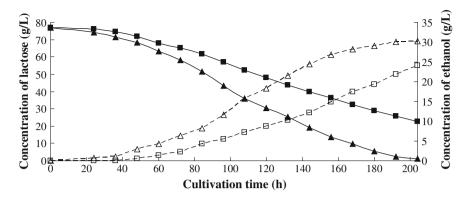


Fig. 16.4 Profiles of lactose consumption (*solid symbols*, *solid lines*) and ethanol production (*open symbols*, *dashed lines*) in RY-510 (*open* and *solid triangles*) and EY-510 (*solid* and *open squares*) during fermentation in CWPS. The duration of fermentation for the RY-510 was 190 h. The residual lactose was 1.98 g/L and the final concentration of ethanol was 30.25 g/L. However, in this duration for EY-510, the residual lactose was 22.45 g/L and the concentration of ethanol was 24.14 g/L

16.4 Conclusion

In *S. cerevisiae*, the presence of glucose affects the galactose metabolism. For the Lac⁺ *S. cerevisiae*, the repression could not only inhibit the galactose utilization on the galactose–glucose mixture, but also affect the lactose consumption as equivalence glucose and galactose were liberated from lactose. While the *MIG1* gene was disrupted, this glucose repression was relieved and the lactose utilization was accelerated. Therefore, the fermentation period on lactose was reduced and the ethanol productive rate was increased. Thus, eliminating glucose repression provides a method for constructing a lactose-consuming *S. cerevisiae* strain.

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Chapter 17 Enzyme Activity Analysis of Protease Produced by Marine Bacteria

Qi Zhang, Xihong He and Hao Liu

Abstract Proteolytic enzymes are ubiquitous in nature, found in all living organisms, and have been applied to many aspects of industry. In this study, 19 strains hydrolyzing protein were screened from deep-sea sediments, of which No. 9 strain has the largest protease activity. Based on morphological characterization and 16S rDNA gene sequences analysis, the strain was identified to be *Bacillus subtilis*. The optimum medium for producing protease was determined through orthogonal test as follows: 2.5 % xylose, 3.5 % peptone, 1.6 % yeast extract and 1 % NaCl, the highest protease activity reached 215.95 U/mL, and increased by 128.06 % than the origin.

Keywords Marine microorganisms • 16S rDNA • Protease • Enzyme activity

17.1 Introduction

Proteases are one of the most important industrial enzymes, accounting for nearly 60 % of all enzyme sales [1], and have been used in various industrial processes, such as pharmaceutical, food, leather, silver recovery, and textiles [2, 3]. However, the use of protease in industrial application is limited by the low activity and instability of the enzymes [4]. In order to improve the stability of the enzyme, the technique of random mutagenesis and site-directed mutagenesis emerged. Pantoliano et al. [5] reported that the rate of inactivation of the combination variant was 300 times slower than that of wild-type subtilisin by six individual amino acid substitution at separate positions. Narhi et al. [6] reported the stabilization of subtilisin from *Bacillus subtilis* by modification of two Asn-Gly sequences. However, this approach is dependent on the number of structural and biochemical data, so the number of variants is rare. While, the microorganisms from exotic

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Q. Zhang et al.

environment are considered to be an important source of enzymes, and they remain catalytically active at the extremes of salinity, pH, and pressure [7, 8]. Therefore, screening of producing stable protease strains and studying the optimum medium for protease production are becoming increasingly important.

In this study, we screened the strains producing protease from deep-sea sediment, and studied the factors on enzyme production to improve the protease activity, so as to provide theoretical support for the screening strains producing stable protease.

17.2 Materials and Methods

17.2.1 Materials and Reagents

Sterile taken deep-sea sediments, R2A Agar (Oxoid) [containing protease peptone 0.5 g, starch 0.5 g, glucose 0.5 g, yeast extract 0.5 g, casein hydrolysate 0.5 g, potassium hydrogen phosphate 0.3 g, sodium pyruvate 0.3 g, magnesium sulfate anhydrous 0.024 g, bacteriological agar 15 g, 1 L distilled seawater], 2216 marine agar [9] (Difco Laboratories), artificial seawater (sodium chloride 24 g, magnesium chloride 10.88 g, sodium sulfate 4 g, calcium chloride 146 g, potassium chloride 0.7 g, sodium bicarbonate 0.2 g, potassium bromide 0.1 g, boric acid 0.027 g, strontium chloride 0.04 g, sodium fluoride 0.003 g, 1 L distilled water) [10].

17.2.2 Isolation and Screening [11]

The strains producing protease were isolated from deep-sea sediments through plate screening. In detail, 1 g of sediments sample was added to 9 mL of sterile artificial seawater, and inoculated, respectively, on R2A and 2216 marine agar plate containing sterile skimmed milk with final concentration of 1 % after gradient dilution $(10^{-2}-10^{-5})$, then cultured for 2 days at 28 °C. Finally, the single colonies that grew well, and obviously formed proteolytic circles were retained for subsequent screening.

The strains with big proteolytic circles were inoculated in fermentation medium, respectively, and cultured in shaken flasks with rotational speed of 200 r/min for 48 h at 28 °C. The enzyme activity was measured by Folin-phenol method [12], and the strain with the highest enzyme activity was as candidate strain.

17.2.3 Bacterial Cultivation and Identification

The candidate strains were cultured in 2216 marine agar plates at 28 °C, and separated single colonies were selected for PCR amplification. The 16S rDNA gene

was amplified by the forward primer 27F (5'-AGAGTTTGATCCTGGCTCAG-3') and the reverse primer 1492R (5'-TACGGCTACCTTGTTACGACTT-3'). The amplification procedure included 5 min at 95 °C, 50 s at 95 °C, 50 s at 52 °C, followed by 30 cycles of 2 min at 72 °C and extend for 10 min at 72 °C. The PCR products were purified and linked with pMD18-T and sequenced with both the forward and reverse primers. The initial nearest neighbor sequences were made by the online BLAST program to the NCBI GenBank database. Sequences were aligned using the CLUS-X program, and phylogenetic tree was constructed with MEGA program.

17.2.4 Protease Activity Analysis

The protease activity was determined by the method of Kembhavi et al. [13]. One unit of protease activity was defined as the amount of enzyme that liberated 1 μ mol of tyrosine in 1 min.

17.2.5 Effects of Different Factors on the Ability of Candidate Strain Producing Protease

Only one factor was changed in each experiment keeping all others constant. The salt content, different carbon and nitrogen sources, and their different concentrations were initially studied at single factor test.

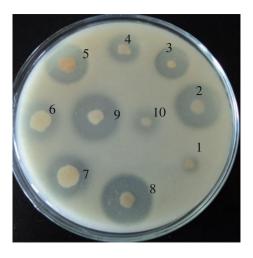
17.3 Result

17.3.1 Screening of Strains Producing Protease and Preliminary Identification of the Ability of Enzyme Production

As shown in Fig. 17.1, four stains displayed relatively larger zone of proteolytic activity when grown on skimmed milk agar medium. Then, by using Folin-phenol method, the ability of enzyme production was measured to screen the candidate strain. The result revealed that the ability of enzyme production of No. 9 is the highest (94.69 U/mL), so No. 9 strain was the candidate strain.

Q. Zhang et al.

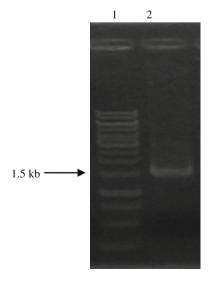
Fig. 17.1 Different strains were inoculated on skimmed milk plate. Different numbers represent different strains, Nos. 2, 5, 8, and 9 strains displayed relatively larger proteolytic *circles* and defined as primarily screening strains



17.3.2 Analysis of 16S rDNA and Physiological and Biochemical Characterization of Candidate Strain

A maximum-likelihood phylogenetic tree (Figs. 17.2 and 17.3) showed that candidate strain (No. 9) had quite high sequence identity to the *B. subtilis* strain Acj 115, and further the result of Gram's method showed that No. 9 strain was a Grampositive bacterium (Fig. 17.4). Identification of physiological and biochemical characterization of candidate strain revealed that V.P. test was positive, while methyl red test was negative, and most of the carbon source could be used. So the candidate strain was identified to be *B. subtilis*.

Fig. 17.2 The amplification of 16S rDNA. *Lane 1* 1 kb DNA ladder, *Lane 2* PCR product of 16S rDNA fragment of 1,465 bp



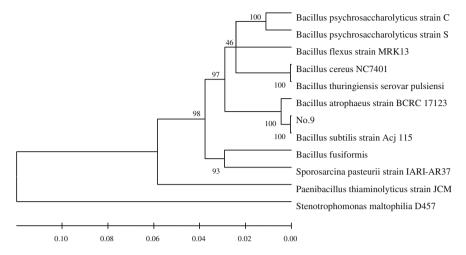
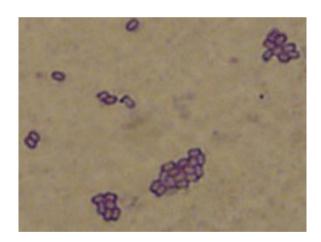


Fig. 17.3 The phylogenetic tree resulting from analysis of the 16S rDNA sequences of candidate strain (No. 9). The tree was created using maximum-likelihood distance clustered by the neighborjoining method and with MEGA program

Fig. 17.4 The result of Gram's dye of bacterium. The strain cultured in the 2216 E medium for 2 days, then Gram's staining and oil microscopy observation, the result showed that the candidate strain is a Gram-positive bacterium



17.3.3 Effect of Different Factors on the Ability of Candidate Strain Producing Protease

17.3.3.1 Carbon Sources

After carbon sources were added to the fermentation medium, compared with the control containing 2 % glucose, the ability of candidate strain producing protease increased by 1.37 fold with xylose (Fig. 17.5). In addition, xylose was added with

Q. Zhang et al.

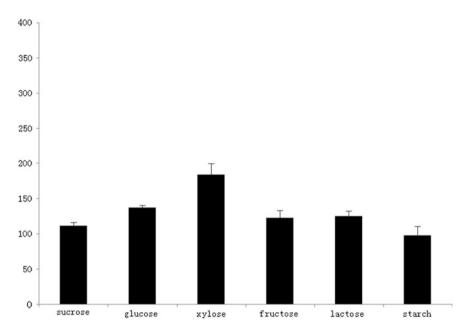


Fig. 17.5 Effects of carbon sources on protease activity. The candidate strain was grown on 2216 E medium at 28 °C and 200 rpm. The protease activity was detected by Folin-phenol method. The highest enzyme activity was 185.78 U/mL, when the carbon source was xylose

concentration of range from 1 to 5 % respectively, and the result showed that the protease activity improved obviously with 2 % of xylose in the medium (185.78 U/mL).

17.3.3.2 Nitrogen Sources

After nitrogen sources were added to the fermentation medium, the result (Figs. 17.6 and 17.7) shows that the protease activity is higher with organic nitrogen sources than inorganic nitrogen sources, and the control has the highest protease activity (123.15 U/mL). Moreover, the supplementation of yeast extract (1.6 %) and peptone (3 %), respectively, resulted in maximal protease activity.

17.3.3.3 NaCl Concentration

Different concentrations of NaCl were added to the fermentation medium, the result showed that the ability of candidate strain producing protease reduced compared with the control containing 2% NaCl (125.81 U/mL).

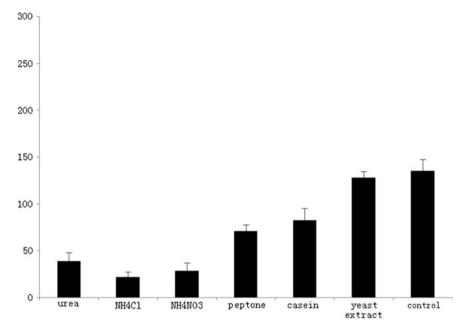


Fig. 17.6 Effects of nitrogen sources on protease activity. The strain grown on 2216 E medium at 28 °C and 200 rpm with 2 % glucose. The protease activity reached the maximum (123.15 U/mL), when peptone and yeast extract are the complex nitrogen sources

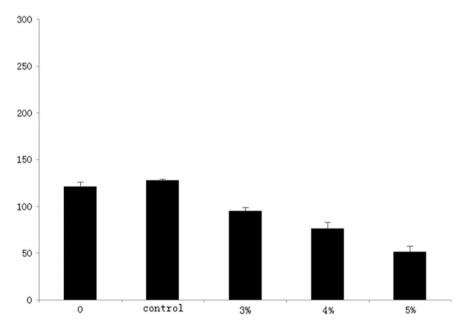


Fig. 17.7 Effects of different concentrations of NaCl on protease activity. The strain grown on 2216 E medium at 28 $^{\circ}$ C with 2 $^{\%}$ glucose, 1 $^{\%}$ peptone and 0.5 $^{\%}$ yeast extract

Q. Zhang et al.

Group number	Xylose (%)	Peptone (%)	Yeast extract (%)	NaCl (%)	Enzyme activity (U/mL)
1	1.5	2.5	1.2	1	174.13
2	1.5	3	1.6	2	190.35
3	1.5	3.5	2	3	192.87
4	2	2.5	1.6	3	174.13
5	2	3	2	1	212.07
6	2	3.5	1.2	2	184.87
7	2.5	2.5	2	2	200.87
8	2.5	3	1.2	3	164.98
9	2.5	3.5	1.6	1	215.95
K_1	557.35	523.98	523.98	602.15	
K_2	571.06	567.41	580.43	576.09	
<i>K</i> ₃	581.81	620.89	605.81	531.98	
F	1.00	23.49	11.68	8.37	

Table 17.1 The result of orthogonal test

F stands for analysis of variance values

17.3.3.4 Orthogonal Test

Orthogonal test with four factors and three levels was performed to analyze the optimal medium for producing protease. The result (Table 17.1) showed that the order of importance that influenced protease activity was found to be peptone > yeast extract > NaCl > xylose, and the optimal combination parameters were 2.5 % xylose, 3.5 % peptone, 1.6 % yeast extract, and 1 % NaCl. Under the optimal culture medium, the activity of protease increased by 128.06 % than the previous culture medium.

17.4 Discussion

By plate screening and enzyme activity determination, No. 9 strain was screened as the experimental strain, followed by construct phylogenetic tree and identification of physiological and biochemical characteristics showed that experimental strain was *B. subtilis*.

Further, through orthogonal test, the best culture medium was determined as: xylose 2.5 %, peptone 3.5 %, yeast extract 1.6 %, NaCl 1 %, and artificial seawater without NaCl. The activity of protease increased by 128.06 % compared to the origin, and peptone was the most important factor. The results of single factor experiment and orthogonal test were different may be due to the interaction between the factors. This paper has studied the conditions for improving the activity of protease, but the stability of protease needs further analysis.

Acknowledgments The authors gratefully acknowledge the support of the National Nature Science Foundation of China (No. 31201282) and the 863 project (No. 2012AA020403).

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Chapter 18

Expression of Gene *uvrA* from *Acetobacter* pasteurianus and Its Tolerance to Acetic Acid in *Escherichia coli*

Yu Zheng, Xingjing Chen, Jing Wang, Haisong Yin, Liqing Wang and Min Wang

Abstract The *uvrA* gene of *Acetobacter pasteurianus* AC2005 coding for subunit A of the excinuclease ABC complex involved in the nucleotide excision repair mechanism was identified. Gene *uvrA* was amplified using *A. pasteurianus* AC2005 genomic DNA as a template. Then the pMV24 plasmid, an expression vector of *Acetobacter*, was used for constructing the recombinant plasmid pMV24-*uvrA*. UvrA was expressed in *Escherichia coli* JM109, and its molecular weight was about 91.1 kDa. With 0.5 % acetic acid shock for 20 and 40 min, the survival rates of recombinant strain *E. coli* JM109/pMV24-*uvrA* were 0.48 and 0.056 %, which increased by 17.5 and 10.2 times, respectively, compared with those of *E. coli* JM109/pMV24. All these demonstrate that the expression of repair excinuclease UvrA could increase the acetic acid tolerance of the strain.

Keywords E. coli · uvrA · Acetic acid · Gene expression

18.1 Introduction

Acetate ion in the microorganism is toxic mainly to itself, and acetobacter can improve its resistance to a high concentration of acetic acid with a unique mechanism. Studies suggest that there are two main mechanisms that confer high acetic acid concentration to the bacteria. One is to maintain the intracellular pH relatively constant, like the proton pump, overoxidation of acetic acid, changes in fatty acid

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164 Y. Zheng et al.

composition in membrane, etc.; all these could affect the permeability of the proton [1, 2]. Another is to repair the damage of cellular components caused by high concentrations of acetic acid, like improvement of the stability of enzyme under high concentrations of acetic acid, correct folding of the protein involved by molecular chaperone, and so on [3–5].

Nucleotide excision repair (NER) is a mechanism commonly used to maintain the integrity of DNA and the proteins involved in this mechanism are mainly UvrA, UvrB, and UvrC [6]. UvrA is the initial induced protein in bacteria that test various structurally unrelated DNA lesions and excise and repair them. In prokaryotic microorganism, repair excinuclease UvrA involves in the excision and repair of DNA, and this repair mechanism is applicable to repair of many DNA damages [7]. According to the literature, in *Thermus thermophilus*, UvrA was overexpressed induced by IPTG and it makes the strain resist high temperature at neutral pH and resist low pH at room temperature. In *Lactobacillus helveticus*, UvrA was activated by exposure to UV radiation and oxidative stress, and the expression of *uvrA* was inducible by pH; UvrA contributes to acid and oxidative tolerance in *L. helveticus* [8].

Acetobacter pasteurianus is an important acid-producing bacterium during solidstate fermentation of vinegar, and is one of the bacteria commonly used in pure liquid fermentation worldwide. As its important application value, the whole-genome of *A. pasteurianus* has been sequenced completely [9, 10]. This research is focused on the repair excinuclease UvrA from *A. pasteurianus* AC2005. Using genetic engineering technology, we study the effects of UvrA on the acetic acid tolerance of *Escherichia coli*, which would lay the foundation for further clarifying the function of repair excinuclease.

18.2 Materials and Methods

18.2.1 Bacterial Strains, Plasmids, and Growth Conditions

Acetobacter pasteurianus AC2005, stored in the laboratory, was cultured in GYE media (2 % glucose, 1.5 % yeast extract, 3.5 % ethanol) at 30 °C. *E. coli* JM109 were used as hosts for the cloning experiments and *E. coli* was grown at 37 °C on Luria–Bertani (LB) broth supplemented with or without 100 μg/ml ampicillin. The pMD19-T simple vector (TaKaRa) was used for both cloning and sequencing analysis. The pMV24 plasmid (Ap^r, lacZ), gifted by Mizkan Group Corporation, Japan, was used for expression of the gene in *E. coli*.

18.2.2 Construction of the Recombinant Plasmid and Expression of the Target Protein

We used expression vector pMV24 (Ap^r, 3,854 bp) and selected two restriction sites (*Eco*RI and *Xba*I) to construct recombinant plasmid. As pMV24 had a lactose promoter, the target protein was expressed induced by IPTG.

18.2.3 Shock Experiments

To test the tolerance of the strains toward extremely high acetate stresses, shock experiments were performed with much higher concentrations of acetic acid. Strains were grown overnight at 37 °C in LB medium containing 100 mg/mL ampicillin, then diluted at a ratio of 1:100 into fresh LB medium containing 100 mg/mL ampicillin, and grown at 37 °C to logarithmic phase reaching an OD₆₀₀ value of 0.6, then added 1 mM IPTG and inoculated for 4–5 h to induce the expression of UvrA. After that acetic acid was added to final concentrations of 0.5 % (v/v). With 40 min incubation at 37 °C, samples were removed and the number of viable bacteria was determined by spread plating serial dilutions onto LB agar containing 100 mg/mL ampicillin. The plates were incubated at 37 °C for about 16 h before enumeration of the colonies. At the same time, the cultures shocked for 20 and 40 min were serially diluted, plated onto LB/agar plates with the Oxford Cup, and incubated at 37 °C for 16 h and then photographed.

18.3 Results

18.3.1 Sequence Analysis of the uvrA in A. pasteurianus AC2005

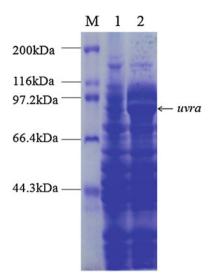
The *uvrA* gene of *A. pasteurianus* AC2005 was cloned and its nucleotide sequence was determined. Analysis of the sequence revealed a gene of 2,514 nt that encoded a protein with 837 amino acids (aa) and a predicted molecular mass of 91.1 kDa. In this study, UvrA was expressed in *E. coli* JM109 induced by 1 mM IPTG. By SDS-PAGE, as shown in Fig. 18.1, a significant band of the protein induced appeared at about 91.1 kDa, and its molecular weight was in line with expectation.

The result of comparison indicates that the protein has high homology with the genus *Acetobacter* and *Gluconobacter*. The protein sequence contains a conserved region of NER enzymes.

The sequence obtained was compared with the *uvrA* gene sequence (GenBank: 8435212) of *A. pasteurianus* IFO3283-01 in GenBank. The result showed that the coding sequence from *A. pasteurianus* AC2005 has a similarity of 93.2 % with the

166 Y. Zheng et al.

Fig. 18.1 Expression of pMV24-*uvrA* induced by IPTG *M* Marker; *I E. coli* JM109/pMV24; *2 E. coli* JM109/pMV24-*uvrA*



corresponding region from *A. pasteurianus* IFO3283-01. Compared with the protein UvrA from *A. pasteurianus* AC2005 speculated by the coding region with UvrA from Acetobacter IFO3283-01, the amino acid sequence has a similarity of 98.6 %. Therefore, we can determine the amplified bands for *uvrA* gene of *A. pasteurianus*.

18.3.2 Effect of uvrA Expression on the Growth of E. coli Under Acidic Conditions

To determine the effect on cell growth caused by the expression of UvrA, the experiment studies the growth curves of the strains with the same initial amount of bacteria cells in LB medium. As acetic acid treatment could significantly reduce the final biomass of different strains, *E. coli* JM109/pMV24 and *E. coli* JM109/pMV24-*uvrA* were grown in LB broth supplemented with or without 0.05 % (vol/vol) acetic acid and their growth curves were determined. As shown in Fig. 18.2, in the presence of acetic acid, the growth of control strain *E. coli* JM109/pMV24 in early logarithmic phase and steady growth was suppressed obviously. However, the growth of recombinant strain *E. coli* JM109/pMV24-*uvrA* in the presence or absence of acetic acid is similar. It showed that the expression of UvrA increased the acetic acid tolerance of recombinant strain, and it initially proved that UvrA was related to the acetic acid tolerance.

As we all know acetic acid is highly toxic to *E. coli* cells, and the concentration of acetic acid for use in *E. coli* is often under 0.1 %. Then acetic acid shock experiments were subsequently performed to test the tolerance of *uvrA* expression strain toward acetic acid at higher concentration. The final concentrations of acid required to adjust cultures of *E. coli* to 0.5 % acetic acid were pH 3. After shocking

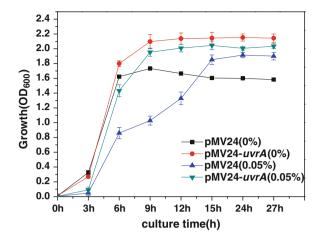


Fig. 18.2 Growth curves of $E.\ coli\ JM109/pMV24$ and $E.\ coli\ JM109/pMV24$ -uvrA induced by IPTG with or without $0.05\ \%$ acetic acid, respectively

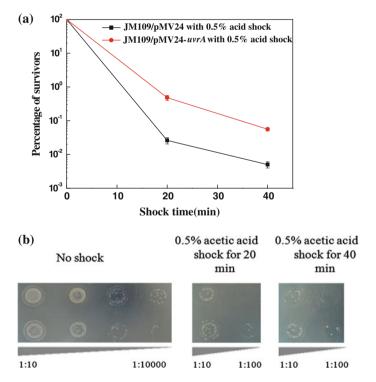


Fig. 18.3 Survival of *E. coli* JM109/pMV24 and *E. coli* JM109/pMV24-*uvrA* under acid-shock conditions, **a** bacterial survival rate at different times treated with 0.5 % acetic acid, **b** colony morphology shocked with 0.5 % acetic acid for 40 min

168 Y. Zheng et al.

with 0.5 % acetic acid for different times, the survival curves for the control strain *E. coli* JM109/pMV24 and the recombinant strain *E. coli* JM109/pMV24-*uvrA* are as shown in Fig. 18.3. The changeable survival rate after the acetic acid treatment reflected that the expression of UvrA affects acetic acid tolerance of *E. coli*. With increasing shock time, the recombinant strain showed a clearly increased tolerance to acetic acid as compared to the control strain exposed to 0.5 % acetic acid. With 0.5 % acetic acid shock for 20 and 40 min, the survival rates of recombinant strain *E. coli* JM109/pMV24-*uvrA* were 0.48 and 0.056 %, which increased by 17.5 and 10.2 times, respectively, compared with those of control strain *E. coli* JM109/pMV24. And with 0.5 % acetic acid shock for 40 min, the plaques of two strains on solid plate were observed. Although the number of colonies of control strain under normal culture condition was slightly higher than the recombinant strain, the number was 10 times higher than control strain after acetic acid shock. All these demonstrate that the expression of repair excinuclease UvrA could increase the acetic acid tolerance of the strain.

18.4 Discussion

We have constructed a recombinant strain of *E. coli* with the help of the pMV24 plasmid which is an expression vector of *Acetobacter*. UvrA was expressed in *E. coli* JM109, and its molecular weight was about 91.1 kDa. Homology analysis of the *A. pasteurianus uvrA* gene product revealed high level homology to the *Acetobacter* and *Gluconobacter* UvrA proteins. The protein sequence contained a conserved region of NER enzymes was found. Comparison of the *A. pasteurianus* AC2005 and *A. pasteurianus* IFO3283-01 UvrA protein demonstrated 98.3 % identity and 98.6 % similarity.

With 0.5 % acetic acid shock for 20 and 40 min, the survival rates of recombinant strain *E. coli* JM109/pMV24-*uvrA* increased by 17.5 and 10.2 times, respectively, compared with those of *E. coli* JM109/pMV24. These demonstrate that the expression of repair excinuclease UvrA could increase the acetic acid tolerance of the strain.

Acknowledgments This work was supported by the National High Technology Research and Development Program of China (2012AA022108, 2013AA102106), the National Natural Science Foundation of China (31201406), the Natural Science Foundation of Tianjin, China, (13JCQNJC10000).

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Chapter 19 Construction of Eschericha coli-Staphylococcus Shuttle Vector for EGFP Expression and Potential **Secretion via Tat Pathway**

Bao-vin Xu, Yi-bing Cheng, Lin Wang, Hao Zhou, Lin Huang, Xiao-van Tang and Oiang Gao

Abstract In this study, the potential for heterologous protein expression and secretion via twin-arginine translocation (Tat) pathway was investigated in Escherichia coli DH5α host using enhanced green fluorescent protein (EGFP) as model protein reporter. To construct the shuttle vector pBT2-ET-5X-EGFP, 17 kinds of PCR-amplified 5x-egfp fragments were, respectively, cloned into plasmid pBT2-Peftu-Tat-EGFP and transformed into E. coli DH5α host. By SDS-PAGE, fluorescence microscope observation and flow cytometry analyzation, EGFP was expressed in an active form in the cells of E. coli DH5α, but failed to translocate to the culture medium.

Keywords E. coli DH5α · Enhanced green fluorescent protein (EGFP) · Twin-arginine translocation pathway • Plasmid pBT2

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19.1 Introduction

Compared to the general protein secretion (Sec) system [1], the twin-arginine translocation (Tat) system can deliver the properly folded proteins to the extracellular environment [2]. Thus, the Tat system can be used in gene engineering to secrete exogenous protein which cannot utilize the Sec system [3].

Green fluorescent protein (GFP) can be expressed in *Escherichia coli* as an ideal reporter [4], but whether it is correctly folded and transported across the plasma membrane depends on the choice of the Tat signal peptide sequence [2]. The wild-type GFP is sensitive to temperature, sometimes less expressed and weakly fluorescent in some cells [5], thus researchers have optimized GFP through site-directed mutagenesis. The enhanced green fluorescent protein (EGFP) is now a widely used GFP mutant with two independent substitutions (F64L, S65T), which has improved the fluorescence intensity as well as stability to be detectable for 16–24 h after excitation [6]. In this study, N-terminal modified EGFP was used as model reporter to study the Tat-dependent secretion pathway in *E. coli* DH5α host.

Shuttle vector has two different replication origins of two microbial plasmids, selection marker gene and a multicloning site, which can be replicated in two different organisms, and is usually used for cloning the amplified cloned gene. In this study, pBT2-ET-5X-EGFP, a shuttle vector of *E. coli* and *Staphylococcus* derived from a shuttle vector pBT2 [7], was constructed to study the expression and secretion of EGFP via Tat-pathway.

19.2 Materials and Methods

19.2.1 Bacterial Strain, Plasmids, and Growth Medium

The bacterial strain and plasmids used in this study are listed in Table 19.1, and all the recombinant DNA manipulations were carried out in *E. coli* DH5 α host. *E. coli* DH5 α was cultured at 37 °C in LB broth (0.5 % yeast extract, 1 % tryptone, and 1 % NaCl) supplemented with 100 µg/mL ampicillin for plasmid selection.

19.2.2 Preparation of the 5x-egfp Fragment

Based on our previous experiments, a linker consisting of five identical amino acids was designed and inserted between the Tat signal peptide and the N-terminal of *egfp* gene to generate a *5x-egfp* fusion gene fragment, where *x* here stands for a certain amino acid codon. Such *5x-egfp* fragment is found to be helpful for secretion from the bacterial host (data not shown). Since 5N-, 5Q- and 5R-EGFP were previously expressed in *E. coli* DH5a host, in order to construct the other 17

Bacterial strain and plasmids	Relevant properties	References
E. coli DH5a	_	Laboratory storage
pMD19-T simple vector	Amp ^r	TaKaRa Biotechnology (Dalian) Co., Ltd.
pBT2	Amp ^r Cm ^r	A gift from University of Tuebingen, Germany
pBT2-Peftu-Tat-EGFP [8] (pBT2-ET-EGFP)	Amp ^r Cm ^r	Laboratory storage

Table 19.1 Strain and plasmids used in this work

kinds of 5x-egfp fragment, the primers (Table 19.2) were correspondingly designed according to the sequence of egfp gene (No. AF302837).

The primers above were used to, respectively, amplificate the 5x-egfp fragment using plasmid pBT2-ET-EGFP as PCR template. The following reaction mixture was set up as: 17.5 μ L ddH₂O, 2.5 μ L parent plasmid, 2.5 μ L each primer, 25 μ L 2× Taq PCR Master Mix. The cycling parameters for the PCR were as follows: 30 cycles of 95 °C for 5 min, 95 °C for 50 s, 70 °C for 30 s, 72 °C for 1 min followed by a single cycle of 72 °C for 10 min. The PCR products were purified by GeneJET Gel Extraction Kit (Thermo, Germany) after 1 % agarose gel electrophoresis.

19.2.3 Transformation of E. coli DH5a

The *E. coli* DH5α competent cells were thawed in ice for 5 min, then a 10-μL ligation product was added to the cells, and the mixture was incubated in ice for 30 min, heat-shocked for 90 s in a 42 °C water bath, immediately transferred to an ice bath for 1–2 min, followed by the addition of 1 mL sterile LB broth and constant shaking at 37 °C and 200 r/min for 1.5 h; a 100-μL culture broth was spread on LB agar plates containing 100 μg/mL chloramphenicol, and incubated at 37 °C for 12 h [9].

19.2.4 Preparation of Plasmid pBT2-ET-5X-EGFP

The resulting PCR products were, respectively, cloned into pMD19-T Simple Vector after overnight ligation at 16 °C and transformed into *E. coli* DH5α host. Plasmid pMD19-T-5X-EGFP prepared from *E. coli* was purified using the TIANprep Mini Plasmid Kit (TIANGEN, Beijing, China). The constructed *5x-egfp* fragment was first digested by double digestion of *Hin*dIII and *Nhe*I (Fermentas, Lietuvos) from plasmid pMD19-T-5X-EGFP, then inserted into the synonymous sites in pBT2-Peftu-Tat-EGFP, a shuttle vector of *E. coli* and *Staphylococcus*, by T4 ligase (Fermentas, Lietuvos). After overnight ligation at 16 °C, the linked products were transformed into *E. coli* DH5α to create the recombinant plasmid

Table 19.2 Primers used in this work

Primer	Sequence	Underline sequence
5A-egfp-F	CAAGCTTGCAGCAGCTGCAGTGAGCAAGGGCGAG	HindIII
5D-egfp-F	CAAGCTTGATGATGATGATGTGAGCCAAGGGCGAGGAGCTGTTCACC	HindIII
5I-egfp-F	CAAGCITATTATATTATTGTGAGCAAGGGCGAGGAGCTGTTCACCGGGG	HindIII
	CAAGCTTGGCGGAGGAGGAGGAGTGAGCAAGGGCGAG	HindIII
5N-egfp-F	CAAGCTTAATAACAATAATAACGTGAGGCGAGGGCGAGGAGCTGTTCACCGGGG	HindIII
5L-egfp-F	CAAGCITCTTCTTACTTCTTGTGAGCAAGGGCGAGGAGCTGTTCACC	HindIII
5F-egfp-F	C <u>AAGCTT</u> TTTTTTTTTTTTTTGTGAGCAAGGGCGAGGAGCTGTTCACCGGGG	HindIII
5P-egfp-F	CAAGCTTCCTCCTCCTCTGTGAGCAAGGGCGAG	HindIII
5S-egfp-F	CAAGCTTTCATCAAGTTCATCAGTGAGCAAGGGCGAGGAGGAGCTGTTCACC	HindIII
	C <u>AAGCTT</u> ACAACAACAACAACAGTGAGCAAGGGCGAGGAGCTGTTCACC	HindIII
5W-egfp-F	CAAGCTTTGGTGGTGGTGGGTGAGCAAGGGCGAG	HindIII
5Y-egfp-F	C <u>AAGCTT</u> TATTATTATTATGTGAGCAAGGCCGAGGAGCTGTTCACCGGGG	HindIII
5V-egfp-F	C <u>AAGCTT</u> GTAGTAGTAGTAGTGAGCAAGGGCGAGGAGCTGTTCACC	HindIII
5H-egfp-F	C <u>AAGCTT</u> CATCATCATCATGTGAGCAAGGGCGAGGAGCTGTTCAC	HindIII
5M-egfp-F	C <u>AAGCTT</u> ATGATGATGATGATGAGCCAAGGCCGAGGAGCTGTTCAC	HindIII
5C-egfp-F	C <u>AAGCTT</u> TGTTGTTGTTGTGAGCAAGGGCGAGGAGCTGTTCAC	HindIII
5K-egfp-F	C <u>AAGCTT</u> AAGAAAAAAAAAGAGGGGGGCGAGGGGGGGTGTTCACCGGGGT	HindIII
5X-egfp-R	GGCTAGCGAATTCATTACTTGTACAGCTCGTCCATGCCGAGAGTGATCC	NheI

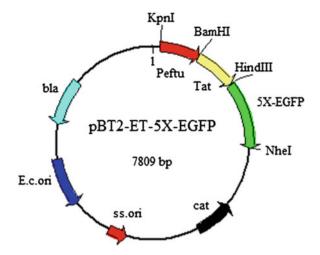


Fig. 19.1 Diagram of plasmid pBT2-ET-5X-EGFP. 5X-EGFP, EGFP with a 5 identical amino acids linker at its N-terminal (X represents for an amino acid of A, C, D, F, G, H, I, K, L, M, N, P, S, T, V, W and Y, respectively); *Tat* a twin-arginine translocation signal peptide of *efeB* gene in *S. carnosus* TM300; *Peftu* a strong promoter of *tufA* gene in *S. carnosus* TM300; *bla* ampicillin resistance gene; *cat* chloramphenicol resistance gene; *E.c.ori* origin of replication in *E. coli*; *ss.ori* origin of replication in *Staphylococcus*

pBT2-ET-5X-EGFP (Fig. 19.1). All plasmid constructions were verified by double restriction enzyme digestion and DNA sequence analysis.

19.2.5 Extraction of EGFP

Single colonies of *E. coli* DH5 α /pBT2-ET-5X-EGFP and *E. coli* DH5 α were, respectively, grown in 50 mL LB broth at 37 °C and 180 r/min for 16 h. The cell pellets were harvested by centrifugation at 12,000 r/min and 4 °C for 15 min. Proteins present in the culture medium were treated with 10 % trichloroacetic acid (TCA) for 12 h at 4 °C. The cells were resuspended in 2 mL PBS buffer, disrupted by sonication, and centrifuged to separate the supernatant from cell fragment. The supernatant was carefully transferred to a new vial and then precipitated by adding 10 % TCA for 12 h at 4 °C. The precipitated proteins were washed successively in 100 and 80 % acetone and dried at room temperature.

19.2.6 Detection of EGFP

After fermentation culture, the expression of the modified 5X-EGFP in *E. coli* DH5α host was observed by fluorescence microscopy (Olympus (China) Co., Ltd., China), the number of fluorescent cells per 10,000 was counted by flow cytometry

176 B. Xu et al.

(Accuri C6, BD, America), and the expression of EGFP in cells and culture medium of *E. coli* DH5α/pBT2-ET-5X-EGFP was also analyzed by SDS-PAGE.

19.3 Result

19.3.1 Construction of 5x-egfp Fragment

The 17 kinds of *5x-egfp* fragment obtained from PCR were examined by agarose gel electrophoresis (Fig. 19.2a, b) and purified with GeneJET Gel Extraction Kit (Fig. 19.2c). Double restriction digestion and sequence alignment of plasmid pMD19-T-5X-EGFP (Fig. 19.2d) showed the same result with the target gene.

19.3.2 Verification of pBT2-ET-5X-EGFP

Recombinant plasmid pBT2-ET-5X-EGFP extracted from recombinant strain was examined by 1 % agarose gel electrophoresis (Fig. 19.3). Their molecular weights were expected as 8,000 bp for pBT2-ET-5X-EGFP and 800 bp for 5X-EGFP.

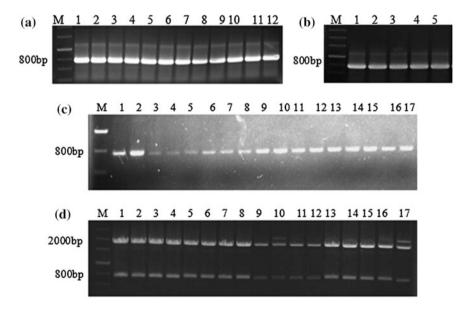


Fig. 19.2 Agarose gel electrophoresis of the PCR product of 5x-egfp fragment. M DNA Marker; a1-a12 PCR product of 5x-egfp (x = codon for A, C, D, F, G, H, I, K, L, M, N or P); b1-b5 PCR product of 5x-egfp (x = codon for S, T, V, W and Y); c1-c17 purified product of 5x-egfp (x = codon for A, C, D, F, G, H, I, K, L, M, N, P, S, T, V, W or Y); d1-d17 pMD19-T-5X-EGFP digested by HindIII and NheI (X = A, C, D, F, G, H, I, K, L, M, N, P, R, S, T, V, W or Y)

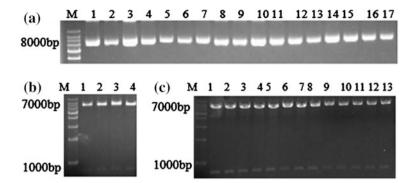


Fig. 19.3 Agarose gel electrophoresis of pBT2-Peftu-Tat-5X-EGFP. M DNA Marker. a1-a17 pBT2-ET-5X-EGFP (X = A, C, D, F, G, H, I, K, L, M, N, P, S, T, V, W or Y); b1-b4 pBT2-ET-5X-EGFP digested by HindIII and NheI (X = A, C, D or F); c1-c13 pBT2-ET-5X-EGFP restricted with HindIII and NheI (X = F, G, H, I, K, L, M, N, P, R, S, T, V, W or Y)

19.3.3 Florescence Detection of EGFP in E. coli DH5a

A 10 μ L culture broth of *E. coli* DH5 α /pBT2-ET-5X-EGFP was used for florescence microscope observation. The *E. coli* DH5 α host strain bearing pBT2-ET-5X-EGFP exhibited strong fluorescence, while the control strain showed no fluorescence (Fig. 19.4).

19.3.4 Detection of EGFP in E. coli DH5a by SDS-PAGE

Dried precipitated proteins in the culture medium and the supernatant were resuspended in 50 μ L PBS. After adding 12.5 μ L loading buffer, the 30 μ L samples were

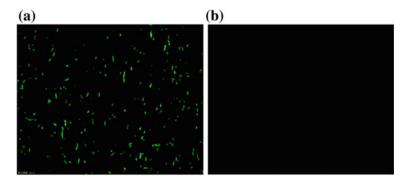


Fig. 19.4 Observation of *E. coli* DH5 α /pBT2-ET-5X-EGFP and *E. coli* DH5 α by fluorescence microscope. a *E. coli* DH5 α /pBT2-ET-5X-EGFP; b *E. coli* DH5 α

178 B. Xu et al.

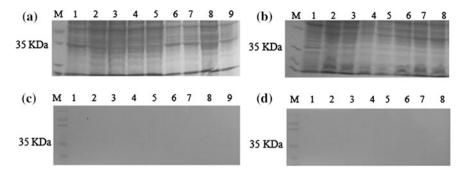


Fig. 19.5 Expression of 5X-EGFP in *E. coli* DH5α validated by SDS-PAGE. *M* Protein Marker (Fermentas, Lietuvos); a1–a9 5X-EGFP extracted from the cytoplasm of *E. coli* DH5α/pBT2-ET-5X-EGFP (X = A, C, D, F, G, H, I, K or L); b1–b8 5X-EGFP extracted from the cytoplasm of *E. coli* DH5α/pBT2-ET-5X-EGFP (X = M, N, P, S, T, V, W or Y); c1–c9 5X-EGFP extracted from the culture medium of *E. coli* DH5α/pBT2-ET-5X-EGFP (X = A, C, D, F, G, H, I, K or L); d1–d8 5X-EGFP extracted from the culture medium of *E. coli* DH5α/pBT2-ET-5X-EGFP (X = M, N, P, R, S, T, V, W or Y)

applied for SDS-PAGE analysis. The molecular weights of the 17 homologous target proteins were expected at about 35 kDa for 5X-EGFP. The results proved that 5X-EGFP were properly expressed in a soluble form in *E. coli*, but could not be translocated outside the cell wall (Fig. 19.5).

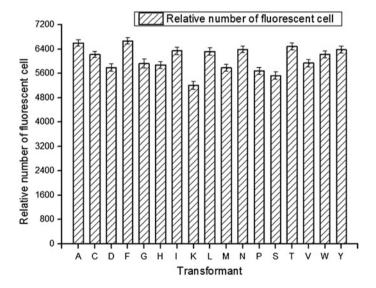


Fig. 19.6 Fluorescence of *E. coli* DH5 α /pBT2-ET-5X-EGFP measured by flow cytometry. *X axis* transformants *A* to *Y* strand for *E. coli* DH5 α /pBT2-ET-5X-EGFP; *Y axis* number of fluorescent cell per 10,000 cells

19.3.5 Expression of EGFP in E. coli DH5a Detected by Flow Cytometry

A 1-mL suspension of *E. coli* DH5 α /pBT2-ET-5X-EGFP was analyzed by flow cytometry. An average of fluorescent cell number indicated that more than half of the cells expressed active 5X-EGFP in *E. coli* DH5 α cells (Fig. 19.6).

19.4 Discussion

Tat pathway first found in *E. coli* was now a focus of protein transportation research around the world due to its feature to help translocate fully folded proteins across the bacterial plasma membrane. Bolhuis et al. [10] found that two integral cytoplasmic membrane proteins TatB and TatC make up a structural and functional unit of the twin-arginine translocases in *E. coli*. Tat-dependent heterologous protein secretion was analyzed in the three different Gram-positive bacteria *Staphylococcus carnosus*, *Bacillus subtilis*, and *Corynebacterium glutamicum* using GFP as reporter [11]. Differences about the final localization and the folding status of the exported GFP proved that the choice of potential bacteria host and suitable microorganism is essential in GFP secretion via Tat pathway.

In this study, the constructed 5X-EGFP was used as a model protein reporter to study the possibility of Tat-dependent heterologous protein secretion in *E. coli* DH5α. By means of SDS-PAGE, fluorescence microscope observation and flow analyzation, all the 17 kinds of 5X-EGFP were successfully expressed in a soluble and active form in *E. coli* DH5α cytoplasm, but failed to be translocated out of the cell wall. One possible explanation is the lack of a series of cofactor-binding-proficient proteins which plays an important role in protein transportation [12], the existence of Gram-negative *E. coli* outer membrane is another reason for this failure.

This research constructed an *E. coli-Staphylococcus* shuttle vector to study the function of Tat-pathway upon exogenous protein secretion. In the future research, the recombinant plasmid pBT2-ET-5X-EGFP is to be transformed into *Staphylococcus carnosus* TM300, a valuable genetic engineering strain in food industry [13] with low extracellular proteolytic activity and no by-products like toxin, hemolysin, coagulase [14], and thereby lays the preliminary experimental basis for further insight into heterologous protein secretion via twin-arginine translocation pathway.

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B. Xu et al.

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Chapter 20 SNP Affects the Mobility of Breast Cancer Cells and the Expression of Metastasis-Related Genes

Juan Hu, Hongpeng He, Hao Zhou, Dandan Wang, Yijie Wang, Xuena Liu, Yongwei Lai and Tongcun Zhang

Abstract Protein S-nitrosylation is a type of posttranslational modification that changes protein stability, intracellular localization, and biological activity. In mammalian cells, NO is synthesized by NOS which is highly expressed in many solid tumors, including breast cancer. NOS is involved in the progress of carcinogenesis by generating NO to modify various tumor-related proteins to enhance or suppress their function. However, the effect of NO on tumor metastasis is less studied. Herein, breast cancer cells MCF-7 and MDA-MB-231 were treated with NO donor SNP, which resulted in altered cell mobility. Meanwhile the expression of several metastasis-related genes was regulated. These results suggest a novel mechanism by which NO promotes the metastasis of breast cancer.

Keywords Breast cancer • Cell migration • SNP • NO

20.1 Introduction

Nitric oxide (NO) is originally identified as a biologic mediator molecule functioning in several processes, including neurotransmission, antimicrobial, and antitumoral activities. Recently, it has been found that NO is able to chemically modify proteins at cysteine or tyrosine residues, which is termed S-nitrosylation or tyrosine nitrosation respectively. Quite a few proteins in mammalian cells were found to be

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182 J. Hu et al.

S-nitrosylated, including HIF, c-Fos, c-Myc, Src, Bcl-2, etc. [1–3]. S-nitrosylation alters the stability or the activity of the modified protein and thereby regulates cell apoptosis, proliferation, angiogenesis, and other biological processes [2].

In mammalian cells, NO is synthesized by nitric oxide synthase (NOS) from L-arginine. There are three types of NOS, namely iNOS (inducible NOS), eNOS (endothelial NOS), and nNOS (neuronal NOS). iNOS was observed to be highly expressed in some tumors such as colon, gastric, prostate, cervical, ovarian, and lung cancers [3–6]. In breast cancer patients, increased expression of iNOS was reported to be associated with a basal-like transcription pattern which is frequently observed in poor survival, ER-negative patients [7]. For ER-positive breast cancer cell, such as MCF7, the disruption of S-nitrosothiol homeostasis altered MCF-7 cell proliferation in a dose-dependent manner [8]. However, the effect of NO on the mobility of breast cancer cell is less studied.

Several NO donors which are able to release NO into cell culture medium have been identified, such as GSNO, DETA/NO, SNAP, and sodium nitroprusside (SNP). To further explore the effects of NO stress on the proliferation and mobility of breast cancer cells, herein, MCF-7 and MDA-MB-231 cells were treated with SNP at different concentrations, the mobility and the proliferation of breast cancer cells were investigated, and the expression of metastasis-related genes was measured.

20.2 Materials and Methods

20.2.1 Cell Culture and Reagents

MCF-7 and MDA-MB-231 human breast cancer cells were cultured in Dulbecco's modified Eagle's medium with L-Glutamine (DMEM; Gibico) supplemented with 10 % fetal bovine serum (Sijiqing, Hangzhou China) and 1 % Antibiotic/Antimycotic (Solarbio, China) at 37 °C in a 5 % CO₂ incubator. SNP was purchased from Beyotime, China.

20.2.2 Wound Healing Assay

Breast cancer cells were seeded in 6-well plates and cultured to confluence, then cells were serum-starved for 24 h. Wound was introduced by scratching the monolayer of cells with a micropipette tip. Cells in 6-well plates were washed with PBS to remove floating cells and cultured in serum-free media containing SNP at different concentrations. The healing of the wound was investigated every 12 h and pictures were taken with a microscope.

Table 20.1 Primer sequences used in this study

Primer	Sequence $(5' \rightarrow 3')$
18S-F	CACGGG AAACCTCACCCGGC
18S-R	CGGGTGGCTGAACGCCACTT
CYR61-F	GTCGTCACCCTTCTCCACTT
CYR61-R	CTTGGCGCAGACCTTACAG
MYL9-F	GAGCCCAAGCGCCTTCT
MYL9-R	GTCAATGAAGCCATCACGGT
MYH9-F	AGCGTTACTACTCAGGGCTCATC
MYH9-R	TCATACTCCTGTAGGCGGTGTCT

20.2.3 *MTT* Assay

Breast cancer cells were seeded in 96-well culture plates and cultured overnight before being treated with SNP at different concentrations. Twenty-four hours after SNP treatment, 20 μ L of MTT (5 mg/mL, Solarbio, China) was added to each well and incubated in an incubator for 4 h, then the medium was removed and formazan crystals were dissolved with 100 μ L of DMSO. Cell viability was analyzed by measuring the absorbance at 590 nm using an ELISA reader (BioTek, USA).

20.2.4 Isolation of Total RNA and RT-PCR

Total RNA was extracted with Trizol reagent (Invitrogen). Reverse transcription was performed with M-MLV reverse transcriptase (Promega). Complementary DNA was quantified by semi-quantitative PCR using EasyTaq DNA Polymerase kit (TransGen Biotech). PCR primers were designed with NCBI online software Primer-BLAST and synthesized by Dingguo Biotech. Sequences of the primers were listed in Table 20.1. Thirty cycles of PCR were carried out and PCR products were visualized by electrophoresis using 2 % agarose gels.

20.3 Results

20.3.1 SNP Promoted the Migration of Breast Cancer Cells

To examine the effect of SNP on the migration of breast cancer cells, wound healing assay was performed for both low mobility breast cancer cell line MCF7 and high mobility breast cancer cell line MDA-MB-231. Pictures were taken with microscopy at different time points. The results of wound-healing assay showed that the breast cancer cells scrambled from both sides of the scratch toward the middle (Fig. 20.1a, b). MCF7 cells treated with SNP at low concentrations, in particular 0.1

J. Hu et al.

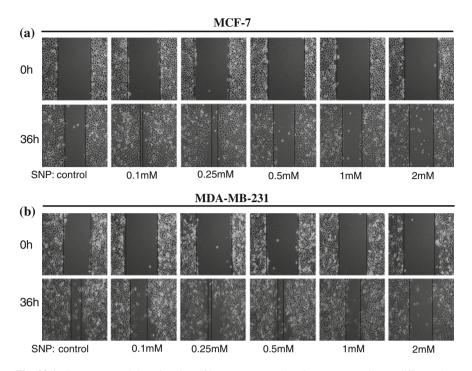


Fig. 20.1 SNP promoted the migration of breast cancer cells. Pictures were taken at different time points after scratch and treatment with SNP. For control group, cells were treated with H_2O , the dissolvent of SNP

and 0.25 mM, moved faster than untreated cells or cells treated with SNP at high concentrations (Fig. 20.1a, compares columns 2 and 3 with other columns in the lower panel). These results suggest that SNP at lower concentration promotes the migration of MCF7 cells.

MDA-MB-231 is another breast cancer cell line with higher mobility. From results of wound healing assay, MDA-MB-231 cell indeed moved faster than MCF-7 (Fig. 20.1a, b). MDA-MB-231 cells treated with SNP at 0.25 and 0.5 mM concentrations moved faster than control group, whereas cells treated with 0.1 and 2 mM of SNP moved slower than control group (Fig. 20.1b). These results suggest that the migration of MDA-MB-231 cells were promoted by SNP at modest concentrations.

20.3.2 SNP Inhibited the Proliferation of Breast Cancer Cells

NO was previously reported to display different biological effects dependent on its concentration [3]. To exclude the possibility that the faster wound healing after SNP

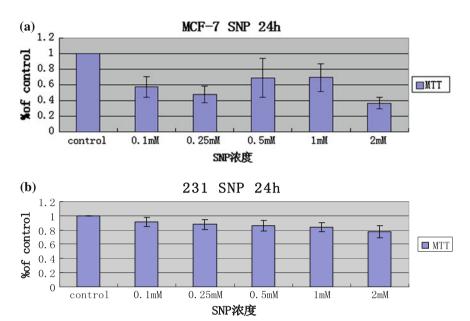


Fig. 20.2 SNP inhibited the proliferation of breast cancer cells

treatment observed for MCF-7 and MDA-MB-231 cells is due to the increased cell proliferation, cell growth was measured with MTT method.

As shown in Fig. 20.2a, SNP inhibited the growth of MCF7 in a dose-dependent manner. For MDA-MB-231, cell growth was inhibited by SNP as well, but the effect was more modest compared to that of MCF-7 (Fig. 20.2b). These results suggest that the faster wound healing observed for MCF-7 and MDA-MB-231 cells was resulted from higher cell mobility but not from cell proliferation. The inhibition of cell growth by SNP at high concentrations is in agreement with a previous report that NO induces apoptosis of MCF-7 cells at high concentration [9].

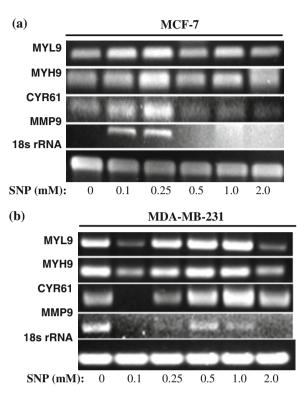
20.3.3 SNP Affected the Expression of Metastasis-Related Genes in a Dose-Dependent Manner

Metastasis is a common phenomenon of malignant tumors and is correlated with the increased mobility of tumor cells. To explore the mechanism by which SNP affects the mobility of breast cancer cells, the expression of a series of genes, such as MYL9, MYH9, CYR61, and MMP9 which are known to be metastasis-related [10, 11], were determined by RT-PCR. As shown in Fig. 20.3a, the expression of the metastasis-related genes in MCF-7 was stimulated by NO at 0.1 or 0.25 mM concentrations (Fig. 20.3a, lane 2 and lane 3), which is in line with the increased

J. Hu et al.

Fig. 20.3 SNP altered the expression of metastasis-related genes in breast cancer cell lines.

a MCF-7 cells were treated with SNP, total RNA was extracted 24 h later. Products of RT-PCR were analyzed with 2 % agarose gel. b
MDA-MB-231 cells were treated for 4 h before being harvested and analyzed



mobility of SNP-treated MCF-7 (Fig. 20.1a). In MDA-MB-231 cells, the expression of these metastasis-related genes was stimulated by SNP at 0.25–1.0 mM concentrations, but suppressed by SNP at 0.1 or 2.0 mM concentrations (Fig. 20.3b), which is also in agreement with the effects of SNP on MDA-MB-231 mobility (Fig. 20.1b). Taken together, these results indicate that SNP, at proper concentrations, stimulates the expression of metastasis-related genes and increased the mobility of breast cancer cells simultaneously.

20.4 Discussion

In malignant solid tumor, NO is released by macrophage and tumor cells to generate a microenvironment inducing angiogenesis which is correlated with the S-nitrosylation of a transcription factor HIF [1]. In addition to the effect on angiogenesis, low level NO was also found to stimulate cell proliferation and high level NO was shown to induce apoptosis. In this study, breast cancer cells were treated with serial diluted NO donor SNP and the results reveal that low mobility breast cancer cell MCF-7 and high mobility breast cancer cell line MDA-MB-293 respond differently to SNP. For MCF-7 cell, SNP obviously inhibited cell proliferation,

meanwhile, at low concentration, SNP promoted MCF-7 cell migration which is coincident with the upregulation of metastasis-related genes. At high concentration, SNP has little effect on both MCF-7 cell migration and metastasis-gene expression. However, for MDA-MB-231 cell, the growth was not significantly affected by SNP. At low or high concentration, SNP inhibited MDA-MB-231 cell migration and the expression of metastasis-related genes. At modest level, SNP promoted cell migration and the expression of metastasis-related genes. These results indicate that SNP stimulates breast cancer cell migration by inducing the expression of metastasis-related genes.

A common property of the genes examined in this study is that they carry CarG-box elements in their promoter regions and were previously shown to be regulated by myocardin-related transcription factors (MRTFs) [11]. MRTFs control cell mobility, cell invasion, and also other cell skeleton-related processes via activating the expression of target genes. The activity of MRTFs is regulated by posttranslational modifications, such as phosphorylation. MRTFs contain cytidine residues in their amino acid sequences. Therefore, it is possible that MRTFs in breast cancer cells were modified by NO released from SNP and this modification stimulates MRTFs transactivity to activate the expression of metastasis-related genes, and in the end, promote breast cancer cell migration. This hypothesis is worth of further study in the future.

Acknowledgments We thank Professor Zhonghua Pang from the Institute of Hematology & Blood Diseases Hospital, Chinese Academy of Medical Sciences & Peking Union Medical College for providing MDA-MB-231 cell line. This work was supported by the National Natural Science Foundation of China (31301073) and Applied Basic Science and Frontier Technology Program of Tianjin (13JCYBJC38000).

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188 J. Hu et al.

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Chapter 21 Molecular Cloning and Characterization of Glycerol Dehydrogenase from *Klebsiella* pneumoniae

Yanhua Liu, Li Zhao, Jianguo Zhang and Yu Zheng

Abstract Glycerol dehydrogenase (GDH) was a key enzyme for 1,3-propanediol fermentation from glycerol. In this report, gene *gdh*, encoding GDH of *Klebsiella pneumoniae* KG1 was gained by PCR with genomic DNA as template. The open reading frame (ORF) of *gdh* consisted of 1,143 bp and encoded 380 amino acids with a deduced molecular mass of 42 kDa. The gene *gdh* was overexpressed in *E. coli* BL21 (DE3) and produced 886-fold higher activity of GDH than that of KG1. The optimum temperature of recombinant GDH was 37 °C and glycerol was the optimum substrate.

Keywords 1,3-Propanediol • 1,3-Propanediol oxidoreductase • Glycerol dehydratase • Glycerol dehydrogenase • *Klebsiella pneumonia*

21.1 Introduction

It is known that certain enteric bacteria are capable of fermenting glycerol to 1,3-propanediol (1,3-PD) as the main product. Fermentation was carried out in two parallel pathways: oxidative and reductive. Through the oxidative pathway, glycerol

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© Springer-Verlag Berlin Heidelberg 2015 T.-C. Zhang and M. Nakajima (eds.), *Advances in Applied Biotechnology*, Lecture Notes in Electrical Engineering 333, DOI 10.1007/978-3-662-46318-5_21 was dehydrogenated by a NAD⁺-linked enzyme, glycerol dehydrogenase (GDH), to dihydroxyacetone (DHA), and then to dihydroxyacetone phosphate (DHAP). The reductive pathway, a coenzyme B₁₂-dependent glycerol dehydratase (GDHt) removed a water molecule from glycerol to form 3-hydroxypropionaldehyde (3-HPA) which was then reduced to 1,3-PD by an NADH-linked 1,3-PD oxidore-ductase (1,3-PDOR) [1, 2]. The GDHt or 1,3-PDOR of *Klebsiella pneumoniae* was sequenced and expressed in *Escherichia coli* successfully [3, 4]. In recent years some messages of construction and characterization of *dha* operon of *K. pneumoniae* were reported, and much more information it contained was available [5–8].

Although many studies on the 1,3-PDOR or GDHt of *K. pneumoniae* and properties of GDH are available on many other organisms, there are few reports on the GDH of *K. pneumoniae* and the corresponding gene. In order to investigate the metabolism of *K. pneumoniae* and improve the productivity of 1,3-PD, we cloned and sequenced the *gdh* gene from *K. pneumoniae*, and then overexpression and characterization was performed in *E. coli*.

21.2 Materials and Methods

21.2.1 Bacterial Strains and Plasmids

Bacterium, *K. pneumoniae* KG1 (kindly provided by Prof. Gong Heng), was used as a *gdh* donor strain. *E. coli* DH5α was used as host for plasmid construction. *E. coli* BL21(DE3) was used as a host strain for expression of pET-*gdh*, which was constructed in this study. pGEM-T (Promega Madison, U.S.A.) a cloning vector and pET-28a(+) were used. *E. coli* was grown aerobically at 37 °C in LB medium supplemented, when necessary, with 25 mg/L kanamycin.

21.2.2 Construction of the Recombinant Vector

Two primers, Pgdh-1 and Pgdh-2, were prepared for PCR to obtain the open reading frame (ORF) of the gene *gdh* according to the sequence of *gdh* of *E. coli* K-12 from Genbank (accession no. NC_000913). Primer Pgdh-1, 5'-ccggaattcatgccgcatttggcactactc-3', consisted of an *Eco*RI site. Primer Pgdh-2, 5'-ccgctcgagTTATTCCCACTCTTGCAGG-3', consisted of an *Xho*I site. PCR was done with the chromosomal DNA of *K. pneumoniae* KG1 as template, and Pfu-DNA polymerase was used (Dingguo Beijing, China). The 1.2 kb product was ligated to pGEM-T to achieve the plasmid pGEM-*gdh* and then Plasmid pGEM-*gdh* was sequenced. Plasmid pGEM-*gdh* was digested with *Eco*RI and *Xho*I (Takara Dalian, China). The resultant was purified and ligated to pET-28a(+) already digested with the same restriction enzymes. The recombinant plasmid pET-*gdh* was obtained.

Other DNA manipulations were carried out according to standard methods. DNA sequencing was done by Invitrogen Co. Ltd. (Shanghai, China).

21.2.3 Preparation of Cell Extract

The cells were harvested by centrifugation $(6,000 \times g, 4 \, ^{\circ}\text{C}, 10 \, \text{min})$. The cell pellet was washed twice with cold kalium phosphate buffer (pH 7.0), then it was suspended in the same buffer, 0.2 (g wet cell)/mL, and dithiothreitol (DTT) was added to 0.1 mmol/L. The cells were treated by sonication with ice bath for 3 s at 200 W, followed by a 5 s pause. This cycle was repeated 60 times. The cell debris was removed by centrifugation $(13,000 \times g, 4 \, ^{\circ}\text{C}, 20 \, \text{min})$ and GDH activity of the supernatant was determined. The protein in supernatant was examined by SDS-PAGE (15 %).

21.2.4 GDH Activity Assay

The activity of GDH was determined spectrophotometrically at 340 nm at different temperatures according to Ahrens et al. [1]. One unit of enzyme activity was defined as the amount of enzyme that consumed 1 μ mol glycerol per minute.

21.3 Results

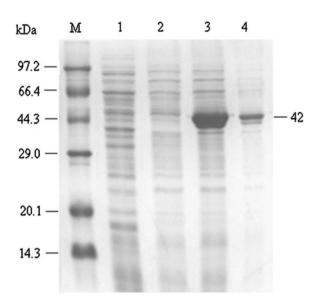
21.3.1 Expression of gdh Gene in E. coli

The gene *gdh* from KG1 was gained by PCR with genomic DNA of *K. pneumoniae* KG1 as template. DNA sequence showed that the open reading frame of *gdh* was 1,143 bp, corresponding 380 amino acid residues.

The expression plasmid, pET-gdh, was transformed into *E. coli* BL21 (DE3) competent cells to express recombinant his-tagged GDH. Expression of the GDH was carried out at different temperatures and IPTG concentrations. The optimum condition for expressing GDH was 1 mmol/L IPTG and at 37 °C (data not shown). Examined by SDS-PAGE, a band corresponding to about 42 kDa protein was observed which was identical to the deduced molecular weight (Fig. 21.1). The SDS-PAGE also revealed that the GDH was soluble in *E. coli* BL21 (DE3), which was important for the high enzyme activity.

192 Y. Liu et al.

Fig. 21.1 SDS-PAGE analysis of GDH protein. Lane M Protein Marker (Takara Dalian, China), Lane 1 KG1, Lane 2 BL21/pET-28, Lane 3 BL21/pET-gdh, Lane 4 the sample of Lane 3 diluted to 10-1 with water



21.3.2 Characteristics of GDH

As depicted in Fig. 21.2, the activity of GDH of *E. coli* BL21/pET28-*gdh* increased as the temperature increased until 37 °C, and then decreased. So 37 °C was its optimum reactive temperature. Hereafter, the enzyme reaction was performed at 37 °C.

Strains of *E. coli* BL21/pET-gdh and *E. coli* BL21/pET-28 were cultured, respectively, at 37 °C to OD600 about 0.5 and then induced for 4 h with 1 mM IPTG. KG1 was cultured at 37 °C. The kinetics of the GDH was compared (Table 21.1).

Surprisingly, the recombinant GDH showed very high dehydrogenate activity, with 774 times and 886 times more than that of $E.\ coli\ BL21/pET-28$ and KG1, respectively (Table 21.1). It was reported that exposure to O_2 of cells growing

Fig. 21.2 Effect of temperature on activity of GDH

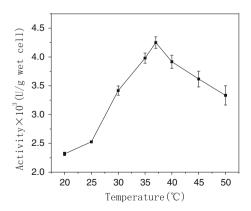


Table 21.1 Activity of GDH in different strains

GDH	Specific activity (U/(g wet cell))
BL21/pET-gdh	$4.17 \pm 0.88 \times 10^3$
BL21/pET-28	5.38 ± 0.64
KG1	4.7 ± 0.42

Table 21.2 Substrate specificity of GDH from *E. coli* BL21/pET-*gdh*

Substrates	Relative activity (%)
Glycerol	100
Glycol	30.7 ± 2.2
1,2-Propanediol	6.7 ± 1.2
2,3-Butanediol	2.3 ± 0.34
1,3-Propanediol	0.6 ± 0.1
1,3-Butanediol	0.3 ± 0.1
1,4-Butanediol	<0.1
Propanol	<0.1
Iso-propanol	<0.1
Butanediol	<0.1

anaerobically resulted in irreversible inactivation of GDH from K. pneumoniae or Klebsiella aerogenes [9]. But both E. coli BL21/pET-gdh and KG1 cultured anaerobically or aerobically showed good activity of GDH in the present of O_2 in this study (data not shown). Therefore, its character to O_2 was different from the GDH of K. pneumoniae reported before [9].

Enzyme activities of GDH with different substrates are summarized in Table 21.2. It revealed that GDH exhibited the most activity with its physiological substrate (glycerol). Moreover, it showed some activity on several diols with adjacent hydroxyls, especially glycol.

21.4 Discussion

By blasting the sequence *gdh* of *K. pneumoniae* KG1 in Genbank (accession no. DQ985161), we found that it had significant similarity to that of bacteria such as *E. coli*, *Salmonella*, *Shigella*, *Klebsiella* and *Citrobacter*. Strains of *E. coli* that were able to convert glycerol to 1,3-PD have been constructed by expressing genes of GDHt (*dhaB*) and 1,3-PDOR (*dhaT*) from *K. pneumoniae* or those from *C. freundii* [10–12]. However, there were at least three enzymes (GDHt, PDOR, and GDH) that composed the key enzyme system of glycerol metabolism for producing 1,3-PD [11]. The GDH from *E. coli* was so similar to that from *K. pneumoniae* (99.7 % similarity) that it could compose the key enzyme system to produce 1,3-PD with GDHt and 1,3-PDOR in recombinant strain containing the gene of *dhaB* and *dhaT*.

GDHt and PDOR were overexpressed both or either in *K. pneumoniae*, however, the recombinant strains did not produce much more 1,3-PD than the wild-type strain

[12]. It was postulated that the accumulation and toxicity of the intermediate 3-HPA was responsible for it [11, 12]. 3-HPA could be reduced to 1,3-PD by 1,3-PDOR, which needed NADH as coenzyme. Moreover, the production of 1.3-PD of E. coli expressing genes of GDHt and 1.3-PDOR was relatively low even after optimizing [12]. The physiological role of 1,3-PD formation was to regenerate the reducing equivalent that was released from the formation of DHA by GDH [13], so NADH was essential for keeping the glycerol reductive pathway active to form 1.3-PD. Considering the high activity of recombinant **GDHt** 1,3-PDOR in engineering strains of K. pneumoniae or E. coli, a large amount of NADH would be needed for the balance of glycerol metabolism. Therefore, the relatively low yield of 1,3-PD in engineering strains was possibly due to the lack of NADH. It was reported that fermentation with glucose as co-substrate could improve the conversion ratio of glycerol to 1,3-PD [14]. A large amount of reducing equivalent and energy contributed from the oxidation of glucose was probably the reason for it. GDH could regenerate NADH by oxidating glycerol [15]. Thus the conversion ratio of glycerol to 1,3-PD would increase if enough reducing equivalents could be regenerated by high activity recombinant GDH or glucose dehydrogenase when glucose was used as co-substrate.

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Chapter 22 Expression of Glucose-6-Phosphate Dehydrogenase and 6-Phosphogluconate Dehydrogenase Improve L-Citrulline Biosynthesis in argG-Deleted Corynebacterium glutamicum

Zhaoxing Liu, Luping Chen, Ning Hao, Lin Xu, Yan Li, Ming Yan and Pingkai Ouyang

Abstract Corynebacterium glutamicum, a well-known producer in the amino acids industry, has become a potential platform organism for synthetic biology in industrial biotechnology. During L-citrulline biosynthesis, NADPH is required as a crucial cofactor. Production of L-citrulline requires 2 mol of NADPH per mole of L-citrulline. The strategy focused on engineering of the pentose phosphate pathway (PPP) flux by different genetic modifications. In this study, two NADPH-supplying strategies based on glucose-6-phosphate dehydrogenase and 6-phosphogluconate dehydrogenase was compared, and their influences on L-citrulline biosynthesis were examined. The zwf and gnd genes were overexpressed in the L-citrulline-producing strain CgAargG. The expression of both genes greatly enhanced intracellular NADPH concentration and L-citrulline production. The concentration of intracellular NADPH was increased by 287 and 363 %, and the production of L-citrulline was increased by 30.8 and 20.5 % in Cg∆argG/pXMJ19-zwf CgΔargG/pXMJ19-gnd, respectively, compared with CgΔargG. The concentration of NADPH in a zwf- and gnd-expressing strain was increased by 287 and 363 %, respectively. These results are helpful for improving L-citrulline biosynthesis and other biosynthetic processes.

Keywords *Corynebacterium glutamicum* · L-Citrulline · L-Ornithine · Glucose-6-phosphate dehydrogenase · 6-Phosphogluconate dehydrogenase

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22.1 Introduction

Corynebacterium glutamicum is the workhorse for the production of amino acids, such as L-glutamate, L-lysine, L-arginine, and L-ornithine [1]. L-Citrulline is also an important amino acid for our health since it is a source of endogenous L-arginine in the body. It is an amino acid of commercial interest that can be used as a food and feed additive and as an intravenous infusion agent. In *C. glutamicum*, L-citrulline is the precursor of L-arginine biosynthesis [2]. The biosynthesis of L-arginine can be improved through overexpressing biosynthetic pathway genes, blocking competing metabolic pathways, improving export systems, and increasing the precursor availability, as researched in recombinant *C. glutamicum* [3–5]. However, cofactor regulation is also important for the biosynthesis of L-arginine and L-citrulline. The most important cofactor for biosynthesis of L-arginine and L-citrulline is NADPH. Two molecules of NADPH are required for generation of one molecule of L-citrulline from ketoglutarate by the enzymes glutamate dehydrogenase (GDH) and *N*-Acetylglutamate 5-semialdehyde dehydrogenase (ArgC, Fig. 22.1). Thus, a sufficient supply of NADPH may be crucial for L-citrulline biosynthesis.

NADPH can be generated through two types of reactions. One is the reductive regeneration of NADPH from NADP+ through various NADP+-dependent dehydrogenases. The important NADP+-dependent dehydrogenases in *C. glutamicum* include glucose-6-phosphate dehydrogenase (Zwf) and 6-phosphogluconate dehydrogenase (Gnd) in the pentose phosphate pathway (PPP) [6, 7] and malic enzyme (MalE) [8]. The overexpression of genes encoding these enzymes or the enhancement of PPP [9] would increase the supply of NADPH. Another reaction for generating NADPH is the phosphorylation of NADH through NADH kinase [10].

To understand the differential influences of Zwf and Gnd on NADPH supply and L-citrulline biosynthesis and to reveal the optimal NADPH-supplying strategy, Zwf expression and Gnd expression were researched in this study.

22.2 Materials and Methods

22.2.1 Strains and Cultivation Conditions

Escherichia coli strains DH5α was used as the hosts for cloning. *E. coli* was grown in Luria-Bertani (LB) medium (tryptone 10 g/L, yeast extract 5 g/L, and NaCl 10 g/L, pH 7.2) at 200 rpm and 37 °C. ArgG-deleted *C. glutamicum* was used for the production of L-citrulline. *C. glutamicum* was grown in LBG medium (LB supplemented with 5 g/L glucose) at 200 rpm and 30 °C. The concentration of chloromycetin was 20 μg/mL for *E. coli* and 10 μg/mL for *C. glutamicum*.

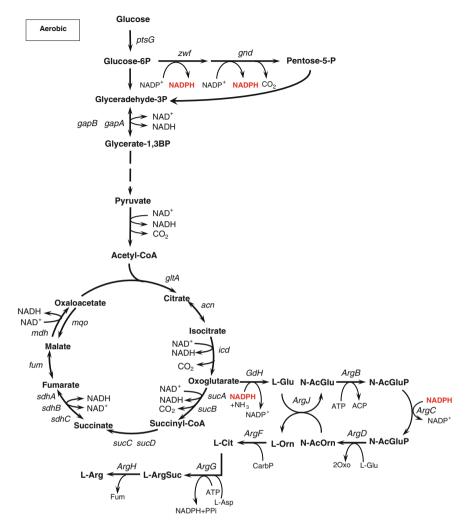


Fig. 22.1 L-Arginine biosynthetic pathway and NADPH metabolism in Corynebacterium glutamicum

22.2.2 Construction of ppnK and zwf Expression Plasmids and Strains

For the expression of *zwf* or *gnd* genes in *C. glutamicum*, the *zwf* gene (1,545 bp) was amplified at first from the genomic DNA of ATCC 13032 using the primers zwf-F and zwf-R and then ligated into vectors pXMJ19 to generate the plasmid, pXMJ19-zwf. Secondly, the *gnd* gene (1,455 bp) was amplified from the genomic DNA of ATCC 13032 using the primers gnd-F and gnd-R and then ligated into

		1
Plasmid or strain	Relevant characteristics	Source
Plasmids		
pXMJ19	Shuttle plasmid	Stored in lab
pXMJ19-zwf	Shuttle plasmid carrying 1545-bp DNA fragment	This study
pXMJ19-gnd	Shuttle plasmid carrying 1455-bp DNA fragment	This study
Strains		
Cg∆argG	Wild type ATCC 13032 + deletion of <i>argG</i> , encoding argininosuccinate synthase	Tang et al. [12]
Cg∆argG/pXMJ19-zwf	Cg∆argG harborring pXMJ19-zwf	This study
Cg∆argG/pXMJ19-gnd	Cg∆argG harborring pXMJ19-gnd	This study

Table 22.1 Plasmids and strains used in this study

vectors pXMJ19 to generate the plasmid, pXMJ19-gnd. Finally, argG-deleted C. glutamicum was transformed with the plasmids pXMJ19-zwf and pXMJ19-gnd to generate two strains, $Cg\Delta argG/pXMJ19$ -zwf and $Cg\Delta argG/pXMJ19$ -gnd (Table 22.1).

22.2.3 Determination of L-Citrulline Production

For the L-citrulline production experiments, C. glutamicum cells were precultured in seed medium (LB medium) at 30 °C and 200 rpm for 12 h. The preculture was inoculated into 50 mL of fermentation medium (glucose 100 g/L, (NH₄)₂SO₄ 20 g/L, urea 5 g/L, K₂HPO₄·3H₂O 1 g/L, KH₂PO₄ 1 g/L, MgSO₄ 0.25 g/L, Mn²⁺, Fe²⁺ 0.01 g/L, ZnSO₄ 1 mg/L, CuSO₄ 0.2 mg/L, CaCl₂ 10 mg/L, biotin 0.1 mg/L, pH 7.0) to a final optical density (OD₆₀₀) of 2.0 and shaken at 30 °C and 200 rpm for 36 h. For the inducible strains of CgΔargG/pXMJ19-zwf and CgΔargG/pXMJ19-gnd, IPTG was added to a final concentration of 1 mM after 12 h of fermentation to induce the expression of the target genes. The amino acid concentration in the fermented broth was measured using high-pressure liquid chromatography (HPLC, DIONEX, UltiMate 3000, USA) after pre-column derivatization with PITC [11]. HPLC separation was performed at 40 °C using a DIONEX C18 column (250 × 4.6 mm). The elution buffer consisted of an aqueous phase (0.1 M sodium acetate, pH 6.5) and an organic phase (80 % acetonitrile). The amino acids were detected at 254 nm, and the spectra were recorded online. The residual glucose concentration in the fermented broth was determined with a biosensor (Institute of Biology, Shandong Academy of Science, China). The cell concentration of fermented broth was determined by measuring the OD₆₀₀ value after dilution with 1 M HCl. The dry cell weight (DCW, grams per liter) was calculated according to an experimentally determined formula: $DCW = 0.25 \times OD_{600}$.

22.2.4 Determination of Intracellular NAD(H) and NADP(H) Concentrations

The intracellular NAD(P)⁺ and NAD(P)H of *C. glutamicum* cells were extracted using HCl and NaOH, respectively, and the amounts of NAD⁺, NADH, NADP⁺, and NADPH in each extract were quantified using a spectrophotometric enzymatic cycling method as described previously [10].

22.3 Results and Discussion

Cg Δ argG is a strain used for L-citrulline production. It can accumulate approximately 13–15 mM L-citrulline after fermentation in a shake flask for 36 h. Because NADPH is an important coenzyme in the L-citrulline biosynthetic pathway, NADPH supply would be crucial. Therefore, in this study, the glucose-6-phosphate dehydrogenase gene (zwf) and 6-phosphogluconate dehydrogenase gene (zwf) from zwf0. zwf1 glutamicum were cloned and expressed in zwf2 and their influence on L-citrulline biosynthesis was investigated.

To elucidate the influence of glucose-6-phosphate dehydrogenase and 6-phosphogluconate dehydrogenase on NADPH supply and L-citrulline biosynthesis, strains expressing zwf and gnd of C. glutamicum by an inducible expression vector pXMJ19 were generated, analyzed, and compared to the initial non-recombinant strain $Cg\Delta argG$ and the control strain carrying the empty vector pXMJ19. Expression of these genes was induced by 1 mM IPTG at 6 h of fermentation. Growth of these strains in fermentation medium revealed that $Cg\Delta argG/pXMJ19$ -zwf and $Cg\Delta argG/pXMJ19$ -gnd exhibited similar growth rates as $Cg\Delta argG$. After fermentation for 36 h, the dry cell weight of the fermented broth from $Cg\Delta argG$, $Cg\Delta argG/pXMJ19$ -zwf, and $Cg\Delta argG/pXMJ19$ -gnd cultures was 10.4 ± 0.2 , 9.4 ± 0.6 , and 11.0 ± 0.2 g/L, respectively. In addition, glucose was consumed in a similar pattern for all the strains. After fermentation for 36 h, approximately 30.2–38.1 g/L (75.5–95.3 %) of the glucose was consumed in these strains (Fig. 22.2).

Next, the intracellular NAD⁺, NADH, NADP⁺, and NADPH concentrations of these strains after fermentation at 30 °C for 24 h were assayed (Table 22.2). The intracellular NADPH concentration in Cg Δ argG/pXMJ19-zwf and Cg Δ argG/pXMJ19-gnd increased by 86 μ M (287 %) and 109 μ M (363 %), respectively. Accordingly, the ratio of NADP(H) to NAD(H) increased in *zwf-* and *gnd-*expression strains.

Finally, the production of L-citrulline in these *C. glutamicum* strains was determined. L-Citrulline production in the fermentation medium of $Cg\Delta argG/pXMJ19$ -zwf and $Cg\Delta argG/pXMJ19$ -gnd increased by 75.6 and 43.6 %, respectively, compared with JHI/pDXW8 and increased by 30.8 and 20.5 %, respectively, compared with $Cg\Delta argG$. This result indicates that the increase of glucose-6-phosphate

Z. Liu et al.

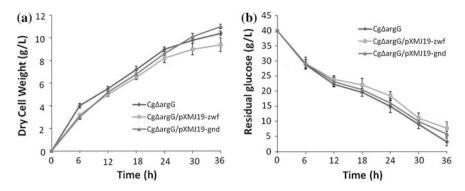


Fig. 22.2 L-Citrulline fermentation of zwf- and gnd-expressing C. glutamicum strains. a Cell growth; b residual glucose

Table 22.2 L-Citrulline production in *argG*-deleted *C. glutamicum* control stains and *zwf*- and *gnd*-expressing strains

Strains ^a	Cg∆argG	Cg∆argG/pXMJ19-zwf	Cg∆argG/pXMJ19-gnd
Stans	Cgdargo	CgZuigGrpZiviJ17-Zwi	Cadana Cripativis 19-gild
$NAD^{+}(\mu M)$	323 ± 37	343 ± 24	357 ± 15
NADH (μM)	44 ± 4	93 ± 23	81 ± 24
NADP ⁺ (μM)	110 ± 7	68 ± 15	84 ± 12
NADPH (μM)	30 ± 8	116 ± 19	139 ± 17
(NADP ⁺ + NADPH)/ (NAD ⁺ + NADH)	0.38	0.42	0.50
(NADH + NADPH)/ (NAD ⁺ + NADP ⁺)	0.17	0.51	0.50

^a The averages of three independent experiments are shown

dehydrogenase and 6-phosphogluconate dehydrogenase activity contributes to Ile biosynthesis, with the effect of glucose-6-phosphate dehydrogenase (Zwf) better than that of 6-phosphogluconate dehydrogenase (Gnd). In addition, the L-citrulline yield on glucose as well as the specific L-citrulline yield by $Cg\Delta argG/pXMJ19$ -zwf and $Cg\Delta argG/pXMJ19$ -gnd improved obviously compared with the control strain $Cg\Delta argG$ (Table 22.3).

Corynebacterium glutamicum is the most widely used bacterium for the production of amino acids, including L-citrulline. During the biosynthesis of these amino acids, NADPH is required as an important cofactor, and thus sufficient supply of NADPH may be helpful for the biosynthesis of amino acids. NADPH can be supplied by reductive regeneration of NADP+ through various NADP+-dependent dehydrogenases or by phosphorylation of NADH through NADH kinase. Several important NADP+-dependent dehydrogenases of C. glutamicum had been examined previously, such as Zwf and Gnd of the PPP pathway. Conversion of the carbon flux from glycolytic pathway to PPP pathway had also been demonstrated to improve the NADPH regeneration and amino acid biosynthesis, such as the expression of the

Strains ^a	CalamaC	CaAanaC/aVMI10 avvf	CaAanaC/nVMI10 and
Strains	Cg∆argG	Cg∆argG/pXMJ19-zwf	Cg∆argG/pXMJ19-gnd
OD ₆₀₀	41.6 ± 0.8	37.5 ± 2.4	44.0 ± 0.8
Glucose consumed (g/L)	36.9 ± 1.2	32.3 ± 2.1	34.3 ± 0.9
Glucose consumed (mM)	204.6 ± 6.7	179.5 ± 11.7	190.2 ± 5.0
Glucose uptake rate (mmol g ⁻¹ (cdw) h ⁻¹)	0.55 ± 0.09	0.53 ± 0.05	0.48 ± 0.07
Glutamate titer (mM)	72.7 ± 3.9	1.2 ± 0.06	1.5 ± 0.08
Citrulline titer (mM)	14.6 ± 0.9	19.1 ± 1.2	17.6 ± 1.1
Arginine titer (mM)	1.55 ± 0.09	0.53 ± 0.03	0.49 ± 0.03
Ornithine titer (mM)	17.1 ± 3.4	24.7 ± 4.9	24.3 ± 4.8
Yield citrulline/glucose (mol/mol)	0.07 ± 0.01	0.11 ± 0.01	0.09 ± 0.01
Citrulline production rate (mmol g ⁻¹ (cdw) h ⁻¹)	0.039 ± 0.005	0.057 ± 0.005	0.044 ± 0.012

Table 22.3 L-Citrulline production in argG-deleted C. glutamicum control stains and zwf- and gnd-expressing strains

fructose 1,6-bisphosphatase gene (fbp) and deletion of the phosphoglucoisomerase gene (pgi).

Among the inducible expression strains, the mere expression of *zwf* and *gnd* gene did not obviously influence the cell growth and glucose consumption. Such results demonstrate coenzyme regulation and NADP(H) supply as a useful metabolic engineering strategy for improving L-citrulline biosynthesis. However, the effect of these NADPH-supplying strategies on other biosynthetic pathways, as well as the systematic metabolic engineering of Cg Δ argG based on such NADPH-supplying strategies and other crucial metabolic engineering strategies shall be researched in the future.

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Part II Optimization and Control of Biological Process

Chapter 23

Production of Amino Acids by Mixed Bacterial Strains-Mediated Solid State Fermentation of Feathers and Dynamic Changes to the Fermentation System

Yu Li, Dunji Hu, Sheng Chen, Xiangnan Lei, Xiangjin Zhang, Xiaoguang Liu and Fuping Lu

Abstract Bioprocessing of chicken feather wastes for production of free amino acids was investigated by solid-state fermentation (SSF) with (1) single strain, (2) double strains, (3) triple strains of B. licheniformis TCCC 11593, B. alcalophilus TCCC 11004 and B. subtilis TCCC 11279. The maximum production of total amino acids amounting to 161.35 mg/g was obtained under optimal SSF conditions: feather concentrations 1.0 g/250 mL, moisture contents 1:9 (w/v), inocula ratios (v/v) of 5:5:3 for *B. licheniformis*: *B. alcalophilus*: *B. subtilis respectively*, 60 h fermentation. Monitoring of mixed bacterial populations in the SSF process by real-time PCR indicated that B. alcalophilus was the dominant microorganism present in the SSF cultures. Moreover, measurements of dynamic changes in multienzyme activities in the SSF process suggested that, in terms of amino acid production, keratinase might play an important role in the initial hydrolysis of keratin and the further hydrolysis was most likely to be accomplished by elastase, collagenase, alkaline protease, and neutral protease elaborated by the three mixed bacterial strains. The above results indicated that mixed-strain SSF could serve as a cost-effective method to utilize chicken feather wastes to produce value-added feed additives or organic fertilizers.

Keywords Feathers waste • Solid-state fermentation • Total amino acids products • Dominant microorganism • Multienzyme activities

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208 Y. Li et al.

23.1 Introduction

Though chicken feathers contain 90 % keratins, they have found limited use because keratins are resistant to degradation by common proteolytic enzymes such as pepsin, trypsin, and papain [1]. Worldwide, millions of tons of feather wastes are generated annually by poultry-processing industries and has become a solid waste management problem [2]. However, feather wastes are a rich source of natural protein and degradation of keratinous substrates represents a potential cost-effective means to produce amino acids that can be used as food additives for animals or a source of nitrogen for plants [3]. Traditional ways of hydrolyzing feathers such as alkali hydrolysis and steam pressure cooking to produce feather meal as animal feed supplement may destroy certain types of valuable amino acids, not to mention that such processes consume large amounts of energy [4] Biodegradation by microorganisms possessing keratinolytic activity represents an alternative attractive method of improving the nutritional value of keratin wastes, as it offers a cheap approach to the production of value-added products [5].

Keratins are easily degraded by keratinases [6] but single keratinase could not efficiently degrade feather even after prolonged incubation. It was shown that complete feather degradation occurred only when bacteria cells were used [7]. However, utilization of substrates by a single strain in pure-culture-based processes has been limited to a narrow range of biomaterials [8]. Mixed cultures are structured by highly diverse microbes and constructed from natural inocula, which facilitate biological processes under nonsterile conditions without altering strain stability. Furthermore, high microbial diversity enables such cultures to utilize complex substrates with complex compositions [9].

Effective use of mixed culture fermentation requires a better understanding of dynamics of microbial communities participating in a specific fermentation process, thus allowing the creation of optimal conditions for desirable population composition of the mixed culture. In recent years, real-time polymerase chain reaction (PCR) approach has been an effective means for analyzing the structure and species composition of complex microbial communities in a wide range of samples from soil [10], compost [11], and solid-state fermentation (SSF) [12].

In this study, a feather-degrading system comprising *Bacillus licheniformis*, *Bacillus alcalophilus*, and *Bacillus subtilis* was employed for SSF of chicken feather meal. Isolates of *B. licheniformis* and *B. subtilis* have shown previously to be capable of effective degradation of feathers, including *B. licheniformis* PWD-1 [13], *B. subtilis* RM-1 [14].

The aim of this study was to improve production of amino acids from chicken feather meal using mixed-culture SSF. Mixed-culture SSF conditions for maximum production of amino acids were determined. Dynamics of bacterial populations and multienzyme activities during SSF were monitored by real-time PCR and specific enzymatic activity assays, respectively, and plausible connections among bacterial population composition, multienzyme activities, and amino acid contents were examined.

23.2 Materials and Methods

23.2.1 Microorganism and Inoculums

The three bacterial strains used throughout this work, *B. licheniformis* TCCC 11593, *B. alcalophilus* TCCC 11004, and *B. subtilis* TCCC 11279 were previously isolated from chicken manure from a chicken farm in Tianjin and identified by the lab. These three strains were maintained on a medium containing (%) yeast extract 0.5, peptone 1.0, NaCl 1.0, agar 2.0 %, pH 7.5.

23.2.2 Solid Substrate Fermentation Process

At the beginning, chicken feathers including barbs, calamuses, and rachises were chopped into 5 mm segments with a grinder. Then, 1 g finely cut feather power and 9 mL culture solution containing (g/L) maltose 1, NH₄Cl 1.5, K₂HPO₄ 1, KH₂PO₄ 0.4, NaCl 0.4, MgSO₄ 0.3 were added into 250 mL Erlenmeyer flasks and mixed thoroughly. After that, the three bacteria strains with a colony concentration of about 1×10^9 CFU/mL were inoculated into the 250 mL Erlenmeyer according to a proportion 1:1:1. Finally, inoculated flasks were incubated in relative humidity of 87 %, at 37 °C for 60 h.

23.2.3 Real-Time PCR

The genomic DNA was extracted from SSF samples by the method previously described with slight modifications [15] and stored at -20 °C. The three pairs of primers used and the predicted sizes of amplicons are shown in Table 23.1.

Real-time PCR amplification of specific sequences from the three bacterial genomes was used to estimate the total number of cells of individual strains. Real-time PCR amplification was conducted on a 7,500 Fast Real-Time PCR System (Applied Biosystems) using the SYBR ®Premix Ex TaqTM (Perfect Real Time) kit (Takara).

Table 23.1	Three pairs of primers	used and the sizes	s of predicted amplic	cons of B. licheniformis,
B. alcalophi	ilus and B. subtilis			

Bacillus sp.	Primer: F	Primer: R	Amplicons (bp)
B. licheniformis	ATTATCAGGAA	GAAACTTTTC	646
	TTTGAACGTGG	AAGCGCTTCT	
B. alcalophilus	GGCACTAC	TCTCCACCTG	678
	GGGCATCG	GAGAGAGGG	
B. subtilis	AATCCCCG	GACCATGTAC	433
	TTTCGTCCA	TCTTGCCATG	

210 Y. Li et al.

23.2.4 Assay of Enzymatic Activity in Solid-State Fermentation

Keratinase activity was determined using 1 % feather powder in Tris/HCl (50 mM, pH 8.0) as a substrate. One unit (U/mL) of keratinolytic production was defined as an increase in the absorbance at 280 nm (A280) [16] of 0.01 per hour under the conditions described above.

Elastolytic activity was assayed by the colorimetric method of Sacher [17]. One unit of elastase activity was defined as the amount of enzyme required to solubilize 20 mg of elastin-Congo red under the tested conditions.

The activity of collagenolytic protease was assayed by the method of Kang [18] with modifications. One unit (U) of collagenase activity was expressed as 1 mmol of glycine equivalent released per mL per minute.

The Anson–Hagihara method was employed for alkaline protease assay [19] using casein as the substrate. The protocol was previously described [20]. One unit of alkaline protease activity (U) was defined as the amount of enzyme that produces 1 µg of tyrosine/min under the assay conditions.

Neutral protease activity was assayed by the method of Chandran Sandhya [21]. One unit of enzyme activity was defined as the amount of enzyme that produced 1 mM tyrosine/min under the assay conditions.

23.2.5 Estimation of Free Amino Acids by HPLC

The quantities of amino acids released during feather degradation by mixed bacterial strains under SSF were estimated by an HPLC method described by Sieberr [22].

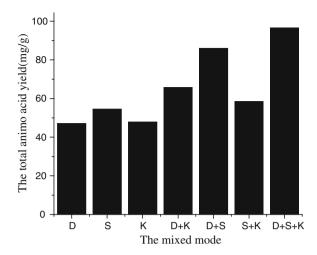
23.3 Results and Discussion

23.3.1 Determination of Optimal Parameters for Production of Amino Acids by SSF

As shown in Fig. 23.1, SSF of feather meal by mixed-culture of the three bacterial strains resulted in higher levels of total amino acids compared with that of SSF by either mixed-culture of two bacterial strains or a single bacterial strain. The highest amino acid yield was 96.61 mg/g.

The optimal combination of variables for the highest yield of total amino acids was determined to be: the inoculation volume for *B. licheniformis*, *B. alcalophilus*, and *B. subtilis* was 5 mL $(5.03 \times 10^9 \text{ CFU/mL})$, 5 mL $(4.95 \times 10^9 \text{ CFU/mL})$ and

Fig. 23.1 Effect of mode of SSF on total amino acid yield (D: *B. licheniformis*, S: *B. alcalophilus*, K: *B. subtilis*)



3 mL $(2.6 \times 10^9 \text{ CFU/mL})$, respectively. The maximum production of total amino acids amounting to 161.35 mg/g was obtained.

Through optimization of mixed-strain SSF conditions, as shown in Fig. 23.2. the highest yield of total amino acids (161.35 mg/g) was obtained when medium feather concentration was 1.0 g/250 mL, moisture content 1:9 (W/V), and fermentation time 60 h. Decreased production of total amino acids was observed at 72 h (114.21 mg/g) and reached the lowest level at 84 h (89.64 mg/g) (Fig. 23.2), which could be explained by the utilization of available free amino acids by the three bacterial strains as carbon and nitrogen sources. However, further fermentation resulted in improved yields of total amino acids (131.47 mg/g) at 96 h (Fig. 23.2).

When the feather concentrations were increased to more than 1 g/250 mL (1.5– 3 %), the total amino acid contents started to decline, most likely due to the substrate inhibition effect caused by the high concentration of feather [23]. Among several important factors affecting microbial growth and enzyme yields under SSF using a particular substrate, moisture level is considered as one of the most critical [24]. SSF processes differ from submerged fermentation culture, since microbial growth and product formation occur at or near the surface of the solid substrate particle [24]. As shown in Fig. 23.2, increases in medium moisture contents resulted in increased levels of total amino acid production and near linearity between moisture content and the total amino acids yield was observed from 1:1 to 1:9 (w/v). Maximum total amino acid yields were achieved at a moisture content of 1:9 (w/v). Thereafter, further increases in moisture level in the fermentation medium resulted in reduced levels of total amino acids (Fig. 23.2). Under solid-state fermentation, levels of moisture higher than the optimum caused a decrease in porosity, an alteration in substrate particle structure, and a lower oxygen transfer, which had a negative effect on Bacillus sp. growth and metabolism [25]. On the other hand, at the lowest and the highest moisture levels, the decomposition rate of the total organic matter decreased [26].

212 Y. Li et al.

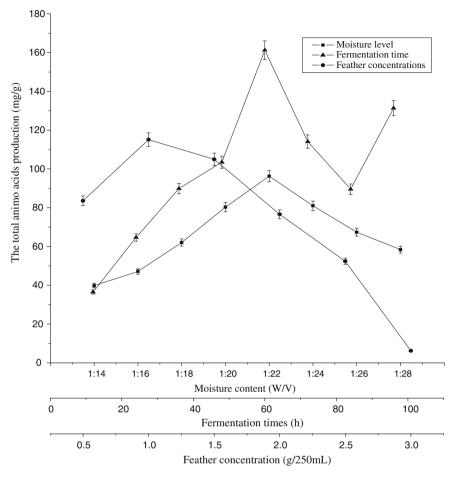


Fig. 23.2 Effect of moisture level (w/v), fermentation time (h) and feather concentrations (g/250 mL) on total amino acid yield

23.3.2 The Dynamic Change in the Populations of the Three Bacillus Species

The three *Bacillus* strains used for SSF of chicken feathers are highly similar in terms of colony morphology. Thus, changes in the bacterial populations of the individual strains during SSF were enumerated by real-time PCR. The standard curves used to quantify the genome copies of the three bacterial strains in SSF were generated by plotting threshold cycles against the logarithm of the initial copy

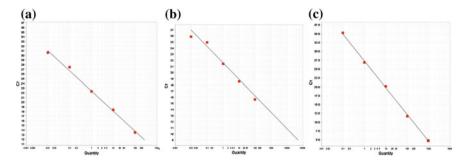


Fig. 23.3 a, b, c The standard curves of B. licheniformis ($R^2 = 0.994$), B. subtilis ($R^2 = 0.991$), and B. alcalophilus ($R^2 = 0.998$)

Table 23.2 The total amino acid yield (mg/g) produced by solid-state fermentation and the numbers of genome template copies of the three *Bacillus* sp. (copies per gram of sample) in the SYBR green real-time PCR assay

Sample (h)	Total amino acids yield	Copy numbers (c	copies/g sample)	
	(mg/g)	B. alcalophilus	B. licheniformis	B. subtilis
12	36.59	2.203E+10 ±3.325E+08	2.515E+10 ±1.056E+08	7.813E+09 ±3.325E+08
24	64.74	2.226E+10 ±2.822E+08	2.526E+10 ±1.161E+08	7.836E+09 ±6.302E+07
36	89.97	2.545E+10 ±2.834E+08	2.645E+10 ±1.854E+08	7.945E+09 ±8.854E+07
48	103.59	2.977E+10 ±6.519E+08	2.927E+10 ±5.319E+08	8.227E+09 ±6.519E+07
60	161.35	3.704E+10 ±3.367E+08	3.502E+10 ±3.372E+08	8.802E+09 ±3.367E+07
72	114.21	3.714E+10 ±1.421E+08	3.634E+10 ±1.431E+08	8.914E+09 ±1.421E+08
84	89.64	3.727E+10 ±1.140E+09	3.687E+10 ±1.140E+08	8.997E+09 ±1.140E+08
96	131.47	3.798E+10 ±4.680E+08	3.703E+10 ±4.680E+08	9.043E+09 ±4.680E+07

number of the target DNA molecules. The known concentrations of PCR products of *B. licheniformis* and *B. subtilis* were diluted from 10^7 copies/ μ L to 10^3 copies/ μ L and of *B. alcalophilus* diluted from 10^8 copies/ μ L to 10^4 copies/ μ L. A high reproducibility between three independent standard dilutions was obtained. The curves showed good linear relationship between the amounts of PCR products and C_T of the target gene (Fig. 23.3a–c). In the range of the detection concentration, the real-time PCR efficiency was calculated as 1.13 based on the equation amplification

214 Y. Li et al.

efficiency = $10^{-1/\text{slope}} - 1$ [27]. In the standard samples, melting curve with a single melting peak was detected at 86.1 °C, and no primer dimers or other unspecific PCR products were amplified, indicating the presence of a single amplicon for each strain. Table 23.2 shows the dynamic change of three strains in gene copies during SSF.

As shown in Table 23.2, it was apparent that the number of *B. alcalophilus* cells was much greater than that of either *B. licheniformis* or *B. subtilis* during 24–60 h. Therefore, it can be concluded that *B. alcalophilus* played a dominant role in the mixed culture SSF using feather meal as a substrate.

In our previous work, *B. licheniformis* TCCC11593 showed a strong ability to degrade feathers, with a maximum keratinase activity of 98.0 U/mL after 48 h of submerged fermentation. The results shown in Table 23.2 indicate that the number of *B. licheniformis* cells was second only to that of *B. alcalophilus* and was closely related to the total amino acid yield. In the meantime, the increase of *B. subtilis* cells number also had a role in promoting production of amino acids. After 60 h, with the consumption of amino acids and the accumulation of harmful metabolites, the total population of the three *Bacillus* strains began to level off.

23.3.3 Dynamic Changes of Multiple Enzyme Activities

A combination of different enzymes could effectively hydrolyze complex substrates of feed stuff [28]. To investigate the relationship between changes in activities of multiple enzymes and the total amino acid contents during feather meal SSF, we measured the respective activities of keratinase, elastase, collagenase, alkaline protease, and neutral protease at indicated time points.

As shown in Fig. 23.4a, during 0–24 h, increased levels of keratinase activity were detected and the highest value reached at 24 h. However, after 24 h, keratinase activity exhibited a dramatic decline and was undetectable at about 60 h, while elastase, collagenase, alkaline protease, and neutral protease activities displayed significant increases in feather meal SSF during 0–60 h. As shown in Fig. 23.4b–d, the keratinase activity was mainly elaborated by *B. licheniformis* and *B. alcalophilus*, while the collagenase was produced by *B. licheniformis* and *B. subtilis*, and all three *Bacillus* species produced high levels of elastase activities.

The fact that total amino acid contents peaked at 60 h, while the levels of keratinase activity started to decline after 24 h and other multiple proteases activities continued to increase until 60 h, suggest that keratinase played an important role during the first 24 h fermentation process for amino acid production, and elastase, collagenase, alkaline protease, and neutral protease used denatured keratins as substrates to produce more amino acids through further hydrolysis during the period of 24–60 h in SSF process.

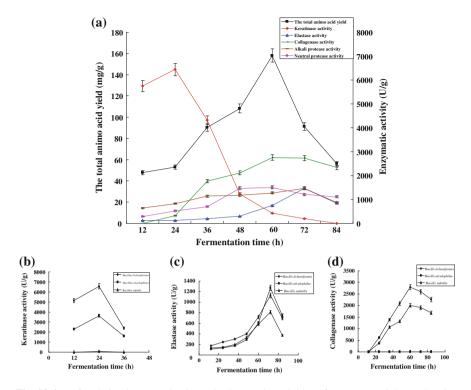


Fig. 23.4 a-d Relation between the dynamic changes in activities of enzymes and the total amino acid yield in SSF

23.3.4 Analysis of Composition of the Total Amino Acids

HPLC method was used to determine the major composition of the hydrolysates of the feather meal mixed-culture SSF. As shown in Table 23.3, under the optimum fermentation conditions, the free amino acids detected in the SSF culture included serine, arginine, tyrosine, valine, cystine, isoleucine, norleucine, and lysine. Interestingly, isoleucine was the most abundant amino acid, reaching 28.01 mg/g, which may be due to the high disulfide content of feather keratin [29]. Previous work showed that the major free amino acids of degradation of feathers by *B. licheniformis* PWD-1 are leucine, tryptophan, valine, isoleucine, phenylalanine, and alanine [13]. The differences could be due to the use of different microorganisms [30].

Table 23.3 Free amino acid analyses in SSF by mixed bacterial strains at 37 °C with a feather concentration of 1.0 g/250 mL, a moisture content of 1:9 (w/v)

in 60 h	•	•				0	,	
Amino acids	Serine	Arginine	Tyrosine	Valine	Cystine	Isoleucine	Norleucine	Lysine
The yield (mg/g)	3.33 ± 0.02	8.05 ± 0.03	6.05 ± 0.04	9.41 ± 0.05	22.09 ± 0.07	28.01 ± 0.06	7.76 ± 0.01	2.52 ± 0.01

23.4 Conclusion

In this work, mixed three-strain SSF of chicken feathers was shown to be superior to either single strain or double-strain SSF for optimal production of free amino acids, which can be used either as a low cost ingredient in animal feed formulation or as an organic fertilizer. Monitoring of the dynamics of the mixed bacterial populations, multienzyme activities, and contents of total amino acids during the course of SSF has provided valuable insights for further designing a cost-effective process to produce amino acids from chicken feather wastes.

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Chapter 24 Effects of Culture Medium on PUFAs Production by Mortierella isabellinas

Dengyue Sun, Cuixia Zhou, Chuanhe Zhu and Yanru Sun

Abstract In the paper, in order to improve the polyunsaturated fatty acids (PUFAs) by *Mortierella isabellina*, the effects of carbon source, nitrogen source, carbon to nitrogen (C/N) ratio, and metal ions on PUFAs production were investigated. The results of the research showed that different kinds of carbon source and nitrogen source had obviously effect on PUFAs production, the glucose and mixture nitrogen (yeast extract: peanut meal at 3:1 (w/w)) were greatly suitable for PUFAs production and cell growth. The glucose concentration and C/N ratios have significant effect on PUFAs production. Meanwhile, the results indicate the metal ions have profound effects on fungus growth and PUFAs production. Ca²⁺ and Mg²⁺ could promote PUFAs production. Almost all the improvement of PUFAs was obtained through increasing thallus biomass, though there were some influences on the metabolism. Under the optimal culture media composition, the yields of cell dry weight, linoleic acid (LA), gama-linolenic acid (GLA), and total PUFAs were 32.48, 2.90, 2.83, and 7.36 g/L, respectively.

Keywords *Mortierella isabellina* • Polyunsaturated fatty acids • Carbon source • Nitrogen source • C/N ratio • Metal ions

24.1 Introduction

Polyunsaturated fatty acids (PUFAs) present a potential industrial value and economic value owning to their specific characteristics [1]. In recent years, there has been an increased interest in nutrient and medical application, especially the ω -3

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D. Sun et al.

and ω -6 series of PUFAs are supposed to be potential food additives and pharmaceuticals for their biological activities [2]. As we all know, the PUFAs could be used to treat heart and circulatory disorders, cancer, or inhibit the inflammatory reaction effectively [13].

Up-to-date, a variety of microorganisms has been found to yield exceptionally large amount of PUFAs under some special conditions. The PUFAs production of microorganisms could be manipulated and changed substantially just by simply changing their growth conditions instead of having to depend on genetic manipulation or extensive breeding efforts [3]. In this study, one of the typically realistic commercial fungal alternatives called *Mortierella isabellina* was chosen as the target microorganism.

Over the past several years, many researchers have put much effort to develop the more economical bioprocesses for PUFAs production. In particular, the production of PUFAs using the fungus *Mortierella* in liquid culture has been extensively studied [4, 5]. But, the cost of PUFAs production was still higher. The mutant gained mutagenesis screening has its own requirement of special conditions for maximum objective product production [6]. In the present study, the effects of the key medium nutrient such as carbon source, nitrogen source, C/N ratio, and metal ions on the LA, GLA, and total PUFAs production of *M. isabellina* were investigated.

24.2 Materials and Methods

24.2.1 Microorganism and Culture Mediums

Microorganism: *M. isabellina* NTG₁₋₁₂₁ was a mutant of *M. isabellina* As 3.3410 obtained through mutagenesis screening in our lab and was kept on potato dextrose agar (PDA) at 4 °C.

Culture mediums: Seed medium contained (g/L): $(NH_4)_2SO_4$, 2; yeast extract, 2; glucose, 100; KH_2PO_4 , 1 and $MgSO_4$, 0.3. Basal submerged culture medium: The carbon source was glucose (100 g/L); nitrogen source was yeast extract (2 g/L); KH_2PO_4 was at 2 g/L; $MgSO_4$ was at 0.3 g/L; and sodium citrate was at 2 g/L. The initial pH of all mediums was adjusted to 6.0 before autoclaving and all cultures were sterilized at 121 °C for 20 min.

24.2.2 Cultivate Methods

All the slants were incubated at 28 ± 1 °C for 6 days. And all research experiments were performed in 500 mL conical flasks, containing 100 mL of liquid medium. The seed medium was added with 1 mL of spore suspension $(1.0 \times 10^5 - 10^6 \text{ cfu/mL})$ before being incubated in an orbital shaker (TENSUC, Shanghai, China) at 28 ± 1 °C

for 2 days at 160 rpm. And the fermentation medium was inoculated with 10 mL seed culture solution, shaken at 160 rpm and incubated at 28 ± 1 °C for 7 days.

24.2.3 Cell Dry Weight Determination and Lipid Extraction

Cell dry weight determination: After 7 days, the mycelia were harvested from the fermentation culture by suction filtration through Whatman-1440-110. Cell dry weight was determined. The dried mycelia were transferred to a 50 mL centrifuge tube after being grinded.

Lipid extraction: Extraction of lipids from mycelia was carried out according to the modified procedure of Bligh and Dyer [7].

24.2.4 Fatty Acid Analysis

Fatty acid methyl ester (FAME) was prepared by the process of boron trifluoride (GB/T 17376-2008) [8]. The fatty acids composition and content were analyzed by using Gas Chromatography-Mass Spectrometer (GC-MS), and the chromatographic column was Stabilwax-DA (30 m × 0.32 mm × 0.25 µm). The initial column temperature (200 °C) was maintained for 8 min, increased to 230 °C with the increasing rate of 5 °C min⁻¹, and the temperature was maintained at 230 °C for 10 min. Helium was used as the carrier gas, and the flow rate was 2.0 mL min⁻¹, with one 10:1 of splitting ratio. Electron impact (EI) mode with a mass scan range of 45-450 amu was applied in the mass spectrum, and the ionization voltage was set at 70 eV. Both the injector port and ion source temperature were maintained at 200 °C. Methyl esters of different fatty acids such as linoleic acid, gama-linolenic acid, palmitic acid, palmitoleic acid, stearic acid, oleic acid, linoleic acid, and so on were used as standard for fatty acid identification and quantitation [9]. There is a linear correlation between the concentration of fatty acids (X) and the peak area (Y) at a specific retention time. In the concentrations between 0 and 4,000 mg/L, the equation and correlation coefficient (r) were Y = 11692x + 80378 and 0.9965, Y = 106803x + 108407 and 0.9995 for gama-linolenic acid and linoleic acid, respectively.

24.2.5 Effects of Carbon and Nitrogen Source, C/N Ratio, and Metal Ions on PUFAs Production

Effect of carbon source on PUFAs production: In the experiment, glucose in the basal fermentation medium was replaced by sucrose (100 g/L), maltose (100 g/L), molasses (100 g/L), and soluble starch (100 g/L) and incubated at 28 °C for 7 days.

D. Sun et al.

For carbon source concentration experiments, the range of carbon source concentrations was from 0 to 140 g/L.

Effect of nitrogen source and C/N ratios on PUFAs production: Yeast extract as the sole nitrogen source in the basal medium was replaced by 2 g/L of beef extract, peptone, urea, (NH₄)₂SO₄, KNO₃, peanut meal, and soybean cake powder. Based on the screening results of nitrogen source, the effect of different ratios of yeast extract and peanut meal (w/w) on PUFAs production was evaluated. The effect of C/N ratio was studied with 100 g/L of carbon source, adjusting the ratio of the mixture of peanut meal and yeast extract at 1:3 (w/w) from 10 to 50.

Effect of metal ions on PUFAs production: Magnesium (Mg^{2+}) , manganese (Mn^{2+}) , iron (Fe^{2+}) , and copper (Cu^{2+}) ions were in the form of sulfates $(MgSO_4\cdot 7H_2O\ 99.0\ \%,\ MnSO_4\cdot H_2O\ 99.0\ \%,\ FeSO_4\cdot 7H_2O\ 99.0\ \%,\ CuSO_4\cdot 5H_2O\ 99.0\ \%)$; while zinc (Zn^{2+}) and calcium (Ca^{2+}) ions in the form of iron II chloride, which are anhydrous and the purity were 98.0 and 96.0 %, respectively. Fe^{2+} , Cu^{2+} , Zn^{2+} , and Ca^{2+} were added into basal fermentation medium containing the optimal carbon and nitrogen source and the concentrations were all 5, 50, and 100 mg/L (concentrations referred to the effective metal ion content), respectively. And the concentration of Mg^{2+} was 50, 100 and 500 mg/L, while the concentration of Mn^{2+} was 2, 10, and 100 mg/L.

24.2.6 Statistical Analysis

All treatments were carried out in triplicate, and the experimental data was analyzed by analysis of variance and Duncan's multiple range tests using the Statistical Analysis System (SAS 8.0) [10].

24.3 Results and Discussions

24.3.1 Effect of Carbon Source on PUFAs Production

The effect of carbon source on PUFAs production was showed in Table 24.1. A high value of cell dry weight was obtained when glucose and maltose were used as carbon source, while molasses proved to be the poorest growth carbon source for the strain and yielded very small detectable biomasses of PUFAs followed by sucrose. Soluble starch was not the poorest carbon source for cell growth, but it was not suitable for LA, GLA, and total PUFAs production. In the study, the results indicated the glucose be the best carbon source for LA, GLA, and total PUFAs production.

According to the above research results, the effect of the different glucose concentrations on PUFAs production was investigated, and the results were

Cell dry weight

 $(g L^{-1})$

Carbon source

Total PUFAs (g L⁻¹)

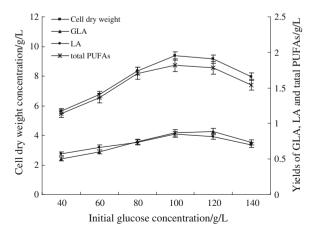
	(8 – 7					
Glucose	9.14 ± 0.37 a	$0.80 \pm 0.07 \text{ a}$	$0.79 \pm 0.13 \text{ a}$	1.62 ± 0.14 a		
Sucrose	$4.38 \pm 0.15 \text{ b}$	$0.16 \pm 0.04 \text{ b}$	$0.16 \pm 0.06 \text{ b}$	$0.33 \pm 0.05 \text{ b}$		
Maltose	6.63 ± 0.29 c	0.41 ± 0.11 c	0.40 ± 0.09 c	$0.82 \pm 0.09 \text{ c}$		
Molasses	1.24 ± 0.07 d	NF	NF	$0.0046 \pm 0.0005 \text{ d}$		
Soluble starch	5.87 ± 0.24 e	$0.14 \pm 0.08 \; d$	$0.13 \pm 0.02 \text{ d}$	0.27 ± 0.01 e		
Data are expressed as means \pm standard deviations of triplicate assays. The different alphabetic superscripts in the same column are significantly different ($p < 0.05$) according to Duncan's multiple range test. NF no found						
represented in Fig. 24.1. As presented in Fig. 24.1, cell dry weight, GLA, LA, and total PUFAs production gradually increased with increasing glucose concentration, and then decreased when the concentration was higher than 120 g/L. When the concentrations of glucose were at 100 and 120 g/L, there was maximum yield of						

 $GLA (g L^{-1})$

Table 24.1 Effect of carbon source on biomass and PUFAs production of NTG₁₋₁₂₁ $LA (g L^{-1})$

concentrations of glucose were at 100 and 120 g/L, there was maximum yield of cell dry weight, GLA, LA, and total PUFAs. The results of statistical analysis showed there are no significant difference at the two concentrations (p > 0.05). The results were similar to those obtained by Papanikolaou et al. [11] who found that the cell dry weight and lipids achieved were remarkably high when the initial glucose concentration was manipulated at 101.9 g/L, though the fermentation conditions were nonoptimized. The detailed explanation of Fig. 24.1 was that at 100 g/L of glucose, the total PUFAs reached to 1.82 g/L including 0.87 g/L of GLA and 0.85 g/L of LA. In the case of 120 g/L of glucose, 1.79 g/L of the total PUFAs containing 0.88 g/L of GLA and 0.82 g/L of LA could be obtained. While the growth condition of fungus was worse when the glucose content was 0 g/L (data not shown in Fig. 24.1) in the fermentation medium. Similarly, the result was also reported by Xain et al. [12]. The relatively optimum concentration of glucose (100 g/L) was used in the following experiments.

Fig. 24.1 The effect of glucose concentration on PUFAs production



D. Sun et al.

24.3.2 Effects of Nitrogen Source and C/N Ratio on PUFAs Production

The effect of nitrogen source on PUFAs production was presented in Table 24.2. Peanut meal gave the highest cell biomass and PUFAs yields including LA, GLA and total PUFAs production, which were 13.59, 1.10, 1.06 and 2.21 g/L, respectively. Peptone, beef extract, and soybean cake powder were the second best nitrogen for cell dry weight, however, the lipids content were obviously lower than yeast extract when they were used as nitrogen source. As we all know, yeast extract is one of the typical nitrogen sources for the cultivation of microorganisms. In the study, although the dry biomass just achieved 10.03 g/L, the yields of LA, GLA, and total PUFAs production all were significantly higher than the case that peptone, beef extract, and soybean cake powder. KNO₃ as one kind of nitrate used as nitrogen source should show an obviously high cell growth and better PUFAs production than (NH₄)₂SO₄. However, in the study the difference of cell dry weight and PUFAs production between KNO₃ and (NH₄)₂SO₄ were not significant. There was almost no cell growth obtained with the urea source as well as with the 0 g/L nitrogen source control. It was not surprising since both glucose and urea does not contain metal ions and other vital micronutrients for cell growth or since one would not expect in the absence of one nitrogen source. Thus, the effect of different ratios of a mixture of yeast extract and peanut meal (w/w) on PUFAs production was carried out and the results were shown in Fig. 24.2.

From Fig. 24.2, the mixtures of yeast extract and peanut meal (3:1, w/w) had higher cell biomass and yields of GLA, LA, and total PUFAs than that of other mixture ratios. And the yield of GLA, LA, and PUFAs gradually declined along with the increase of peanut meal except the cell dry weight had slight rise. In view of the same aim, many other literatures have also investigated the effect of different ratios of various nitrogen sources on one certain PUFA or PUFAs production [13, 14].

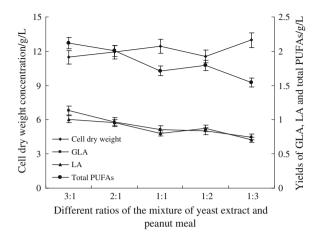
Table 24.2 Effect of nitrogen source on biomass and PUFAs production of NTG₁₋₁₂₁

			-	
Nitrogen source	Cell dry weight (g L ⁻¹)	LA (g L ⁻¹)	GLA (g L ⁻¹)	Total PUFAs (g L ⁻¹)
	(8 2)			
Yeast extract	10.03 ± 0.31 a	0.79 ± 0.06 a	$0.78 \pm 0.07 \text{ a}$	1.59 ± 0.11 a
Beef extract	$11.45 \pm 0.44 \text{ b}$	$0.78 \pm 0.13 \text{ a}$	$0.73 \pm 0.09 \text{ b}$	1.53 ± 0.07 a
Peptone	11.59 ± 0.27 b	$0.74 \pm 0.05 \text{ b}$	0.71 ± 0.11 c	1.47 ± 0.06 a
Urea	4.48 ± 0.16 c	0.20 ± 0.02 c	$0.24 \pm 0.03 \text{ d}$	$0.45 \pm 0.09 \text{ b}$
(NH ₄) ₂ SO ₄	$7.87 \pm 0.23 \text{ d}$	$0.34 \pm 0.05 \text{ d}$	0.36 ± 0.04 e	$0.71 \pm 0.12 \text{ c}$
KNO ₃	$7.81 \pm 0.26 \text{ d}$	0.38 ± 0.08 e	0.36 ± 0.06 e	$0.76 \pm 0.13 \text{ d}$
Peanut meal	13.59 ± 0.57 e	$1.10 \pm 0.13 \text{ f}$	$1.06 \pm 0.17 \text{ f}$	2.21 ± 0.18 e
Soybean meal	11.64 ± 0.49 b	0.77 ± 0.11 a	$0.74 \pm 0.08 \text{ b}$	$1.53 \pm 0.08 \text{ f}$

Data are expressed as means \pm standard deviations of triplicate assays

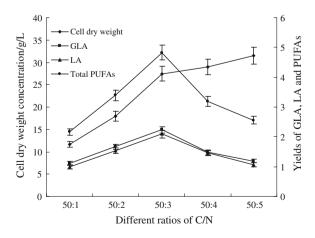
The different alphabetic superscripts in the same column are significantly different (p < 0.05) based on Duncan's multiple range test

Fig. 24.2 The effects of different ratios of the mixture of yeast extract and peanut meal on PUFAs production



As can be seen in Fig. 24.3, the cell dry weight sharply increased from the ratio of 50/1–50/3, then increased slightly with reducing the C/N ratio, and the difference among ratios of 50/3, 50/4, and 50/5 were not significant. While, the total PUFAs presented the maximum yield (up to 4.83 g/L) at the ratio of 50/3, and the yield of GLA and LA nearly had the same trend showing a peak at the ratio of 50/3. The results are consistent with the result reported by Kioke (2001). A number of researchers have found that the medium with higher carbon source and lower nitrogen source are relatively suitable for lipids accumulation [15]. Some eukaryotic oleaginous species like yeast and mold are able to accumulate a mass of oil taking up more than 70 % of the biomass [16]. And the accumulative process of acetyl CoA in eukaryotic oleaginous organisms will be activated when the assimilable nitrogen source is exhausted and the assimilable carbon source is abundant. As a result, it promotes the synthesis of PUFAs in fungus [11]. The test results obtained

Fig. 24.3 The effects of different C/N ratios on PUFAs production



D. Sun et al.

in the experiment were also corresponded to the theory, so the C/N ratio at 50/3 was selected for the following research.

24.3.3 Effects of Metal Ions on PUFAs Production

The effects of some metallic ions on the PUFAs production were shown in the following Table 24.3. The metal ions containing $\mathrm{Mn^{2+}}$, $\mathrm{Fe^{2+}}$, $\mathrm{Cu^{2+}}$, and $\mathrm{Zn^{2+}}$ had strong inhibitory effect (data not shown) on the cell growth and PUFAs production. As illustrated in Table 24.3, in the case of $\mathrm{Ca^{2+}}$, each investigation index increased to the maximum and then decreased with the concentration increasing (had maximal values at 50 mg/L). Similarly, $\mathrm{Mg^{2+}}$ in the medium appeared to be the same condition and the concentration of 100 mg/L had the maximum. Compared with the controlled test metal ion, a certain amount of $\mathrm{Ca^{2+}}$ and $\mathrm{Mg^{2+}}$ were capable of promoting fungi growth and lipid synthesis. In order to improve one or more PUFAs yields, many experts examined the effects of metallic ions on PUFAs production and gained some achievements [17, 18]. The growth and lipid production of the *M. isabellina* $\mathrm{NTG_{1-121}}$ were also strongly influenced by metal ions (50 mg/L $\mathrm{Ca^{2+}}$ and 100 mg/L $\mathrm{Mg^{2+}}$).

24.3.4 Production of PUFAs Under the Optimal Medium

On the basis of the above results, the main culture medium composition was optimized by central composite design, and an optimal cultivation was given as follows: carbon source 116 g/L, nitrogen source 10.8 g/L, Ca²⁺ 30.8 mg/L, and Mg²⁺

Table 24.5 Elects of field followings and FeF745 production of TVFG ₁₋₁₂₁								
Metal ions	Concentration (mg L ⁻¹)	Cell dry weight (g L ⁻¹)	GLA (g L ⁻¹)	LA (g L ⁻¹)	Total PUFAs (g L ⁻¹)			
Ca ²⁺	5	$26.87 \pm 0.14 \text{ a}$	1.96 ± 0.06 a	1.90 ± 0.10 a	$4.03 \pm 0.16 \text{ a}$			
	50	$28.24 \pm 0.27 \text{ b}$	$2.35 \pm 0.11 \text{ b}$	$2.27 \pm 0.13 \text{ b}$	$4.70 \pm 0.12 \text{ b}$			
	100	27.64 ± 0.19 c	$1.93 \pm 0.09 \text{ a}$	1.89 ± 0.07 a	$3.94 \pm 0.32 \text{ a}$			
Control	0	$26.06 \pm 0.17 d$	$2.24 \pm 0.12 \text{ b}$	2.21 ± 0.11 b	4.51 ± 0.24 b			
	50	$26.42 \pm 0.26 \text{ a}$	$1.74 \pm 0.15 \text{ a}$	$1.69 \pm 0.09 \text{ a}$	$3.5 \pm 0.19 \text{ a}$			
Mg ²⁺	100	$27.70 \pm 0.31 \text{ b}$	$2.48 \pm 0.20 \text{ b}$	$2.240 \pm 0.17 \text{ b}$	$4.80 \pm 0.30 \text{ b}$			
	500	25.43 ± 0.21 c	$1.87 \pm 0.08 \text{ c}$	1.71 ± 0.14 a	$3.64 \pm 0.18 \text{ a}$			
Control	0	26.06 ± 0.17 a	$2.24 \pm 0.12 \text{ b}$	2.21 ± 0.11 b	$4.51 \pm 0.24 \text{ b}$			

Table 24.3 Effects of metal ions on biomass and PUFAs production of NTG₁₋₁₂₁

Data are expressed as means \pm standard deviations of triplicate assays

The different alphabetic superscripts in the same column are significantly different (p < 0.05) based on Duncan's multiple range test

96.8 mg/L. Meanwhile, the fermentation conditions were optimized and the results will be discussed in another article.

In conclusion, the optimum medium composition for PUFAs production with the mutant NTG₁₋₁₂₁ of *M. isabellina* AS3.3410 were 2 g/L of KH₂PO₄ and sodium citrate, 116 g/L of glucose as carbon source, 10.8 g/L of the mixture of yeast extract and peanut meal at 3:1 (w/w) as nitrogen source at C/N ratio of 50/3, supplements with Ca²⁺ 30.8 mg/L and Mg²⁺ 96.8 mg/L, respectively. Under the optimum medium condition, the yields of cell dry weight, GLA, LA, and total PUFAs were 32.48, 2.90, 2.83 and 7.36 g/L, respectively. And the changes of PUFAs yields were almost by improving the biomass, though the factors had some effect on the metabolism.

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228 D. Sun et al.

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Chapter 25 A Simple Mechanochemical Cycle Model for Dynein

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Abstract Dynein is the largest and most morphologically complex of biological molecular motors. By analyzing and summarizing the structure and experimental parameters, we develop a mechanochemical cycle model for dynein hydrolyzing ATP. Based on the model, we discuss the relationship between the load and the step size of the dynein, the result derived fit well with the experimental results.

Keywords Dynein • Mechanochemical cycle • Load • Power stroke • Step size

25.1 Introduction

Dynein is a kind of motor protein that uses the energy from an ATP hydrolysis to make work [1, 2]. Dyneins are subdivided into two major groups: axonemal and cytoplasmic dyneins [3]. The cytoplasmic dynein is responsible for intracellular transport including nuclear migration, organization of the mitotic spindle, chromosome separation during mitosis, and the positioning and function of many intracellular organelles [4]. The axonemal dyneins are anchored in linear arrays along microtubules inside cilia and flagella. These dyneins coordinate their motion to cause the rhythmic beating of the cilia and flagella [5, 6].

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²²⁹

Every axonemal dynein has 1–3 heavy chains, where each heavy chain consists of three major parts: the AAA ring head, the stalk, and the tail. The head serving as the catalytic engine of the motor contains six AAA modules (AAA1–AAA6) tandemly arranged in a ring, which is structurally related to hexameric ATPases in the AAA + superfamily. The first four AAA modules (AAA1–4) contain nucleotide-binding and hydrolysis Walker A motifs. The stalk emerging from between AAA4 and AAA5 is a ~15 nm anti-paralled coiled-coil. A globular microtubule-binding domain is located at its tip [7]. Linker, which has been proposed to serve as a mechanical element, is a structure connecting AAA1 and the main part of the tail.

In axonemes, the tail of dynein is fixed on the A-tubule of a doublet microtubule; the stalk interacts cyclically with the B-tubule of a neighboring doublet microtubule [8]. In the mechanochemical cycle of dynein hydrolyzing ATP, the hydrolyzing ATP in the head causes conformational change of the head; the change is transmitted through the stalk and the tail to the neighboring doublet microtubules leading to a translocation of the adjacent doublet. The mechanism of the relative motion is not clear.

The electron micrographs of a single dynein molecular show that in different biochemical states dynein has distinct conformations. In the absence of nucleotide (apo-dynein) the stem and stalk are closer together than in the ADP·Pi molecule state. In either state, the stalk is always perpendicular to the head [9]. Between the tail and the head is the linker docked onto the head, which swings relative to the head ring to make the motion between the tail and the head [9, 10].

Some models for dynein assumed that the stalk acted as a lever arm, in apo-nucleotide state the stalk bind to B-tubule and tilted toward the minus end of the microtubule, and then rotated toward the plus end of the microtubule to make the relative motion of the adjacent doublet. However, these models had some contradictions with recent experiments [9–16]. (1) By analyzing the electron micrographs Ueno et al. [11] found that the stalk angles relative to microtubules did not change significantly either in the ADP·Pi or the apo-nucleotide states. The stalk always tilted towards the plus end of the microtubule. They thought that dynein pulled microtubules without rotating its stalk; the stalk was used as a tether rather than as a lever arm, and the dynein moved by shortening the distance between its head/stalk and the tail-micrtubule attachment. (2) By calculating the stiffness of the tail and stalk, Linderman et al. [1] found that if the head was supported only by the tail, the stalk and the tail were in series in transmitting the force generated by dynein to the microtubule. The maximum stiffness of the stalk was 0.47 and 0.1 pN/nm for the tail, the flexibility of the tail limited the force that can be transferred by the entire dynein heavy chain to 0.6 pN. The force was much smaller than experimental value 5 pN. And in the following experiment, dynein could generate force even without the tail. This implies that the head and the stalk are the functional bridge for force production, and that there is a second anchoring mechanism to hold the head. It is that the head anchors on the A-tubule. (3) Kon et al. [13] found in experiment that the tail had undergone twice ATP-induced motions relative to the head domain during the mechanochemical cycle of dynein. When ATP binds to the head, the tail rotates relative to the head immediately. The release of the hydrolysis products induced the second rotation of the tail relative to the head; this rotation corresponded to the power stroke.

Based on the researches above, we proposed a simple mechanochemical cycle for dynein hydrolyzing ATP. Based on the model, we discussed the relationship between the load and the step size of the dynein; the result we derived fits well with the experimental results.

25.2 The Mechanochemical Cycle for Dynein Hydrolyzing ATP

Based on the researches above we developed a simple mechanochemical cycle for a dynein single head hydrolyzing ATP, as shown in Fig. 25.1

In the model, in the dynein exist two states while binding to B-tubule and a state while dissociating from the B-tubule, as shown in Fig. 25.1. The transition between these states includes three steps: (1), (2), (3). (1) ATP binding to AAA1 of the head causes the stalk to dissociate from the B-tubule, the head moves laterally relative to the microtubule. ATP bound to the dynein forming dynein ATP complex. (2) ATP hydrolysis to ADP and Pi. The hydrolysis products bind to dynein. In this process the conformational of the dynein changes, dynein binds tightly to B-tubule again. (3) ADP and Pi are released and the rotation of the linker relative to the head makes

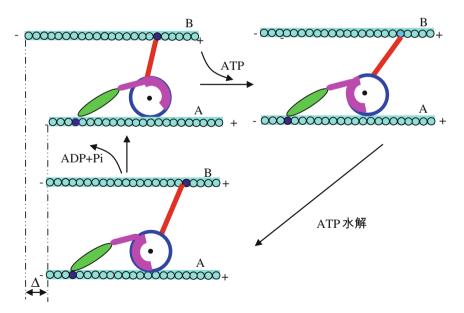


Fig. 25.1 The mechanochemical cycle for dynein hydrolyzing ATP. The *blue bars* represent the doublet microtubules. + and - represent the minus and plus end of the microtubule respectively

X. Zhao et al.

the tail rotate relative to the head and the stalk. The rotation leads a relative displacement Δ , which is regarded as the value of the power stroke.

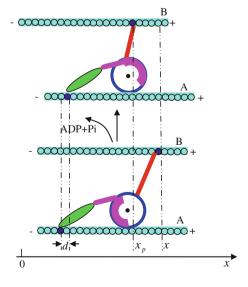
Considering the size of the dynein, the stalk is about 25 nm long, the diameter of the head is about 13 nm, the tail is about 10 nm long, and the linker is about 10 nm long. The overall length of the dynein is 45–50 nm, and the distance between the A-tubule and B-tubule of a relaxed axoneme is 20–25 nm [9]. The cross-bridge formed by dynein in the inner of the neighbor doublet microtubule should be like the illustration in Fig. 25.1.

In the model, the power stroke is caused by the relative rotation between the tail and the head, this fit well with the experimental results of Hironori Ueno et al. When ATP binds to the head, the tail rotates relative to the head, and when the hydrolysis products release, the tail rotates again relative to the head; this fit well with the experimental researches of Takahide Kon et al.

25.3 The Step Size of the Dynein Responds to Load

Assume that in ADP·Pi state, the position of the binding site for dynein to the microtubule is x, in apo-state the position is x_p , $x - x_p$ is the value of the step size. But in this process there exist a displacement between the doublet microtubules, and the value of the displacement is d. So dynein moves $x - x_p - d$ along the microtubule in completing one power stroke. The relative position of the dynein and the microtubules is shown in Fig. 25.2.

Fig. 25.2 The schematic illustration for power stroke of the dynein



So, the length change between the ADP·Pi state and apo-state is $k(x - x_p - d)$, k is proportional constant. The energy stored in dynein is

$$E = \frac{1}{2} K_{ps} \left[k \left(x - x_p - d \right) \right]^2$$
 (25.1)

 $K_{\rm ps}$ is the elastic coefficient of the dynein. On the load $F_{\rm ext}$, the work did by dynein is

$$E = \frac{1}{2} K_{ps} [k(x - x_p - d)]^2 + F_{ext} x$$
 (25.2)

When the energy is smallest, x is also the smallest. Deformation Eq. (25.2),

$$E = \frac{1}{2}K_{ps}k^2x^2 - \left(K_{ps}k^2x_p + K_{ps}k^2d - F_{ext}\right)x + \left(\frac{1}{2}K_{ps}k^2x_p^2 + K_{ps}k^2x_pd\right)$$
(25.3)

When the energy is the smallest,

$$x_{\min} = \frac{K_{\rm ps}k^2x_{\rm p} + K_{\rm ps}k^2d - F_{\rm ext}}{K_{\rm ps}k^2}$$
 (25.4)

So in this situation the average value of the step size of dynein is

$$x_{\min} - x_{\rm p} = d - \frac{F_{\rm ext}}{K_{\rm ps}k^2}$$
 (25.5)

When $F_{\text{ext}} = 0$, the step size of the dynein is 32 nm, the value of the d is 32 nm. The elastic coefficient, K_{ps} , is not a constant. According to the WLC model for wormlike polymer, the elastic force between two heads of dynein can be expressed as

$$F = \frac{K_{\rm B}T}{A} \left[\frac{1}{4} \left(1 - \frac{z}{L} \right)^{-2} - \frac{1}{4} + \frac{z}{L} \right]$$
 (25.6)

where F is the elastic force, $K_{\rm B}$ is the Boltzman's constant, T is absolute temperature (300 K in our model), A is the persistence length, L is the chain's contour length, and z is the elongation value. Note that $\frac{z}{L}$ denotes the fractional extension of the chain.

From Eq. (25.6), we can get

$$K_{\rm ps} = \frac{dF}{dz} = \frac{K_{\rm B}T}{AL} \left[1 + \frac{1}{2} \left(1 - \frac{z}{L} \right) - 3 \right]$$
 (25.7)

X. Zhao et al.

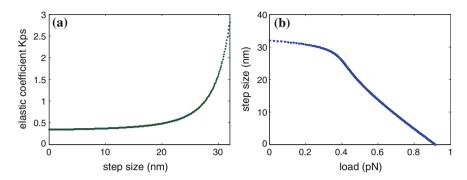


Fig. 25.3 a The change in K_{ps} with the step size. **b** The step size of dynein calculated using Eq. (25.11)

When z = 0, the value of K_{ps} is about 0.48 pN/nm, therefore,

$$K_{\rm ps} = 0.32 \left[1 + 0.5 \left(1 - \frac{z}{L} \right)^{-3} \right]$$
 (25.8)

Because z is equal to

$$z = x_{\min} - x_{\mathrm{p}} - L \tag{25.9}$$

Then Eq. (25.7) can be written as

$$K_{\rm ps} = 0.32 \left[1 + 0.5 \left(2 - \frac{x_{\rm min} - x_p}{L} \right)^{-3} \right]$$
 (25.10)

Substituting Eq. (25.10) into Eq. (25.5), we can derive the step size-force relation, shown as

$$F_{\text{ext}} = 0.32k^2 \left[d - \left(x_{\text{min}} - x_{\text{p}} \right) \right] \cdot \left[1 + 0.5 \left(2 - \frac{x_{\text{min}} - x_{\text{p}}}{L} \right)^{-3} \right]$$
 (25.11)

in which the term $x_{\min} - x_p$ represents the step size of dynein. Based on the literature [9], the value of 20 nm is used for L. Since the step size is 32 nm at $F_{\rm ext} = 0$, the value of d is 32 nm too. Taking k = 0.29, we obtained the change in $K_{\rm ps}$ with step size and the relationship between the step size and load, see Fig. 25.3

25.4 Discussion

Based on the structure and the kinetic theory of dynein, we proposed a simple mechanochemical cycle for dynein hydrolyzing ATP. Based on the model, we calculate the step size of the dynein response to load. The relationship between the step size and the load fits well with the experimental results; this implies that our model is credible.

In the calculation, we note that the relationship between the external load and the step size is related to the elastic coefficient of the dynein; we considered the nonlinear effect of the dynein on the step size, this point was always ignored in the previous model. From Fig. 25.3a we note that the larger the step size the bigger the value of $K_{\rm ps}$. Thus, the nonlinear effect of the dynein cannot be ignored when the step size is bigger.

But in our model we assume the proportional constant between the length of the entire dynein and the step size of the dynein. This factor has not been justified by experiments and needs to be tested in the future.

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Chapter 26 The Research on Biotransformation Pathway of Digitoxin by Aspergillus ochraceus and the Analysis of Products Activity

Jianmei Luo, Ting Song, Fangfang Cui, Yanbing Shen and Min Wang

Abstract In order to analyze the biotransformation pathway of digitoxin by *Aspergillus ochraceus* Rn405, a product—digitoxigenin was used as the substrate to biotransform. The substrate consumption and product synthesis were monitored by HPLC during the bioconversion process. The preliminary results were as follows: Firstly, hydrolysis reaction happened and digitoxin was converted into digitoxigenin; then, with extension of the reaction time, the hydroxyl group was introduced to the C-11 of digitoxigenin by the role of hydroxylase and another product named 11α -hydroxydigitoxigenin was generated in a large amount. The influences of these two products on myocardial apoptosis and their antineoplastic activities were further analyzed. It was found that compared with the substrate digitoxin and reference substance digoxin, the apoptosis effects of digitoxigenin and 11α -hydroxydigitoxigenin on rat myocardial cells remarkably decreased; but no obvious antineoplastic activities on the four tumor cells, including human breast cancer cell MCF-7, rat breast cancer cell 4T1, human cervical cancer cell Hela, and rat melanoma cell B-16 were observed.

Keywords *Aspergillus ochraceus* • Digitoxin • Bioconversion • Biotransformation pathway • Activity analysis

Digitaloid drug is an exogenous cardiac glycoside compound, which has a good effect on the treatment of congestive heart failure diseases [1, 2] and antitumor activities [3–6]. However, its clinical application is seriously limited by defects like slow metabolism and therapeutic dose approximates to toxic dose. In order to solve these problems, lead compounds with new structures would be developed continuously,

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and the structure of the existing cardiac glycoside drugs would be further modified. Cardiac glycosides have quite complicated structures and contain multiple chiral centers. When chemical approach is used for structural modification, poor selectivity, side reactions like oxidization, cyclization, and polymerization, and the complex steps are easy to happen. Besides, it is hard to reform the structure of some special loci [7]. Bioconversion is another effective way to modify and reform the structure of native compound. It is equipped with advantages of strong stereoselectivity, mild reaction condition, few by-products, low cost, and environmentally friendly nature. What is more, this method can conduct chemical reactions that traditional chemical synthesis cannot or can hardly achieve. At present, bioconversion has become an important method to gain a large number of new-type lead compounds [8].

In our previous research work, two converted products (digitoxigenin and 11α -hydroxydigitoxigenin) have been separated, purified, and identified from the bioconversion liquid of digitoxin by *Aspergillus ochraceus* [9]. In order to discuss the biotransformation pathway, the bioconversion reaction was conducted by using digitoxigenin as the substrate. Meanwhile, the influences of converted products on rat myocardial apoptosis ratio and their antineoplastic activities were preliminarily studied. This work would lay a foundation for the development of lead compounds of efficient antineoplastic drugs with low toxicity.

26.1 Materials and Methods

26.1.1 Experimental Materials

26.1.1.1 Experimental Strain

Aspergillus ochraceus Rn405 was kept by Microbial Pharmaceutical Laboratory, College of Biotechnology, Tianjin University of Science & Technology.

26.1.1.2 Culture Medium

The strain was kept with potato juice slant medium. The culture medium consisted of 20 g/L glucose, 20 g/L yeast cream, and 20 g/L peptone; pH was adjusted to 4.5 before autoclaving. 50 mL culture medium was loaded in 250-mL shake flasks and sterilized at 115 °C for 20 min.

26.1.1.3 Cell Lines

Human breast cancer cell MCF-7, rat breast cancer cell 4T1, human cervical cancer cell Hela, and rat melanoma cell B-16 were kept by Tianjin International Joint Academy of Biomedicine. Rat myocardial cell H9c2 was kept by College of Biotechnology, Tianjin University of Science & Technology.

26.1.1.4 Main Reagents and Instruments

Digitoxin (99.0 % purity, Dr Company, Germany); digoxin (99.0 % purity, National Institutes for Food and Drug Control); acetonitrile (chromatographical purity, Real & Lead Chemical Co., Ltd., China); other reagents (analytical purity, Real & Lead Chemical Co., Ltd., China); ESI-MS mass spectrometer (Bruker Daltonics Inc., USA); AV400 nuclear magnetic resonance analysis meter (Bruker Optics, USA); the preparative liquid phase analysis meter (Agilent Technologies, USA).

26.1.2 Experimental Methods

26.1.2.1 Strain Culture

The strain was inoculated on the slant medium and cultured under 28 °C for 5–7 days. When the slant bottom turned into maroon and a large number of golden yellow spores appeared on the surface, it could be applied to bioconversion reaction.

26.1.2.2 Bioconversion Process

The spore suspension of *A. ochraceus* Rn405 was inoculated into 250-mL shake flask containing 50 mL culture medium to make its final concentration of 10⁶ cells/mL. After cultivation at 28 °C for 28 h on a rotary shaker (180 r/min), the digitoxin, which was previously dissolved in DMF (0.1 mL) was added into each flask at the final concentration of 200 mg/L. The bioconversion was carried out under the same cultivation conditions for 96 h.

26.1.2.3 Separation and Purification of Products

The culture supernatant and mycelium were separated by filtration with suction. The filter liquor was extracted with ethyl acetate in 3 volumes for 3 times. The organic phase was collected and further concentrated by the rotary evaporator. Then silica gel column chromatography was used for separation and purification according to the literature [9].

26.1.2.4 Analysis of Bioconversion Pathway

Digitoxigenin (compound 2) was used as the substrate for bioconversion according to the bioconversion conditions described in Sect. 26.1.2.2. The samples were taken every 24 h and the substrate conversion ratio and product formation ratio were

J. Luo et al.

monitored by HPLC. One flask culture of organism without exogenous substrate (for the identification of endogenous metabolites) was also performed as a control. HPLC analysis conditions were as follows: A Kromasil C18 (4.6 mm \times 250 mm, 5 μ m particle size) was used and acetonitrile—water (40:60, v/v) was employed as the mobile phase at 1.0 mL/min. The system was operated at 30 °C and the detection wavelength was set as 220 nm; the injection volume was 10 μ L.

26.1.2.5 Analysis of Pharmacological Activities of Products

Influences of Converted Products on Myocardial Apoptosis

Rat myocardial cell H9c2 was cultivated on the 16-well microplate and highglucose DMEM culture solution was added. After incubation on 5 % CO2 incubator at 37 °C for 24 h, a certain amount of digitoxin and various converted products were, respectively, dissolved in DMSO to prepare the mother liquor. Six concentration gradients of 10^{-9} , 10^{-8} , 10^{-7} , 10^{-6} , 10^{-5} , and 10^{-4} mol/L were set for each compound, and three parallel samples were used for each concentration. After the induction of 24 h, 4 % paraformaldehyde was used to fix the cell at 4 °C for 20 min. Then PBS buffer solution was applied to wash for 2 times (3 min for each time). After washing with PBS buffer solution by shaking hands for several times, the cells were dyed with Hoechst 33258 staining solution at a final concentration of 10 µg/mL on incubator at 37 °C for 30 min and then washed with PBS buffer solution for 2 times (3 min for each time). Finally, 200 cells were randomly selected for observing under fluorescence microscope and 340 nm exciting light conditions. The apoptosis ratio was calculated by the percentage of apoptotic cells in total cells. During the experiment, one group only adding DMSO was set as the blank control, and staurosporine with a final concentration of 5 µmol/L was used as positive control (the induction time of positive control group was 40 min).

Antineoplastic Activities of Converted Products

The in vitro inhibitory activities of digitoxin and its converted products on human breast cancer cell MCF-7, rat breast cancer cell 4T1, human cervical cancer cell Hela, and rat melanoma cell B-16 were detected via MTT method.

(1) Tumor cell culture

The tumor cell freezing tube was taken out from the liquid nitrogen container, and put at 37 °C on water bath until the cell suspension in freezing tube was thawed. The cell suspension was put into 25-mL culture flask and 5 mL of DMEM culture medium with 10 % serum was added. It was cultivated on 5 % $\rm CO_2$ incubator at 37 °C until it filled the culture flask.

(2) Cytotoxicity screening

The cultivated tumor cells were spread out onto a 96-well microplate at the concentration of 1×10^6 CFU and $100~\mu L$ of diluted compound solution was added. At the same time, $100~\mu L$ of blood serum medium was added into the well as a blank control group. After cultivation on 5 % CO $_2$ incubator at 37 °C for 48 h, MTT solution was added. After incubation at 37 °C for 2 h, the purple crystal appeared and the supernatant was then taken out. DMSO was added to dissolve purple crystal and the microplate was vibrated at room temperature for 15 min. OD value was detected at the wavelength of 570 nm with microplate reader.

The tumor cell suppression ratio was calculated according to the following formula:

$$\begin{split} & \text{Suppression ratio (\%)} \\ &= \frac{\text{Average OD value of blank control group} - \text{Average of OD value of test group}}{\text{Average OD value of blank control group}} \times 100~\% \end{split}$$

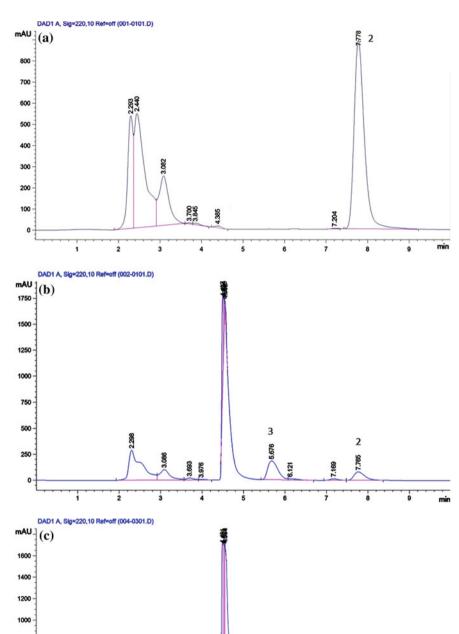
26.2 Results and Analysis

26.2.1 Analysis of Biotransformation Pathway

According to the structures of two products (Fig. 26.2), 11α -hydroxydigitoxigenin (compound 3) had a hydroxyl group at the C-11 when compared with digitoxigenin. Therefore, it was speculated that during the bioconversion of digitoxin by *A. ochraceus*, hydrolysis reaction firstly occurred and the substrate digitoxin (compound 1) was converted into digitoxigenin (compound 2). After that, with extension of the reaction time, the hydroxyl group was introduced to the C-11 of digitoxigenin by the role of hydroxylase. Another product named 11α -hydroxydigitoxigenin (compound 3) was then generated in a large amount. In order to verify the above conjecture, the product named digitoxigenin (compound 2) which was obtained by separation and purification from the broth of *A. ochraceus* Rn405 was used as substrate to carry out bioconversion. The conversion process was monitored by HPLC.

According to Fig. 26.1, when digitoxigenin (compound 2) was set as the substrate, a retention peak 3 with the retention time of 5.676 min appeared in the reaction liquid after conversion for 48 h. It was almost the same with the retention time of the standard substance of 11α -hydroxydigitoxigenin [9]. It was speculated that 11α -hydroxydigitoxigenin was generated during the bioconversion process. Thereby, it was preliminarily suggested that the biotransformation pathway on bioconversion of digitoxin by *A. ochraceus* 405 was presented in Fig. 26.2.

242 J. Luo et al.



◄Fig. 26.1 HPLC diagram of bioconversion of digitoxigenin (compound **2**) by *Aspergillus ochraceus* 405, **a** conversion for 0 h; **b** conversion for 48 h; **c** conversion for 96 h, *I* digitoxin (compound **1**); 2 digitoxigenin (compound **2**); 3 11α-hydroxydigitoxigenin (compound **3**)

$$\begin{array}{c} CH_3 \\ CH_3 \\ OH \\ 1 \end{array}$$

Fig. 26.2 Reaction formula of bioconversion of digitoxin (compound 1) by A. ochraceus 405, I digitoxin (compound 1); 2 digitoxigenin (compound 2); $3 \, 11\alpha$ -hydroxydigitoxigenin (compound 3)

26.2.2 Influences of Digitoxin and Relevant Products on Cell Apoptosis

Digitoxin (compound 1), digitoxigenin (compound 2), 11α-hydroxydigitoxigenin (compound 3), the reference substance digoxin, and the positive control group of staurosporine were used as reaction medicines to induce apoptosis of rat myocardial cell H9c2, respectively. As shown in Fig. 26.3, similar impacts on myocardial apoptosis ratio were observed among the reference substance digoxin, digitoxin (compound 1), digitoxigenin (compound 2), and 11α-hydroxydigitoxigenin (compound 3). The apoptosis ratio of rat myocardial cell H9c2 declined with the decreasing concentration of the four compounds from 10⁻⁴ to 10⁻⁹ mol/L. When the concentration of the two converted products decreased from 10^{-4} to 10^{-5} mol/L, the apoptosis ratio of myocardial cell H9c2 fluctuated between 99 and 100 %. When the concentration of the two converted products decreased to 10⁻⁶ mol/L, the apoptosis ratio declined slightly and reached 98.7 and 96.3 %, respectively. The concentration of the two converted products further reduced from 10^{-7} to 10^{-9} mol/L, the apoptosis ratio reduced from 33.8 and 29.1 % to 13.4 and 11.5 %, respectively, which was obviously lower than those of digitoxin (compound 1) and the reference substance digoxin (Fig. 26.3). It was preliminarily concluded that the removal of carbohydrate chain from digitoxin (compound 1) or the insertion of the hydroxyl group at C-11 position would weaken the apoptosis ratio of rat myocardial cell H9c2.

J. Luo et al.

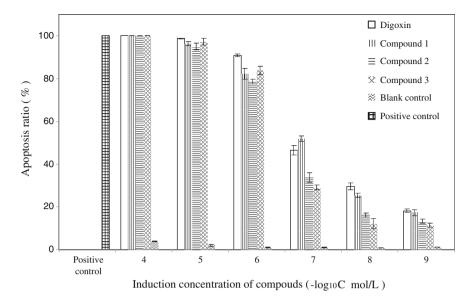


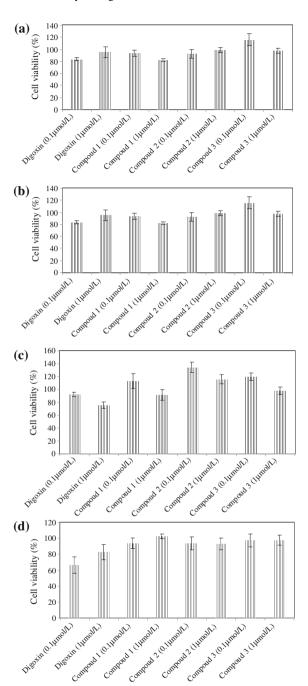
Fig. 26.3 Influences of four compounds on apoptosis ratio of rat myocardial cell H9c2. Digoxin was the control group of pharmacological activity; compound 1, 2, and 3 were test groups of digitoxin, digitoxigenin, and 11α -hydroxydigitoxigenin, respectively; the group only with DMSO was set as the blank control; staurosporine at the final concentration of 5 μ mol/L was the positive control

26.2.3 Suppression Effects of Digitoxin and Relevant Products on Tumor Cells

The suppression effects of digitoxin (compound 1), digitoxigenin (compound 2), 11α -hydroxydigitoxigenin (compound 3) and the reference substance digoxin on human breast cancer cell MCF-7, rat breast cancer cell 4T1, human cervical cancer cell Hela, and rat melanoma cell B-16 were analyzed by MTT experiment.

As shown in Fig. 26.4, a certain suppression effect of the reference substance digoxin on all the four tumor cells were detected. Digitoxin (compound 1) at 0.1 μ mol/L promoted the proliferation of rat breast cancer cell 4T1 and human cervical cancer cell Hela while no suppression effects on human breast cancer cell MCF-7 and rat melanoma cell B-16 were observed. When the concentration of Digitoxin (compound 1) increased to 1 μ mol/L, no suppression effects on the four tumor cells were observed. Almost no suppression effects of Digitoxigenin (compound 2) at the concentration of 0.1 and 1 μ mol/L on human breast cancer cell MCF-7 and rat melanoma cell B-16 were detected but the facilitation effects on the proliferation of rat breast cancer cell 4T1 and human cervical cancer cell Hela were observed. The proliferative effects of 11α -hydroxydigitoxigenin (compound 3) at the concentration of 0.1 μ mol/L on human breast cancer cell MCF-7, rat breast cancer cell 4T1, and human cervical cancer cell Hela were observed while neither

Fig. 26.4 Inhibitory activities of four compounds on tumor cells. Digoxin was the control group of pharmacological activity; compound 1, 2, and 3 were test groups of digitoxin, digitoxigenin, and 11α-hydroxydigitoxigenin, respectively. a Human breast cancer cell MCF-7; b rat breast cancer cell 4T1; c human cervical cancer cell Hela; d rat melanoma cell B-16



J. Luo et al.

suppression effects nor proliferative effects on rat melanoma cell B-16 was presented. When the concentration of 11α -hydroxydigitoxigenin (compound 3) increased to 1 µmol/L, neither suppression effects nor proliferative effects on the four tumor cells were observed. In conclusion, no inhibitory activities of digitoxigenin (compound 2) and 11α -hydroxydigitoxigenin (compound 3) at the concentration of 0.1 and 1 µmol/L on all the four tumor cells were observed.

26.3 Conclusion

- (1) The biotransformation pathway of digitoxin by *A. ochraceus* 405 was preliminarily speculated by the bioconversion experiment using one product named digitoxigenin (compound 2) as the substrate. It was suggested that hydrolysis reaction happened first and the substrate digitoxin (compound 1) was converted into digitoxigenin (compound 2); After that, with extension of the reaction time, the hydroxyl group was introduced to the C-11 of digitoxigenin by the role of hydroxylase, and another product named 11α -hydroxydigitoxigenin (compound 3) was generated in a large amount.
- (2) The influences of digitoxigenin (compound 2) and 11α -hydroxydigitoxigenin (compound 3) on apoptosis ratio of myocardial cells and their antineoplastic activities were analyzed with Hoechst dyeing and MTT methods, respectively. The results showed that when concentration of digitoxigenin (compound 2) was ranged from 10^{-7} to 10^{-9} mol/L, the apoptosis ratio of rat myocardial cells was obviously lower than those of digitoxin (compound 1) and the reference substance digoxin while no suppression effects on human breast cancer cell MCF-7, rat breast cancer cell 4T1, human cervical cancer cell Hela, and rat melanoma cell B-16 were observed at the concentration of 0.1 and 1 μmol/L. When the concentration of 11αhydroxydigitoxigenin (compound 3) was ranged from 10⁻⁷ to 10⁻⁹ mol/L, the apoptosis ratio of rat myocardial cells was obviously lower than those of the reference substance digoxin and digitoxin (compound 1) while no suppression effects on the above four tumor cells were presented at the concentration of 0.1 and 1 µmol/L. In conclusion, the apoptosis ratio of rat myocardial cells could be reduced by the hydrolysis of digitoxin carbohydrate chain, but no suppression effects on the above four tumor cells were observed. The apoptosis ratio of rat myocardial cells could be decreased by C-11 hydroxylation, but no suppression effects on the above four tumor cells were detected.

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Chapter 27

Astragalus membranaceus Polysaccharide-Enhanced Lymphocytes Proliferation of Yellow Drum Nibea albiflora In Vitro

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Abstract To quickly evaluate the immunostimulatory effect of *Astragalus membranaceus* polysaccharide (AMP) on yellow drum *Nibea albiflora*, the head kidney lymphocytes proliferation was tested with/without concanavalin A (Con A) or lipopolysaccharide (LPS) in vitro. The lymphocytes were incubated within 0, 25, 50, 100, 200, 400, and 800 µg/mL AMP for 48 h, with/without Con A (2.3 µg/mL) or LPS (18.2 µg/mL). Results showed that there was a dose-dependent relationship between lymphocytes proliferation and the AMP concentration (P < 0.05). The AMP-enhanced lymphocytes proliferation at 50–200 µg/mL significantly (P < 0.05), with peak value at 100 µg/mL. Low dose (0–25 µg/mL) and high dose (400–800 µg/mL) of AMP did not stimulate lymphocytes proliferation significantly (P < 0.05). Synergetic effects were observed between AMP and Con A/LPS on lymphocytes proliferation. Data in this chapter implied that the AMP might contribute to both cellular and humoral immunity of N. *albiflora* in a dose-dependent manner.

Keywords Astragalus membranaceus \cdot Polysaccharide \cdot Lymphocytes proliferation \cdot Nibea albiflora

27.1 Introduction

The yellow drum *Nibea albiflora*, which belongs to Perciformes, Sciaenidae, and Nibea, is naturally distributed in the coastal waters of China, Korea, and south Japan. Yellow drum is a well-known fishing species in China and loved for its

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delicious taste and trophic value. In the recent decades, because of overfishing and environmental deterioration, the wild population of yellow drum declined rapidly. The culture of yellow drum developed rapidly these years [1]. With the increase of culture scale and density, yellow drum diseases have become more and more serious [2].

More and more antibiotics and chemotherapeutics that previously widely used to control fish diseases have been banned because of their unwanted residues in fish and environmental pollution. A large number of studies showed that plant extracts are a promising alternative to chemotherapy in fish aquaculture [3]. The *Radix Astragali*, which is the dry root of *Astragalus membranaceus* var. *mongholicus* and *A. membranaceus* var. *membranaeceus*, is a commonly used Chinese herbal medicine for its tonic effects [4]. The *A. membranaceus* polysaccharide (AMP), which is extracted from the root of *A. membranaceus*, showed multiple pharmacological effects on livestock and poultry, such as antivirus, antiradiation, antistress, antioxidation, and immunostimulatory activity [5, 6]. The AMP also showed promising application prospects in aquaculture, such as its immunostimulation [7], growth promotion [8], antivirus [9], and hepatocyte protection [10] efficacy.

To date, the main means to study the immunomodulatory mechanism of immunostimulants on fish is via dietary administration, which needs long-term culture period. The in vitro test means, which have the advantages of easy controllability, well repeatability, less material requirement, and short time consumption, have been widely employed in the research and development of veterinary and human immunostimulants. But in vitro test means were rarely employed in fish species. This paper evaluated the immunostimulatory effect of AMP on *N. albiflora* by means of testing its lymphocytes proliferation capacity in vitro, in order to provide a rapid reference that whether the AMP can be used as an immunostimulant for *N. albiflora* healthy culture.

27.2 Materials and Methods

27.2.1 AMP Purification

The crude AMP (purchased from Shengwang Pharmacy Company in Shandong province) was dissolved in distilled water. Then ethanol was added to the supernatant to reach 80 % content, in order to precipitate crude AMP. The protein in crude AMP was moved out by Sevag method [11]. Then the AMP was obtained by lyophilization. Total sugar, reducing sugar, uronic acid, and protein in the resulting AMP were assayed. Total sugar was measured by phenol-sulfuric acid method [12] with glucose as standard. Reducing sugar was measured as described by Wen et al. [13]. Uronic acid was measured as described by Meseguer [14] and Yu et al. [15]. Protein was measured using Bradford [16] method, using bovine serum albumin as the standard.

27.2.2 Lymphocytes Preparation

The head kidney lymphocytes from healthy *N. albiflora* (80 ± 5 g) were separated according to Wang et al. [17].

27.2.3 Lymphocytes Proliferation

The RPMI-1640 proliferation medium was prepared as following: RPMI-1640 culture medium contained 10 % heat inactive (56 °C water bath for 30 min) FBS, 2 % heat inactive (45 °C water bath for 30 min) *N. albiflora* serum, 100 IU/mL ampicillin, and 0.1 mg/mL streptomycin sulfate. The AMP was dissolved in RPMI-1640 culture medium at 0, 50, 100, 200, 400, 800, and 1,600 μ g/mL, and filtered (0.22 μ m) for later use.

Lymphocytes proliferation was assayed according to the method described by Wang et al. [17] with some modifications. A 90 µL aliquot of proliferation medium containing 7×10^5 cells was added to wells of a 96-well flat bottom plate. Then the AMP was added as one of the following: (1) 100 µL AMP solution and 10 µL of proliferation medium (four wells each fish); (2) 100 µL AMP solution and 10 µL lipopolysaccharide (LPS, Sigma, 200 μg/mL, four wells each fish); and (3) 100 μL AMP solution and 10 μL concanavalin A (Con A, Sigma, 25 μg/mL, four wells each fish). The final AMP concentration in 96-well plate was 0, 25, 50, 100, 200, 400, and 800 μg/mL, respectively. The 96-well plate was cultured in humidified 5 % CO₂ incubator at 28 °C for 48 h. Then 20 μL of 3-[4,5-dimethylthiazol-2-yl]-2,5-diphenyl tetrazolium bromide (MTT, 1 mg/mL HBSS) was added to each well and incubated for additional 4 h at 28 °C. The plate was centrifuged at 500×g for 10 min at 4 °C. The supernatant was discarded, the formazan crystals in each well were dissolved by adding 200 μL DMSO and 25 μL of glycine buffer (0.1 M glycine, 0.1 M NaCl, pH 10.5). The contents of the wells were then thoroughly mixed, and the O.D. at 550 nm of the resulting suspension was measured using a microplate reader (Varioskan Flash) 10 min later. The stimulation index (SI) was calculated as OD_{550 (AMP added group)}/OD_{550 (control group)}.

27.2.4 Statistical Analysis

Data were analyzed by one-way analysis of variance (ANOVA), and any significant difference was determined at the P < 0.05 level by Duncan's multiple range tests. The analyses were carried out with the SPSS 18.0 software. Data were presented as means \pm standard error, n = 6.

27.3 Results

Total sugar, reducing sugar, uronic acid, and protein in resulting AMP were 74.63 ± 1.96 , 4.94 ± 0.13 , 15.38 ± 1.25 , and 4.73 ± 0.14 %, respectively.

Data in Table 27.1 indicated that there was a dose-dependent relationship between lymphocytes proliferation and the AMP concentration (P < 0.05). A medium dose of AMP (50–200 µg/mL) stimulated lymphocytes proliferation significantly (P < 0.05), with peak O.D. value at 100 µg/mL, while low dose (0–25 µg/mL) and high dose (400–800 µg/mL) of AMP did not stimulate lymphocytes proliferation significantly (P < 0.05). Synergetic effects were observed between AMP and stimuli (Con A and LPS). The lymphocytes proliferation reached peak value in group AMP and AMP + LPS at 100 µg AMP/mL, and in group AMP + Con A at 200 µg AMP/mL. The SI also showed the same trend in each group.

27.4 Discussion

The thymus-, kidney-, spleen-, and mucosa-associated lymphoid tissues are believed to be the primary immune organs in fish. The thymus degenerates with the growth of fish. In the development of fish kidney, head kidney lost urination function and become the immune and hemopoietic organ. The head kidney develops to be the second immune organ after thymus. Fish spleen is generally smaller in size than head kidney. The mucosa-associated lymphoid tissue is not easy to separate and the immune related cells in it are small in amount. So fish head kidney becomes the most important source of immune cells for in vitro studies [18].

Table 27.1	The effect of	AMP on	lymphocytes	proliferation	of N.	albiflora,	with/without	the
presence of	Con A or LPS							

AMP	AMP		AMP + Con A		AMP + LPS	
(μg/ mL)	O.D. ₅₇₀	SI	O.D. ₅₇₀	SI	O.D. ₅₇₀	SI
0	0.107 ± 0.006 a		0.134 ± 0.006 a		0.123 ± 0.004 a	
25	0.125 ± 0.006 ab	1.17	0.139 ± 0.012 a	1.04	0.120 ± 0.011 a	0.98
50	$0.145 \pm 0.013 \text{ b}$	1.36	0.144 ± 0.010 a	1.07	0.135 ± 0.011 a	1.10
100	0.171 ± 0.017 c	1.60	0.254 ± 0.036 c	1.90	0.241 ± 0.044 c	1.96
200	0.141 ± 0.016 b	1.21	0.257 ± 0.016 c	1.92	$0.185 \pm 0.012 \text{ b}$	1.50
400	0.128 ± 0.007 ab	1.20	$0.190 \pm 0.012 \text{ b}$	1.42	0.147 ± 0.006 a	1.20
800	0.117 ± 0.007 a	1.09	0.125 ± 0.007 a	0.93	0.122 ± 0.012 a	0.99

The optical density data are represented as mean \pm standard error; Data within the same column sharing the same letters do not differ significantly (P > 0.05), whereas those with different letters differ significantly (P < 0.05)

Lymphocytes proliferation is an important indicator of immune defense in aquatic animals. The Con A-responsive lymphocytes belong to the B-cell subpopulation and the LPS-responsive lymphocytes belong to the T-cell subpopulation. The B-cell subpopulation involves cellular immunity while the T-cell subpopulation involves humoral immunity [19]. In this paper, Con A and LPS were used in lymphocytes proliferation assay to examine whether the two mitogens have synergy or antagonism effect with AMP on head kidney lymphocytes, and whether AMP contributes to the humoral or cellular immunity of *N. albiflora*. Results in Table 27.1 showed that Con A and LPS have synergy effect with AMP. Data in AMP + Con A group and AMP + LPS group implied that the AMP might contribute to both cellular and humoral immunity of *N. albiflora* in a dose-dependent manner (Table 27.1). This paper provided a rapid reference that the AMP is a potential immunostimulant for the healthy culture of *N. albiflora*. Further oral administration tests are needed to verify the immunostimulant efficacy of AMP.

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254 H. Shi et al.

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Chapter 28

Aspergillus niger Pellets Absorbed Bacillus sp. Isolated from Soybean Wastewater Sludge

Ningning Diao, Xiaowei Wu and Jianguo Zhang

Abstract More efficient soybean wastewater treatment approach is necessary because of the increasing soybean processed in the worldwide. *Aspergillus niger* pellets formed during the wastewater treatment was a promising method from the views of easy harvest, safety authorized by FDA, and lower sludge obtained. For further COD removal on the base of *A. niger* pellets, a bacteria (*Bacillus* sp.) was isolated from sludge of classical soybean wastewater treatment. And then the absorption ratio was tested in different conditions for potential application in wastewater treatment. In this research, 76.24 % of *Bacillus* sp. was absorbed from the 3.0×10^8 /mL initial cell concentration at pH 5.0. These results confirmed the possibility of further COD removal by the combination of *A. niger* and bacteria without losing the advantages of fungal pellets.

Keywords Wastewater • Fungal pellet • Absorption

28.1 Introduction

Soybean food is a kind of typical food in Asian area and is nourishable for human body. Soybean food was approved by more and more people in worldwide as the result of more flavors of soybean food provided and cultural communication. Therefore, the requirement of soybean food is huge and increasing. The total soybean processed for soybean foods production was 9,753,000 ton in China in 2010 [1]. Therefore, there is a huge amount of wastewater need to be treated also.

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One promising method for soybean wastewater treatment was fungal pellet formation [2]. Fungal pellet is one kind of morphologies, which like a ball consisted of entangled fungal mycelium. Fungal pellet was formed well in soybean wastewater because soybean wastewater has high content of organic compounds and varieties of other nutrients [3]. Fungal pellet has the advantages of easy harvest, low fermentation broth viscosity [4], and high yield of some proteins [5]. The diameter of fungal pellet is about 0.5 cm, which is harvested by simple filter with low cost. *Aspergillus niger* is the most used fungal species for pellet formation because *A. niger* was authorized as "Generally Regarded As Safe" by the American Food and Drug Administration (FDA) [6]; and *A. niger* genome was sequenced [7, 8]; *A. niger* has solid foundation of research [9]. Also, *A. niger* has a long history in food industry. *A. niger* hyphae in the form of pellet absorbed Cu²⁺, Zn²⁺, Ni²⁺ at pH 4.0–6.0 [10]. *A. niger* pellet was the right condition for citric acid production for food industry [11].

The crystal clear supernatant was obtained after 72 h cultivation after inoculation of A. niger spores. However, the chemical oxygen demand (COD) of supernatant was not low enough [2]. Therefore, it is necessary to modify the fungal pellet approach for further low COD. The dominated microbial species in typical soybean wastewater sludge was considered as the major contribution for COD removal. In this research, the A. niger pellets cooperated with this major species by absorption as assumpted was a potential method for further COD decreasing. The dominated species in soybean wastewater sludge was isolated and identified as Bacillus sp. The absorption of *Bacillus* sp. to *A. niger* pellets at different conditions was tested. The forces of A. niger and Bacillus sp. absorption were attributed to electrostatic interaction, hydrophobicity, and spatial interaction. Within these forces, electrostatic interaction was the major force. Krull reported that A. niger charge increased with the pH value increasing from 2.5 [12]. Wargenau showed the wide range of pH for negative charge [13]. This negative charge came from the carboxyl groups of cell wall. The carboxyl group contents in cell wall changed at different physiological conditions of cell [13]. The cell-cell interaction was also affected by the cation ion concentration based on the Derjaguin-Landau-Verwey-Overbeek mode (DLVO) of colloid science [14]. Bacillus had negative charge also [15]. Bacillus sp. attached to A. niger as that of microalgae cell attached onto fungal hyphae [16]. Different cells with same charge aggregated together because of differen charges at different area of cell surface [17] and the salt bridge affection [18]. Hydrophobins of fungal cell also played an important role during cells getting together. Hydrophobin was a kind of amphiphilic protein comprising 100-150 amino acids. Hydrophobin improved the cell aggregated by decreasing the Gibbs free energy [19]. A. niger co-pellets with microbial species provided the new route for bioenergy production [16] and wastewater treatment [20].

28.2 Materials and Methods

28.2.1 Fungal Species

Aspergillus niger spores was stored in lab at -70 °C.

28.2.2 Bacillus sp. Isolation

The soybean sludge came from Shanghai Tramy Green Food Co., Ltd. The sludge was diluted with sterilized water. And the liquid was taken to spread on PDA agar plates. These plates were cultivated at different temperature for microbial growth. After several days of cultivation, colonies were taken for identification.

28.2.3 Bacillus sp. Cultivation

The Bacillus sp. was cultivated in PDA (potato 200 g/L, dextrose 20 g/L, agar 15 g/L) medium at 37 °C, and then was harvested by centrifugation (4,000 rpm, 5 min). The cell was washed twice by sterilized water through suspension and centrifugation. Finally, the Bacillus sp. cell was added with phosphate buffer solution (PBS) for absorption.

28.2.4 Fungal Pellets Preparation

The *A. niger* spore was inoculated in 250 mL flask with 100 mL Czapek's medium (glucose 25.0 g/L, NaNO₃ 3.0 g/L, K₂HPO₄ 1.0 g/L, KCl 0.5 g/L, MgSO₄·7H₂O 0.5 g/L, FeSO₄·7H₂O 0.01 g/L). The final pH was adjusted by 1 mol/L HCl and 1 mol/L NaOH to 5.6. The *A. niger* was cultivated at 30 °C, 200 rpm. The *A. niger* pellets were formed after 48 h cultivation and was harvested at 72 h cultivation.

28.2.5 Aspergillus niger Pellets Absorbed Bacillus sp. Cell

Twenty *A. niger* pellets were mixed with *Bacillus* sp. in PBS buffer (50 mmol/L) with different pH, the sample was taken at interval for *Bacillus* sp. cell number determination.

N. Diao et al.

28.2.6 Cell Number Calculation

The *A. niger* spores and *Bacillus* sp. cell numbers were determined by the method of blood counting chamber. The cell number was calculated by the Eq. 28.1.

Cell (per mL) =
$$C_1/80 \times 400 \times 1,0000 \times N$$
 (28.1)

 C_1 Cell numbers in 80 small chambers N dilution ratio

28.2.7 Microbial Photos

Pull 100 mL fermentation broth with *A. niger* pellets in a petri dish. Photos were taken by digital camera (Sony Dsc-t70). The *Bacillus* sp. colonies photos were also taken by the same digital camera.

28.3 Results

28.3.1 Bacillus sp. Isolation from Soybean Sludge

The COD removal by *A. niger* pellets worked in aerobic condition. Therefore, the associated microbial isolation process was carried in aerobic condition. Several microbial species were obtained after cultivation. One bacterium was selected for further research because bacteria were the major contribution in aerobic digestion of wastewater treatment (Fig. 28.1). The selected bacteria was white, bulging, and identified as *Bacillus* sp.

28.3.2 Aspergillus niger Pellets Used for Soybean Wastewater Treatment

After 72 h cultivation, *A. niger* spores grew and became cell pellets in soybean wastewater. And the yellow and cloudy soybean wastewater turned into crystal clear (Fig. 28.2). The COD of soybean wastewater reduced from 32,600 to 4,360 mg/L (Table 28.1). Therefore, 86.66 % of COD was removed by *A. niger* pellets.

Fig. 28.1 The *Bacillus* sp. colonies from soybean wastewater sludge



Fig. 28.2 Aspergillus niger pellets after 72 h cultivation in soybean wastewater



Table 28.1 Chemical oxygen demand of soybean wastewater and wastewater after treatment

	Soybean wastewater	Treated
COD (mg/L)	32,600	4,360

28.3.3 Aspergillus niger Pellets Preparation in Czapek's Medium

Figure 28.3 showed the *A. niger* pellets after 72 h cultivation in Czapek's medium. There were about 50 pellets in every 100 mL medium. And the diameter of 90 % pellets was 0.5–1.0 cm, which showed the relatively constant diameter of pellets. The *A. niger* pellets are white, smooth.

N. Diao et al.

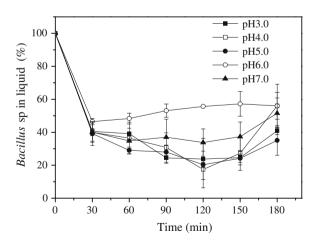
Fig. 28.3 The *A. niger* pellets formed in Czapek's medium



28.3.3.1 Effect of pH on the Bacillus sp. Absorption into Aspergillus niger Pellets

For all the pH values (pH 3.0, 4.0, 5.0, 6.0, 7.0), the *Bacillus* sp. cell number in liquid phase reduced sharply within first 30 min of mixing. And then, *Bacillus* sp. cell number decreased slowly until 150 min except that of pH 6.0. Finally, *Bacillus* sp. cell number increased from 150 to 180 min. The *Bacillus* sp. cell number kept constant from 30 to 180 min (Fig. 28.4). The *A. niger* pellets kept intact during all the process. The minimum *Bacillus* sp. cell in liquid phase was obtained at pH 5.0. Therefore, pH 5.0 was set as the default condition for further research.

Fig. 28.4 Effects of pH on the *Bacillus* sp. absorption into *A. niger* pellets



28.3.4 Effect of Ions on the Bacillus sp. Absorption into Aspergillus niger Pellets

With all different cation ion and different concentrations, the *Bacillus* sp. cell number decreased as time went on (Figs. 28.5, 28.6 and 28.7). The same phenomenon occurred as that in Fig. 28.4. After 30 min, *Bacillus* sp. cell numbers with addition of cation ions had higher values than that of control. *Bacillus* sp. cell number with addition of Na⁺ showed an increase during the period from 30 to 60 min. And then *Bacillus* sp. cell number decreased slowly until 180 min. For the Ca²⁺, the *Bacillus* sp. cell number at 0.1 mol/L was lower than that of 0.25 and 0.5 mol/L. There was no significant difference among 0.1, 0.25, 0.5 mol/L Na⁺. The *Bacillus* sp. cell number at different Mg²⁺ concentrations showed difference after 120 min mixing. Finally, the *Bacillus* sp. cell number at 0.1 mol/L was lower than that of 0.25 and 0.5 mol/L with the addition of Mg²⁺.

Fig. 28.5 Effects of Na⁺ on the *Bacillus* sp. absorption into *A. niger* pellets

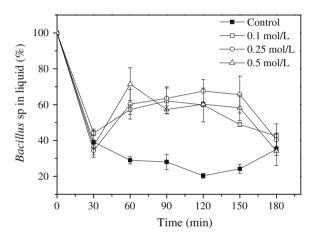
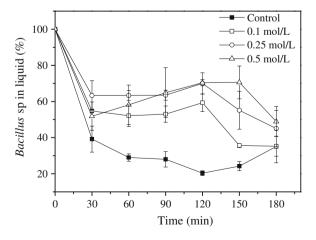
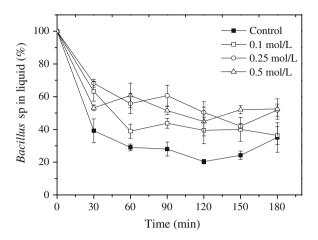


Fig. 28.6 Effects of Ca²⁺ on the *Bacillus* sp. absorption into *A. niger* pellets



N. Diao et al.

Fig. 28.7 Effects of Mg²⁺ on the *Bacillus* sp. absorption into *A. niger* pellets



28.4 Discussion

The classic approach of soybean wastewater treatment was filtration, anaerobic digestion (Upflow anaerobic sludge blanket, UASB), aerobic reaction (Sequencing Batch Reactor, SBR), and sediment sequencely. This process was contributed by large amount of microbes of different species to enhance the COD removal, fungal pellets formation in soybean wastewater with other microbial species as an association was considered as a potential approach. *Bacillus* sp. was selected because it was an important microbe producing biofilm and enzymes for COD removal in sludge [21, 22]. The *Bacillus* sp. produced biofilm and enzymes which help with organic compounds degradation and attachment to cells.

Bacillus sp., belong to bacteria, grew faster than A. niger. Therefore, the Bacillus sp. cell number increased significantly, even some of them were absorbed by A. niger pellets when these two species were mixed in medium. To test the Bacillus sp. absorption into A. niger, these two species were mixed in PBS without any nutrients. The Absorption between cells was derived by electrostatic interaction and hydrophobicity mainly [23]. For given filamentous fungi (A. niger) in this research, the hydrophobicity was defined because the hydrophobicity was provided by the hydrophobins from filamentous fungi. However, the electrostatic interaction was affected by the ion concentration, and the cation valence [24]. The Na⁺, Ca²⁺, and Mg²⁺ were not only major cations for the electrostatic interaction, but also necessary for cell growth as nutrients [13]. The effects of cation concentrations on the absorption between Bacillus sp. and A. niger showed the same trend of absorption ratio.

The results of this research confirmed that the *Bacillus* sp. cell was absorbed by *A. niger* pellets with the maximal value of 76.24 % absorbed. These two species playing the role of COD removal together provided a further COD reduction compared to the 86.66 % COD removed by *A. niger* pellets only. The *Bacillus* sp. cell number fluctuated during the absorption process especially in Figs. 28.5 and

28.6. This phenomenon was caused by the ion interaction among microbial cell and buffer, which was the same as that of the ion exchange. Therefore, high ion concentration decreased the absorption ratio of *Bacillus* sp. These results gave the clue that ion concentration decreasing from 50 mmol/L was the right direction for *Bacillus* sp. absorption. Further more, immobilization bacteria like *Bacillus* sp. by chemical methods on the base *A. niger* pellets absorption is another method to stabilize the co-pellets for long-time utilization.

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Chapter 29 Expression, Purification and Characterization of Maltase from "Quick" Baker's Yeast

Cui-Ying Zhang, Hai-Yan Song, Xue Lin, Xiao-Wen Bai and Dong-Guang Xiao

Abstract The "quick" baker's yeasts are capable to rapidly metabolize maltose, thereby improving leavening ability. Given that maltase is the determining factor in maltose fermentation for "quick" baker's yeast, it is hence necessary to research the properties of the maltase independently for the "quick" baker's yeast. In this study, the heterogeneous expression of MAL62 encoding for maltase and purification of maltase were well completed. Furthermore, the enzymatic properties of maltase concentrated on the effects of substrate (maltose) and end product (glucose) were investigated as well as the physic-chemical properties and transglycosylation activity. The substrate maltose did not affect the activity of maltase, while the maltase activity was inhibited by the end product. This study provides guidance for the research of maltose metabolism for "quick" baker's yeast.

Keywords "Quick" baker's yeast · Maltase · Enzymatic properties

29.1 Introduction

Baker's yeast (*Saccharomyces cerevisiae*) is the key biological leaven using in baking industry, and is of crucial importance in dough leavening [1]. Maltose is the most abundant fermentable sugar in lean dough [2–5]. According to the level of utilizing maltose, the baker's yeast could be classified into "quick" and "slow." "Quick" baker's yeasts are capable to rapidly metabolize maltose, thereby improving leavening ability. The ability of a baker's yeast strain to ferment maltose depends on the presence of three gene products; they are maltose permease encoded by MALx1 (where x is locus) gene, maltase encoded by MALx2 gene, and a

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266 C.-Y. Zhang et al.

positive regulatory protein activating the two enzymes encoded by MALx3 gene [6–8]. Maltose is transported across the cell membrane via maltose permease and cleaved intracellularly into two units of glucose by maltase. Given that maltase is the determining factor in maltose fermentation for "quick" baker's yeast [5, 9], it is hence necessary to research the properties of the maltase independently for the "quick" baker's yeast.

Maltase (α -D-Glucoside glucohydrolase; EC 3.2.1.20) exists in a large amount of organisms and can be classified into three types (I, II, and III) according to the substrate specificities [10]. The α -glucosidase from Thermoanaerobacter tengcongensis, which belongs to type II, preferentially utilizes maltose and has low specificity toward aryl glucosidase [11]; Type III is similar to type II, except that they hydrolyze di- and oligosaccharides and starch with comparable rates. The enzymes from *Mucor racemosus* and Penicillium species belong to the Type III [12]. The maltase of yeast is classified into type I. Some properties of maltase from different yeasts have been studied [13, 14]. However, the comprehensive properties of maltase for the "quick" baker's yeast are little known, particularly the properties involving in maltose metabolism.

In this study, the MAL62 gene from a "quick" baker's yeast was successfully expressed in *Escherichia coli* BL21 (DE3) and the maltase was well purified. To investigate the enzymatic properties of maltase, we focus on the effects of substrate (maltose) and end product (glucose), in addition to the physic-chemical properties and transglycosylation activity.

29.2 Materials and Methods

29.2.1 Strains and Plasmids

The "quick" industrial baker's strain BY14 was obtained from the Yeast Collection Center of the Tianjin Key Laboratory of Industrial Microbiology. *E. coli* DH5 α was stored in our laboratory. The expression vector pET-28a and the strain BL21 (DE3) were provided by Prof. Fu-Ping Lu (Tianjin University of Science and Technology, P.R. China).

29.2.2 Construction of MAL62 Expression System

Yeast genomic DNA was prepared from industrial baker's yeast strain BY14 using a yeast DNA kit (D3370-01, Omega, USA). The MAL62 fragment was amplified via PCR using the genomic DNA of BY14 as template and the primers (CGCGGATCC ATGACTATTTCTGATCATCCAG, the restriction site of BamHI is underlined and CCCAAGCTT GTTATTTGACGAGGTAGATTCT, the restriction site of HindIII

is underlined). The PCR product was purified with a Gel Extraction Kit (Sigma, USA) and cloned to the vector pET28a after digestion with BamHI and HindIII (TaKaRa, Japan), thereby generating pET28a-MAL62. DNA sequencing was performed (Beijing Genomic institution, China). For gene expression, the plasmid pET28a-MAL62 was transformed into *E. coli* BL21 (DE3), resulting in the strain BL21-M.

29.2.3 Purification of Maltase

The strain BL21-M carrying the plasmid pET28a-MAL62 was cultured in 50 mL of Luria-Bertani medium (LB medium, 10 g/L tryptone, 5 g/L yeast extract, and 10 g/L NaCl), which was supplemented with kanamycin (60 µg/mL) [11]. When the OD₆₀₀ reached to 1.0, IPTG was added with the final concentration of 0.4 mM, and the cells were continuously cultivated for 12 h at 26 °C. 60 mL of cells were harvested by centrifugation (3,500×g, 4 °C, 20 min) and washed by PBS buffer (0.8 % NaCl, 0.02 % KCl, 0.142 % Na₂HPO₄, 0.027 % KH₂PO₄; pH 7.0) [15]. The cell precipitation was suspended by 7 mL PBS buffer. The suspended cells were disrupted through sonication (power 50 %, on 4 s, off 9 s, total time 40 min) and the crude enzyme was centrifuged at 8,000×g for 20 min. The collected supernatant was loaded onto Ni-NTA affinity column (Kang Wei Company, China) and recombinant protein was eluted with a liner gradient of imidazole (0.1–0.5 M) PBS buffer. The collected enzyme was dialyzed in PBS buffer at 4 °C for 12 h. Protein concentration was determined by the method of Bradford [16], and the bovine albumin was used as standard.

29.2.4 Standard Maltase Activity Assays

Maltase activity was tested using maltose as the substrate. The reaction mixture contained 0.5 mL of 0.4 M maltose, 0.4 mL of PBS buffer (pH 7.0), and 0.1 mL of the maltase. After incubation at 40 °C for 20 min, the reaction was stopped by boiling for 10 min. The freed glucose was assayed with biosensor. When the *p*-nitrophenyl-a-D-glucoside (*p*-NPG) was used as the substrate, maltase activity was measured in a mixture containing 0.9 mL PBS buffer (pH 7.0), 2 mL of 10 mM substrate solution, and 0.1 mL of the maltase. After incubating at 40 °C for 10 min, the reaction was stopped with 2 mL of 1 M Na₂CO₃, and the released *p*-nitrophenol was quantified with spectrophotometer at 410 nm. A unit of maltase activity was defined as the amount of enzyme liberating 1 μmol of glucose or 1 μmol of *p*-nitrophenol in 1 min.

268 C.-Y. Zhang et al.

29.2.5 Effects of Maltose and Glucose on Maltase Activity

To investigate the effect of substrate on the enzyme activity, different concentrations (from 40 to 300 mM) of maltose were used. The $K_{\rm m}$ and $V_{\rm max}$ values were calculated from the Hanes plots relating [s]/v to [s]. The effect of end production was determined using p-NPG as the substrate, and the kinetic parameters for the maltase under different concentrations of glucose were obtained from the Lineweaver–Burk plots relating 1/v to 1/s. The $K_{\rm i}$ value was determined by the graphical method of Dixon. The date was analyzed using Origin Software.

29.2.6 Effects of pH, Temperature, and Chemicals on Maltase Activity and Stability Tests

The optimum temperature of maltase activity was assayed at 20–55 °C for 20 min using 200 mM maltose. After incubation at various temperatures for 2 h, thermal stability was analyzed by assessing residual activity. The optimum pH of maltase activity was tested in PBS buffer with different pH (pH 3.0–9.0) at 40 °C for 20 min. The effect of pH on enzyme stability was analyzed after incubation in PBS buffer with different pH (pH 3.0–9.0) at 4 °C for 24 h. The residual activity was measured under the standard hydrolytic condition.

The effect of metal ions (Cu²⁺, Fe³⁺, Co²⁺, Mn²⁺, Ni²⁺, Ca²⁺, Pb²⁺, and Mg²⁺) was examined with a final concentration of 5 mM, and that of the chelating agent EDTA was 5 and 25 mM. All tests were performed in PBS buffer (pH 7.0) and the enzyme activity tested under no ions was defined as 100 %.

29.2.7 Transglycosylation Activity Analysis

The reaction mixture (1 mL), which consists of 0.9 mL of 1 M maltose solution (from pH 7.0 PBS buffer) and 0.1 mL of maltase, was incubated at 30 °C. Every 1 h, sample was taken out and boiled at 100 °C for 10 min for maltase inactivation and filtered through 0.45 μ m pore size cellulose acetate filters (Millipore Corp, Danvers, MA, USA). High-performance liquid chromatography (HPLC) with a refractive index detector and Aminex[®] HPX-87H column (Bio-Rad, Hercules, CA, USA) was utilized at 65 °C with 5 mM H₂SO₄ as the mobile phase at a flow rate of 0.6 mL/min [17] to analyze the sugars.

29.3 Results and Discussion

29.3.1 Purification of Maltase

Maltase from "quick" industrial baker's strain BY14 is an intracellular enzyme, which made us difficult to study the enzymatic properties directly. In this study, the MAL62 gene was successfully expressed in *E. coli* BL21 (DE3). In general, *E. coli* is the preferred host for the property studies of recombinant protein expression due to the easily genetic manipulation, the relatively inexpensive of cultivation, and the rapid target expression, typically producing protein in 1 day. The importance of *E. coli* for heterologous protein production is perhaps best highlighted by the wide variety of commercial products available for the *E. coli* expression system [18]. In the present study, the maltase of "quick" baker's yeast was well purified. A relatively sharp band at 68 kDa was observed on SDS-PAGE (Fig. 29.1). Purification steps are represented in Table 29.1. The final enzyme solution was purified 6.2-fold compared with the crude extract. The molecular mass of the purified enzyme estimated by SDS-PAGE was 68 kDa (Fig. 29.1).

29.3.2 Effect of Different Concentrations of Substrate on Maltase Activity

Different concentrations of maltose were incubated with the same amounts of recombinant enzyme and the maltase activity was represented by the maltose

Fig. 29.1 SDS-PAGE of purified recombinant Histagged maltase. *M* marker; *I* purified protein

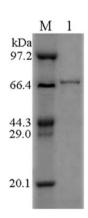


Table 29.1 Purification of recombinant enzyme from BL21-M

Step	Protein (mg)	Total activity (µ)	Specific activity (µ/mg)	Purification fold
Crude	43.3	549.4	12.7	
Ni ²⁺ affinity	3.0	237.0	79.0	6.2

270 C.-Y. Zhang et al.

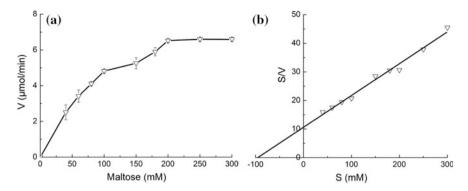


Fig. 29.2 Effect of maltose concentration on maltase activity. **a** 1 mL of reaction mixture consisted of different concentrations of maltose, 0.1 mL of enzyme, and PBS buffer reacted at 40 °C for 5 min; **b** Hans plot of maltase activity was determined using maltose as the substrate. Data were shown as mean \pm SD (n = 3)

hydrolysis rate. With the increase of substrate concentration, no decrease trend was found in hydrolysis rate (Fig. 29.2a), which suggested that maltose did not restrain the maltase activity. Furthermore, the apparent $K_{\rm m}$ (Michaelis constant) value of maltase for maltose calculated from Fig. 29.2b was 95.507 mM. These findings suggested that the enzyme could not be inhibited by substrate (maltose). Simultaneously, the $K_{\rm m}$ for maltose was 95.507 mM, much higher than that from *Saccharomyces logos* (the $K_{\rm m}$ for maltase was 7.7 mM) [19], so maltose may not be the optimum substrate for the maltase from the "quick" baker's yeast.

29.3.3 Effect of End Product on Maltase Activity

To investigate the effect of end product on maltase activity, different concentrations of glucose were added to the identical reaction mixtures to test the maltase activity. With the increase of glucose concentration, the maltase activity was gradually decreased (Fig. 29.3a). When the concentration of glucose was increased to 180 mM, 84.6 % of the maltase activity was left. Various substrate (p-NPG) concentrations were used to determine the type of glucose inhibition under different concentrations of glucose. Results were obtained after incubation at 40 °C for 10 min. According to Fig. 29.3b, the $K_{\rm m}$ and $V_{\rm max}$ were 0.661 mM and 0.018 μ mol/min, respectively, without the presence of glucose in reaction mixture. Notably, when the 20 and 180 mM of glucose were added, the $K_{\rm m}$ and $V_{\rm max}$ increased to 0.901 mM and 0.017 μ mol/min, and 1.204 mM and 0.014 μ mol/min, respectively. These results indicate that the type of glucose affecting on maltase was linear mixed inhibition (Fig. 29.3c). The graphical method of Dixon was used to determine the $K_{\rm i}$, which is the dissociation constant of glucose, and the $K_{\rm i}$ value for glucose was 139.176 mM. Enzyme activity is inhibited by the end product, which corresponds to the previous

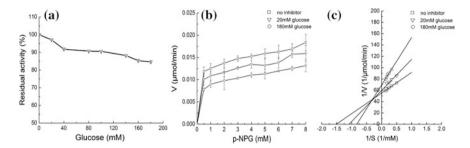


Fig. 29.3 Effect of end product on maltase activity. a 3 mL of reaction mixture consisted of different concentrations of glucose, 0.1 mL of enzyme, p-NPG and PBS buffer reacted at 40 °C for 20 min; b 3 mL of reaction mixture consisted of different concentrations of glucose, 0.1 mL of enzyme, p-NPG and PBS buffer reacted at 40 °C for 10 min; 2 mL of 1 M Na₂CO₃ was added to stop the reaction; c Lineweaver–Burk plot of p-NPG hydrolysis. Data were shown as mean \pm SD (n = 3)

studies [20–22]. Compared with the maltase from Saccharomyces italicus exhibiting intense inhibition of end product (glucose) with 1.2 mM of K_i for glucose [23], the effect of end product (glucose) on maltase activity from "quick" baker's yeast was not significant (K_i = 139.176 mM). Furthermore, the glucose liberated from maltose could be rapidly oxidized in "quick" baker's yeast cells. Little or no glucose was accumulated in yeast cells, so that the end product (glucose) inhibition could be neglected. For this reason, "quick" baker's yeast had the capacity of rapidly metabolizing maltose.

29.3.4 Physic-Chemical Properties of Maltase

The physical and chemical factors, which are the basic elements, are also of great importance to influence the enzyme activity. The physic-chemical properties of

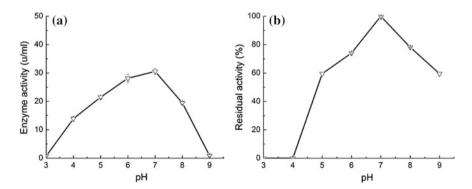


Fig. 29.4 Effect of pH on maltase activity. **a** The maltase activity was measured at various pH. **b** Effect of pH on the stability of maltase. After incubation at different pH at 4 °C for 24 h, the residual activity was measured at standard condition. The highest activity was defined as 100 %. Data were shown as mean \pm SD (n = 3)

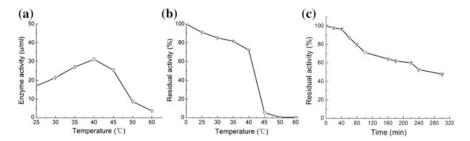


Fig. 29.5 Effect of temperature on maltase activity. **a** The maltase activity was measured at various temperatures at pH 7.0. **b** Thermostability of the recombinant enzyme. After 2-h incubation at different temperature, the residue activity was measured at standard condition. The initial activity was defined as 100 %. **c** Effect of temperature on the stability of maltase. The maltase was incubated at 40 °C for different times and residual activity was assayed at standard condition. Data were shown as mean \pm SD (n = 3)

maltase, the effects of pH, temperature, metal ions, and EDTA on maltase activity were assayed. The maximum of maltase activity was shown at pH 7.0 (Fig. 29.4a). More than 60 % of residual maltase activity was observed from pH 5.0 to pH 9.0, while almost no residual activity was found at pH 4.0 or below it (Fig. 29.4b). As shown in Fig. 29.5a, the highest maltase activity was observed at 40 °C. Therefore, the thermotolerance of maltase was determined at 40 °C. The maltase activity significantly decreased by 52.3 % compared with that at the beginning, and the half-life of maltase at 40 °C was 244.6 min (Fig. 29.5c). After 2-h incubation at various temperatures, the maltase displayed different levels of activity decrease. The residual activity of maltase decreases slowly when incubation before 40 °C. However, the residual activity exhibited a much greater decline from 40 to 45 °C, and hardly any maltase activity was found at a higher temperature (50 and 60 °C) (Fig. 29.5b). The presences of Ca²⁺ and Mg²⁺ slightly increased the enzyme activity by 4.3 and 8.3 %, respectively (Table 29.2). By contrast, all the six other metal ions

Table 29.2 Effect of different chemical reagent on activity of the purified maltase

Relative activity (%)
100 ± 0.24
11.2 ± 0.12
32.6 ± 0.12
104.3 ± 0.31
84.2 ± 0.23
61.4 ± 0.31
108.3 ± 0.20
_
_
-100 ± 0.22
72.3 ± 0.21

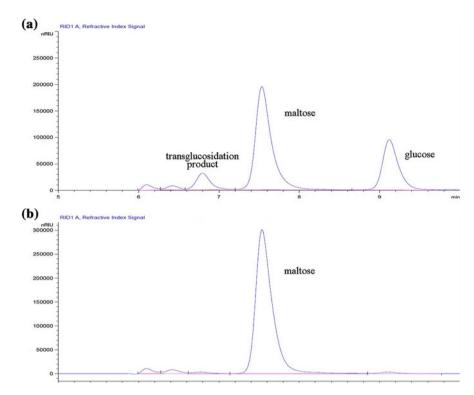


Fig. 29.6 Analysis of transglucosidation products by HPLC. Reaction mixture: 0.9 mL of 1 M maltose, 0.1 mL of maltase. **a** Reaction mixture with enzyme; **b** reaction mixture without enzyme

(Ni²⁺, Co²⁺, Mn²⁺, Pb²⁺, Cu²⁺, and Fe³⁺) decreased the maltase activity at different levels. Evidently, no maltase activity was found at the presence of Cu²⁺ and Fe³⁺ (the final concentration of ions was 5 mM) (Table 29.2). Inaction by Cu²⁺ indicates that the thiol groups, carboxyl groups, or histidine residues, which were important to the function of maltase, existed in the enzyme molecule [24–26]. Addition with 5 mM of EDTA did not evidently affect the maltase activity, but 27.7 % of maltase activity was notably lost when the concentration of EDTA was increased to 25 mM (Table 29.2). The evident inhibition with EDTA, even at 25 mM, indicates that a metal cation is not needed for maltase activity [22].

29.3.5 Transglycosylation Activity of Maltase

The transglycosylation activity of the recombinant maltase was assayed using maltose as the substrate. The reaction mixture incubated with the maltase for 5 h and that without enzyme were comparatively analyzed through HPLC. Compared

274 C.-Y. Zhang et al.

with the reaction mixture without enzyme, a new transglycosylation substance was detected at 6.7 min in reaction mixture with enzyme (Fig. 29.6). These results reveal that maltase had the capacity of transglycosylation.

29.4 Conclusion

In this study, the maltase was cloned and purified successfully. The enzyme was not inhibited by the substrate (maltose), however, it was inhibited by end product (glucose) and the type of glucose affecting on maltase was linear mixed inhibition. This study also suggested that the maltase had transglycosylation activity, while the physic-chemical properties of enzyme may limit its use in industry.

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Chapter 30 Evaluation of an Ethanol-Tolerant Acetobacter pasteurianus Mutant Generated by a New Atmospheric and Room Temperature Plasma (ARTP)

Xiaoying Wu, Yuqiao Wei, Zeming Xu, Lingpu Liu, Zhilei Tan and Shiru Jia

Abstract A novel atmospheric and room temperature plasma (ARTP) which used helium as the working gas was employed to generate mutants of *Acetobacter pasteurianus* to improve the ethanol tolerance, which is a poorly characterized industrial strain. The best strain U1-1 was selected after mutagenesis. U1-1 could grow in liquid medium with 11 % ethanol. The production of acetic acid reached 32.83 ± 0.75 g/L, 385.7 % higher than that of the parent strain, meanwhile, U1-1 has stable production. Moreover, the cell membrane permeability were measured by PI assay and the results show that the cell membrane permeability of starting strain (AP-1.01) is significantly higher than U1-1. The better ethanol tolerance of strain U1-1 was maybe due to the decreased membrane permeability.

Keywords ARTP • *Acetobacter pasteurianus* • Mutant • Ethanol tolerance • Membrane permeability

30.1 Introduction

Acetic acid bacteria (AAB) are gram-negative strictly aerobic bacteria that are widely used for the production of vinegar, gluconate, bacterial cellulose, and so on. AAB are divided into three species: *Acetobacter*, *Gluconobacter*, and *Gluconacetobacter*. *Acetobacter pasteurianus*, which belongs to the *Acetobacter* sp. is widely used in the production of vinegar in many countries [1].

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During the production of vinegar, AAB subjected to many adverse environmental factors, such as high temperature in the solid fermentation, high concentration of acetic acid, and ethanol. There are many researchers who focus on the acetic acid tolerance mechanisms and thermotolerance mechanisms of AAB, and how to improve these tolerances [1–5]. However, few researchers pay attention to the ethanol tolerance. Ethanol is the main carbon source of AAB and the substrate of acetic acid fermentation by AAB. Ethanol has been verified harmful to the growth of microorganism; ethanol would change the permeability of cell membrane, resulting in cell lysis that ultimately leads to cell death [6-8]. The high concentration of ethanol can seriously affect AAB in the fermentation process. Wei et al. [9] used genome shuffling to improve the ethanol tolerance of acetic acid bacterium to 13 %. Yu et al. [10] used continuous ethanol stress adaptation culture succeed in improving the ethanol tolerance of A. pasteurianus CGMCC 3089. Atmospheric and room temperature plasma (ARTP) is a novel plasma generating system driven by a radiofrequency (RF) power supply with water-cooled, bare-metallic electrodes which has the advantages of low costs, low plasma temperature, flexible operations, and security and thus shows promising applications in biotechnology [11]. Recently, many researchers used ARTP for their own strain breeding [11–14].

In this paper, the ethanol tolerance of *A. pasteurianus* was obtained by using ARTP generated with helium gas. Specially, we focused on the distinction in the cell membrane permeability between starting strain and mutant strain.

30.2 Materials and Methods

30.2.1 Microorganisms

The acetic acid bacterium strain, A. pasteurianus AP-1.01 was used as starting strain.

30.2.2 Culture Medium

The media used in this study were as follows. The compositions of the GY medium were (g/L): glucose 20, yeast extract 10, adding 3 % (v/v) ethanol after sterilization, solid medium with ager power 20 and $CaCO_3$ 10. The composition of $CaCO_3$ selection medium were (g/L): glucose 20, yeast extract 10, $CaCO_3$ 10, agar power 20, adding 11 % (v/v) ethanol after sterilization. The composition of stress medium was the same as GY medium, but the ethanol concentration was 3 % (v/v), 9 % (v/v), 11 % (v/v), 13 % (v/v), respectively. The liquid medium used for shaker screening was GY medium, except the ethanol concentration was 11 % (v/v). The initial pH of the medium mentioned above was adjusted to 6.8.

30.2.3 Culture Conditions

The cells from slant cultures were transferred to 50 mL GY medium in a 250 mL Erlenmeyer flask and shakened with 180 rpm at 30 °C until the optical density (0) of the culture at 600 nm reached 0.8, and then the culture was used as the seed for the following experiment.

5 % (v/v) inoculums were inoculated into the 50 mL medium in a 250 mL Erlenmeyer flask and incubated with shaking at 180 rpm in 30 °C for 96 h. The yield of acetic acid was measured by acid-base titration [2].

30.2.4 Mutation with ARTP

The ARTP mutation breeding system was purchased from Si Qing Yuan Biotechnology Co., Ltd, Beijing, China. It consisted of a plasma generator, a helium gas supply and control subsystem, and a sample plate made of stainless steel which can be moved up and down [11]. The operating parameters of the helium-based ARTP were the following: the helium gas flow rate of Q He = 10.0 L/min and RF power input of 120 W, and the treating time ranged from 0 to 180 s, the treatment distance was 2 mm.

The specific operating steps were according to the methods of Wang [13] with some modifications. The strain A. pasteurianus was cultivated at 30 °C and 180 rpm to logarithmic phase (about 24 h). Then 3 mL of culture was moved into sterile tubes and the absorbance of culture was adjusted to $OD_{600} = 0.80 \pm 0.07$ with sterile saline. Then 10 μ L of culture was removed and spread on each sterilized steel plate and dried in sterile air for few minutes. This plate was later exposed to the ARTP system's nozzle exit. The working parameters were mentioned above. After mutation, the plate was placed in a new sterile tube containing 1 mL of sterile water, and then the suspension was spread on the GY solid medium. After 48 h of cultivation at 30 °C, colonies that exhibited large transparent zones were selected and then inoculated with sterilized toothpick onto plates of CaCO₃ selection medium. After cultivation at 30 °C, colonies that grown fast and exhibited large transparent zones were selected for further shaker screening.

30.2.5 Calculation of Lethality Rate and Shaker Screening

The lethality rate of the bacterial strain was evaluated based on the following equation:

Lethality rate (%) =
$$\frac{U - T}{U} \times 100$$

280 X. Wu et al.

where U is the total colony count of the sample without treatment, and T is the total colony count after treatment by the plasma. All the colony numbers were obtained by the CFU (Colony-Forming Units) method on the solid medium.

For shaker screening, the cells isolated from $CaCO_3$ selection medium were transferred to 50 mL GY medium in a 250 mL Erlenmeyer flask at 30 °C and 180 rpm for 24 h, and then with 5 % (v/v) inoculums were inoculated into the 50 mL medium containing 11 % (v/v) ethanol in a 250 mL Erlenmeyer at 30 °C and 180 rpm for 96 h. After cultivation, the acetic acid yield was measured as above. The strain with high acetic acid at high ethanol concentration (11 % (v/v)) was used for further research.

30.2.6 Genetic Stability of the Mutant Strain

The stability of the mutant was evaluated by the serial subcultivation. The mutant strain was incubated on the solid medium with 3 % (v/v) ethanol for 48 h (the first subculture). Then the cell from the plate was incubated on the same medium for further subculture. The cells from the first to sixth subculture were transferred to 50 mL GY medium containing 11 % (v/v) ethanol in a 250 mL Erlenmeyer flask at 30 °C and 180 rpm for 96 h. The yield of acetic acid was measured with the method mentioned above.

30.2.7 Determination of Membrane-Disrupting Activity of Starting Strain and Mutant Strain Under Different Concentration of Ethanol

The membrane-disrupting activity of starting strain and mutant strain was determined by measuring the fluorescence enhancement of PI (Sigma Corp., USA) in different concentration of ethanol-treated cells according to the modified method of Bo et al. [15]. Cells incubated under different ethanol concentration were harvested by centrifugation at 12,000 rpm for 2 min and washed twice with phosphate buffer saline (PBS). Cells were then resuspended in PBS, and the final cell density was adjusted to OD₆₀₀ = 0.15 - 0.25. 100 μL of the cell suspension was mixed with 100 μL of PI (initial concentration was 12 $\mu g/mL$) and was then incubated at room temperature for 10 min. The fluorescence intensity was measured at excitation and emission wavelengths of 535 and 617 nm, respectively, by a SYNERGY4 multifunctional enzyme mark instrument (Bioteck). Each experiment was conducted in triplicate.

Membrane-disrupting activity
$$=\frac{\text{fluorescence}}{\text{OD}_{600}}$$

30.3 Results and Discussions

30.3.1 Lethality Rate Curve and Choice of Treatment Time

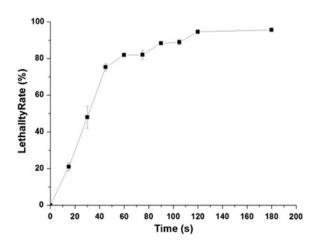
The effect of various plasma treatment times on the lethality rate is shown in Fig. 30.1. There is a clear dose–response relationship between the plasma treatment time and the *A. pasteurianus* lethality rate. As shown in Fig. 30.1, treatment with ARTP for 75 s resulted in a lethality rate of 82.01 %. According to the breeding strategies, we selected treatment time 75, 90, 105 s, so that the lethality rate was from 80 to 90 %. Thus, we could get more positive mutation.

30.3.2 Screening for Ethanol-Tolerant Strains

In this study, more than 90 colonies grown on the $CaCO_3$ selection petri dishes after ARTP treatments were selected randomly based on growth rate and transparent ring diameter. These colonies were further selected based on the production of acetic acid under the high concentration (11 % (v/v)) of ethanol. Part of the screening date was shown in Fig. 30.2. As shown in Fig. 30.2, the production of starting strain was 6.760 g/L, comparing with the starting strain, there are 25 colonies that have higher yield. Among them, we selected 14 colonies for rescreening. The rescreening results were shown in Table 30.1.

As shown in Table 30.1, the production of some colonies decreased comparing with preliminary screening. The instability of production resulted from the instability of mutagenesis. But we could see from Table 30.1, colonies A1-10, A1-9, A2-11, U1-1, 7-26, 7-40 still have high yield. Comparing with starting strain, production was increased by 347.3, 321.8, 289.8, 331.4, 92.67, 216.3 %,

Fig. 30.1 Effect of various plasma treatment times on the lethality rate. Each value is the mean of three replicates ±SD



282 X. Wu et al.

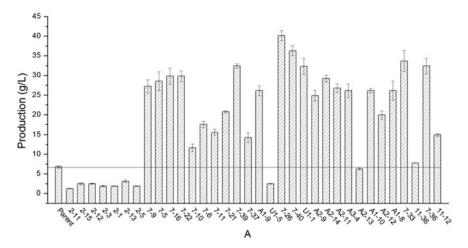


Fig. 30.2 Part of the screening data

Table	30.1	Rescreening	results
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Strain	Production (g/L)	Improving percentage (%)	Strain number	Production (g/L)	Improving percentage (%)
U1-1	29.16 ± 1.44	331.4	A1-8	14.69 ± 2.70	117.3
A2-11	26.35 ± 1.50	289.8	A2-14	25.92 ± 3.89	283.4
A1-10	30.24 ± 1.50	347.3	A1-9	28.51 ± 1.30	321.8
7-5	3.080 ± 0.71	-54.44	7-33	2.590 ± 0.92	-61.66
7-9	8.600 ± 1.92	27.22	7-39	4.540 ± 0.92	-32.90
7-22	11.66 ± 4.49	72.49	7-40	21.38 ± 0.92	216.3
7-16	7.340 ± 1.50	8.580	7-26	13.02 ± 1.30	92.67

respectively. These six colonies were selected for further screening. The screening result was presented in Table 30.2.

As shown in Table 30.2, the production of A1-10, A1-9, A2-11, 7-26, 7-40 decreased. However, the production of colony U1-1 remained the same compared with former. So U1-1 was selected as the ethanol-tolerant strain.

30.3.3 Stability of the Mutant

Mutant phenotype may occur in delays in the passage, which would lead to recession of production traits of high-yielding strains. So it is necessary to evaluate the stability of the mutant. The genetic stability of U1-1 was confirmed by subcultivation. The genetic stability of U1-1 was shown in Table 30.3. As shown in

Strain number	Production (g/L)	Improving percentage (%)	Strain number	Production (g/L)	Improving percentage (%)
number	1.6 /	1 0 . ,	number	1	1 2 7
A1-10	8.750 ± 0.46	29.41	U1-1	33.70 ± 1.83	398.5
A1-9	10.37 ± 0.37	53.37	7-26	3.890 ± 0	-42.49
A2-11	19.67 ± 3.20	190.8	7-40	4.540 ± 0.92	-32.90

Table 30.2 The second rescreening results

Table 30.3 Genetic stability results

Passage number	Production (g/L)	Improving percentage (%)
The first generation	31.10 ± 1.30	360.1
The second generation	33.26 ± 1.98	392.0
The third generation	32.83 ± 0.75	385.7
The fourth generation	30.67 ± 1.50	353.7
The firth generation	29.81 ± 1.30	341.0
The sixth generation	32.24 ± 1.41	376.9

Table 30.3, the first to sixth generation, the production was 31.10 ± 1.30 , 33.26 ± 1.98 , 32.83 ± 0.75 , 30.67 ± 1.50 , 29.81 ± 1.30 , 32.24 ± 1.41 g/L, respectively. It was shown that U1-1 has stable property and could be used for the following study.

30.3.4 Membrane-Disrupting Activity of Starting Strain and Mutant Strain Under Different Concentration of Ethanol

The membrane-permeabilizing activity was investigated by using the PI assay (Fig. 30.3). It was based on the principle that the PI enters only membrane-compromised cells, after which the fluorescence of this probe is enhanced by 20–30-fold due to its binding to nucleic acids. A significant increase in fluorescence intensity was observed in the ethanol-treated groups. The fluorescence intensity increased significantly as the ethanol concentration gradually added, simultaneously the fluorescence intensity increased as the treatment time extended.

The membrane-permeabilizing activity of starting strain and mutant strain were compared under different concentration of ethanol. As shown in Fig. 30.4, no matter what the ethanol concentration is, the membrane-permeabilizing activity of starting strain is significantly higher than the mutant strain. Shafiei et al. [16] found ethanol and acetic acid could affect the cell envelope integrity of *Acetobacter senegalensis* by fluorescence staining. Therefore, the ethanol tolerance of U1-1 may be partly

284 X. Wu et al.

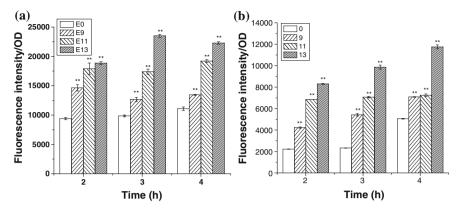


Fig. 30.3 Membrane-disrupting activity as determined in the PI assay. a Starting strain; b mutant strain. Each value is the mean of three replicates \pm SD. The *asterisk* (*) indicates significant difference from equivalent control point (t test, P < 0.05)

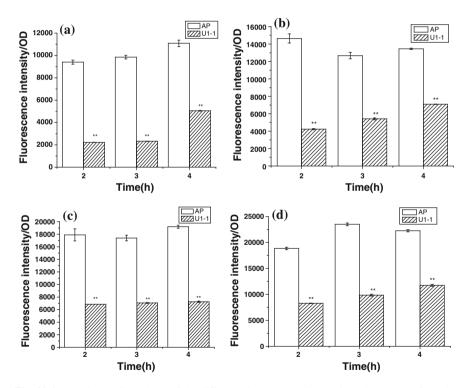


Fig. 30.4 Membrane-disrupting activity difference between starting strain and mutant strain under different concentration of ethanol. **a** Without addition of ethanol; **b** addition of 9 % (v/v) ethanol; **c** addition of 11 % (v/v) ethanol; **d** addition of 13 % (v/v) ethanol. Each value is the mean of three replicates \pm SD. The *asterisk* (*) indicates significant difference from equivalent control point (*t* test, P < 0.05)

because of the change of membrane-permeabilizing activity, so that high concentration of ethanol will not make membrane permeability increased, leading to cell death.

30.4 Conclusion

The present study has revealed that the helium-based ARTP altered certain biochemical characteristics of *A. pasteurianus* parent type strain. As a result, a mutant strain named U1-1 with ethanol tolerance and high yield was isolated due to mutagenesis. The yield of acetic acid reached 32.83 ± 0.75 g/L, 385.7 % higher than that of the starting strain. Furthermore, the membrane permeability of mutant strain is less than the starting strain by PI assay, which may be the reason why U1-1 has ethanol tolerance. In order to confirm this inference more research is needed, for instance, lipidomics and metabonomics.

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Chapter 31

A Thermotolerant *Acetobacter* pasteurianus T24 Achieving Acetic Acid Fermentation at High Temperature in Self-Adaption Experiment

Yuqiao Wei, Xiaoying Wu, Zeming Xu, Zhilei Tan and Shiru Jia

Abstract The aim of the present work was to improve the thermotolerance of acetic acid bacteria. In this study, a thermotolerant *Acetobacter pasteurianus* T24 was obtained through adaptive experiments. The strain T24 exhibited better growth at 40 °C in solid medium. The strain T24 exhibited rapid growth and the wild stain HN101 exhibited a longer lag phase at 40 °C in acetic acid fermentation than strain T24. Under the condition of low ethanol concentration, the highest acetic acid concentration produced by T24 at 40 °C increased by 18.05 % over the wild stain HN101. These results indicated that the adapted strain had acquired thermotolerance over the course of adaptation. Thus, this strain was used for acetic acid fermentation at high temperature. This work reveals the potential value for improvement in industrial vinegar production.

Keywords Acetobacter pasteurianus · Thermotolerance · Acetic acid fermentation

31.1 Introduction

Acetic acid bacteria (AAB) comprise a group of gram-negative or gram-variable, ellipsoidal to rod-shaped cells that have an obligate aerobic metabolism with oxygen as the terminal electron acceptor. In the first classification of AAB, two main genera were determined as *Acetobacter* and *Gluconobacter*, but nowadays 12 genera are recognized and accommodated to the family *Acetobacteraceae*, the *Alphaproteobacteria*: *Acetobacter*, *Gluconobacter*, *Acidomonas*, *Gluconacetobacter*, *Asaia*, *Kozakia*, *Swaminathania*, *Saccharibacter*, Neoasaia, *Granulibacter*,

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288 Y. Wei et al.

Tanticharoenia, and Ameyamaea [1]. Isolation, purification, identification, and preservation of AAB are very difficult. Since Acetobacter prefers to oxidize ethanol more strongly than glucose and Gluconobacter prefers glucose more than ethanol, most strains useful in vinegar manufacture belong to Acetobacter [2]. Acetobacter pasteurianus is a typical example for acetic acid fermentation.

AAB during acetic acid fermentation expose the acetic acid bacteria to various stresses such as high temperature, ethanol, and acetic acid. Temperature is one of the most important environmental factors affecting growth and survival of microbes [3]. Although temperature plays an important role in the fermentation process, large-scale industrial fermentation could not be carried out in strict temperature control. In recent years, temperature in summer has gradually been increasing in many countries. Almost all domestic vinegar manufacturers can produce vinegar for only half a year. Furthermore, the fact that fermentation temperature was improved can reduce cooling costs. Therefore, more thermotolerant microbes will be required to achieve stable fermentation.

Some researches attempted to achieve this goal. Ndoye et al. [4] have isolated a thermotolerant acetic acid bacterium from mango fruit. This strain can survive at 40 °C. Lu et al. [2] have obtained a thermotolerant and high acetic acid producing bacterium *Acetobacter* sp. I14-2. This bacterium is thermotolerant and retained 97 and 68 % of acetic acid producing activity after 3d incubation at 35 and 37 °C, respectively, compared with that when incubated at 30 °C. Akiko et al. [5–7] researched two groups of chaperones GroES/GroEL and dnaK/dnaJ of AAB. After the modification of molecular tools, the strains of two groups of chaperones overexpressed can survive at 42 °C.

In this study, a thermotolerant *A. pasteurianus* T24 was obtained through adaptive experiments. Comparison of growth and acetic acid production between *A. pasteurianus* HN101 and *A. pasteurianus* T24 at high temperature was researched. As a result, *A. pasteurianus* T24 enhanced acetic acid production at 40 °C.

31.2 Materials and Methods

31.2.1 Bacterial Strains

A wild strain *A. pasteurianus* HN101 and a thermotolerant strain *A. pasteurianus* T24 were used in this study. *A. pasteurianus* HN101 was used for a comparison of thermotolerance with this adapted strain.

31.2.2 Culture Media

Two different media were used in this study: GYC solid medium (2 % glucose, 1 % yeast extract, 1 % CaCO₃, 3 % ethanol and 2 % agar, pH 6.8) and GY liquid

medium (2 % glucose, 1 % yeast extract and 3 % ethanol, pH 6.8). All the chemicals used in this study were purchased from No. 1 Chemical Reagent Factory, Tianjin, China.

31.2.3 Culture Conditions

Shaking-flask fermentation for vinegar production was performed in 250-mL Erlenmeyer flasks. For the preculture, *A. pasteurianus* HN101 was inoculated into the Erlenmeyer flask containing 50 mL GY medium and cultured at 30 °C, 180 rpm for about 24 h, until the $\rm OD_{600}$ reached 0.6–0.8. For the main culture, 2.5 mL seeds were mixed with 47.5 mL fresh GY medium and grown at 180 rpm at different temperatures (30 and 40 °C) for 96 h.

For plate assay, each strain was cultured in GY medium at 30 °C, 180 rpm for 24 h, and then diluted with sterilized 0.9 % NaCl solution $(10^{-1}, 10^{-2}, 10^{-3}, 10^{-4}$ and 10^{-5}). The diluted solutions (5 μ L) were spotted onto GYC agar plates. The plates were cultured in an incubator at various temperatures (30 and 40 °C).

31.2.4 Biomass Estimation

Bacterial growth was measured using a spectrophotometer by absorbance maxima (600 nm). Acetic acid concentrations of culture medium at several growth phases were measured by alkali titration with 0.1 M NaOH using 1 % phenolphthalein as a pH indicator.

31.3 Results and Discussions

31.3.1 A Thermotolerant Strain Adaptation to High Temperature

The thermal adaptation was carried out in an unconventional way (Fig. 31.1): one involved repeating the cultivation under fermentation conditions at 40 °C. The cultivation was repeated 12 times. Then, the culture broth was diluted to 10^{-1} , 10^{-2} , 10^{-3} , 10^{-4} , and 10^{-5} using sterilized 0.9 % NaCl solution. The diluted solutions (100 µL) were spread onto GYC agar plates. The plates were cultured in an incubator at 40 °C for 60 h. At last, all single colonies were spotted onto GYC agar plates in the same conditions. As a result, a thermotolerant *A. pasteurianus* T24 was screened (Fig. 31.2). As shown in the Fig. 31.2, the strain T24 exhibited rapid acetic acid fermentation ability under the same cultivation time. Thus, this strain was used

290 Y. Wei et al.

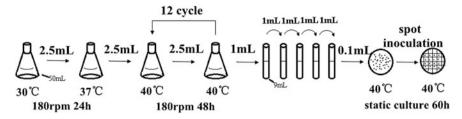


Fig. 31.1 Scheme for self-adaption treatment to high temperature

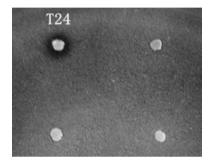


Fig. 31.2 The thermotolerant *A. pasteurianus* T24 has been screened in GYC medium at 40 °C. The clear zone surrounding bacterial colony indicated a hydrolysis of CaCO₃ to H₂CO₃ by acetic acid, which is a product obtained from ethanol oxidation

for acetic acid fermentation at high temperature. Yoshinao et al. [8] and Illeghems et al. [9] revealed genetic instability of *A. pasteurianus* through whole-genome analyses. To obtain a thermotolerant stain, this method is available. Qi et al. [10] also used self-adaption under acidic stress to obtain a mutant with higher vinegar production.

31.3.2 Functional Characterization of a Thermoadapted Strain

To confirm the thermotolerance of the mutant, growth comparisons were carried out on a GYC plate (Fig. 31.3). The growth of the strain T24 was not significantly different from wild-type strain HN101 at 30 °C, while the strain T24 exhibited better growth at 40 °C. These results indicate that the strain T24 had acquired thermotolerance over the course of adaptation. Under the same dilution ratio, it is reliable to prove the thermotolerance of the mutant. Hiromi et al. [11] also used this method to confirm the thermotolerance of *Gluconobacter frateurii* CHM43.

To examine the difference in fermentation ability between the wild and adapted strains, the growth was compared in a flask shaking culture with GY medium until

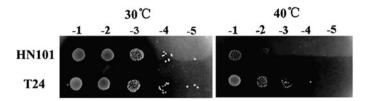


Fig. 31.3 Growth comparison of A. pasteurianus HN101 and A. pasteurianus T24 strains at 30 and 40 $^{\circ}\mathrm{C}$

acetic acid was accumulated (Fig. 31.4). As shown in the Fig. 31.4, the growth and acetic acid concentration of the strain T24 were not significantly different from the wild-type strain HN101 at 30 °C. In the acetic acid fermentation, AAB exhibit three growth phases: first growth proceeds by completely oxidizing ethanol to acetic acid (ethanol oxidation phase), then growth stops after consuming the ethanol as the respiratory substrate and the viable cell number gradually decreases in this stationary phase (acetic acid resistance phase). When the viable cell number decreases to some threshold, cell growth starts again by utilizing acetic acid accumulated (acetate overoxidation phase). In the ethanol oxidation phase, cells oxidize ethanol

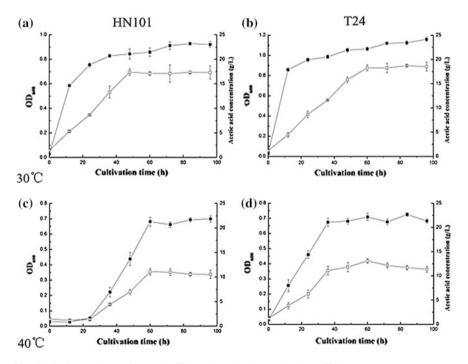


Fig. 31.4 Comparison of growth (\blacksquare) and acetic acid production (\square) between *A. pasteurianus* HN101 and *A. pasteurianus* T24 in flask-shaking cultures at 30 °C (\mathbf{a} , \mathbf{b}) and 40 °C (\mathbf{c} , \mathbf{d})

to acetic acid via acetaldehyde, which is carried out by two sequential reactions with membrane-bound alcohol and aldehyde dehydrogenases linked to the respiratory chain [12]. In the acetic acid resistance phase, the cells resist against acetic acid produced themselves by several mechanisms without assimilating the acetic acid, but finally the cells start to assimilate the acetate by oxidizing it to CO₂ via TCA cycle, of which the phase is called as "overoxidation phase" because of this phenomenon is nuisance for acetic acid fermentation [13, 14]. Thus, it was presumed that the substrate ethanol addition influenced the growth and acetic acid production. The next step was to examine the influence of ethanol addition change.

However, the strain T24 exhibited rapid growth and the wild stain exhibited a long lag phase at 40 °C. Besides, the strain T24 achieved the highest acetic acid production at 60 h. The highest acetic acid concentration produced by two strains at 40 °C was 11.08 g/L (HN101, 60 h) and 13.08 g/L (T24, 60 h), respectively. It has been shown that some acetic acid bacteria grew well at 37 °C, but exhibited largely delayed fermentation at 39 °C [15]. The results match with the previous study. The strain T24 will proceed in higher temperature and is suitable for industrial vinegar production without strict temperature control.

31.4 Conclusions

The thermotolerant *A. pasteurianus* T24 was obtained through self-adaption at high temperature. This method is available. The strain T24 exhibited better growth at 40 °C in GYC medium. The strain T24 exhibited rapid growth and the wild stain HN101 exhibited a longer lag phase at 40 °C in acetic acid fermentation than strain T24. The strain T24 had acquired thermotolerance over the course of adaptation. Under the condition of low ethanol concentration, the highest acetic acid concentration produced by T24 at 40 °C increased by 18.05 % over the wild stain HN101. Certainly, further studies on culture conditions and thermotolerant mechanisms of the adapted strain will be taken into consideration in future work.

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Chapter 32 Optimization of Cultural Conditions for Extracellular Polymeric Substances (EPS) Production by *Burkholderia* Using Response Surface Methodology

Baojiang Sun, Peipei Han, Ruyu Tao, Qixiu Pang and Shiru Jia

Abstract Extracellular polymeric substances (EPS) are complex biopolymers secreted by a variety of microorganisms. Except for the basic protecting cells from rigorous environment, EPS also influenced the bioflocculation and settleability of sludge. There were reports about the application of EPS in minimizing harbor siltation. On the basis of this point, a series of experiment were proceeded. Based on single factor experiment, the cultural conditions for EPS production were investigated and optimized by response surface methodology (RSM) at the Box–Behnken design. Optimal EPS yield (6.40 g/L) was obtained at a combination of sucrose 21.0 g/L, NH₄Cl 1.1 g/L, MgSO₄ 0.2 g/L. And the influence of NH₄Cl for EPS yield was the most significant, as the rest in order was sucrose and MgSO₄. A verification experiment confirmed the reliability of the model.

Keywords Optimization • Extracellular Polymeric substances (EPS) • Response surface methodology (RSM)

32.1 Introduction

Extracellular polymeric substances (EPS), a complex high molecular weight mixture of biopolymers, are secreted by a variety of microorganisms during their growth to protect the cells from severe environment, and have been found widely

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296 B. Sun et al.

present in pure cultures, activated sludge, granular sludge, and biofilms [1–3]. EPS are composed of a series of organic materials [4], carbohydrate and protein were the primary constituents in many EPS of activated sludge [5] and pure cultures [6], with nucleic acids, lipids [7] and other polymers being present in lower concentration. EPS have been reported with a significant influence on the physicochemical properties of microbial aggregates and activated sludge, such as the structure, floc size, surface charge, bioflocculation, and settleability [8, 9]. Recently, a new point was brought up that the application of EPS in minimizing harbor siltation [10]. Briefly, EPS secreted by microorganisms from the harbor could interact with the mud. As a result, some characters of the mud changed, sequentially the settleability of the mud was changed. In order to verify whether the EPS produced by isolated microorganism had reduced the settling velocity and finally delayed the settling, and furthermore to find out the mechanism of EPS delaying the settling, amounts of EPS would be needed. In consideration of the truth that there was little available medium and cultural conditions for the new isolated microorganism to produce EPS, it was necessary to find a suitable medium and optimize the cultural conditions for EPS production.

Statistical experimental design methods provide a systematic and efficient means of reaching particular goals and simultaneously studying several control factors [11]. Hence, these methods can be used to examine and optimize the operational variables. Response surface methodology (RSM) was one of applied experimental design frequently. And RSM had been used in broad fields, to optimize the growth and proteolytic activity [12] and production of xanthan gum [13], all acquiring a good job.

In the present study, optimization of EPS production in batch experiments was attempted through RSM, where the effects of three variables (sucrose, NH_4Cl , $MgSO_4$) were investigated.

32.2 Materials and Methods

32.2.1 Bacterial Strain and Cultural Conditions

Burkholderia, used throughout the present study, was previously screened from the mud, Port of Lianyungang China. The organism was grown on seed culture medium [14] at 28 °C for 24 h in 160 r/min, then inoculated to fermentation culture at the inoculum size of 10 %, incubated for 3 days at 28 °C in 160 r/min. The biomass was represented by determining the absorbance of culture at 600 nm.

The seed culture medium (pH 7.2) contained: 5 g/L Glucose, 10 g/L Peptone, 3 g/L Beef extract, 2 g/L MgSO₄· 7 H₂O.

The composition of basic fermentation medium [15] utilized for the EPS production was as follows: 25 g/L glucose, 1 g/L NH₄Cl, 0.2 g/L MgSO₄·7H₂O, 2 g/L Na₂HPO₄, 1 g/L KH₂PO₄, and 1.5 g/L CaCO₃. The initial pH of media was adjusted to 7.0.

32.2.2 EPS Harvesting

After incubation (3 days), the medium was centrifuged at 4,000 r/min for 15 min after diluted with distilled water to separate the supernatant and cell pellet. The EPS in the supernatant was precipitated with 3 volumes of prechilled ethanol, and further separated by centrifugation. Subsequently, the quantity of EPS was expressed by dry weight [16].

32.2.3 Optimization of Cultural Conditions for EPS Production

32.2.3.1 The Single Factor Experiment

On the basis of basic fermentation medium, carbon source (glucose, sucrose, maltose, lactose), nitrogen source (NH₄Cl, (NH₄)₂SO₄, peptone, yeast extract) and MgSO₄ on EPS production were determined to acquire the optimum carbon source, nitrogen source and their respective central point concentration. Magnesium used as the activator of lots of enzymes for metabolism in organism was crucial for the synthesis of various products and was the necessary microelement. As a result, MgSO₄ was chosen as the mineral salt to optimize for EPS production.

32.2.3.2 Response Surface Methodology Design

Based on above experiments, the variables used for further optimization studies by RSM were sucrose, NH_4Cl , $MgSO_4$, each at three-code levels (-1, 0, 1), as shown in Table 32.1. Design-Expert 8.0 was used to analyze the data of RSM, the treatment schedule for the model is given in Table 32.2.

RSM is an empirical statistical modeling technique employed for multiple regression analysis using quantitative data obtained from properly designed experiments [17]. Box–Behnken statistical design is one type of RSM design that is an independent quadratic design having the treatment combinations at the midpoints of the edges of the process space and at the center [18].

Table 32.1 Factors and levels of Box–Behnken design

Levels	Factors		
	Sucrose	NH ₄ Cl	MgSO ₄
	A (g/L)	B (g/L)	C (g/L)
-1	15	1.0	0.15
0	20	1.2	0.20
1	25	1.4	0.25

298 B. Sun et al.

Runs	Sucrose (g/L)	NH ₄ Cl (g/L)	MgSO ₄ (g/L)	EPS yield (g/L)
1	0	0	0	6.45
2	0	0	0	5.96
3	0	0	1	5.53
4	-1	-1	0	4.44
5	1	1	-1	5.02
6	-1	1	0	2.44
7	0	0	0	5.76
8	0	0	0	5.42
9	-1	0	1	3.41
10	-1	0	-1	2.93
11	1	0	1	4.55
12	1	-1	0	4.85
13	0	1	-1	2.02
14	0	0	0	6.35
15	0	1	1	2.98
16	0	-1	-1	5.78
17	1	1	0	3.28

Table 32.2 Box-Behnken design arrangement and the experimental data

32.2.4 Statistical Analysis

All experiments were carried out in triplicate and results represent mean standard \pm error.

32.3 Results and Discussions

32.3.1 Single-Factor Experiment

Results of optimal carbon source and nitrogen source were shown in Fig. 32.1. EPS production was not detected while maltose and lactose were used as carbon source, respectively. The same result could be seen while (NH₄)₂SO₄, peptone, yeast extract were chosen for nitrogen source, but the biomass was high in each condition indicating that these nitrogen sources just promoted the growth of organisms, not the production of EPS. Some conclusions could be drawn that sucrose and NH₄Cl were the suitable carbon and nitrogen sources, respectively, and the optimal concentration was 20 g/L sucrose, 1.2 g/L NH₄Cl and 0.2 g/L MgSO₄ (Fig. 32.1b, d, e).

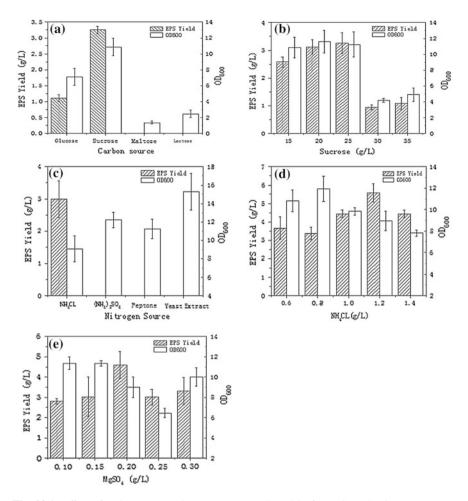


Fig. 32.1 Effect of carbon source, nitrogen source, and MgSO₄ for EPS production

32.3.2 Response Surface Methodology Design

The actual yield of EPS (response) obtained was listed in Table 32.2. A second-order polynomial model was fitted for the production of EPS (Y), given an equation as follows:

$$Y = 5.99 + 0.56A - 1.24B + 0.090C + 0.11AB - 0.24AC + 0.29BC - 1.17A^2 - 1.07B^2 - 0.84C^2$$

The quadratic terms A^2 , B^2 , C^2 of that model were significant, and the *F*-value was 10.74. All these could be shown obviously in Table 32.3. The significant level *p* of the model was less than 0.005 which indicated that the model was significant.

300 B. Sun et al.

Source	Sum of squares	DF	Mean square	F-Value	Prob > F	Significance
A	2.51	1	2.51	7.95	0.0258	*
В	12.20	1	12.20	38.65	0.0004	*
С	0.065	1	0.065	0.21	0.6642	
AB	0.046	1	0.046	0.15	0.7133	
AC	0.23	1	0.23	0.71	0.4258	
BC	0.34	1	0.34	1.16	0.3173	
A^2	5.74	1	5.74	18.19	0.0037	*
B ²	4.80	1	4.80	15.21	0.0059	*
C ²	2.99	1	2.99	9.47	0.0179	*
Model	30.50	9	3.39	10.74	0.0025	*
Residual	2.21	7	0.32			
Lack of fit	1.49	3	0.50	2.76	0.1760	
Pure error	0.72	4	0.18			
Total	32.71	16				

Table 32.3 Variance analysis of regression model

In general, the lack of fit represent the conditions that regression model failed to describe experimental point which was out of the regression scope. In this study, lack of fit was not significant, indicating that the equation was well fitting the experiment.

The response surface slope and contour graphically reflected the interaction between various factors. The gentle and steepness of the response surface directly showed the response sensitivity level, for example, the response was slow when the slope was gentle [19]. The point on the corners and center of the contour represented the experimental point, and the contour of ellipse demonstrated the interaction between two factors was significant. Figure 32.2 showed the NH₄Cl has a more negative influence for EPS production than sucrose and MgSO₄, inferring that it needs a high ratio between carbon and nitrogen source to produce EPS. When the concentration of sucrose approximately ranged from 19 to 23 g/L (Fig. 32.2b, d), effects of NH₄Cl and MgSO₄ for EPS yield was low. The significant level p of C, AB, AC, BC was more than 0.05, which indicate that MgSO₄ was not the important factor for EPS production, and there were little interaction between the factors. It may be explained that the magnesium was the microelement which the cells needed. The range of MgSO₄ concentration of experiment designed was sufficient for EPS production and the concentration of MgSO₄ did not inhibit the EPS production. It could be summarized that the optimum range of sucrose and NH₄Cl were around 20 and 1.1 g/L, respectively, and the influence of MgSO₄ would be very low when the other two were in optimum range. In addition, NH₄Cl and sucrose were two significant factors for EPS production, according to Table 32.3.

The optimum levels of the variables were analyzed by Box–Behnken design using the function of "Numerical" in the Design Expert (8.0). The predicted concentration of each factors were sucrose 21.10~g/L, $NH_4Cl\ 1.08~g/L$, $MgSO_4\ 0.2~g/L$,

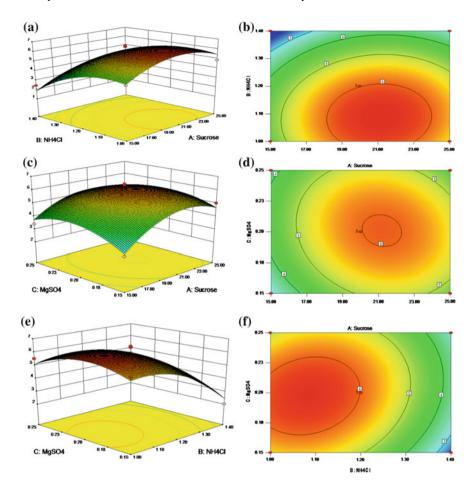


Fig. 32.2 Response surface plots and contour plots

giving a highest EPS yield of 6.40 g/L. A verification experiment was carried out to confirm the reliability of the model under optimal conditions. Maximum EPS yield was 6.26 ± 0.15 g/L, closely agree with the predicted value. To make the actual operation easier, the cultural condition was amended to the condition that were sucrose 21.0 g/L, NH₄Cl 1.1 g/L, MgSO₄ 0.2 g/L. The final EPS yield of optimization process increased obviously compared to the initial cultural condition.

32.4 Conclusions

This paper presented a study of optimizing the cultural conditions associated with EPS production by RSM. Results of regression analysis indicated that the influence of NH_4Cl for EPS yield was the most significant, as the rest in order was sucrose

B. Sun et al.

and MgSO₄. The interaction between any two variables was not significant. The results predicted by Box–Behnken design in RSM analysis showed that a combination of sucrose 21.0 g/L, NH₄Cl 1.1 g/L, MgSO₄ 0.2 g/L, gave rise to the maximum EPS yield. Furthermore, this study laid a good foundation for the subsequent mechanism research of reducing the settling velocity of mud by EPS, as well as the study of characteristic of EPS.

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Chapter 33 Effects of pH, Temperature, Storage Time, and Protective Agents on Nisin Antibacterial Stability

Zhilei Tan, Jing Luo, Fang Liu, Qian Zhang and Shiru Jia

Abstract Nisin as a kind of bacteriocin and peptides can inhibit the growth of gram-positive bacteria effectively, which is easily influenced by heat, alkali, and storage time significantly. In this study, it is concluded that adding substances such as chitosan, ascorbic acid, tyrosine, and FeSO₄ do well in the enhancement of nisin antibacterial stability. The protective agents have been proposed to improve stability and long-term effectiveness of nisin. Above all chitosan was of great efficiency, followed by FeSO₄ and tyrosine, but ascorbic acid was better in acidic conditions. Similar phenomenon occurred when nisin was at elevated pH and high temperature. At pH 6.0, room temperature, nisin titer got rise to 154.7 IU/mL after added chitosan compared to the original 105.6 IU/mL, and even after the 121 °C heating for 20 min, the titer was 2.8 times higher than control.

Keywords Nisin • Titer • Antibacterial stability • Protective agents

33.1 Introduction

Nisin is a 34-amino acid lantibiotic, containing 3.5 kDa polypeptide with antimicrobial activity [1], and has been approved by WHO as a preservative in food [2]. Several studies have reported that the antimicrobial efficacy of nisin was reduced when it was applied in foods. In its free form, nisin can react with reducing sugars through Maillard reaction and can nonspecifically bind with lipids and proteins, and hence decrease its antibacterial activity [3]. In addition, seen from the structure of nisin, it contains unsaturated amino acids Dhb2, Dha5 and Dha33, which make it

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Z. Tan et al.

easy to occur intermolecular nucleophilic addition reaction under alkaline pH conditions, making the reductions of solubility and antibacterial activity [4].

In order to overcome these problems, alternative procedures that target the sustained release of nisin have been proposed. Many studies applied liposomes, polymeric capsules, or films to improve nisin biological activity especially for long storage or heat treatment [3, 5]. However, some drawbacks are attributed to liposomes as a nisin carrier including inactivation and fast leakage due to interaction of nisin with liposomes, low stability, and high cost [6, 7]. Our study was aimed to find out a more effective but simple way to solve the poor antibacterial stability of nisin. Chitosan (CS), ascorbic acid (VC), tyrosine (Tyr), and FeSO₄ were chosen as protective agents to promote the antibacterial stability of nisin. There exist a strong intra- and intermolecular hydrogen bonds in aqueous media in CS molecular [8], which may result in a solid combination between nisin and CS. And biocompatible organic acids could be interacted with nisin to form hydrogen bond, which may lead to increased solubility and retention of nisin activity at various pH [9]. Besides, Tyr belongs to a phenolic group, and the presence of it accounts for a great binding between surfactant head group and lipid [10], perhaps giving rise to a better antibacterial activity of nisin. In addition, particulate iron has important catalytic effects on reactions that produce reactive oxygen species [11], which may prevent the intermolecular nucleophilic addition reaction of nisin.

33.2 Materials and Methods

33.2.1 Materials and Equipment

Nisin was gifted by New Silver Elephant Biological Engineering Co., Ltd. Tyr, VC, and CS were purchased from Solarbio Science and Technology Co., Ltd. Other drugs were all analytically pure chemical reagents.

33.2.2 Agar Diffusion Method of Nisin Titer Test [12]

Standard nisin is 1,000 IU/mg (dissolved in the sterile 0.02 N HCl, and was diluted to some titers ranging from 5 to 100 IU/mL). Measured 100 μ L Nisin standard solution in every Oxford cup, incubated overnight at 33 °C, and then measured the diameters of bacteriostatic rings. Drew the standard curve which takes diameters (mm) of bacteriostatic rings as the abscissa and logarithms of nisin titers as the ordinate (y = 0.2618x - 1.3979, $R^2 = 0.9934$). The calculation formula for titer loss rate [13] as formula (33.1) shows:

Nisin titer loss rate =
$$\frac{C_1 - C_2}{C_1} \times 100 \%$$
 (33.1)

In the formula, C_1 is for untreated nisin titer, C_2 is for nisin titer measured after processing.

33.2.3 Effects of pH, Temperature, and Storage Time on Nisin Titer

Due to the rare applications of nisin in the alkaline pH condition, the research focused on the pH ranging from 2.0 to 6.0. After nisin solution with different pH (2.0, 4.0 and 6.0, 0.5 g/L) treated at 25, 85, 100 and 121 °C for 20 min, the nisin titers were tested and the titer loss rates were calculated, respectively. Nisin solution (0.5 g/L) with different pH (2.0, 4.0, 6.0) were heated at 121 °C for 20 min and stored in 4 °C for 30 days, then the titers were tested and compared against unheated control solution.

Furthermore, to protect nisin antibacterial stability under different heating (25, 85, 100, and 121 $^{\circ}$ C) and pH (2.0, 4.0, and 6.0) treatments, VC, Tyr, CS, and FeSO₄ were added at the mass ratio of nisin 1:1, 2:1, 1:1, and 2:1 before such treatments, respectively. And then the mixtures were incubated with shaking 200 rpm in 25 $^{\circ}$ C for 30 min to assure the reaction between them.

33.3 Results and Discussions

33.3.1 Effects of pH, Temperature, and Storage Time on Nisin Titer

33.3.1.1 Effects of pH and Temperature on Nisin Titer

The stability of nisin was studied in the pH range of 2.0–6.0 and at 25, 85, 100 and 121 °C, optimal stability was observed at pH 2.0, 25 °C. As shown in Table 33.1, the antibacterial activity continuously decreased with increasing pH and temperature, showing a greater than 5-fold reduction when the pH was increased from 2.0 to 6.0 at RT (room temperature).

Synergistic effects of temperature and pH on nisin titers were revealed. The titer loss rate was below 40.0 % after nisin solution (pH 2.0) was heated under temperature 100 °C. However, compared to 489.3 IU/mL at RT, the highest titer loss rate was 68.6 % (declined to 153.5 IU/mL) after an elevated temperature (121 °C) heated. The phenomenon is in line with the report of Yun et al. [14]. Besides, pH has greater impact on nisin activity than temperature. The nisin titer loss rate at RT was 29.5 % (pH 4.0) and 78.4 % (pH 6.0) in contrast with control (pH 2.0), respectively. And with the temperature increased from RT to 121 °C, the analogous results were observed that the higher the temperature and pH of nisin, the lower titer

308

Table 33.1 Increase rates of nisin titers on the condition of different pH and temperatures while adding CS, Tyr, VC, and FeSO₄

$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$						•			() () () () () () () () () ()		•	
25 489.31 ± 61.05³ - 519.88 ± 41.90³ 0.06° 569.68 ± 30.23³ 0.16° 556.49 ± 88.91³ 0.14° 85 295.25 ± 48.61 39.66³ 605.70 ± 43.04 1.05 669.45 ± 39.98 1.27 798.73 ± 71.47 1.71 100 197.15 ± 15.50 59.71 485.49 ± 49.25 1.46 485.49 ± 49.25 1.46 572.26 ± 24.63 1.90 121 153.51 ± 18.42 68.63 234.05 ± 58.20 0.52 253.43 ± 21.01 0.65 204.83 ± 8.04 0.33 25 344.94 ± 53.36 29.51 410.09 ± 22.87 0.19 440.85 ± 31.70 0.28 442.92 ± 25.24 0.28 100 99.88 ± 9.26 79.59 235.52 ± 12.91 1.35 317.02 ± 114.51 2.17 128.23 ± 6.02 0.28 121 35.08 ± 4.12 92.83 72.21 ± 9.83 1.06 136.40 ± 29.44 2.89 152.15 ± 22.10 0.28 25 105.64 ± 46.10 78.41 168.04 ± 6.38 0.59 140.83 ± 5.51 0.33 154.72 ± 14.42 0.47 <td>hd</td> <td>(°C)</td> <td>Nisin</td> <td></td> <td>Nisin-VC</td> <td></td> <td>Nisin-Tyr</td> <td></td> <td>Nisin-CS</td> <td></td> <td>Nisin-FeSO₄</td> <td></td>	hd	(°C)	Nisin		Nisin-VC		Nisin-Tyr		Nisin-CS		Nisin-FeSO ₄	
85 295.25 ± 48.61 39.66b 605.70 ± 43.04 1.05 669.45 ± 39.98 1.27 798.73 ± 71.47 1.71 100 197.15 ± 15.50 59.71 485.49 ± 49.25 1.46 485.49 ± 49.25 1.46 572.26 ± 24.63 1.90 121 153.51 ± 18.42 68.63 234.05 ± 58.20 0.52 253.43 ± 21.01 0.65 204.83 ± 8.04 0.33 25 344.94 ± 53.36 29.51 410.09 ± 22.87 0.19 440.85 ± 31.70 0.28 442.92 ± 25.24 0.28 85 137.15 ± 12.14 71.97 264.57 ± 10.54 0.93 437.41 ± 92.77 2.19 277.55 ± 47.14 1.02 100 99.88 ± 9.26 79.59 235.52 ± 12.91 1.35 317.02 ± 114.51 2.17 128.23 ± 6.02 0.28 121 35.08 ± 4.12 92.83 72.21 ± 9.83 1.06 136.40 ± 29.44 2.89 152.15 ± 22.10 3.34 25 105.64 ± 46.10 78.41 168.04 ± 6.38 0.59 140.83 ± 5.51 0.33 154.72 ± 14.42 0.47	2.0	25	489.31 ± 61.05^{a}	I	519.88 ± 41.90^{a}	0.06°	569.68 ± 30.22^{a}	0.16^{c}	556.49 ± 88.91^{a}	0.14^{c}	592.48 ± 34.60^{a}	0.21^{c}
100 197.15 ± 15.50 59.71 485.49 ± 49.25 1.46 485.49 ± 49.25 1.46 572.05 ± 24.63 1.90 121 153.51 ± 18.42 68.63 234.05 ± 58.20 0.52 253.43 ± 21.01 0.65 204.83 ± 8.04 0.33 25 344.94 ± 53.36 29.51 410.09 ± 22.87 0.19 440.85 ± 31.70 0.28 442.92 ± 25.24 0.28 100 99.88 ± 9.26 79.59 235.52 ± 12.91 1.35 317.02 ± 114.51 2.19 277.55 ± 47.14 1.02 121 35.08 ± 4.12 92.83 72.11 ± 9.83 1.06 136.40 ± 29.44 2.89 152.15 ± 22.10 3.34 25 105.64 ± 46.10 78.41 168.04 ± 6.38 0.59 140.83 ± 5.51 0.33 154.72 ± 14.42 0.47 85 42.33 ± 12.82 91.35 85.03 ± 0.51 1.01 110.66 ± 3.37 1.61 113.28 ± 19.39 1.68 100 35.76 ± 2.058 92.69 50.63 ± 19.66 0.42 77.68 ± 3.97 1.17 89.82 ± 1.70 1.51		85	295.25 ± 48.61	39.66 ^b	605.70 ± 43.04	1.05	669.45 ± 39.98	1.27	$ 798.73 \pm 71.47 $	1.71	$ 412.57 \pm 76.28 $	0.40
121 153.51 ± 18.42 68.63 234.05 ± 58.20 0.52 253.43 ± 21.01 0.65 204.83 ± 8.04 0.33 25 344.94 ± 53.36 29.51 410.09 ± 22.87 0.19 440.85 ± 31.70 0.28 442.92 ± 25.24 0.28 85 137.15 ± 12.14 71.97 264.57 ± 10.54 0.93 437.41 ± 92.77 2.19 277.55 ± 47.14 1.02 100 99.88 ± 9.26 79.59 235.52 ± 12.91 1.35 317.02 ± 114.51 2.17 128.23 ± 6.02 0.28 121 35.08 ± 4.12 92.83 72.11 ± 9.83 1.06 136.40 ± 29.44 2.89 152.15 ± 22.10 3.34 25 105.64 ± 46.10 78.41 168.04 ± 6.38 0.59 140.83 ± 5.51 0.33 154.72 ± 14.42 0.47 85 42.33 ± 12.82 91.35 85.03 ± 0.51 1.01 110.66 ± 3.37 1.61 113.28 ± 19.39 1.68 100 35.76 ± 2.058 92.69 50.63 ± 19.66 0.42 77.68 ± 3.97 1.17 89.82 ± 1.70 1.51		100		59.71	485.49 ± 49.25	1.46	485.49 ± 49.25	1.46	572.26 ± 24.63	1.90	278.82 ± 55.62	0.41
25 344.94 ± 53.36 29.51 410.09 ± 22.87 0.19 440.85 ± 31.70 0.28 442.92 ± 25.24 0.28 85 137.15 ± 12.14 71.97 264.57 ± 10.54 0.93 437.41 ± 92.77 2.19 277.55 ± 47.14 1.02 100 99.88 ± 9.26 79.59 235.52 ± 12.91 1.35 317.02 ± 114.51 2.17 128.23 ± 6.02 0.28 121 35.08 ± 4.12 92.83 72.21 ± 9.83 1.06 136.40 ± 29.44 2.89 152.15 ± 22.10 3.34 25 105.64 ± 46.10 78.41 168.04 ± 6.38 0.59 140.83 ± 5.51 0.33 154.72 ± 14.42 0.47 85 42.33 ± 12.82 91.35 85.03 ± 0.51 1.01 110.66 ± 3.37 1.61 113.28 ± 19.39 1.68 100 35.76 ± 2.058 92.69 50.63 ± 19.66 0.42 77.68 ± 3.97 1.17 89.82 ± 1.70 1.51 121 15.43 ± 1.01 96.85 20.31 ± 2.76 0.32 17.18 ± 0.80 0.11 58.52 ± 2.00 2.79		121	153.51 ± 18.42	68.63	234.05 ± 58.20	0.52	253.43 ± 21.01	0.65	204.83 ± 8.04	0.33	225.72 ± 21.05	0.47
85 137.15 ± 12.14 71.97 264.57 ± 10.54 0.93 437.41 ± 92.77 2.19 277.55 ±47.14 1.02 100 99.88 ± 9.26 79.59 235.52 ± 12.91 1.35 317.02 ± 114.51 2.17 128.23 ± 6.02 0.28 121 35.08 ± 4.12 92.83 72.21 ± 9.83 1.06 136.40 ± 29.44 2.89 152.15 ± 22.10 3.34 25 105.64 ± 46.10 78.41 168.04 ± 6.38 0.59 140.83 ± 5.51 0.33 154.72 ± 14.42 0.47 85 42.33 ± 12.82 91.35 85.03 ± 0.51 1.01 110.66 ± 3.37 1.61 13.28 ± 19.39 1.68 100 35.76 ± 2.058 92.69 50.63 ± 19.66 0.42 77.68 ± 3.97 1.17 89.82 ± 1.70 1.51 121 15.43 ± 1.01 96.85 20.31 ± 2.76 0.32 17.18 ± 0.80 0.11 58.52 ± 2.00 2.79	4.0		344.94 ± 53.36	29.51	410.09 ± 22.87	0.19	440.85 ± 31.70	0.28	442.92 ± 25.24	0.28	368.92 ± 48.18	0.07
100 99.88 ± 9.26 79.59 235.52 ± 12.91 1.35 317.02 ± 114.51 2.17 128.23 ± 6.02 0.28 121 35.08 ± 4.12 92.83 72.21 ± 9.83 1.06 136.40 ± 29.44 2.89 152.15 ± 22.10 3.34 25 105.64 ± 46.10 78.41 168.04 ± 6.38 0.59 140.83 ± 5.51 0.33 154.72 ± 14.42 0.47 85 42.33 ± 12.82 91.35 85.03 ± 0.51 1.01 110.66 ± 3.37 1.61 113.28 ± 19.39 1.68 100 35.76 ± 2.058 92.69 50.63 ± 19.66 0.42 77.68 ± 3.97 1.17 89.82 ± 1.70 1.51 121 15.43 ± 1.01 96.85 20.31 ± 2.76 0.32 17.18 ± 0.80 0.11 58.52 ± 2.00 2.79		85	137.15 ± 12.14	71.97	264.57 ± 10.54	0.93	437.41 ± 92.77	2.19	277.55 ±47.14	1.02	136.90 ± 21.28	0.00
121 35.08 ± 4.12 92.83 72.21 ± 9.83 1.06 136.40 ± 29.44 2.89 152.15 ± 22.10 3.34 25 105.64 ± 46.10 78.41 168.04 ± 6.38 0.59 140.83 ± 5.51 0.33 154.72 ± 14.42 0.47 85 42.33 ± 12.82 91.35 85.03 ± 0.51 1.01 110.66 ± 3.37 1.61 113.28 ± 19.39 1.68 100 35.76 ± 2.058 92.69 50.63 ± 19.66 0.42 77.68 ± 3.97 1.17 89.82 ± 1.70 1.51 121 15.43 ± 1.01 96.85 20.31 ± 2.76 0.32 17.18 ± 0.80 0.11 58.52 ± 2.00 2.79		100	99.88 ± 9.26	79.59		1.35	317.02 ± 114.51	2.17	$ 128.23 \pm 6.02 $	0.28	141.09 ± 7.65	0.41
25 105.64 ± 46.10 78.41 168.04 ± 6.38 0.59 140.83 ± 5.51 0.33 154.72 ± 14.42 0.47 85 42.33 ± 12.82 91.35 85.03 ± 0.51 1.01 110.66 ± 3.37 1.61 113.28 ± 19.39 1.68 100 35.76 ± 2.058 92.69 50.63 ± 19.66 0.42 77.68 ± 3.97 1.17 89.82 ± 1.70 1.51 121 15.43 ± 1.01 96.85 20.31 ± 2.76 0.32 17.18 ± 0.80 0.11 58.52 ± 2.00 2.79		121	35.08 ± 4.12	92.83	72.21 ± 9.83	1.06	136.40 ± 29.44	2.89	$ 152.15 \pm 22.10 $	3.34	96.74 ± 11.22	1.76
42.33 ± 12.82 91.35 85.03 ± 0.51 1.01 110.66 ± 3.37 1.61 113.28 ± 19.39 1.68 10 35.76 ± 2.058 92.69 50.63 ± 19.66 0.42 77.68 ± 3.97 1.17 89.82 ± 1.70 1.51 11 15.43 ± 1.01 96.85 20.31 ± 2.76 0.32 17.18 ± 0.80 0.11 58.52 ± 2.00 2.79	0.9	25	105.64 ± 46.10	78.41	168.04 ± 6.38	0.59	140.83 ± 5.51	0.33	$ 154.72 \pm 14.42$	0.47	211.30 ± 13.17	1.00
35.76 ± 2.058 92.69 50.63 ± 19.66 0.42 77.68 ± 3.97 1.17 89.82 ± 1.70 1.51 15.43 ± 1.01 96.85 20.31 ± 2.76 0.32 17.18 ± 0.80 0.11 58.52 ± 2.00 2.79		85	42.33 ± 12.82	91.35	85.03 ± 0.51	1.01	110.66 ± 3.37	1.61	$ 113.28 \pm 19.39 $	1.68	106.41 ± 14.49	1.51
$ 15.43 \pm 1.01 $ $ 96.85 $ $ 20.31 \pm 2.76 $ $ 0.32 $ $ 17.18 \pm 0.80 $ $ 0.11 $ $ 58.52 \pm 2.00 $ $ 2.79 $		100	35.76 ± 2.058	92.69	50.63 ± 19.66	0.42	77.68 ± 3.97	1.17	89.82 ± 1.70	1.51	87.11 ± 22.07	1.44
		121	15.43 ± 1.01	96.85	20.31 ± 2.76	0.32	17.18 ± 0.80	0.11	58.52 ± 2.00	2.79	80.30 ± 6.94	4.20

 $^{\rm a}$ Nisin titer (IU/mL), $^{\rm b}$ Nisin titer loss rate (%), $^{\rm c}$ Increase rates of nisin titer (times)

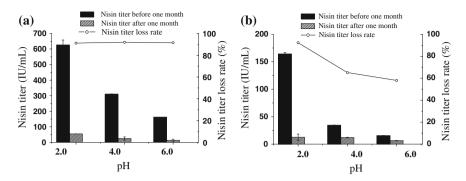


Fig. 33.1 Variation trends of nisin titers before and after a month. a, b Represent nisin titer and titer loss rate of those stored in 4 °C a month after RT or 121 °C treated for 20 min, respectively

would be. Therefore, the results indicated that in the neutral-to-elevated pH (above 6.0) region as well as high temperature (100–121 °C), the long-term stability of nisin is limited.

33.3.1.2 Effects of Storage Time on Nisin Titer

In addition to temperature and pH, effect of storage time on antibacterial activity of nisin is shown in Fig. 33.1. After nisin solution with different pH (2.0, 4.0, and 6.0) stored at 4 °C for 1 month, all the nisin titer loss rates of different heating treatments (at RT or 121 °C, 20 min) were higher than 90.0 %. Interestingly are the results of Fig. 33.1b that nisin titers decreased below 200 IU/mL after the 121 °C retreating for 20 min before storage, and upon storage for a month at 4 °C, the loss rates were declined to 65.2 and 58.4 % in pH 4.0 and 6.0 conditions, respectively, which were both lower than the 92.3 % loss rate of control (pH 2.0). But seen from the overall titer variation trends, nisin antibacterial activity was seriously decreased to less than 25 IU/mL after a month. In these cases, the storage time could well be a limiting factor with respect to practical applications.

33.3.2 Effects of Different Protective Agents on Nisin Titer

Compared to the control which has not been added any protective agent, those were of great effect on antibacterial stability. Moreover, VC, Tyr, CS, and FeSO₄ were all negative on titer-tested plates, suggesting that they do not have evident bactericidal effect under heating and pH treatments.

The results of different treatments are shown in Table 33.1. Considering the four kinds of protective agents, the effect of CS was the best, even after heating at 121 °C for 20 min, the increased titer of nisin remained upon 0.3, 3.3, and 2.8 times

Z. Tan et al.

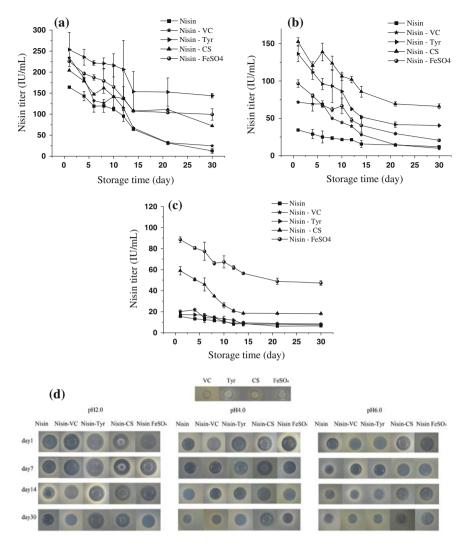


Fig. 33.2 a, b, and **c** Show nisin titer variation trends of nisin with pH 2.0, 4.0, and 6.0 via heating at 121 °C for 20 min. **d** Evidently represents antimicrobial activity of nisin at pH 2.0, 4.0, and 6.0, 121 °C for 20 min revealed by bacteriostatic rings

of pH 2.0, 4.0 and 6.0, respectively. Besides, the protective effects of VC and Tyr were better in acid conditions, especially in pH 4.0 with a 121 $^{\circ}$ C 20 min heating, the titers of nisin-VC complex and nisin-Tyr complex were as 1.1 and 2.9 times as nisin, respectively, but in pH 6.0 at the same treatment, they were both without effect. FeSO₄ was ideal in neutral environment (pH 6.0), the increased rates of nisin titer was 4.2 times at 121 $^{\circ}$ C.

As the statistical data shown in Fig. 33.2, after a monthly storage, the titer variation trends of different pH nisin solution via heating or not were both declined, but those with protective agents were higher than controls seen from Fig. 33.2, CS and FeSO₄ were always the most effective agents of the four, especially pH 6.0, in which the other two VC and Tyr expressed little effect. The same results were shown in Fig. 33.2d, which evidently revealed the antimicrobial activity of Nisin pretreated in pH 2.0, 4.0 and 6.0 and at 121 °C for 20 min revealed by bacteriostatic rings. Therefore, it becomes evident that the antibacterial stability of nisin, particularly at elevated temperatures, pH and long-term storage, were improved by the protective agents.

33.4 Conclusions

pH and temperature have synergistic effects on nisin titer. In the weak acidic region (pH 6.0 in this paper) and at an elevated pH (pH 9.0), a much lower stability is observed than that at low pH values (pH 2.0 and 4.0). When it was in weak acid and alkaline environments, the influence of temperature on nisin antibacterial stability was significant. Different pH nisin solutions were stored in 4 °C for 1 month, whose titers showed a downward trend and all the loss rates exceeded 90 %. The results implied that nisin has, especially at low pH, a good short-term stability, sufficient to survive in most heat treatments. CS was the most effective agent on nisin antibacterial stability, especially when it was at room temperature that the titer increased from 105.6 IU/mL of control to 154.7 IU/mL, even if the heating temperature was up to 121 °C, still better than other protective agents (except FeSO₄ at pH 6.0). Followed by FeSO₄ especially on weak acidic condition (pH 6.0) which can stabilize nisin antibacterial stability after heating to 121 °C for 20 min. The protective effects of Tyr and VC to nisin were somehow good, whose titers can be higher than that of the control group especially in acidic region. But with the increase of temperature and pH, the antibacterial activity was still difficult to maintain. What is more, with the increase of storage time, CS, Tyr, VC, and FeSO₄ on nisin antibacterial stability were effective all the same.

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Chapter 34 Enhanced Solvent-Stable Alpha Glycosidase Production by *Bacillus licheniformis* JXC-1 by Optimization of Feeding Strategies

Jun Fang, Qunfang Tang, Long Liu and Jianghua Li

Abstract In the previous work, we screened a strain (*Bacillus licheniformis* JXC-1) producing solvent-stable (10 % *N*,*N*-Dimethylformamide, short for DMF) α -glucosidase from several different soil samples. In this work, we attempt to improve the α -glucosidase production by optimization of feeding strategies in a 3-L fermenter. Specifically, the key factors of solvent-tolerant α -glycosidase production were investigated first, and the optimal conditions (pH 7.0, the initial maltose concentration 25 g/L, and agitation speed 600 rpm) were obtained; the enzyme activity reached 444.7 U/L under the optimal conditions. Then, four feeding strategies with different feeding rates for 4–8 h to feed maltose or both maltose and tryptone were carried out, and it was found that feeding maltose and tryptone at a rate of 2.25 mL/h (4–5 h), 6.75 mL/h (5–6 h), 9 mL/h (6–7 h), and 15 mL/h (7–8 h) significantly increased the α -glycosidase production, from 444.7 to 872.5 U/L.

Keywords Bacillus licheniformis JXC-1 \cdot Alpha glycosidase \cdot Fed-batch fermentation \cdot Solvent-stable enzymes

34.1 Introduction

The α -glucosidase (α -D-glucoside glucohydrolase; EC 3.2.1.20) is an amylase that acts on α -1,4 bonds from nonreducing end [1, 2], preferentially from short saccharides such as α -glucosides, maltooligosaccharides, and α -glucan, releasing α -glucose [3]. In addition, the released α -glucose can be transferred to another glucose, maltose, or maltotriose molecule by α -1,6 linkage [4, 5]. Because of the capacity of transglycosylation, the α -glucosidase has applications in the industrial production of isomalto oligosaccharides (IMO, isomaltose, panose, isomaltotriose,

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J. Fang et al.

and so on) and in the combination of sugars to biologically useful materials [6]. Beyond that, it is also being used as a guide to structure-based design of anti-HIV inhibitors [7].

The solvent-stable enzymes adapt to mediate cellular and metabolic processes in a solvent-rich environment and are stable in the presence of organic solvents [8, 9]. Besides, it is reported that enzyme catalysis in a nonaqueous/low-water media makes it possible to synthesize novel compounds which are difficult to synthesize conventionally and to obtain the biologically active enantiomer for which the racemic resolution is either very complex or difficult [10–12].

As a new type of efficient biological agent, glycosidases can also be used to synthesize oligosaccharides and several important glycosidic derivatives efficiently. Using high concentration of organic solvent system to reduce the concentration and activity of water can significantly reduce the hydrolysis rate of products, and prompt the reaction for the synthesis of glycosidic compounds [13]. As far as we know, the solvent-stable α -glucosidase has potential applications for synthesis of flavonoid glycosides compounds in solvent-water media.

In previous studies, we screened a strain (*Bacillus licheniformis* JXC-1) which can produce solvent-tolerant (10 % DMF) α -glucosidase from several different soil samples [14]. However, the production of α -glucosidase was very low. In this work, we attempt to establish an efficient fermentation process to improve α -glucosidase production. First, the influence of pH (uncontrolled and constant 6.0, 7.0, and 8.0), initial maltose concentration (20, 25, 30, and 35 g/L), and agitation speed (500, 600, 700, and 800 rpm) on α -glycosidase production was investigated in a 3-L fermenter. Based on the optimal fermentation conditions, four different feeding strategies which fed maltose or both maltose and tryptone at different rates for 4–8 h were investigated to improve the production of α -glucosidase.

34.2 Methods

34.2.1 Microorganism and Fermentation

The microorganism used in this work was *B. licheniformis* JXC-1, which was screened from soil in our previous work [14]. The seed culture was prepared by growing cells in 250-mL shaker flasks containing 30 mL Luria-Bertani (LB) medium inoculated with 3 % (v/v) feedstock and then cultivated for 12 h at 37 °C on a shaking flask incubator (200 rpm).

Fermentation medium contained (g/L): maltose 30, tryptone 39, yeast extract 3, $CaCl_2$ 0.4, $FeSO_4 \cdot 7H_2O$ 0.164, NaCl 0.076, and distilled water 1.0 L with natural pH at 7.0–7.5.

Cultivations were done in a 3-L fermentor (LiFlus GM BioTRON, Korea) with a working volume of 1.5 L. The inoculation amount was 3 % (v/v). The temperature in the batch culture was kept at 37 °C. During the whole fermentation process, pH was controlled automatically by a 25 % (v/v) ammonia solution and a 25 % (v/v)

hydrochloric acid solution. The agitation speed was set as 500-800 rpm while keeping aeration rate at 1vvm. Samples were taken every 2 h to measure the α -glycosidase production, reducing sugar concentration, and dry cell weight (DCW).

In batch culture, the maximum specific growth and production rate were shown approaching to 4 h while the reducing sugar was 18 g/L (Fig. 34.3). At the same time, the DCW and the production of α-glucosidase raised rapidly for 4–8 h. When the fermentation time was 6 and 8 h, the reducing sugar concentration was 15 and 7 g/L, respectively. To obtain higher production, we designed four feeding strategies at different feeding rates to keep the reducing sugar concentration at 18 g/L and changed the rate using a computer-controlled pump. The feeding rates were set as A 8.25 mL/h (4–8 h), B 4.5 mL/h (4–6 h), and 12 mL/h (6–8 h), C 4.5 mL/h (4–6 h), 9 mL/h (6–7 h), and 15 mL/h (7–8 h), D 2.25 mL/h (4–5 h), 6.75 mL/h (5–6 h), 9 mL/h (6–7 h), and 15 mL/h (7–8 h). Maltose (500 g/L) or maltose (500 g/L) and tryptone (780 g/L, determined by the optimal ratio of carbon and nitrogen sources in the fermentation medium) were fed by different rates according to the four feeding strategies.

34.2.2 Analysis Method

34.2.2.1 DCW Measurement

The fermentation broth was centrifuged at $10,000 \times g$ and 4 °C for 10 min. The clear supernatant was used for the analysis of reducing sugar. The DCW was determined by drying the pellets at 105 °C in the oven till constant weight. The specific growth rate was determined from the slope of the semi-logarithmic plot of DCW versus culture time.

34.2.2.2 Enzyme Activity Assay

The samples were centrifuged at $10,000 \times g$ and 4 °C for 10 min, and then the cell pellets were washed with Na₂HPO₄ (0.2 M) and citric acid (0.1 M) buffer solution (pH 6.8) and resuspended in the same buffer. The cell suspension was sonicated to break up the cell walls and centrifuged at $10,000 \times g$ and 4 °C for 15 min to remove cell debris. The hydrolytic activity of the enzyme was measured basically according to the modified method [15]. Briefly, the reaction mixture in a test tube containing 0.9 mL Na₂HPO₄ (0.2 M) and citric acid (0.1 M) buffer solution (pH 6.8), 0.1 mL of 5 mM *p*-nitrophenyl- α -D-glucopyranoside (α -PNPG, Sigma-Aldrich), 5 min at 37 °C, then adding 1 mL of the cell suspension to them. After 20 min of incubation, the reaction was stopped with 1 mL of 1 M Na₂CO₃, and the *p*-nitrophenolate released was quantified by a spectrophotometer-722 (Third Analytical Instrument Factory, Shanghai, China) at 410 nm. One unit (U) of enzyme activity was defined as the amount of enzyme that releases 1 µmol of p-nitrophenolate per minute per mL reaction mixture at pH 6.8, 37 °C.

J. Fang et al.

34.2.2.3 Reducing Sugar Concentration Determination

The determination of reducing sugar concentration was modified according to what was previously reported [16]. To 1 mL of the sugar solution in a test tube 1 mL of the DNS reagent was added. Tubes were placed in boiling water bath for 15 min, transferred to ice to rapidly cool down to room temperature. Each sample was added with 8 mL of distilled water to stop the reaction. The absorbance was measured at 540 nm, using a spectrophotometer-722. (Third Analytical Instrument Factory, Shanghai, China.)

34.3 Results and Discussion

34.3.1 Influence of pH on α-Glucosidase Production in Batch Culture

The influence of pH (uncontrolled and constant 6.0, 7.0, and 8.0) on α -glucosidase production in batch culture is shown in Fig. 34.1. In our previous shake flask experiments, the initial pH was a key factor influencing the production of α -glucosidase significantly (data not shown). The maximum production of α -glucosidase was

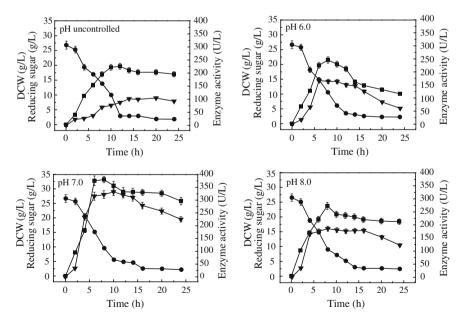


Fig. 34.1 Influence of pH on α-glucosidase production in batch culture. ■: DCW (g/L), \blacktriangledown : activity of α-glucosidase (U/L), \bullet : reducing sugar concentration (g/L)

104.6, 167.1, 332.6, and 186.1 U/L with pH was uncontrolled, constant 6.0, 7.0, and 8.0, respectively, and meanwhile, the maximum DCW was 19.7, 21.5, 33.2, and 23.7 g/L, respectively. As shown in Table 34.1, with the pH of constant 7.0, α -glucosidase production per DCW and productivity reached 10.02 U/g and 13.86 U L⁻¹ h⁻¹, much higher compared to the other three pH controlled modes. It was concluded that the constant pH at 7.0 was optimal for the production of α -glucosidase.

34.3.2 Influence of Initial Maltose Concentration on a-Glucosidase Production in Batch Culture

The influence of different initial maltose concentrations on cell growth and α -glucosidase production was further examined (Fig. 34.2). The maximum α -glucosidase production was obtained with an initial maltose concentration of 25 g/L with 444.7 U/L while the production was 332.6 U/L with 30 g/L. Though DCW increased with the increase of maltose concentration (the DCW with 30 g/L was higher than that with 25 g/L), increasing maltose concentration in medium did not enhance α -glucosidase production. Thus, feeding maltose in fermentation process may be an efficient approach to obtain higher production. So the initial maltose concentration was set as 25 g/L.

34.3.3 Influence of Agitation Speed on α-Glucosidase Production in Batch Culture

Dissolved oxygen is one of the essential nutrients for aerobic microbial growth as well as a key environmental factor affecting the material production and energy metabolism. In general, under the condition of the air flow normally, agitation speed directly affects the efficiency of oxygen transfer in fermentation liquid, and then affects the cell growth and production of α -glucosidase. We investigated the effect of agitation speed on cell growth and α -glucosidase production in batch culture and the results are shown in Fig. 34.3. Dissolved oxygen has a positive impact on cell growth since the DCW was 44.6 % higher when the agitation speed was increased to 800 rpm (33.2 g/L) from 500 rpm (18.4 g/L). In contrast to that, a good cell growth was not on behalf of a high enzyme production. The maximum α -glucosidase production with 600 rpm (444.7 U/L) was 33.7 % higher than that with 800 rpm (332.6 U/L). At the same time, Table 34.1 shows that 600 rpm had the highest production per DCW (22.1 U/g) as well as the α -glucosidase productivity (18.53 U L⁻¹ h⁻¹). Thus, we chose 600 rpm as a condition of further research.

Table 34.1 Comparison of α-glucosidase fermentation parameters on different control conditions by Bacillus licheniformis JXC-1 in batch culture

Parameters	hН				Initial m	Initial maltose concentration (g/L)	centration	(g/L)	Agitation	Agitation speed (rpm)	(md	
Conditions	Natural	0.9	7.0	8.0	20	25	30	35	500	009	700	800
Fermentation time (h)	24	24	24	24	24	24	24	24	24	24	24	24
Maximum DCW (g/L)	19.7	21.5	33.2	23.7	18.9	23.1	33.2	25.1	18.4	23.1	24.2	33.2
Maximum production (U/L)	104.6	167.1	332.6	186.1	176.1	444.7	332.6	198.9	289.7	444.7	359.2	332.6
Maximum specific growth rate [g/(g h)]	0.95	0.73	6.15	1.88	0.63	0.82	6.15	4.93	9.92	0.82	0.75	6.79
Maximum specific production rate [U/(g h)]	4.56	39.8	17.5	714.3	171.8	135.3	17.5	31.9	13.67	135.3	265.7	17.5
Alpha-glucosidase production per dry cell weight (U/g)	5.9	9.4	10.02	7.85	17.6	22.1	10.7	8.8	16.2	22.1	21.2	10.7
Alpha-glucosidase productivity [U/(L h)]	4.36	96.9	13.86	7.75	7.34	18.53	13.86	8.29	12.1	18.53	15.0	13.9

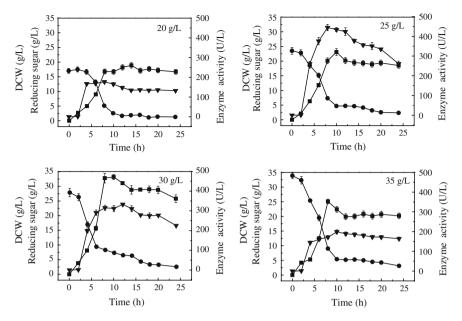


Fig. 34.2 Influence of initial maltose concentration on α -glucosidase production in batch culture. \blacksquare : DCW (g/L), \blacktriangledown : activity of α -glucosidase (U/L), \bullet : reducing sugar concentration (g/L)

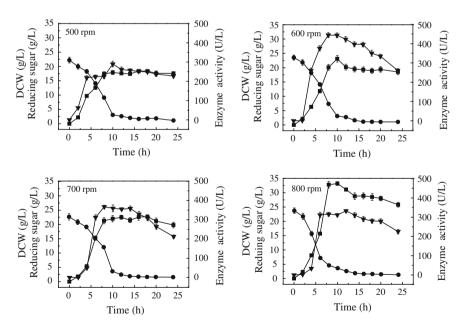


Fig. 34.3 Influence of agitation speed on α -glucosidase production in batch culture. \blacksquare : DCW (g/L), \blacktriangledown : activity of α -glucosidase (U/L), \bullet : reducing sugar concentration (g/L)

J. Fang et al.

34.3.4 Four Different Strategies of Feeding Maltose for α-Glucosidase Production

In batch culture, the maximum specific growth and production rate were shown approaching to 4 h while the reducing sugar was 18 g/L (data not shown). At the same time, the DCW and the production of α -glucosidase rised rapidly for 4–8 h. Thus, to obtain higher production, we put forward four strategies to feed maltose for 4–8 h at different rates to keep the reducing sugar concentration at 18 g/L. The influence of different fed-batch strategies with the optimal conditions feeding maltose is shown in Fig. 34.4. The maximal DCW in fed-batch fermentation (strategy C) was 45.6 g/L and the maximum α -glucosidase production (strategy D) was 633.8 U/L. Fed-batch fermentation (strategy D) resulted in higher α -glucosi-

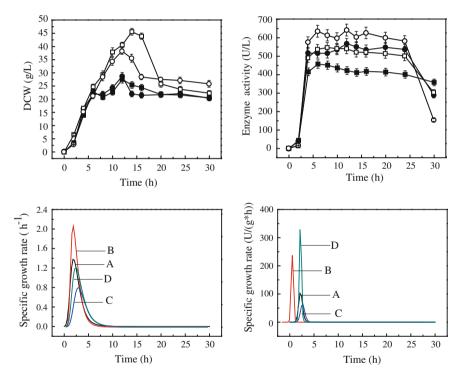


Fig. 34.4 Influence of four fed-batch culture strategies on α-glucosidase production by feeding maltose. ■: strategy A, the feeding rate was set as 8.25 mL/h (4–8 h), \blacksquare : strategy B, the feeding rate was set as 4.5 mL/h (4–6 h) and 12 mL/h (6–8 h), \square : strategy C, the feeding rate was set as 4.5 mL/h (4–6 h), 9 mL/h (6–7 h), and 15 mL/h (7–8 h), \square : strategy D, the feeding rate was set as 2.25 mL/h (4–5 h), 6.75 mL/h (5–6 h), 9 mL/h (6–7 h), and 15 mL/h (7–8 h)

Parameters	Feedin	g maltos	e		Feeding maltose and tryptone			
Strategy	A	В	С	D	A	В	C	D
Fermentation time (h)	30	30	30	30	30	30	30	30
Maximum DCW (g/L)	27.4	28.5	45.6	38.3	30.7	29.5	35.2	23.5
Maximum production(U/L)	457.3	568.9	545.5	633.8	724.4	616.9	618.2	872.5
Maximum specific growth rate [g/(g h)]	0.76	0.8	0.76	1.23	0.68	0.68	1.3	0.4
Maximum specific production rate [U/(g h)]	201.3	271.6	85.8	429.5	151.2	168.5	272.2	128.8
Alpha-glucosidase production per dry cell weight (U/g)	35.7	20.0	18.6	56.6	31.4	27.1	22.3	43.0
Alpha-glucosidase productivity [U/(L h)]	15.24	18.96	18.18	21.13	24.15	20.56	20.61	29.08

Table 34.2 Comparison of α -glucosidase fermentation parameters between different carbon and organic nitrogen sources feeding approaches by *B. licheniformis* JXC-1 in fed-batch cultures

dase production and productivity compared with the other three feeding strategies. With this strategy, the α -glucosidase titer per DCW and productivity reached 56.6 U/g and 21.13 U L⁻¹ h⁻¹ at 14 h (Fig. 34.4 and Table 34.2). More than that, the production with strategy D increased 42.5 % compared to that in batch culture.

34.3.5 Four Different Strategies Feeding Maltose and Tryptone for α -Glucosidase Production

Nitrogen sources are metabolized to produce amino acids, nucleic acids, proteins, and cell-wall components in bacteria. In order to maintain the maximum cell growth rate, not only sufficient and suitable carbon source should be provided, but the corresponding organic nitrogen source which is beneficial to the growth of cell also should be supplied. With the four feeding strategies, we further investigated the effect of feeding maltose and tryptone at different rates in fed-batch culture, shown in Fig. 34.5. The maximal DCW in fed-batch fermentation (strategy C) was 35.2 g/L and the maximum α -glucosidase production (strategy D) was 872.5 U/L. Though the maximum DCW was lower than 45.6 g/L, it is also worth mentioning that the production of α -glucosidase with feeding maltose and tryptone by four strategies was much higher than that with feeding maltose, the maximum α -glucosidase production increased 37.7 %.

322 J. Fang et al.

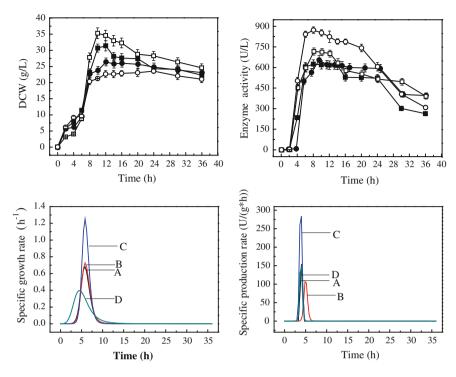


Fig. 34.5 Influence of four fed-batch culture strategies on α-glucosidase production by feeding maltose and tryptone. ■: strategy A, the feeding rate was set as 8.25 mL/h (4–8 h), •: strategy B, the feeding rate was set as 4.5 mL/h (4–6 h) and 12 mL/h (6–8 h), □: strategy C, the feeding rate was set as 4.5 mL/h (4–6 h), 9 mL/h (6–7 h), and 15 mL/h (7–8 h), ○: strategy D, the feeding rate was set as 2.25 mL/h (4–5 h), 6.75 mL/h (5–6 h), 9 mL/h (6–7 h), and 15 mL/h (7–8 h)

34.4 Conclusions

In this work, the influence of pH, initial maltose concentration, and agitation speed on α -glucosidase production was investigated in a 3-L fermenter. We found that the constant pH 7.0, initial maltose concentration of 25 g/L, and agitation speed of 600 rpm were the better conditions, which raised the production of α -glucosidase 3.25 times from 104.6 to 444.7 U/L. We further proposed four feeding strategies to feed maltose or both maltose and tryptone at different rates for 4–8 h. With the strategy D where the maltose and tryptone feeding rate was as 2.25 mL/h (4–5 h), 6.75 mL/h (5–6 h), 9 mL/h (6–7 h) and 15 mL/h (7–8 h), we finally obtained the highest α -glucosidase production (872.5 U/L) which was 96.2 % higher than that in batch culture. These results indicated that the feeding strategy was effective for the production of α -glucosidase.

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Chapter 35

The Effect of Different Activated Carbon and Bleaching Temperature on Kojic Acid Bleaching

Wu Meng, Cuiying Zhang and Dongguang Xiao

Abstract In this paper, the effects of six different types of activated carbons on decolorization of kojic were studied and two kinds of them were picked out based on the decoloration rate or kojic acid yield, respectively. After mixing in different proportions, the effects of different ratios of the two activated carbons at different bleaching temperature conditions on kojic bleaching were determined by experiments. The results show that the decoloration rate was the highest (79.84 %) at 70 °C when the ratio of 767 was 20 % (total 0.2 g of activated carbon), and it had a higher yield of kojic acid (91.26 %), while the highest kojic acid rate (94.06 %) was obtained at 80 °C when the ratio of 767 was 100 %; the decoloration rate was also better (73.26 %) under these conditions. In contrast, the better bleaching effect with minimum energy consumption was at 50 °C when the ratio of 767 was 40 %, the kojic acid yield reached more than 90 %. These three combined ratios have their own advantages, according to the different needs for various occasions.

Keyword Activated carbon • Kojic • Decoloration rate • Respectively rate

35.1 Introduction

Kojic acid is an organic acid with good antibacterial effect against pathogenic bacterium [1–3], it can also inhibit flies and other pests [4, 5]. Besides its use in the medical field, kojic acid has various applications in several other fields such as food, cosmetics, agriculture field, etc. [6–9]. The primary way to produce kojic acid in the industry was submerge fermentation with glucose as carbon source [10, 11];

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W. Meng et al.

however, the kojic acid is obtained by fermentation often complexing with ferric ion [12], so the kojic obtained form crude extract must go through bleaching and refining.

Currently, the kojic acid extraction bleaching process in industry still mainly relies on an activated carbon ZX-767 [13]; through optimization experiments, upto 78 % of decoloration rate and 90 % of kojic acid yield can be achieved, but the quality of kojic acid has not been increased significantly and the production cost has not been reduced effectively yet. With the increasing competition of kojic acid production in the domestic market, it is urgent to reduce the cost of kojic acid extraction process and improve the quality of kojic acid's end product. However, bleaching is the most important step in kojic acid extraction process. The decoloration of kojic acid material in industry is mainly activated carbon [14] and the ion-exchange resin [15], both of them have many different types of materials, and between the different types they also have significant differences.

In this paper, we experimented with six different types of activated carbon for decoloration experiments of kojic and used decoloration rate and kojic acid yield as two important indicators to pick out two kinds of activated carbon. Then after combining in different proportions, we studied the effects of different ratios at different carbon bleaching temperature conditions on kojic bleaching.

35.2 Materials and Methods

35.2.1 Activated Carbon

Activated carbons, used in the experiments and shown in Table 35.1, were purchased from Jiangsu Liyang activated carbon plant.

35.2.2 The Choice of Activated Carbon

Different varieties of activated carbon have different effects on bleaching. In this study, Chengdu Jin Kaisheng Ltd. kojic crude material was crystallized as experimental subjects. Six different types of activated carbons (Table 35.1) were used to

Tubic 33.1	Severar mam parar	neces of the ser	cetion activated v	caroon	
Name	Species	Particle size (mash)	Moisture (%)	Iron (%)	Specific surface area (m²/g)
ZX-767	Injection type	200	10	0.02	-
ZXK-AR	Reagents	200	10	0.02	_
ZX-210	Nutshell	φ1.0 mm	5	_	$1,100 \pm 50$
ZX-3011	Sugar class	200	10.0	0.10	~1,300
ZX-720	Medicinal class	200	15	0.10	_
ZX-507	Powdery	200	_	0.02	~1,400

Table 35.1 Several main parameters of the selection activated carbon

Name	Species	Subjects	Bleaching temperature (°C)	Bleaching time (min)	Dosage of the activated carbon (g)
ZX-767	Injection type	Crude kojic acid solution mL	50	30	0.200
ZXK-AR	Reagents	Crude kojic acid solution mL	50	30	0.200
ZX-210	Nutshell	Crude kojic acid solution mL	50	30	0.200
ZX-3011	Sugar class	Crude kojic acid solution mL	50	30	0.200
ZX-720	Medicinal class	Crude kojic acid solution mL	50	30	0.200
ZX-507	Powdery	Crude kojic acid solution mL	50	30	0.200
ZA-307	Powdery	Crude Rojic acid solution inL	30	30	0.200

Table 35.2 The experiment scheme of five activated carbons at 50 °C

do the experiment. We took 50 g crude kojic acid crystals and dissolved these crystals in 500 mL of distilled water. Then 50 mL of the solution was measured and placed in six 250 mL Erlenmeyer flasks. 0.200 g of the different types of activated carbons was added in the flasks at 50 °C temperature for 30 min for bleaching. The effects of all activated carbon bleaching were compared. The design of the experiments are shown in Table 35.2.

35.2.3 The Ratio of Different Carbon Decoloration Experiments at Different Temperatures

We took 50 g crude kojic acid crystals, dissolved these crystals in 500 mL of distilled water, and stirred and mixed it. Then 50 mL of the solution was measured and placed in six 250 mL Erlenmeyer flasks, adding 0.200 g of the different types of activated carbons in the flasks at 50 °C temperature for 30 min for bleaching. All activated carbon bleaching effects were compared. The design of the experiments are shown in Table 35.3. We selected the highest rate of decoloration activated carbon X and the highest kojic acid yield carbon Y through the different proportions of decoloration experiments. Finally, we obtained the optimal ratio combinations.

Table 35.3 The decolorized experiment scheme of two activated carbons in different proportions at different temperatures

Activated carbon	Subjects	Bleaching temperature (°C)	Dosage of the activated carbon (g)	The proportion of X (%)
X, Y	Crude kojic acid solution of 50 mL	50, 60, 70, 80, 90	0.2	0
X, Y	Crude kojic acid solution of 50 mL	50, 60, 70, 80, 90	0.2	20
X, Y	Crude kojic acid solution of 50 mL	50, 60, 70, 80, 90	0.2	40
X, Y	Crude kojic acid solution of 50 mL	50, 60, 70, 80, 90	0.2	60
X, Y	Crude kojic acid solution of 50 mL	50, 60, 70, 80, 90	0.2	80
X, Y	Crude kojic acid solution of 50 mL	50, 60, 70, 80, 90	0.2	100

W. Meng et al.

35.2.4 Bleaching Methods of the Activated Carbon

Activated carbon is the most commonly used method for decolorizing, which relies on the van der Waals forces to make the pigment adsorb to the surface of the activated carbon. The smaller the particles of the activated carbon are, the greater the surface area and the stronger the adsorption capacity.

There are two important indicators in the kojic acid bleaching process, the decoloration rate, and the kojic acid's final quality.

$$DR = \frac{OD_1 - OD_2}{OD_1} \times 100\%$$
 (35.1)

In the formula:

OD₁ absorption value before bleaching of the solution of kojic acid; OD₂ absorption value after bleaching of the solution of kojic acid;

absorption value after bleaching of the solution of kojic acid

Another is kojic acid yield, related to the loss of kojic acid

$$Y = \frac{m_2}{m_1} \times 100\% \tag{35.2}$$

In the formula:

 m_1 the quality of kojic acid in the solution before bleaching; m_2 the quality of kojic acid in the solution after bleaching.

35.3 Results and Discussion

35.3.1 Comparison of Different Types of Activated Carbon Decolorization Experiments

Comparison of the experiment can be made from Fig. 35.1: Although the carbon ZX-210's kojic acid yield is up to 98 %, it has almost no bleaching effect on kojic acid, the bleaching rate of ZX-767 activated carbon injection is the highest, it can reach 74 %; the kojic acid rate of medicinal activated carbon ZX-720 is higher, which can reach 92 %. Thus, in order to obtain the optimal ratio combinations we used carbon ZX-767 and ZX-720 in different proportions and at different temperatures to perform the kojic acid decolorization experiments.

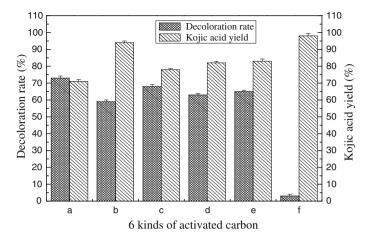


Fig. 35.1 The decolorized result of six kinds of activated carbon. *Note a, b, c, d, e, g* represent carbon ZX-767, ZX-720, ZXK-AR, ZX-3011, ZX-507, ZX-210

35.3.2 The Combined Ratio of Kojic Acid Bleaching Experiments at Temperature of 50 °C

It can be seen from the experimental results (Fig. 35.2) that at temperature of 50 °C, the combined ratio of: 0.2 g of activated carbon, the 767 is 40 %, the decolorization rate is 74 %, the kojic acid yield is 90.10 %, which is better than any other combination.

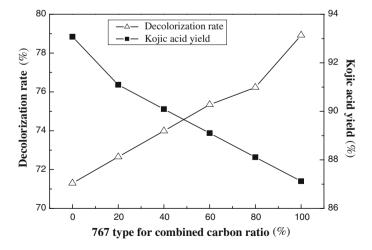


Fig. 35.2 The decolorized result of combined activated carbon at 50 °C

W. Meng et al.

35.3.3 The Combined Ratio of Kojic Acid Bleaching Experiments at Temperature of 60 °C

It can be seen from the experimental results (Fig. 35.3) that at temperature of 60 °C, the combined ratio of: 0.2 g of activated carbon, the 767 is 40 %, the decolorization rate is 75.12 %, the kojic acid yield is 88.04 %, which is better than any other combination.

35.3.4 The Combined Ratio of Kojic Acid Bleaching Experiments at Temperature of 70 °C

It can be seen from the experimental results (Fig. 35.4) that at temperature of 70 °C, the combined ratio of: 0.2 g of activated carbon, the 767 is 20 %, the decolorization rate is 79.84 %, the kojic acid yield is 91.26 %, which is better than any other combination.

35.3.5 The Combined Ratio of Kojic Acid Bleaching Experiments at Temperature of 80 °C

It can be seen from the experimental results (Fig. 35.5) that at temperature of 80 °C, the combined ratio of: 0.2 g of activated carbon, the 767 is 0 %, the decolorization rate is 73.62 %, the kojic acid yield is 94.06 %, which is better than any other combination.

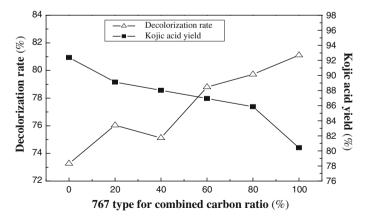


Fig. 35.3 The decolorized result of combined activated carbon at 60 °C

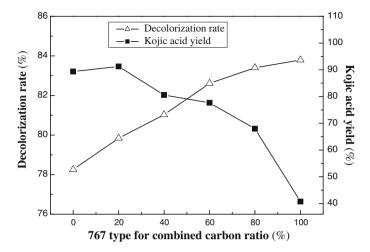


Fig. 35.4 The decolorized result of combined activated carbon at 70 °C

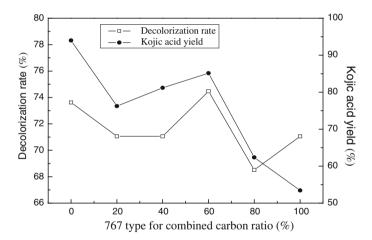


Fig. 35.5 The decolorized result of combined activated carbon at 80 °C

35.3.6 The Combined Ratio of Kojic Acid Bleaching Experiments at Temperature of 90 °C

It can be seen from the experimental results (Fig. 35.6) that at temperature of 90 $^{\circ}$ C, the combined ratio of: 0.2 g of activated carbon, the 767 is 80 %, the decolorization rate is 79.64 %, the kojic acid yield is 73.16 %, which is better than any other combination.

W. Meng et al.

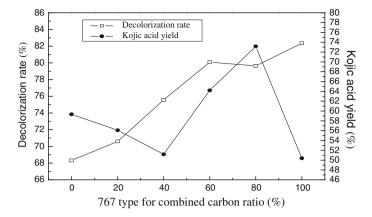


Fig. 35.6 The decolorized result of combined activated carbon at 90 °C

35.4 Conclusion

From the comprehensive comparison of the experimental results, we can see that at temperature of 70 °C, the combined ratio of: 0.2 g of activated carbon, the 767 is 20 %, the decolorization rate is the highest (79.84 %), kojic acid yield is (91.26 %). At temperature of 80 °C, the combined ratio of: 0.2 g of activated carbon, the 720 type activated carbon is (100 %), kojic acid yield is the highest (94.06 %), the decolorization rate is (73.26 %). At temperature of 50 °C, the combined ratio of: 0.2 g of activated carbon, the 767 is 40 %, it required the minimum temperature (low energy consumption), the kojic acid yield is over (90.10 %), the decolorization rate is (74 %) (Table 35.4). Thus, the ratio of the combination of these three have their own advantages. This are the ideal results of the experiments.

Table 35.4 The best proportion of different activated carbons composed at different temperatures

Activated carbon	Temperature (°C)	The amount of activated carbon portfolio (g)	767 The optimum ratio of activated carbon (%)	Decolorization rate (%)	Kojic acid yield (%)
ZX-767, ZX-720	50	0.2	40	74.00	90.10
ZX-767, ZX-720	60	0.2	40	75.12	88.04
ZX-767, ZX-720	70	0.2	20	79.84	91.26
ZX-767, ZX-720	80	0.2	0	73.62	94.06
ZX-767, ZX-720	90	0.2	80	79.64	73.16

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Chapter 36 Effect of Iodine on the Growth and Quality of *Nostoc flagelliforme*

Honglei Fu, Yujie Dai, Yue Han, Lifang Yue, Feng Xia and Shiru Jia

Abstract This study was carried out to evaluate the effect of diverse iodine form-iodide and iodate on the biomass as well as the content of soluble protein, pigment, and iodine in Nostoc flagelliforme cells. In our study, tested factors (iodine form-iodide and iodate) significantly influenced the growth and content of soluble protein, pigment, and iodine in N. flagelliforme cells. Moreover, results obtained have shown that N. flagelliforme cultured with I take up greater amounts of iodine than those treated with IO_3^- . Besides, at the initial iodine concentration of 600 mg/L in the form of KI and KIO₃, the biomass of N. flagelliforme significantly increased by 2.6 % and decreased by 3.2 %, respectively. In addition, the highest content of soluble protein was, respectively, 2.56 and 2.46 mg/g when the concentration of iodine in the form of iodide and iodate was at a dose of 600 mg/L. With respect to chlorophyll a content, the results of our work have shown that N. flagelliforme cells cultured with KI contained higher amounts of chlorophyll a than those treated with KIO₃. All these studies suggest that the introduction of iodine in the form of KI proved to be much more effective in respect to iodine accumulation of N. flagelliforme than that of KIO₃.

Keywords Iodine • Bioaccumulation • Nostoc flagelliforme

36.1 Introduction

Iodine is an essential micronutrient for humans. It acts as an essential component of thyroid hormones which is of importance for the human metabolism [1]. Iodine deficiency causes many diseases such as endemic goiter, cretinism, fetal

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336 H. Fu et al.

abnormalities, and Kaschin-Beck disease, the last one being related to simultaneous deficiency of iodine and selenium [2, 3]. Almost 35.2 % of the global population has inadequate iodine nutrition [4] and it has been calculated that 1.6 billion people are at a risk of iodine deficiency [5]. In order to conquer iodine deficiency in the human population, biofortification of microalgae with iodine can be an alternative method of introducing this microelement into the human diet.

Iwamoto and Shiraiwa [6] found that six species of microalgae such as *Bellerochea sp.*, *Chlamydomonas reinhardtii*, *Dunaliella tertiolecta*, *Isochrysis galbana*, *Skeletonema costatum*, and *Tetraselmis sp.* exhibited high iodine tolerance. Additionally, *E. huxleyi* was observed to accumulate 10 times more iodine than that found in seawater, indicating that microalgae can be used in iodine-related industries for extracting iodine from seawater and iodine-containing wastewater. The relationship between iodine and microalgae was investigated by some studies. Some researches on iodine in microalgae revealed that microalgae participate in the iodate (IO₃⁻)-to-iodide (I⁻) speciation and emit iodine as monoiodomethane into the marine environment [7–9]. Moreover, microalgae are supposed to function in the global iodine cycle [10, 11]. All these studies strongly suggest that the physiologic characteristics of microalgae are closely associated with their iodine usage.

Nostoc flagelliforme, an edible terrestrial filamentous cyanobacterium, has been famous for its food and herbal value for more than 400 years [12]. Cultivation of free cells isolated from colonial filaments in liquid medium has also been performed for reasons of low growth rate of *N. flagelliforme* under natural conditions and the great demand from the industry [13, 14]. So the increased cell growth rate in the liquid culture led to a new technique for the culture of *N. flagelliforme*. Some studies found that *N. flagelliforme* was a good candidate for heavy metal ions enrichment [15]. However, very limited information on the iodine-enrichment is available. The aim of this work is to study the effect of iodine on the growth and quality of *N. flagelliforme*, so as to provide basic data for the mechanism of iodine uptake by cabbage and an effective way to reduce iodine deficiency disorders.

36.2 Materials and Methods

36.2.1 Materials and Experimental Design

The *N. flagelliforme* cells used in liquid suspension cultures were obtained from the Tianjin Key Lab of Industrial Microbiology (Tianjin, China). The cells were cultured in BG11 medium with and without iodine in 500-mL shake flasks containing 200 mL medium at 25 °C under continuous illumination of 60 μ mol photon/(m² s). The solution was formulated by adding iodine in the form of iodide (μ) or iodate (μ) at five concentrations (200, 400, 600, 800, 1000 mg/L) to BG11 medium. The formula without iodine served as control.

36.2.2 Sampling and Chemical Analyses

The cell growth was measured by weighing the harvested samples that had been previously dried to a constant weight at 80 °C [16]. The total iodine of the dried samples were quantified by the iodide-starch photometry method [17] following the pretreatment of oxygen flask combustion method [18]. Methanol-soluble pigments (chlorophyll a and carotenoids) were extracted in methanol at 4 °C in the dark and the amount was calculated according to Li et al. [19]. The soluble protein was extracted through the method of ultrasonication and the contents were determined via the Bradford method [20].

36.2.3 Statistical Analysis

Obtained results were statistically verified using ANOVA module of SPSS Statistics 19.0 program for significance level P < 0.05. In case of significant changes, homogeneous groups were distinguished on the basis of Duncan test.

36.3 Results and Discussion

36.3.1 Effect of Iodine on the Growth of N. flagelliforme

As shown in Fig. 36.1, biomass of N. flagelliforme increases with iodine concentration in the culture solution when the concentration remains low, but decreases with the increase in concentration. These results demonstrate that iodine concentrations affect algal growth, including iodine-induced stimulation and inhibition. Moreover, the growth of N. flagelliforme was inhibited when the initial potassium iodide in the medium was higher than 600 mg/L (Fig. 36.1a), while the growth of N. flagelliforme was inhibited when the initial potassium iodate in the medium was higher than 200 mg/L (Fig. 36.1b), indicating that iodide shows much less toxicity to the N. flagelliforme than iodate. Furthermore, at the exogenous iodine level of 200 mg/L, the biomass of N. flagelliforme was significantly increased compared with the control group cultivated without additive iodine (Fig. 36.1c, d). Besides, at the initial iodine concentration of 400 and 600 mg/L in the form of KI, the biomass of N. flagelliforme significantly increased by 5.1 and 2.6 %, respectively, compared with the control groups cultivated without exogenous iodine addition (Fig. 36.1c). In addition, the biomass of N. flagelliforme was correspondingly decreased with the increase in the concentration of iodine in both forms, suggesting that there was a negative correlation between biomass yield and initial iodine addition.

H. Fu et al.

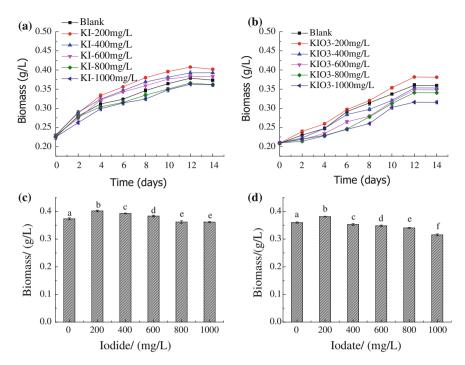


Fig. 36.1 Effect of iodine on the growth of *N. flagelliforme* in the BG11 medium after adding potassium iodide (**a**) or potassium iodate (**b**). Effect of iodine on biomass production (g/L) of *N. flagelliforme* in the BG11 medium after adding potassium iodide (**c**) or potassium iodate (**d**) at the end of the cultivation. Data with the same letters are not significantly different (P > 0.05)

36.3.2 Effect of Iodine on the Content of Pigments and Soluble Protein of N. flagelliforme

Particular attention should be given to the fact that a significant increase in the amount of soluble protein was detected in *N. flagelliforme* grown in the presence of KI within the range of 200–1,000 mg/L and KIO₃ within the range of 200–800 mg/L (Table 36.1). Moreover, the content of soluble protein reached the highest point when the iodine concentration was 600 mg/L. The possible explanation for the obtained results could be that some protein synthesis was promoted during the process of iodine accumulation. Fu and Xie [21] hold another view that iodine is the cofactor or activator in the synthesis of some amino acids.

In this study, a higher content of pigment was found in *N. flagelliforme* treated with exogenous iodine (KI and KIO₃) compared with the control groups cultivated without exogenous iodine addition (Table 36.1). It should be mentioned that *N. flagelliforme* cells cultured with KI contained more of chlorophyll a than those treated with KIO₃. However, there was no observed difference in carotenoid after cultivation of *N. flagelliforme* treated with KI and KIO₃.

Treatments	Iodine concentration in the medium (mg/L)	Soluble protein (mg/g)	Chlorophyll a (mg/L)	Carotenoid (mg/L)	Carotenoid: chlorophyll a
Exogenous	-	2.22 ± 0.01 a	3.36 ± 0.03 a	0.87 ± 0.01 a	0.259
iodine in the	200	$2.40 \pm 0.03 \text{ b}$	$3.64 \pm 0.02 \text{ bc}$	$0.95 \pm 0.01 \text{ b}$	0.260
form of iodide	400	2.41 ± 0.04 b	3.71 ± 0.04 c	$0.99 \pm 0.03 \text{ bc}$	0.267
	600	2.56 ± 0.01 c	$3.81 \pm 0.02 \text{ d}$	1.02 ± 0.06 c	0.269
	800	$2.47 \pm 0.04 \text{ d}$	3.68 ± 0.06 bc	1.02 ± 0.06 bc	0.276
	1,000	2.33 ± 0.03 e	$3.61 \pm 0.06 \text{ b}$	1.01 ± 0.01 bc	0.279
Exogenous	_	2.22 ± 0.01 a	3.36 ± 0.03 a	0.87 ± 0.01 a	0.259
iodine in the form of iodate	200	$2.39 \pm 0.01 \text{ b}$	3.55 ± 0.05 bc	$0.92 \pm 0.02 \text{ b}$	0.260
	400	$2.40 \pm 0.06 \text{ b}$	$3.68 \pm 0.06 \text{ d}$	0.99 ± 0.01 c	0.268
	600	$2.46 \pm 0.04 \text{ b}$	3.76 ± 0.01 e	1.02 ± 0.04 c	0.271
	800	$2.42 \pm 0.05 \text{ b}$	3.61 ± 0.03 cd	1.00 ± 0.04 c	0.278
	1,000	2.26 ± 0.01 a	$3.47 \pm 0.05 \text{ b}$	0.97 ± 0.01 c	0.281

Table 36.1 Soluble protein and pigment contents of *N. flagelliforme*

Data with the same letters are not significantly different (P > 0.05)

The ratio of carotenoid: chlorophyll a has long been considered a valuable parameter for assessing adverse environmental conditions [22]. As shown in Table 36.1, there was a gradual increase in the ratio of carotenoid: chlorophyll a of *N. flagelliforme* exposed to iodine (KI and KIO₃). It should be emphasized that iodine addition might result in higher increase in carotenoid synthesis compared with chlorophyll a. From these observations, the high ratio of carotenoid: chlorophyll a may be related to the reduction of biomass in excess iodine doses.

36.3.3 Accumulation of Iodine in N. flagelliforme Cells

In general, iodine contents in the cells of *N. flagelliforme* increase significantly with the amount of applied iodine in the medium, however, there is a turning point at the concentration of 600 mg/L in the solution (Fig. 36.2); below this point iodine contents in the cells of *N. flagelliforme* increase with the initial iodine concentration, while above this point this phenomenon is the opposite. It is supposed that the toxic effect of the high iodine concentration on the growth of *N. flagelliforme* results in a change in the iodine contents in the cells of *N. flagelliforme* in response to the increase in iodine concentration in the culture solution.

Studies have shown that *Spirulina maxima* cultured with iodine in the form of I^- take up greater amounts of iodine than *S. maxima* treated with IO_3^- [23]. This relation has been already confirmed in our study with cultivation of *N. flagelliforme*, which is another kind of blue-green microalga. Iodine contents of *N. flagelliforme*

340 H. Fu et al.

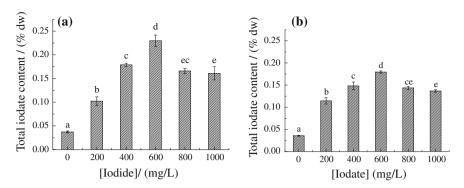


Fig. 36.2 Variation in iodine content in the cells of *N. flagelliforme* as a function of the additive concentration of exogenous potassium iodide (**a**) or potassium iodate (**b**) in the medium. Data with the same letters are not significantly different (P > 0.05)

cultured with the initial iodine concentration of 600 mg/L in the form of KI could be increased to 6.18 times as high as that of control (Fig. 36.2a), while the contents of those treated with the initial iodine concentration of 600 mg/L in the form of KIO₃ could be raised to 5.04 times as high as that of control (Fig. 36.2b). It is observed that a higher accumulation of iodine was found after KI application rather than KIO₃. Introduction of iodine in the form of KI proved to be much more effective in respect to iodine accumulation of *N. flagelliforme* than that of KIO₃.

36.4 Conclusions

The results of this study show that low concentration of iodine accelerates the growth of *N. flagelliforme* and raises the content of soluble protein, pigment, and iodine. 600 mg/L is found to be the optimum concentration of iodine in the cultivated medium, while the positive effects turn into toxic effects when the concentration exceeds 800 mg/L. In addition, more iodine was accumulated in *N. flagelliforme* cells grown in KI than in those grown in KIO₃. Data obtained demonstrated that the content of soluble and pigment in *N. flagelliforme* cells cultured with KI contained more than those treated with KIO₃. All these studies reveal that *N. flagelliforme* can be served as an alternative to accumulate this element and the introduction of iodine in the form of KI proved to be much more effective in respect to iodine accumulation of *N. flagelliforme* than that of KIO₃.

Acknowledgments This work was supported by the National Natural Science Foundation of China under Grant number 31271809.

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Chapter 37 The Replacement of Phe28 by Ser Enhances the Stability of the GLP-1 Analog During Fermentation

Peng-Yan Li, Xue-Gang Luo, Qian Li, Wei Zhao, Hao Zhou and Tong-Cun Zhang

Abstract Glucagon-like peptide-1 (GLP-1) is an incretin, which can effectively lower blood glucose levels, increase insulin secretion, and improve insulin sensitivity in patients with diabetes. Therefore, it is a potent endogenous insulin-stimulating hormone. Unfortunately, since GLP-1 can be degraded by the enzyme DPP-IV and NEP 24.11, its therapeutic benefit and fermentation yield are limited by the instability. Previous study has shown that replacement of the Lys34 with Arg could enhance the stability of GLP-1 in the plasma. In this study, to further improve the stability of the GLP-1, an additional mutation, replacement of Phe28 by Ser, was constructed, and then the protein was expressed in the recombinant *Pichia pastoris*, and analyzed by HPLC. The results show that this mutation could improve the stability of GLP-1 during fermentation.

Keywords GLP-1 · Degradation · Gene modification · Stability

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37.1 Introduction

Glucagon-like peptide-1 (GLP-1) is a 31 amino-acid polypeptide hormone produced by intestinal L-cells in response to nutrient ingestion [1]. Effects of GLP-1 can promote glucose-mediated insulin secretion, lower plasma glucagon, delay gastric emptying, stimulate β -cell growth, and insulin secretion in a glucose-dependent manner, and therefore help maintain glucose homeostasis [2, 3]. In addition, GLP-1 has also been shown to reduce body weight [4, 5], which is favorable for many people with type 2 diabetes. Hence, GLP-1-based therapies are becoming increasingly attractive for the treatment of type 2 diabetes [6, 7].

However, native GLP-1 is not a viable therapeutic agent because it has a short half-life of <2 min, resulting from rapid degradation by the enzyme dipeptidyl peptidase IV (DPP-IV) and rapid renal elimination [8]. GLP-1(7–36) is rapidly metabolized by DPP-IV to the major metabolite GLP-1(9–36) amide, often thought to be inactive. Besides, the neutral endopeptidase NEP 24.11 (also known as neprilysin, CD10 and CALA antigen [9]) is another endoprotease implicated in GLP-1 degradation. NEP 24.11 cleaves GLP-1 at several internal sites aminoproximal to bulky hydrophobic amino acids such as phenylalanine at residue 28 and leucine at residue 32 (Fig. 37.1) [10]. Therefore, an intracellular form of NEP 24.11, or an endopeptidase with similar specificity, might be responsible for internal

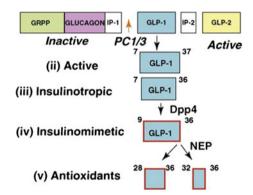


Fig. 37.1 The schematic diagram for the production and degradation of GLP-1. The prohormone convertases PC1/3 selectively cleave proglucagon to produce GLPs. The insulinotropic GLP-1 peptides consist of GLP-1(7–36) amide and a glycine-extended form GLP-1(7–37). They are released from the intestine in response to feeding and stimulate glucose-dependent insulin secretion. After their secretion, these insulinotropic GLP-1s are rapidly converted to insulinomimetic hormones GLP-1(9–36) amide and GLP-1(9–37) by removal of two N-terminal amino acids by Dpp4, and then might be further cleaved by NEP to produce small C-terminal nonapeptide GLP-1(28–36) and pentapeptide GLP-1(32–36), and the corresponding decapeptide GLP-1(28–37) and hexapeptide GLP-1(32–37). It is proposed that these small peptides might target mitochondria and modulate oxidative phosphorylation, glucose and fatty acid metabolism, and energy expenditure, resulting in attenuation of oxidative stress (ROS formation) and promotion of cell survival

cleavage of GLP-1. To solve these problems, some pharmacological mimetics have been developed with improved stability [11, 12], including of the analogs of GLP-1 with extended or truncated N- or C-terminal and analogs with exchanged residues in positions Gly10, Asp15, Val16, Ser17, Ser18, Glu21, Phe27, and Trp31 [13].

Previous study has shown that replacement of the Lys34 with Arg could enhance the stability of GLP-1 in the plasma [14]. In this study, to further improve the stability of GLP-1, an additional mutation, replacement of Phe28 with the Ser, was constructed, and then they were expressed in the recombinant *Pichia pastoris*, and analyzed by HPLC.

37.2 Materials and Methods

37.2.1 Strain and Plasmids

The methylotrophic yeast *P. pastoris* strain GS115 (deficient in histidinol dehydrogenase activity and yps1⁻, His⁻) in this study was offered by Shuqing Chen from Zhejiang University. The host strain *E. coli* DH5α and plasmid pPIC9K in this study were maintained in our laboratory.

37.2.2 Enzymes and Reagents

The restriction endonucleases, T4 DNA ligase, Pfu DNA polymerase, Taq DNA polymerase, and agar gel DNA extraction kit were purchased from Takara (Dalian, China). Primer synthesis and DNA sequencing were performed by Invitrogen (Carlsbad, CA, USA). All other reagents were purchased from local companies in Tianjin of China.

37.2.3 Construction of pPIC9K GLP-1 Analog-1 and pPIC9K GLP-1 Analog-2

Plasmid for the expression of GLP-1 analogs (pPIC9K GLP-1 analog-1 and pPIC9K GLP-1 analog-2) were constructed as shown in Fig. 37.2. The GLP-1 analog gene was obtained from the coding DNA of GLP-1 and optimized based on the codon bias of *P. pastoris* GS115. The recombination genes were synthesized with overlapped PCR. First, GLP-1 analog gene fragments were obtained by using primers P1/F, P1/Z, and P1/R (Table 37.1). Second, the product in step was further extended to get the full length of GLP-1 analogs with the primers of P2/F and P2/R.

346 P.-Y. Li et al.

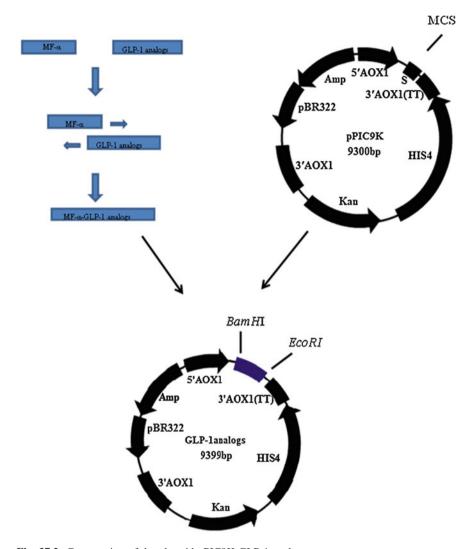


Fig. 37.2 Construction of the plasmid pPIC9K GLP-1 analogs

Third, the α -factor signal sequence was amplified from the pPIC9K plasmid using the primers of P3/F and P3/R. Finally, the α -factor signal sequence was ligated with the GLP-1 analog by PCR using the products of the second and third steps as templates and the primers of P4/F and P4/R. The final PCR products were digested by *Bam*HI and *Eco*RI and then inserted into the pPIC9K vector, generating the recombinant plasmid pPIC9K-GLP-1 analog-1 (K34R) and pPIC9K-GLP-1 analog-2 (K34R/F28S).

Primer name	Primer sequence (5'–3')
P1/F	$\frac{\text{CCG}\underline{\text{CTCGAG}}\text{AAAAGACACGCTGAGGGTACTTTCACTTCCGACGTTTCCTCC}}{(Xho\mathbf{I})}$
P1/Z	GCGATGGACTCCTTAGCAGCTTGACCCTCCAAGTAGGAGGAAACGTCGGA
P1/R	$\frac{CTAG\underline{TCTAGA}}{(Xba\mathbf{I})}\underline{TTAACCTCTAACCAACCAAGCGAT} \underline{GGACTCCTTA}$
P2/F	GTATCTCTCGAGAAAAGACACGCTGAGGGTACT
P2/R	GCCGAATTCTTATTAACCTCTACCTCTAAC (EcoRI)
P3/F	CGCGGATCCAAACGATGAGATTTCCTTC (BamHI)
P3/R	AGTACCCTCAGCGTGTCTTTTCTCGAGAGATAC
P4/F	AGCGGATCCAAACGATGAGATTTCCTTC (BamHI)
P4/R	$\frac{GCC\underline{GAATTC}}{(EcoRI)}TTATTAACCTCTACCTCTAACCAACCAAGCGAT \textit{GAACTCC}}{(EcoRI)}$

Table 37.1 Primers used for construction of pPIC9K GLP-1 analogs

The letters in italic and bold are the mutated gene. GGA was replaced by GAA

37.2.4 Transformation and Screening for Multicopy Transformants by G418 Selection

To obtain the recombinant yeast strain of GLP-1 analogs, *P. pastoris* GS115 was transformed by electroporation at 1.5 kV with *Sal*I-cut fragment of pPIC9K-GLP-1-analogs. The cells were plated on MD His-plates for 3–5 days at 30 °C. Single colonies from the MD plate were carefully removed under sterile conditions and patched to YPD plates containing different concentrations of G418 (0.5, 1.0, 2.0, 3.0 or 4.0 mg/mL) and incubated for 3–5 days at 30 °C in order to select multicopy transformants. The transformant with the fastest growth rate on YPD plates containing the highest concentration of G418 was screened and identified. PCR amplification was performed to confirm whether the gene of GLP-1 was integrated into the *P. pastoris* GS115 genome, according to the specifications of the Multicopy Pichia Expression Kit.

37.2.5 Expression of the Recombinant Protein

The transformant was inoculated into a 250 mL baffled Erlenmeyer flask with 50 mL of buffered minimal glycerol medium (BMGY, 100 mM potassium phosphate, pH 6.0, 1.34 % yeast nitrogen base (YNB), 0.002 % biotin and 1 % glycerol) and was grown at 30 °C overnight with vigorous agitation at 220 rpm. When the optical density (OD $_{600}$) reached 5.0, the cells were collected by centrifugation at 12,000 rpm for 10 min. The cell pellet was resuspended in BMMY medium

348 P.-Y. Li et al.

(BMGY with 0.5 % methanol instead of 1 % glycerol) at a starting OD_{600} . Methanol, as an inducer, was added to a final concentration of 0.5 % every 24 h. Samples were collected for analysis after 48 h.

37.2.6 HPLC Analysis

After lyophilization and dissolution of the samples in 1 mL PBS for 10 mL supernatant, analytical HPLC was performed on a Jupiter 5u C18 300A with solvents A (11.5 g ammonium diacid phosphate in 900 mL water and 100 mL acetonitrile, pH 3.7) and B (960 mL acetonitrile in 270 mL water) at a flow rate of 1.0 mL/min. The peptides were eluted with a linear gradient of solvent B from rate of 20–60 % for 30 min and 20 % solvent B eluted for 10 min.

37.3 Results

37.3.1 Construction of the GLP-1 Analogs Recombinant Plasmid

The recombinant yeast *P. pastoris* GS115 strains containing pPIC9K GLP-1 analog-1 and pPIC9K GLP-1 analog-2 used for expressing were constructed. To secretly express the authentic forms of the peptide, the genes of GLP-1 analogs were ligated seamlessly into the downstream of α -factor signal sequence in pPIC9K vector. The results of restriction enzyme digestion and DNA sequencing showed that the expression vectors of two GLP-1 analogs were both constructed successfully (Fig. 37.3).

37.3.2 Establishment of GLP-1 Analogs Recombinant Yeast Strain

Following amplification in the $E.\ coli$ DH5 α strain, the recombinant plasmids were linearized with SalI and transformed into $P.\ pastoris$ GS115. After screening, the genomic DNA of the potential positive clones was extracted and PCR was performed to detect the integration of the extraneous gene. As shown in Fig. 37.4, the correct band could be observed in Lane 1, indicating that the target gene has been integrated into the genome of recombinant strains.

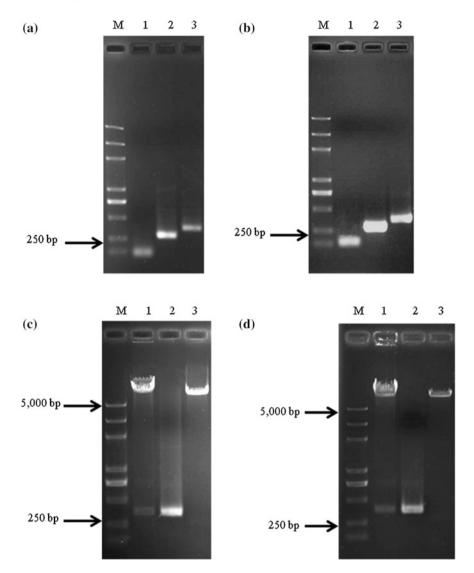


Fig. 37.3 Construction of recombinant plasmid pPIC9K-GLP-1-analogs. **a** The PCR products during the amplification of GLP-1 analog-1. **b** The PCR products during the amplification of GLP-1 analog-2. *Lane M* DNA ladder marker; *Lane 1* the PCR product of GLP-1 analog; *Lane 2* the PCR product of α-factor signal sequence; *Lane 3* the PCR product of α-factor signal sequence fused with GLP-1 analogs. **c** Identification of recombinant plasmid pPIC9K-GLP-1-analog-1. **d** Identification of recombinant plasmid pPIC9K-GLP-1-analog-2. *Lane M* DNA ladder marker; *Lane 1* recombinant plasmid pPIC9K-GLP-1-analogs digested by *Eco*RI and *Bam*HI; *Lane 2* the PCR product; *Lane 3* the vector pPIC9K digested by *Eco*RI and *Bam*HI

350 P.-Y. Li et al.

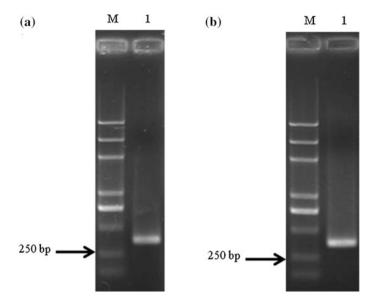


Fig. 37.4 Analysis of integration of GLP-1 analogs in the genome of *Pichia pastoris* GS115. **a** The analysis of the integration of GLP-1 analog-1. **b** The analysis of the integration of GLP-1 analog-2. *Lane M* DNA ladder marker; *Lane 1* the PCR product of GLP-1 analog gene

37.3.3 Analysis of Recombinant GLP-1 Analog-1 and GLP-1 Analog-2 Protein

Expression of the GLP-1 analog gene was induced by the activation of AOX1 promoter with methanol. To confirm GLP-1 analog protein, the purified protein was analyzed by HPLC. As shown in Fig. 37.5, the results showed that the recombinant GLP-1 analog-2 could be detected in the supernatant of fermented broth, whereas

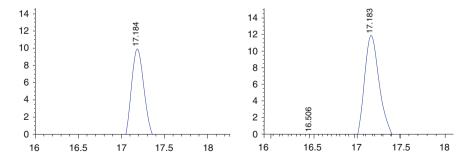


Fig. 37.5 HPLC results of GLP-1 analog proteins. ${\bf a}$ The standard sample; ${\bf b}$ the recombinant GLP-1 analog-2 product

the GLP-1 analog-1 could not be found, indicating that replacement of the 28 amino acid phenylalanine by serine could improve the stability of GLP-1(7–36) amide.

37.4 Conclusion

GLP-1 can be degraded by the action of DPP-IV in vitro and in vivo. In addition, similar to Dpp4, NEP 24.11 is currently considered to function as another degrading enzyme for the destruction and disposal of GLP-1 [10, 15]. Such degradation could not only limit the clinical use of GLP-1, but also hinder the fermented production of this drug. In this study, two mutated GLP-1 analogs which have K34R and K34R/F28S replacements, respectively, were successfully constructed and stably transformed into yeast strains. Using HPLC analysis, we found that compared with K34R mutation, the additional replacement of Phe28 with Ser could significantly further increase the stability of GLP-1.

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Chapter 38 Optimization of Fermentation Condition of Man-Made Bee-Bread by Response Surface Methodology

Chuanren Duan, Yongfen Feng, Hao Zhou, Xiaohua Xia, Yaning Shang and Yamin Cui

Abstract In research and development for drugs with thrombolytic function, natto kinase is commonly used for the treatment of thrombosis drug. The objective of this research is to use the Box–Behnken design to optimize the four fermentation parameters for activity of nattokinase to create bee-bread by a mixed culture with *Lactobacillus plantarum* MRS3, *Bacillus subtilis natto* ATCC15245, and fresh beepollen. The maximum yield of 801.29 IU/mL nattokinase activity was obtained after optimization experiments of culture conditions such as inoculum size 6.6 %, water size 35 %, fermentation temperature 33 °C, culture time 9 days (216 h), and pH stay at the natural range, and incubated the mixture in OMWS Oumai with a solid state cultivation, and finally obtained the new bee-bread products with high nattokinase activity and rich nutrient. The results demonstrated that exploiting functional food of bee-bread as both medicine and food is viable.

Keywords Response surface methodology · Nattokinase · *Lactobacillus planta-rum* · *Bacillus subtilis* · Bee-bread · Optimization

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C. Duan et al.

38.1 Introduction

In recent years, nattokinase has drawn the increasing attention of investigators because of its important physiological functions in the heart and cerebral vessels. Nattokinase is a potent fibrinolytic enzyme that is considered to be a new promising agent for thrombolytic therapy, while the main drugs of urokinase (UK) and streptokinase (SK) are too expensive and their half-life is short. Compared to the clinical thrombolytic drugs (urokinase and streptokinase), nattokinase have several advantages such as safe, low cost, and easy oral administration [1]. This enzyme has been found in many resources, including Japanese natto [2], Chinese food douchi [3], and doufuru [4]. It is also present in various microorganisms, the most important strain is the genus *Bacillus* [5]. Therefore, many researchers have focused on the healthy food which was fermented by probiotics.

Pollen is the male gamete of seed plant, which plays an important role in sexual reproduction, which is rich in essential nutrients like protein, carbohydrates, vitamins, growth hormone, and other trace elements, and therefore has high nutritional value and special health function. So, bee-pollen is known as the "longevity beauty food", "micro-nutrient library" [6]. Bees do not consume pollen as collected by foragers, the pollen pellets are stored in cells, and honey, nectar, and glandular secretions are added to the mass of stored pollen. Stored pollen generally has a specific bacterial flora associated with it. Pollen stored in this way undergoes lactic acid fermentation and becomes what is called "bee bread". Bee bread is stored in combs and differs from fresh bee-collected pollen, but fermentation may be responsible for increasing stabilization of the product or may lead to chemical changes that increase digestibility and nutritive value for the bee pollen; the worker bees and larvae bees choose the bee-bread as their food but not pollen. Researchers show that at the appropriate period, the nutrient content in bee-bread is higher than pollen [7].

Because the natural output of bee-bread is limited, cannot meet the demand of the market and the price is high. We must pave new roads to make artificial bee-bread. The aim of this study was to maximize nattokinase activity in bee bread, and develop a functional food used as both medicine and food. Optimization was performed as follows: selection of the most significant factors, searching the optimal level using the response surface methodology (RSM), and a confirmation experiment to verify the optimal conditions.

38.2 Materials and Methods

38.2.1 Strains and Incubate Conditions

Lactobacillus plantarum MRS3 was separated and identified from rape bee-bread from Yunyang, Chongqing, China, by Duan et al. [8]. Bacillus subtilis natto

ATCC15245 was purchased from American Type Culture Collection. These strains were maintained as a spore suspension in 75 % glycerol and stored at -20 °C.

Nutrient agar medium for *Bacillus subtilis natto* (in g/L distilled water) [9]: peptone, 10; beef extract, 30; NaCl, 5; agar, 15; and pH 7.0.

MRS medium for *L. plantarum* (in g/L distilled water) [10]: peptone, 10; beef extract, 10; yeast extract, 5; citric acid diammonium hydrogen, 2; glucose, 20; K₂HPO₄, 2; sodium acetate, 5; MgSO₄, 0.5; MnSO₄, 0.25; Tween 80, 1 mL; agar, 15; and pH 6.2–6.4.

The bacteria were maintained on 15 % agar slants, the slants were kept at 4 $^{\circ}$ C and sub-cultured every 20 days in aforementioned media. The inoculum used in the following fermentation was developed by transferring 1 loop-full of the organism from slant to 100 mL Erlenmeyer flasks with 30 mL of liquid medium (aforementioned media without agar), and the inoculated flasks were incubated at 32 $^{\circ}$ C in orbital shaker at 180 rpm for 24 h.

38.2.2 Fermentation Conditions

Fermentation experiments were performed with 100 mL Erlenmeyer flask containing 15 g fresh pollen (procured from Yunyang, Chongqing, China), pollen sterilized by autoclaving for 15 min at 90 °C in order not to damage the nutrient, then inoculated the two bacteria successively, controlling the temperature of OMWS Oumai and solid state cultivation mixture for a few days with the normal pH.

38.2.3 Optimization of Culture Conditions for Inoculum Size, Water Size, Fermentation Temperature, Culture Time

RSM is a powerful technique for testing multiple process variables, because fewer experimental trials are required compared with the study of one variable at a time. Also, interactions between variables can be identified and quantified by this technique [11]. RSM helps to explore the optimum concentrations of each variable. RSM has been successfully used for the enhancement of fibrinolytic enzyme production by *Bacillus* sp. strain AS-S20-1 [12], *Bacillus subtilis* [13], and *Bacillus natto* [14].

In this article, the effects of inoculum size, water size, fermentation temperature, and culture time were researched [15]. Primarily, one of these factors changed while the others maintained. Based on these studies, these four independent factors were chosen for the further optimization experiments to obtain maxima nattokinase activity using RSM. The low, middle, and high values were coded as -1, 0, +1, respectively, as present in Table 38.1. The software Design-Expert (Version 8.0.6,

C. Duan et al.

Levels	Variables				
	X_1 Inoculum size	X ₂ Water size	X ₃ Temperature	X ₄ Culture time	
	(%)	(%)	(°C)	(day)	
-1	3	30	30	8	
0	5	35	34	9	
1	10	40	37	10	

Table 38.1 Factors and levels in the response surface design

Stat-Ease Inc., Minneapolis, USA) was used for experimental design, the treatment schedule for the model is given in Table 38.2. The response values (Y) in each trial was the average of triples.

38.2.4 Enzyme Separation

At the culture end, the separation of enzyme was according to Wen et al. [16]. About 3 g of sample was dissolved in 45 mL of sterilized normal saline (SS). This mixture stood overnight at 4 $^{\circ}$ C and was centrifuged at 5,000 rpm for 20 min in a cooling centrifuge at 4 $^{\circ}$ C. Supernatant was collected as the crude enzyme, and 10 μ L of aliquot was prepared on a fibrin plate to assay the fibrinolytic activity.

38.2.5 Enzyme Assay

Fibrinolytic activity of the extracted enzyme was determined by the Fibrin Plate assay [17]. Agarose, fibrinogen, and thrombin were prepared, 15 mL of fibrinogen solution (0.89 mg/mL) was kept in a water bath for 15 min at 46 °C. The heated fibrinogen solution was then mixed with 20 mL of 1 % agarose solution (w/v) and mixed well by shaking, and 20 min later, the resulting media was poured into a dish (9 cm in diameter) with 1 mL of thrombin solution (15 IU thrombin in 10 mmol/L PBS buffer, pH 7.5). The dishes were in static cultivation for an hour to form fibrin clots, and 10 holes (1 mm in diameter) were punched on the dishes.

Aliquots of 10 μ L of lumbokinase at 200, 400, 600, 800, 1000, 1200, 1400, 1600, 1800 and 2000 IU/mL were separately added in each hole and incubated at 37 °C for 18 h (each with one test and one control). The diameter of the cleared zone was measured and the thrombolytic circular area was calculated. The standard curve showing the relationship between lumbokinase activity (IU/mL) and thrombolytic circle area (mm²) was established.

The fibrinolytic activity of samples was determined using the curve through adding 10 μ L of sample supernatant prior on each hole of the fibrin plate to incubation for 18 h at 37 °C, and then measuring the lytic area.

Runs X₁ Inoculum X2 Water X_3 X_4 Culture Nattokinase size (%) Temperature activity (IU/mL) size (%) time (day) (°C) 427.19 -1 629.20 784.74 -1-1646.32 491.43 -1804.37 -1 -1 551.93 -1538.49 -1-1 601.80 -1526.00 532.69 -1 772.35 -1464.41 -1416.29 555.89 506.57 -1 574.35 -1-1583.92 618.40 -1 590.08 -1796.31 561.29 -1-1573.84 611.90 -1 506.03 -1-1531.46 -1432.19 767.39 583.06

Table 38.2 Box-Behnken response surface experimental design and results

38.3 Results and Discussion

38.3.1 The Standard Curve of Lumbokinase Activity

The applications of 10 μ L lumbokinase, at 200, 400, 600, 800, 1000, 1200, 1400, 1600, 1800, and 2000 IU/mL, all produced clear fibrinolytic zones in the fibrin plates. The lumbokinase activity (IU/mL) and the lytic circle area (mm²), averaged with triples, the activity and the area (mm²) were exponentially correlated ($R^2 = 0.986$, P < 0.001) (Figs. 38.1 and 38.2, Table 38.3).

358 C. Duan et al.

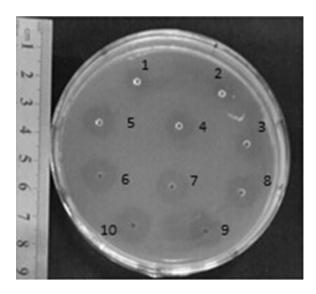


Fig. 38.1 Fibrin plate showing *clear zones* with the lumbokinase 200 IU/mL $\it{1}$, 400 IU/mL 2, 600 IU/mL $\it{3}$, 800 IU/mL $\it{4}$, 1,000 IU/mL $\it{5}$, 1,200 IU/mL $\it{6}$, 1,400 IU/mL $\it{7}$, 1,600 IU/mL $\it{8}$, 1,800 IU/mL $\it{9}$ and 2,000 IU/mL $\it{10}$

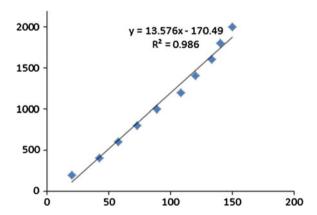


Fig. 38.2 The *standard curve* of the relationship between lumbokinase activity (*Y*) and thrombolytic circle area (*X*) established based on the data of 200, 400, 600, 800, 1000, 1200, 1400, 1600, 1800, and 2000 IU/mL of lumbokinase

38.3.2 Response Surface Methodology

The significant factors (inoculum size, water size, fermentation temperature, and culture time) were examined further by RSM using Box–Behnken design. The nattokinase activity (response) is presented in Table 38.2.

Runs	Thrombolytic circle area (mm ²)	Lumbokinase activity (IU/mL)				
1	20.5084	200				
2	42.4292	400				
3	57.5489	600				
4	73.1382	800				
5	89.2253	1,000				
6	108.2495	1,200				
7	120.1410	1,400				
8	133.5504	1,600				
9	140.6054	1,800				
10	150.4396	2,000				
Based on the ANOVA of the model (Table 38.4), the Model F -value of 44.49 implies the model is significant. There is less than 0.01 % chance that the "Model F -Value" is not significant. Values of "Prob $> F$ " less than 0.0500 indicate model terms are significant. In this model, C , AB , AC , BD , CD , A^2 , B^2 , C^2 , D^2 are						

Table 38.3 The actual values of the thrombolytic circle area

Table	38.4	Analysis	s of	variance	table

Source	Sum of squares	df	MeanSquare	F-value	P-value prob > F	
Model	3.279E + 005	14	23422.55	44.49	<0.0001	Significant
A	1665.69	1	1665.69	3.16	0.0970	1
В	826.51	1	826.51	1.57	0.2308	
C	9133.39	1	9133.39	17.35	0.0010	1
D	1291.48	1	1291.48	2.45	0.1396	1
AB	8542.38	1	8542.38	16.23	0.0012	1
AC	3888.15	1	3888.15	7.39	0.0167	1
\overline{AD}	566.92	1	566.92	1.08	0.3170	1
BC	1483.79	1	1483.79	2.82	0.1154	1
BD	32130.56	1	32130.56	61.03	<0.0001	1
CD	5323.89	1	5323.89	10.11	0.0067	1
A^2	75379.24	1	75379.24	143.17	< 0.0001	1
B^2	66328.38	1	66328.38	125.98	< 0.0001	1
C^2	74170.93	1	74170.93	140.88	< 0.0001	1
D^2	1.794E + 005	1	1.794E + 005	340.78	< 0.0001	1
Residual	7370.87	14	526.49			1
Lack of fit	6397.56	10	639.76	2.63	0.1822	Not significant
Pure error	973.31	4	243.33			1
Cor total	3.353E + 005	28				1

360 C. Duan et al.

Table 38.5 Analysis of variance (ANOVA) table for response-surface quadratic model

Parameter	Value
Std. dev.	22.95
Mean	585.51
R-squared	0.9780
Adj R-squared	0.9560
Pred R-squared	0.8856
Press	383075
Adeq precision	22.769

significant model terms. The equation in terms of the coded factors is given as follows (38.1):

Nattokinase activity(IU/mL) =
$$+785.03 + 11.78 \times A + 8.30 \times B - 27.59 \times C - 10.37 \times D$$

 $-46.21 \times A \times B + 31.18 \times A \times C + 11.90 \times A \times D - 19.26 \times B \times C$
 $+89.62 \times B \times D - 36.48 \times C \times D - 107.80 \times A^2 - 101.12 \times B^2$
 $-106.93 \times C^2 - 166.31 \times D^2$ (38.1)

where A is inoculum size, B is water size, C is fermentation temperature, and D is culture time. Values greater than 0.1000 indicate the model terms are not significant. In Table 38.5, the "Lack of Fit F-value" of 2.63 implies that lack of fit is not significant as the pure error. There is a 18.22 % chance that the "Lack of Fit F-value" is nonsignificant. Nonsignificant lack of fit is good—we want the model to fit.

The fit of the model was expressed by the coefficient of regression R^2 , which was found to be 0.9780, indicating that the model gives predictions in 97.8 % confidence level. The "Pred R^2 " of 0.8856 is in reasonable agreement with the "Adj R^2 " of 0.9560. "Adeq Precision" measures the signal to noise ratio. A ratio greater than 4 is desirable. The ratio of 22.769 indicated an adequate signal. In conclusion, this model can be used to navigate the design space.

Accordingly, three-dimensional surface response plots were generated for the pairwise combination of the four factors, while keeping the other two factors at their center point levels. The shape of the contour can reflect the interaction significance, the circular means interaction was not significant, while the oval showed significant interaction. Graphs for the four significant interactions are given here to highlight the roles played by these factors (Fig. 38.3a–d).

As shown in Fig. 38.3a, when keeping the inoculation amount at a certain level, the nattokinase activity increases while the water added increases, but when the water content is greater than 36 %, nattokinase enzyme activity is on the decline. As the water content increases, the quantity of nattokinase activity increases, but when the inoculation amount continues to increase, nattokinase activity is gradually reduced. A similar profile was observed in Fig. 38.3c, d. It is evident from Fig. 38.3b that when the interaction between inoculum size and fermentation temperature was studied while keeping the other parameters such as water size and

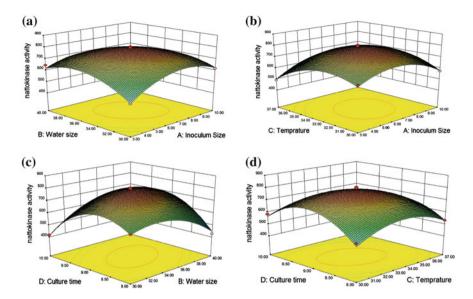


Fig. 38.3 Response surface plot for nattokinase activity, the interaction between inoculum size and water size (a), inoculum size and fermentation temperature (b), water size and culture time (c), and fermentation temperature and culture time (d)

culture time at their zero levels, the effect of temperature was significant on the response of nattokinase activity.

From the three-dimensional surface response plots, the optimal process parameters were obtained. It was observed that the fermentation condition of man-made bee-bread was inoculum size 6.59 %, water size 35.23 %, fermentation temperature 33.05 °C, and culture time 9 days. In combination with the results of actual operation convenient and variance analysis, we adjusted the parameters as: inoculum size 6.6 %, water size 35 %, fermentation temperature 33 °C, and culture time 9 days, yielded a maximum nattokinase activity of 801.29 IU/mL which exactly matched the predictions of 787.169 IU/mL.

38.4 Conclusion

From the experiments above, we present a preparation technology for a new nattokinase functional food—bee-bread from several new points as follows. First, a new fermentation substrate of pollen was used; second, we developed a study involving *L. plantarum* MRS3 and *Bacillus subtilis natto* ATCC15245 for production of nattokinase from pollen by the mixed culture; lastly, RSM was adopted to optimize the variables and to study their influence on nattokinase activity. The results showed a combination of concentration of inoculum size 6.59 %, water size 35.23 %, 362 C. Duan et al.

fermentation temperature 33.05 °C, and culture time 9 days, which would yield a maximum nattokinase activity of 801.29 IU/mL, which exactly matched the predicted value of 787.169 IU/mL. Finally the new bee-bread products were obtained with high nattokinase activity and rich nutrient, these data proved that it is possible to create a functional food of bee-bread, which is used as both medicine and food.

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Chapter 39 **Prediction of Lysine Acetylation Sites** in *Porcine Pancreas* Lipase Modified by the Ionic Liquids Using Molecular **Dynamics Simulations**

Yi-Gang Jia, Yang Zhang, Hong-Man Zhang, He Huang, Lu-Jia Zhang and Yi Hu

Abstract The molecular dynamics (MD) simulations method was used to study the lysine acetylation sites of *Porcine Pancreas* lipase (PPL) modified by ionic liquids [HOOCBMIm][C1] and [HOOCMMIm][C1]. By analyzing the effects impacting on the difficulty of lysine modifications upon different sites, including the solventaccessible surface area, hydrogen bonds, and salt-bridges, a prediction model was achieved. The prediction acquired the exact number of modified lysine (4 and 9 respectively) and the specific modification sites in the ionic liquids [HOOCBMIm] [Cl] and [HOOCMMIm][Cl] modification systems, respectively, which are consistent with the results of our previous studies.

Keywords Porcine pancreas lipase • Molecular dynamics simulation • Ionic liquids modification • Lysine acetylation sites • Solvent-accessible surface area

39.1 Introduction

Lipases (EC3.1.1.3) are a kind of enzymes as important biocatalysts due to their high selectivity, high stability, mild reaction conditions, with no need for coenzyme, and so on. Owing to their ability to catalyze various reactions including hydrolysis, esterification, transesterification, alcoholysis, and aminolysis, lipases are

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366 Y.-G. Jia et al.

widely used in pharmaceutical, chemical, food, feed, and energy industries [1–3]. However, lipases still have many shortcomings, especially the vulnerability of being inactive under unnatural conditions such as organic solvent surrounded, high temperature, and extreme pH [4]. Therefore, in order to improve their catalytic properties and make them adapt to further practical applications, quite a number of technological measures can be done to have lipases modified, such as genetic engineering, chemical modification, and immobilization [5, 6].

Recently, as a novel chemical modification reagent, ionic liquids (ILs) have been adopted to improve the properties of immobilized enzyme and modified free enzyme [7–12]. The results showed that the catalytic properties of lipases modified by ILs, such as thermal stability, hydrolytic activity, and enantioselectivity were improved to a different extent. Doumèche's group chose functionalized ionic liquids, e.g., imidazole and pyrrolidine, to modify the formate dehydrogenase (FDH), leading to prolong the half-life ($T_{1/2}$) from 1.5 \pm 0.6 days to 4.7–9 days in the carbonate buffer [7]. The significant improvement in the catalytic properties of enzyme was closely related to the excellent properties of ILs and the modification degree upon the lysine in enzymes [10]. However, due to the poor cognitive of the mechanism of how modification improves the catalytic performance of enzymes, it is difficult to offer rational guidance for the modification of various enzymes with different structures.

The identification of the modification sites was essential for uncovering the mysterious veil on the enhancement of catalytic properties after enzyme modification.

With the combination of molecular simulation and experimental methods, Mogharrab et al. attempted to clarify the mechanism regarding traditional chemical modification of the enzymes. AQ-modified HRP was constructed so that carboxylic acid anthraquinone (AQ) was covalently bound to the side-chain ε-nitrogens of these residues (Lys¹⁷⁴, Lys²³² and Lys²⁴¹); the results revealed the mechanism of structural stabilization and functional improved compared with native Horseradish peroxidase (HRP) [13]. However, ionic liquids, as a novel chemical modification reagent, take great advantages over the traditional modification reagent. Its mechanisms upon improving the catalytic properties of enzymes still remain unclear.

In our previous work, the modification numbers of PPL on two different ionic liquids ([HOOCBMIm] [Cl] and [HOOCMMIm] [Cl]) (3–6 and 7–13, respectively) can be obtained by MALDI-TOF mass spectra experiment [10]. However, the location of modification sites and mechanisms of catalytic properties need to be further elaborated. In this study, molecular dynamics (MD) simulations were performed to investigate the ILs modification sites of PPL in the following three solvents: water (system 1), [HOOCBMIm][Cl] (system 2), and [HOOCMMIm][Cl] (system 3), respectively. The specific modification sites of PPL in the three systems were analyzed and further discussed based on the results of MD simulations. These simulation results will be helpful to understand the mechanism of the enhancement of catalytic properties after enzyme modification and provide a theoretical basis for the design of ILs modifiers and the option of proper modification degrees in enzyme engineering.

39.2 Experimental

39.2.1 Acquisition of Protein Structure

The initial conformation of PPL solved 2.8 Å at resolution was taken from the NMR structure (PDB entry code: 1ETH) [14]. PPL is composed of 448 amino acid residues and 7 disulfide bonds. Ligands and inhibitors in three-dimensional model of PPL were removed prior to MD simulation. 128 bound water molecules in the crystal structure were reserved in the MD simulations.

39.2.2 Structure Determination

The geometric structure of ILs molecules (Fig. 39.1) were derived from Chemoffice (Cambridge Soft Corporation) program and then minimized and the cation charge was +1. The force field parameters of [HOOCBMIm][CI] and [HOOCMMIm][CI] were obtained through the Automated force field Topology Builder (ATB) online server [15]. Meanwhile, the coordinate files of those ILs were also derived from online server. Atom partial charges were obtained with the RESP method at the 6-31G(d) level by Gaussian 03 program [16].

39.2.3 Molecular Dynamics Simulations

All MD simulations were performed using GROMACS 4.5.4 package [17–19] at Hp Z800 workstation, employing the GROMOS96 53a6 united atom protein force field [20]. Protonation states of titratable groups Arg, Lys, Asp, His, and Glu were computed by an online server named PROPKA at pH 7.0 [21]. Each model was solvated in a cubic boxes filled with SPC water molecules [22] and neutralized by adding appropriate NA ions. The number of ILs in systems (1, 2 and 3) were 0, 100, and 100, respectively. The 100:1 molar ratio of the modifiers to PPL in systems

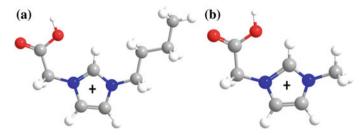


Fig. 39.1 The 2D geometry structural of [HOOCBMIm][Cl] and [HOOCMMIm][Cl], respectively. **a** [HOOCBMIm][Cl]. **b** [HOOCMMIm][Cl]

368 Y.-G. Jia et al.

2 and 3 were in line with the experimental section in previous studies [10]. The LINCS algorithm was used to constrain all bond lengths [23]. The Particle Mesh Ewald (PME) method was used to evaluate the long electrostatic interactions [24, 25]. The temperature of each system was maintained by velocity rescaling method [26]. Similarly, the pressure of each system was maintained by the Parrinello-Rahman method [27]. The cut-offs for short-range nonbonded interactions (van der Waals and electrostatics interactions) were set to 14 Å. The snapshots of each system were saved every 1 ps for data analysis.

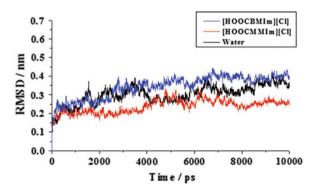
Each initial system was subjected to energy minimization (1,000 steepest descents and 1,000 conjugate gradient) and 100 ps equilibration with position restrained. Then the MD simulations were carried out for 10 ns with a timestep of 2 fs at constant temperature (300 K) and pressure (1 atm). The MD trajectories were analyzed and calculated by GROMACS 4.5.4 program including the $C\alpha$ root mean square deviations ($C\alpha$ -RMSD) and solvent-accessible surface area (SASA). Visual Molecular Dynamics 1.9.1 program (VMD) was used for the occupy percentages of hydrogen bonds and salt-bridges analysis [28]. The figures of protein structure were carried out by PyMOL program [29].

39.3 Results and Discussion

39.3.1 Structural Investigation

The equilibrium of three systems were evaluated using $C\alpha$ -RMSD compared to the simulation starting structure (Fig. 39.2). As can be seen from Fig. 39.2, the conformations of protein in systems 1 and 2 maintained stable throughout MD simulations. However, the protein structure in system 2 had larger float compared to systems 1 and 3 (Fig. 39.2). In the last 3 ns of the simulation in systems 1, 2, and 3, the $C\alpha$ -RMSD values of systems 1, 2, and 3 tended to relatively constant, leveling off around 0.3, 0.35 and 0.2 nm respectively. This is consistent with experimental results that reported the stability of the enzyme in hydrophobic and hydrophilic ILs [30].

Fig. 39.2 The Cα-RMSD of PPL respectively in systems 1 (*black*), 2 (*blue*) and 3 (*red*) and against time



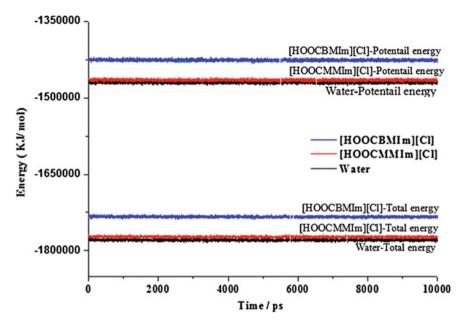


Fig. 39.3 The energy distribution of PPL, respectively, in systems 1 (black), 2 (blue), and 3 (red) and against time

Meanwhile, the overall stability of the total energy and potential energy were examined in three systems over the course of simulation. For all three systems, the total energy and potential energy were conservative and suggested the models in three systems to be well equilibrated (Fig. 39.3).

39.3.2 Calculation of Solvent Accessible Surface Area

SASA is the surface area of a protein molecule that is accessible to a solvent. The greater the degree of the same amino acids exposed to the solution phase, the greater the probability of its reaction with the reagents within the same solution phase. And the main reactive sites of lysine were concentrated in the side chain ε -amino group of lysine. Furthermore, the modification degree of lysine ε -amino groups affected by ILs was mainly determined by the SASA values. Some authors validated this concept for the correlation of residue relative reactivities with the accessibility of lysine residues on the protein surface [31–34].

The main chain structure, surrounding residues and solution phase influenced the variation of exposure rate of lysine residues in PPL. PPL contains 22 lysine residues located on the protein surface (amino acids position 26, 33, 70, 81, 92, 96, 108, 137, 198, 233, 240, 269, 296, 318, 342, 350, 351, 374, 398, 400, 418, and 427) (Fig. 39.4). The SASA values of residues are determined with a sphere of a 1.4 Å

370 Y.-G. Jia et al.

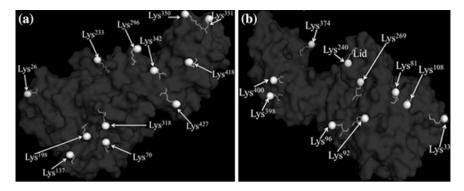


Fig. 39.4 Structure of PPL shown through surface representation. Lysine shown through *white stick* model representation (*Sphere* represents N atom). The *black* of PPL (**b**) is seen by rotating 180 around the Z-axis

radius to probe the surface of protein [35]. The SASA values of ε-amino group of lysine residues in PPL are listed in Table 39.1. As seen in Table 39.1, there are large differences in the SASA values of ε-amino group of each lysine residue in these three systems. In addition, it was found that the lysines with larger surface area in system 1 were different from those in systems 2 and 3 throughout MD simulation. This may be partially due to charge redistribution at the protein surface laid in the ILs system, leading to a different unfolding pattern. Although the lysine residues with higher exposure rate could be easily modified in system 1, the same residues could not be achieved in other systems. During the 10 ns simulations, compared with the lysine residues the status of protein structures obtained in system 1 (Fig. 39.5a) deviated substantially from that in systems 2 and 3 (Fig. 39.5b, c, respectively), it could be noted that the lysine residues status of protein structures in systems 1, 2, and 3 showed a significant difference. In order to seek for the precise sites of ILs modification, further studies involving protein structures should be performed with real ILs systems.

To identify the specific sites of ILs modification, the mean accessibility of the side chain ϵ -amino group of each lysine residue in PPL was analyzed throughout 10 ns MD simulation. There were 11 lysine residues with a SASA over 0.27 nm² in system 2, namely: Lys²6 (0.323 nm²), Lys³0 (0.272 nm²), Lys³1 (0.316 nm²), Lys³6 (0.295 nm²), Lys²69 (0.321 nm²), Lys²96 (0.290 nm²), Lys³50 (0.275 nm²), Lys³51 (0.322 nm²), Lys³74 (0.290 nm²), Lys⁴00 (0.293 nm²) and Lys⁴18 (0.277 nm²), and eight residues were with a SASA over 0.2 nm² and less than 0.27 nm², including Lys³2, Lys¹08, Lys¹37, Lys¹98, Lys²40, Lys³18, Lys³42 and Lys⁴27. Additionally, three residues were with low SASA, which was Lys³3, Lys³98, and Lys²33, respectively. While in system 3, the highest accessibilities for residues were Lys²6 (0.287 nm²), Lys³3 (0.332 nm²), Lys³1 (0.294 nm²), Lys³6 (0.315 nm²), Lys¹98 (0.307 nm²), Lys³31 (0.314 nm²), Lys²40 (0.317 nm²), Lys²50 (0.309 nm²), Lys³18 (0.304 nm²), Lys³50 (0.331 nm²), Lys³51 (0.327 nm²), Lys³74 (0.309 nm²), Lys³40 (0.293 nm²), Lys³418 (0.291 nm²) and Lys⁴27 (0.294 nm²), and

Residue Water^a (nm²) [HOOCBMIm][Cl]^a (nm²) [HOOCMMIm][Cl]^a (nm²) Lys²⁶ 0.160 0.323 0.287 Lys³³ 0.328 0.153 0.332 Lys⁷⁰ 0.285 0.272 0.253 Lys⁸¹ 0.306 0.316 0.294 Lys⁹² 0.246 0.252 0.224 Lys⁹⁶ 0.307 0.295 0.315 Lvs¹⁰⁸ 0.263 0.213 0.243 Lys¹³⁷ 0.098 0.231 0.247 Lys 198 0.304 0.260 0.307 $\overline{Lys^{233}}$ 0.168 0.045 0.314 Lys²⁴⁰ 0.268 0.245 0.317 Lys²⁶⁹ 0.321 0.309 0.306 Lys²⁹⁶ 0.313 0.290 0.322 Lvs³¹⁸ 0.220 0.263 0.304 Lys³⁴² 0.223 0.252 0.225 Lvs³⁵⁰ 0.275 0.331 0.333 Lvs³⁵¹ 0.322 0.327 0.328 $\overline{Lys^{374}}$ 0.210 0.290 0.309 Lvs³⁹⁸ 0.287 0.193 0.193 Lys⁴⁰⁰ 0.293 0.293 0.268 Lys⁴¹⁸ 0.289 0.277 0.291 Lvs⁴²⁷ 0.294 0.261 0.294

Table 39.1 The SASA of the ϵ -amino group in lysine residues on the PPL surface in systems 1, 2, and 3, respectively

five residues were with a SASA over $0.2~\rm nm^2$ and less than $0.27~\rm nm^2$, including Lys⁷⁰, Lys⁹², Lys¹⁰⁸, Lys¹³⁷ and Lys³⁴². The lowest SASA value lie in the residue Lys³⁹⁸ (0.193 nm²). Accordingly, based on the principle of accessibilities correlation, together with the average modification degrees (systems 2 and 3) which were obtained in preliminary experiments ($22.6 \pm 1.2~\%$ and $53.4 \pm 2.8~\%$ in systems 2 and 3, respectively), we define that lysine with a SASA value over $0.27~\rm nm^2$ can be easily modified [10]. Furthermore, we preliminarily supposed the corresponding modification sites to be easily modified by [HOOCBMIm][C1] were Lys²⁶, Lys⁷⁰, Lys⁸¹, Lys⁹⁶, Lys²⁶⁹, Lys²⁹⁶, Lys³⁵⁰, Lys³⁵¹, Lys³⁷⁴, Lys⁴⁰⁰ and Lys⁴¹⁸, while that easily modified by [HOOCMMIm][C1] were Lys²⁶, Lys³³, Lys⁸¹, Lys⁹⁶, Lys¹⁹⁸, Lys²³³, Lys²⁴⁰, Lys²⁶⁹, Lys²⁹⁶, Lys³¹⁸, Lys³⁵⁰, Lys³⁵¹, Lys³⁷⁴, Lys⁴⁰⁰, Lys⁴¹⁸ and Lys⁴²⁷.

^a The mean SASA values of equilibrium stages (7-10 ns) during 10 ns MD simulations

372 Y.-G. Jia et al.

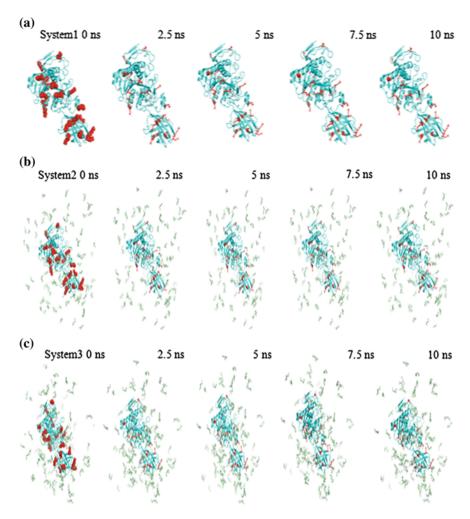


Fig. 39.5 The lysine residues status of PPL obtained in systems 1–3, respectively. a Water. b [HOOCBMIm][Cl]. c [HOOCMMIm][Cl]. All water in solvent was removed in systems 1–3

39.3.3 Effects of Bond Occupancy on the Modification Sites

The other effects impacting on the difficulty of modification of lysine residue at different sites in PPL were examined, respectively, including hydrogen bonds and salt-bridges. Combining the above results with SASA values, the modification sites of modified lysine were ultimately determined.

When Lys on the protein surface is being modified by chemical modifiers, its ϵ -NH₂ groups can react with other amino acids and cause an occupation of its atomic orbital by other atoms to form a new state, which makes the ϵ -NH₂ difficult

to be modified. We define this as "bond occupancy". The interaction between amino acids on the protein surface is most likely to form two noncovalent strong forces: hydrogen bonds and salt-bridges. Thus, we could partly determine whether Lys can be easily modified by analyzing the bond occupancy (hydrogen bonds and salt-bridges).

In the 10,000 frames of extracted structure from the 10 ns MD simulations, if with over 450 frames of the Lys ϵ -NH₂ occupied, the Lys was then in low possibility of attack by the modifying agent in the solution. Therefore, we believe the Lys with over 450 frames occupied is difficult to be modified.

39.3.3.1 Effects of Hydrogen Bonds on the Modification Sites

In order to investigate the effects of hydrogen bonds on the modification sites, hydrogen bonds formed between lysine residues and its adjacent residues were analyzed. Based on the experiment results of PEGylation of Hirudin on ionexchange column, relative reactivities of lysine residues were also concerned with hydrogen bonds [31]. During the equilibrium stages (7–10 ns) of 10 ns MD simulations, the formation of hydrogen bonds in systems 2 and 3 are listed in Tables 39.2 and 39.3, respectively. In system 2 (Table 39.2), the lysine with hydrogen bonds frames over 450 were Lys²⁶, Lys⁷⁰, Lys⁸¹, Lys⁹², Lys¹³⁷, Lys²³³, Lys²⁶⁹ and Lys⁴⁰⁰, and we supposed that the ε-amino group on the side chain of these residues were difficult to be modified. Even though the e-amino group of lysine residue 26 had accessibility as high as 0.323 nm², it was involved in the formation of hydrogen bonds to affect its reactivity. Nevertheless, in system 3, the lysine with strong bond occupancy, i.e., Lys²⁶, Lys⁷⁰, Lys⁹², Lys¹⁰⁸, Lys¹³⁷, Lys²³³, Lys³⁴², Lys³⁷⁴, Lys³⁹⁸, Lys⁴⁰⁰, Lys⁴¹⁸ and Lys⁴²⁷, were identified with over 450 occupied frames in the trajectories (Table 39.3). Similarly, Lys²³³, with accessibility of 0.314 nm², whose ε -amino helped in the formation of hydrogen bonds, is difficult to be modified.

39.3.3.2 Effects of Salt-Bridges on the Modification Sites

To explore whether salt-bridges would affect modification sites, salt-bridges between lysines and its adjacent residues in each MD trajectory were analyzed for information about occupied percentages in detail. Shuvaev et al. found that the distance between the N atom of Lys⁷⁵ and carboxyl O atoms of Glu⁷⁹ in apoE may affect glycosylation rate [36]. The distance of salt-bridge between Lys⁷⁵ and Glu⁷⁹ in apoE2 were larger than that identified in apoE2 and apoE3 by molecular simulation methods, thus the 75 lysine residues in apoE2 was more easily glycosylated. Similarly, O'Brien et al. found that the nitrogens of Lys⁶⁵ and Lys¹⁴⁹ with moderately accessibilities were engaged in salt pairing with carboxylate side chain of Glu⁶⁴ and Asp²⁵⁸, respectively [32]. Therefore, the Lys⁶⁵ and Lys¹⁴⁹ in HRP had less possibility to react with ethylene glycol bis (succinimidylsuccinate) (EGNHS).

374 Y.-G. Jia et al.

Table 39.2 Hydrogen bonds formed at the side chain ϵ -amino group of lysine and its around residue in [HOOCBMIm][CI] system environment

System	Donor	Acceptor	Snapshots occupancy/frame
[HOOCBMIm][Cl] ^a	Lys ²⁶	Asp ¹³	926
	Lys ²⁶	Asp ¹⁴	1,409
	Lys ³³	Ser ³¹	201
	Lys ⁷⁰	Ser ¹⁴⁵	500
	Lys ⁷⁰	Arg ⁷²	257
	Lys ⁸¹	Arg ¹¹²	738
	Lys ⁹²	Ser ⁸⁸	500
	Lys ⁹⁶	Asn ⁹³	387
	Lys ¹⁰⁸	Arg ¹¹²	423
	Lys ¹³⁷	Ser ¹⁴³	119
	Lys ¹³⁷	Thr ¹⁶⁹	402
	Lys ¹⁹⁸	Glu ¹⁷¹	210
	Lys ²³³	Tyr ²⁸⁹	729
	Lys ²⁴⁰	Asn ²⁴¹	50
	Lvs ²⁶⁹	Phe ³³⁶	739
	Lys ²⁶⁹	Tyr ²⁶⁸	208
	Lys ²⁹⁶	Ala ²⁹⁴	39
	Lys ³¹⁸	Thr ³¹⁹	9
	Lys ³¹⁸	Phe ¹⁹⁹	90
	Lys ³⁴²	Phe ²⁹⁸	308
	Lys ³⁵¹	Gln ³⁷⁸	6
	Lys ³⁷⁴	His ³⁵⁵	391
	Lys ³⁹⁸	Phe ³⁶¹	44
	Lys ³⁹⁸	Asn ⁴⁴⁶	26
	Lys ⁴⁰⁰	Thr ⁴²⁰	464
	Lvs ⁴¹⁸	Asp ⁴³⁰	270
	Lys ⁴²⁷	Arg ⁴²³	34
	Lys ⁴²⁷	Tyr ⁴²⁹	278

^a Hydrogen bonds were formed based on three criteria: (1) the distance between hydrogen atom and H-bond receptor ranges from 1.2 to 3.5 Å; (2) an angle constraint (H-bond donor—hydrogen atom—H-bond receptor is equal to or larger than 120°)

In system 2, Lys²⁶, Lys³³, Lys⁷⁰, Lys⁸¹, Lys¹⁰⁸, Lys¹³⁷, Lys¹⁹⁸, Lys²⁴⁰, Lys²⁶⁹, Lys³¹⁸, Lys³⁴², Lys³⁷⁴, Lys⁴¹⁸ and Lys⁴²⁷ participated in salt form with acidic amino acids (Asp and Glu) to a different extent (Table 39.4). The relatively high "bond occupancy" were identified as Lys²⁶, Lys⁷⁰, Lys¹⁰⁸, Lys³⁴², Lys³⁷⁴, Lys⁴¹⁸ and Lys⁴²⁷, which had low reactivities. Thus, while Lys⁷⁰ and Lys⁴¹⁸, in system 2 had high accessibilities, the "bond occupancy" analysis of them suggested that these two residues had little chance to react with ILs. In addition, the salt-bridges analysis of lysines in system 3 showed that Lys⁷⁰, Lys¹⁹⁸, Lys³⁴² and Lys⁴¹⁸ had

System	Donor	Acceptor	Snapshots occupancy/frame
[HOOCMMIm][Cl]	Lys ²⁶	Asp ¹³	926
	Lys ²⁶	Ala ¹⁸	100
	Lys ³³	Ser ³¹	201
	Lys ⁷⁰	Asn ¹⁴⁶	976
	Lys ⁷⁰	Glu ⁹⁸	1,083
	Lys ⁸¹	Glu ⁸³	143
	Lys ⁹²	Ser ⁸⁸	1,914
	Lys ⁹⁶	Asn ⁹³	413
	Lys ¹⁰⁸	Asp ³⁶	806
	Lys ¹³⁷	Thr ¹⁶⁹	583
	Lys ¹⁹⁸	Glu ¹³⁷	210
	Lys ²³³	Tyr ²⁸⁹	908
	Lys ²⁴⁰	Asn ²⁴¹	12
	Lys ²⁶⁹	Tyr ²⁶⁸	208
	Lys ²⁶⁹	Asp ²⁵⁸	83
	Lys ³¹⁸	Asp ³¹³	435
	Lys ³⁴²	Ser ³⁸⁴	109
	Lys ³⁴²	Glu ⁴²²	1,012
	Lys ³⁵¹	Gln ³⁷⁸	122
	Lys ³⁷⁴	Glu ³⁷¹	484
	Lys ³⁹⁸	Asn ⁴⁴⁶	806
	Lys ⁴⁰⁰	Asn ³⁶⁶	842
	Lys ⁴⁰⁰	Phe ³⁶¹	385
	Lys ⁴¹⁸	Asp ⁴³⁰	881
	Lys ⁴²⁷	Tyr ⁴²⁹	259
	Lys ⁴²⁷	Asp ⁴²⁵	971

Table 39.3 Hydrogen bonds formed at the side chain ϵ -amino group of lysine and around its residue in [HOOCMMIm][Cl] system environment

formed relatively firm salt-bridges (Table 39.5). Hence, the corresponding ε -amino group of those lysines were difficult to be modified.

By analyzing the "bond occupancy", the ϵ -amino amino groups of lysine residues easily reacting with the functionalized ionic liquid [HOOCBMIm] [CI] were Lys³³, Lys⁵6, Lys¹98, Lys²40, Lys²96, Lys³18, Lys³50, Lys³51 and Lys³98; while the most likely sites to be modified by [HOOCMMIm] [CI] were Lys³³, Lys⁵1, Lys⁵6, Lys²40, Lys²69, Lys²96, Lys³18, Lys³50 and Lys³51. And then, combining SASA with the modification numbers measured in the experiment (3–6 and 7–13, respectively) [10], we thought that the most likely modification sites in systems 2 and 3 were, respectively, (Lys³6, Lys²96, Lys³50 and Lys³51) (Fig. 39.6a) and (Lys³3, Lys³1, Lys³6, Lys²40, Lys²40, Lys²40, Lys²40, Lys³40, Lys³40, Lys³418, Lys³50, and Lys³51) (Fig. 39.6b).

^a Hydrogen bonds were formed based on three criteria: (1) the distance between hydrogen atom and H-bond receptor ranges from 1.2 to 3.5 Å; (2) an angle constraint (H-bond donor—hydrogen atom—H-bond receptor is equal to or larger than 120°)

Table 39.4 Salt bonds formed at the side chain ε-amino group of lysine and its around residue (Asp or Glu) in [HOOCBMIm][Cl] ionic liquid system environment

System	Donor	Acceptor	Snapshots occupancy/frame
[HOOCBMIm][Cl] ^a	Lys ²⁶	Asp ¹³	876
	Lys ²⁶	Asp ¹⁴	706
	Lys ³³	Asp ³⁶	27
	Lys ⁷⁰	Glu ⁹⁸	1,210
	Lys ⁸¹	Glu ⁸³	76
	Lys ¹⁰⁸	Asp ³⁶	717
	Lys ¹³⁷	Glu ¹³⁴	45
	Lys ¹⁹⁸	Glu ¹⁷¹	107
	Lys ²⁴⁰	Asp ²⁵⁸	3
	Lys ²⁶⁹	Asp ²⁵⁸	100
	Lys ³¹⁸	Glu ¹⁷¹	43
	Lys ³¹⁸	Asp ³¹³	271
	Lys ³⁴²	Glu ⁴²²	1,058
	Lys ³⁷⁴	Glu ³⁷¹	488
	Lys ⁴¹⁸	Asp ⁴³⁰	942
	Lys ⁴²⁷	Asp ³⁹⁵	2
	Lys ⁴²⁷	Asp ⁴²⁵	1,052

 $^{^{\}overline{a}}$ Ion pair formed: the distance between acidic amino acid (RCOO $^-$) and a basic amino acid (RNH $_3$) was less than 4 Å

Table 39.5 Salt bonds formed at the side chain ε-amino group of lysine and its around residue (Asp or Glu) in [HOOCMMIm][Cl] ionic liquid system environment

System	Donor	Acceptor	Snapshots occupancy/frame
[HOOCMMIm][Cl]	Lys ²⁶	Asp ¹³	140
	Lys ³³	Asp ³⁴	28
	Lys ⁷⁰	Glu ⁹⁸	1,344
	Lys ⁸¹	Asp ¹⁰⁶	34
	Lys ⁸¹	Glu ⁸³	184
	Lys ¹⁰⁸	Asp ¹⁰⁶	197
	Lys ¹³⁷	Glu ¹³⁴	12
	Lys ¹⁹⁸	Glu ¹⁷¹	502
	Lys ²⁴⁰	Asp ²⁵⁸	36
	Lys ²⁶⁹	Asp ²⁵⁸	5
	Lys ³⁴²	Glu ⁴²²	1,625
	Lys ³⁷⁴	Glu ³⁷¹	2
	Lys ⁴¹⁸	Asp ⁴³⁰	1,603
	Lys ⁴²⁷	Asp ³⁹⁵	199
	Lys ⁴²⁷	Asp ⁴²⁵	131

 $[\]overline{^a}$ Ion pair formed: the distance between acidic amino acid (RCOO $^-$) and a basic amino acid (RNH $_3$) was less than 4 Å

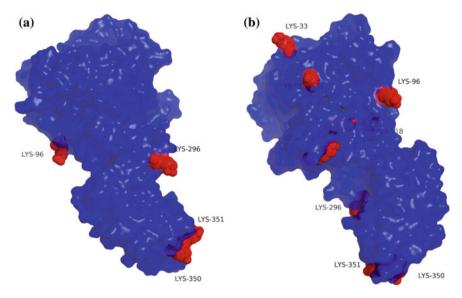


Fig. 39.6 The most likely modification sites of PPL modified by [HOOCBMIm][Cl] (a) and [HOOCMMIm][Cl] (b), respectively

39.4 Conclusion

Molecular dynamics simulations were performed on ILs modified PPL to investigate the role of structure-function on protein modification sites. In this study, a simple and novel method was established to predict the modification site. First, the number of modified lysine and the modification sites were preliminarily determined by the calculation of SASA. Second, "bond occupancy" including hydrogen bonds and salt-bridges were successfully applied to exclude the low reactivity of lysine residues and further to identify the modification sites. Finally, we found that the most possibly modification sites in systems 2 and 3 were (Lys⁹⁶, Lys²⁹⁶, Lys³⁵⁰ and Lys³⁵¹) and (Lys³³, Lys⁸¹, Lys⁹⁶, Lys²⁴⁰, Lys²⁶⁹, Lys²⁹⁶, Lys³¹⁸, Lys³⁵⁰ and Lys³⁵¹), respectively. These results of the number of modificated lysine residues are in line with the experimental results reported previously. Further research is underway to develop a more sophisticated method to verify the modification sites and elucidate the modification mechanism.

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Chapter 40 Effect of Oxygen on Fermentation **Characteristics of Three** Non-Saccharomyces from Hengshui Laobaigan

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Abstract In this paper, effects of different oxygen concentrations on the fermentation characteristics of three non-Saccharomyces from Hengshui Laobaigan fermented grains were studied. Three non-Saccharomyces strains, Pichia kudriavzevii (Y3), Pichia anomala (Y4), and Wickerhamomyces anomalus, were isolated from fermenting grains in previous study. Because it is difficult to control oxygen concentration in solid state fermentation, the amount changes of fluid in containers were utilized to control oxygen. At the initial fermented stage (0 h), the volumes of liquid 50, 100, 150, 200, and 250 mL were corresponding to the oxygen concentrations 6.90, 6.71, 6.25, 4.32, and 4.50 mg/L, respectively. Three strains were separately incubated for 3 % yeast inoculation quantity and cultured in different amounts of sterilized medium for 6 days. The data shows that oxygen had been largely consumed in the stage of rapid yeast reproduction. Significance test concluded that different oxygen concentrations have significant effects on ethyl acetate. Reducing sugar was increased generally with increasing liquid volume; while alcohol yield of Y3 had no significant difference except for 50 mL, but Y4 and Y6 had remarkable differences in every liquid volume. 1-propanol was unaffected by liquid volume while different oxygen concentration has significant effect on isobutanol.

Keywords Hengshui laobaigan · Non-Saccharomyces · Fermentation · Oxygen concentration

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381

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382 H. Zhu et al.

Chinese liquor is produced by solid state fermentation with grains and distilled mainly from fermented cereals. Grains in containers can be quickly fermented by many kinds of microbes in sophisticated biological transformation systems. Yeasts have played a major role in the process of fermentation [1].

In fermented systems, yeasts are divided into two groups: Saccharomyces cerevisiae and non-Saccharomyces, depending on different functions. The former is mostly completed ethanol fermentation and has better ethanol biosynthetic ability than the latter. But non-Saccharomyces can produce multiple enzymes which turn the precursor substance of grains into main flavor components. The flavor components are important for improving full-bodied and well-structured Chinese liquor [2, 3]. For example, the major ingredients of Hengshui Laobaigan-flavor Liquor included ethyl acetate, ethyl lactate, ethanol, acetaldehyde, 1-propanol, isopentyl, and isobutanol.

Most of the esters in the process of fermentation are produced by non-Saccharomyces, especially Pichia Saccharomyces [4]. For instance, Pichia anomala is the prevalent strain which can exhibit biocatalytic effect under various extreme environments. These extreme conditions include low pH, low water activity, high osmotic pressure, and anaerobic conditions [5]. Various non-Saccharomyces exhibit their ability to adapt to environments differently and the stages of metabolism do not mesh. Meanwhile, heat and oxygen transfer are difficult in solid state fermentation and are obviously different among different layers [6–8]. Main flavor components in earlier stage are seriously affected by oxygen. So research on the effect of different oxygen conditions on non-Saccharomyces has an important guiding sense to the actual production.

Three non-Saccharomyces strains, Pichia kudriavzevii, P. anomala and Wickerhamomyces anomalus, were isolated from fermenting grains in the previous study. Because it is difficult to control oxygen concentration in solid state fermentation, the amount changes of fluid in containers were utilized to control oxygen [9]. So the research on the effect on non-Saccharomyces fermentation characteristics in different blocks under different oxygen conditions will be an important guiding sense to the actual production.

40.1 Materials and Methods

40.1.1 Materials

Pichia kudriavzevii (Y3), P. anomala (Y4), and Wickerhamomyces anomalus (Y6) were stored in Hebei Hengshui Laobaigan liquor Co., Ltd.

40.1.2 Methods

The fermentation medium was prepared from sorghum. Shattered sorghum was proportionately added to water and boiled; starch was hydrolyzed by heat-resistant α -amylase and glucoamylase, in 1 h at 100 °C and 3 h at 65 °C, respectively. All matter was filtered and the filtrate was adjusted to 13 Brix by water. The fermentation medium was the adjusted filtrate and was separately took 50, 100, 150, 200, and 250–250 mL Erlenmeyer flasks at 121 °C for sterilization.

Three non-Saccharomyces strains, P. kudriavzevii, P. anomala, and W. anomalus, were separately incubated for 3 % yeast inoculation quantity and culture in different amount of sterilized medium. Erlenmeyer flasks were sealed by fermentation bung and stored static cultivation at 28 °C. Oxygen concentrations and yeast counts were monitored in the process of fermentation. The sample of Erlenmeyer flask was never reused. Reducing sugar, alcohol, ethyl acetate, 1-propanol, isopentyl, and isobutanol were determined on the sixth day. Reducing sugar of the fermenting liquor was determined and the other distillates were determined by means of gas—liquid chromatography.

Dissolved oxygen analyzer (SG9) is produced by Shanghai Mettler-Toledo Co., Ltd and have a probe which can drive 1 cm under the surface of medium. In this paper, Erlenmeyer flasks were sealed by a rubber stopper with two holes, one for the probe and other for the fermentation bung. All units were hermetically sealed. CO_2 produced during fermentation escaped through the fermentation bung.

Titrimetry was used to determine the reducing sugar and hydrometer method was to mensurate alcohol. In this paper, the initial reducing sugar level of the medium was 13.88 g/100 mL.

Gas chromatography (GC 2010) unit was purchased from Shimadzu Corporation, Japan and has an SPB-1701 capillary column. Column temperature: initial temperature 45 °C for 2 min, 4 °C/min risen to 65 °C, 10 °C/min risen to 220 °C; injector temperature: 200 °C; detector temperature: 240 °C; carrier gas (N₂): 0.085 Mpa (1.08 mL/min); make-up gas: 50.0 mL/min; H₂ gas: 30 mL/min; atmosphere: 300 mL/min.

10.00 mL distillates were added to 0.10 mL 3 g/100 mL internal standard and blended injection. Yeast count was measured by hemocytometer [10]. Statistical analysis of data was performed using SPSS 19.0 software and was significant at 0.05 level.

40.2 Results and Discussion

40.2.1 Changes of Oxygen Concentration in Static Fermentation

In this paper, all samples were cultured by liquid static fermentation, so oxygen of medium was obtained from the atmosphere by simple diffusion in Erlenmeyer flasks. While more liquid was in the flask, the less atmosphere was in it. A lot of 384 H. Zhu et al.

flavor components, such as alcohol, CO_2 , ethyl acetate, 1-propanol, isopentyl and isobutanol, were biosynthesized in the fermentation process. And more and more CO_2 from various fermentation stages escaped through the fermentation bung. As a result, fewer and fewer oxygen was kept in flask (Fig. 40.1).

Let us take *P. anomala* (Y4), for example. At the initial fermented stage (0 h), the volumes of liquid 50, 100, 150, 200, and 250 mL were corresponding to the oxygen concentrations 6.90, 6.71, 6.25, 4.32, and 4.50 mg/L, respectively. The fastest sample of oxygen decline rate was 50 mL liquid and had been reduced to 1.22 mg/L at 3 h, while the oxygen concentration of 100 mL liquid volume was reduced fewer. Oxygen of 200 and 250 mL decreased from 4.32 and 4.50 mg/L to 0.24 and 0.26 mg/L in 5.5 h, respectively; meanwhile, 150 mL had the lowest declined rate.

After 36 h at the oxygen concentration of zero, the tendency of all samples is consistent. But different liquid volume of every strain was markedly different starting point and trend of fermentation. This demonstrates that the oxygen concentration of medium controlled by liquid volume was correct.

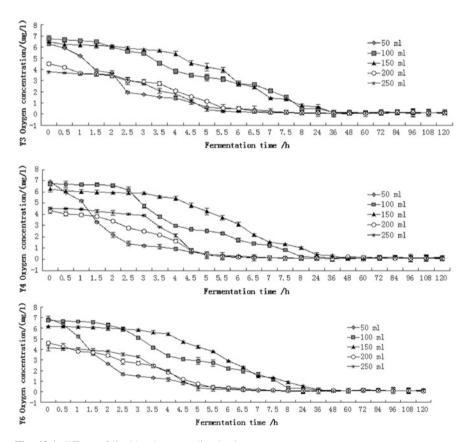


Fig. 40.1 Effects of liquid volume on dissolved oxygen

40.2.2 Effect of Oxygen on Yeast Counts of Three Strains

Effects on yeast counts of the three strains by different liquid volumes were clearly different (Fig. 40.2). Yeast counts of Y3 and Y6 in the early fermented part were closely related with liquid volumes of flask. The fastest growing period of yeast counts in the early stages was 0–24 h and was matched the fastest reducing period of oxygen concentration in Fig. 40.1. The data shows that oxygen had been largely consumed in the stage of yeast rapid reproduction. The maximum yeast counts of

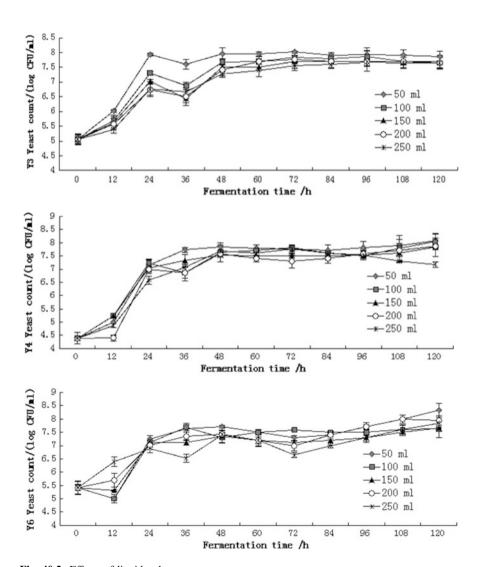


Fig. 40.2 Effects of liquid volume on yeast counts

386 H. Zhu et al.

three strains were the samples with 50 mL liquid medium and it was also proved this rule.

Y4 yeast counts were not significantly affected by liquid volumes (Fig. 40.2) in 0–96 h. While the count of 50 mL was up to 1.16×10^8 cfu/mL and 250 mL was 1.50×10^7 cfu/mL in final period.

It concluded that different oxygen concentrations have significant effects on the metabolic modulation mechanisms. This problem will need to be further gone into.

40.2.3 Effect of Oxygen on Reducing Sugar of Three Strains

After a 6 day static cultivation, the reducing sugar of Y3 every liquid volume was reduced to 2.50 g/100 mL and the utilization rate was high (Fig. 40.3). Y3 strain reducing sugar increased with liquid volume. The sugar of 50 and 250 mL was 7.21 and 12.10 g/100 mL while the utilization rates were 48.05 and 12.82 %, respectively. Trends of Y6 reducing sugar were significantly parallel with Y4, but the utilization rates were higher than the latter. In sum, reducing sugar was increased generally with increasing liquid volume.

40.2.4 Effect of Oxygen on Alcohol of Three Strains

Alcohol yield of Y3 increased generally with increasing liquid volume and Y4 was descended; Y6 alcohol had been declining in waves (Fig. 40.4).

From the results analysis, the less oxygen concentration was in the medium, the more alcohol yield of Y3; Y4 was just the opposite. The maximum alcohol yield of

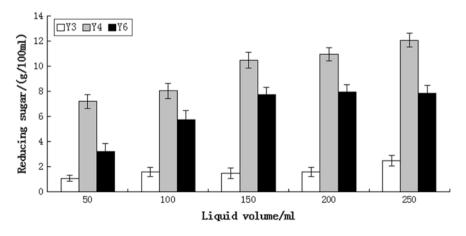


Fig. 40.3 Effects of liquid volume on reducing sugar

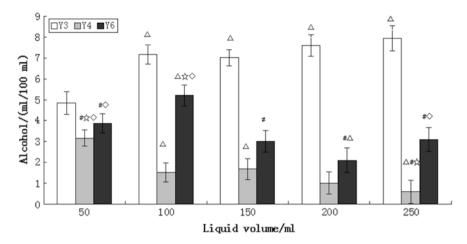


Fig. 40.4 Effects of liquid volume on alcohol. " \triangle " Significant differences with 50 mL; "#" Significant differences with 100 mL; " \triangle " Significant differences with 150 mL; " \triangle " Significant differences with 200 mL (P < 0.05)

Y3 was for 100 mL liquid while the minimum was for 200 mL. Significant test of alcohol yield registered that Y3 had no significant difference by liquid volume except for 50 mL, but Y4 and Y6 had remarkable differences in every liquid volume.

40.2.5 Effect of Oxygen on Ethyl Acetate of Three Strains

Ethyl acetate yields of 3 strains were depressed generally with increasing liquid volume (Fig. 40.5). Ethyl acetate yield of Y4 was highest to 385.74 mg/100 mL for 50 mL liquid volume, which was 3.27 times than Y3, 5.80 times than Y6 in the same condition. Significance test of ethyl acetate showed that Y3 had significant difference between 50 and 100 mL liquid volume and others, while Y4 and Y6 had remarkable differences in every liquid volume. It concluded that different oxygen concentrations have significant effects on ethyl acetate.

40.2.6 Effect of Oxygen on 1-Propanol, Isobutanol, and Isopentanol of Three Strains

1-Propanol yields of Y3 were depressed generally with increasing liquid volume (Fig. 40.6). 1-propanol of Y4 and Y6 altered little when the concentration of oxygen increased. Significance test of 1-propanol was not being performed because 1-propanol was unaffected by the liquid volume.

388 H. Zhu et al.

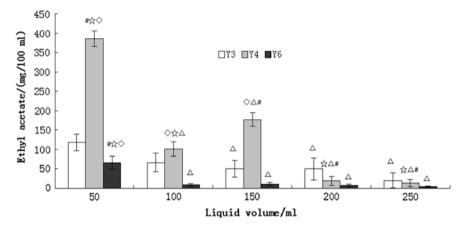


Fig. 40.5 Effects of liquid volume on ethyl acetate. " Δ " Significant differences with 50 mL; "#" Significant differences with 100 mL; " Δ " Significant differences with 150 mL; " Δ " Significant differences with 200 mL (P < 0.05)

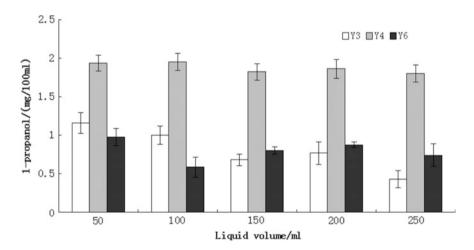


Fig. 40.6 Effects of liquid volume on 1-propanol

Isobutanol yields of Y3 and Y4 were also reduced with increasing liquid volume (Fig. 40.7). For every volume, Y6 isobutanol changed less than Y3 and Y4. Three strains had remarkable differences in every liquid volume and it concluded that different oxygen concentrations have significant effects on isobutanol.

Isopentanol yields of Y6 increased generally with increasing liquid volume (Fig. 40.8), but the Y3 and Y4 were just the opposite. In the same way, three strains had remarkable differences in every liquid volume and it concluded that different oxygen concentrations have significant effect on isopentanol.

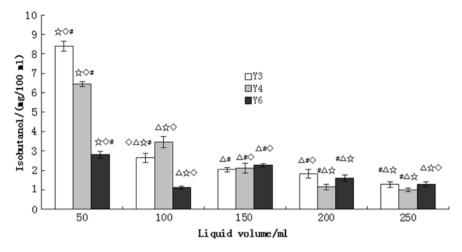


Fig. 40.7 Effects of liquid volume on isobutanol. " \triangle " Significant differences with 50 mL; "#" Significant differences with 100 mL; " \triangle " Significant differences with 150 mL; " \triangle " Significant differences with 200 mL (P < 0.05)

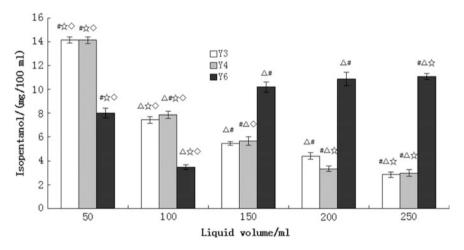


Fig. 40.8 Effects of liquid volume on isopentanol. " \triangle " Significant differences with 50 mL; "#" Significant differences with 100 mL; " \triangleq " Significant differences with 150 mL; " \triangleq " Significant differences with 200 mL (P < 0.05)

40.3 Conclusion

At the initial fermented stage (0 h), the volumes of liquid 50, 100, 150, 200, and 250 mL were corresponding to the oxygen concentrations of 6.90, 6.71, 6.25, 4.32, and 4.50 mg/L, respectively. Three strains were separately incubated for 3 % yeast inoculation quantity and cultured in different amounts of sterilized medium for

390 H. Zhu et al.

6 days. The data shows that oxygen had been largely consumed during the stage of rapid yeast reproduction. Significance test concluded that different oxygen concentrations have significant effect on ethyl acetate. Reducing sugar was increased generally with increasing liquid volume; while alcohol yield of Y3 had no significant difference except for 50 mL, but Y4 and Y6 had remarkable differences in every liquid volume. 1-propanol was unaffected by liquid volume while different oxygen concentrations have significant effect on isobutanol.

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Chapter 41 Microbial Transformation of Antitumor Isatin Derivatives by Fungi

Xiaolin Peng, Kailin Han, Yan Wang, Peng Yu and Hua Sun

Abstract The biotransformation of HKL-2c (1) and HKL-2h (2), the antitumor lead compounds found by our lab, were investigated. Compounds 1 and 2 were individually submitted to incubations with selected fungi *Trichoderma koningii* AS 3.4290, *Trichoderma viride* AS 2.2942, *Aspergillus flavus* AS 3.3950, *Aspergillus sydowii* AS 3.4258, *Aspergillus* AS 3.3885, and *Aspergillus niger* AS 3.3928. The products of hydrolysis from 1 and 2 by *Aspergillus* AS 3.3885 were identified based on spectroscopic methods. Those fungi, therefore, are useful for mild, selective hydrolysis of an ester of isatin substrates.

Keywords Isatin · Microbial transformation · Antitumor

41.1 Introduction

Isatin, an indole derivative, was found in many plants, such as *Isatis tinctoria*, *Calanthe discolor*, and *Couroupita guianensis* [1]. In recent years, many isatins have marked anticancer effects toward various types of cancer cell lines in vitro and in vivo. Some of them had been successfully developed to treat human cancer diseases in some therapeutic areas, including vinblastine [2], vincristine [3], and sunitinib [4]. Sunitinib (sunitinib, SU11248) is a novel multitargeted oral anticancer

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392 X. Peng et al.

Fig. 41.1 The structures of isatin, sunitinib, HKL-2c, and HKL-2h

drug, approved by the United States FDA in 2006. It was the first cancer drug simultaneously approved for two different indications, renal cell carcinoma (RCC) and imatinib-resistant gastrointestinal stromal tumor (GIST) [5].

Our previous study successfully identified the novel isatin derivatives 1 and 2 (Fig. 41.1) as the anticancer agents with inhibitory concentrations (IC₅₀) at nanomolar level [6]. The molecular mechanism analysis showed that compound 2 treatment downregulated the expression of CDK1 and cyclin B but upregulated the level of phosphorylated CDK1. Besides, compound 2 treatment caused the dissipation of mitochondrial membrane potential, activated caspase-3, and lowered the Bcl-2/Bax ratio in K562 cells [7].

To further expand the structural diversity of isatin and investigate the potential application, compounds **1** and **2** were subjected to biotransformation by the fungi *Trichoderma koningii* AS 3.4290, *Trichoderma viride* AS 2.2942, *Aspergillus flavus* AS 3.3950, *Aspergillus sydowii* AS 3.4258, *Aspergillus* AS 3.3885, and *Aspergillus niger* AS 3.3928. This paper describes the isolation, structural elucidation, and characterization of the transformed products, as well as a discussion of their antitumor activity.

41.2 Materials and Methods

41.2.1 Reagents and Strains

All reagents and solvents were purchased from commercial suppliers and used without further purification. D(+)-sucrose, D(+)-glucose, $NaNO_3$, $MgSO_4$, KCl, $FeSO_4 \cdot 7H_2O$, K_2HPO_4 , Na_2SO_4 anhydrous, and agar powder were purchased from Sinopharm Chemical Reagent Co., Ltd (China). The strains used in biotransformation were purchased from the China General Microbiological Culture Collection Center (CGMCC, Beijing, China), and stored and maintained on 20 % (v/v) glycerol stock solutions at -20 °C. It inoculated on a slant medium 28 °C cultured for 3–4 days and maintained at 4 °C.

NMR spectra were obtained at 400 MHz for ¹H and 100 MHz for ¹³C on a Bruker Avance 400 spectrometer, in DMSO-d₆, with solvent peaks used as

references. ESI–MS data were measured on an Agilent Technologies 6210 LC/TOF instrument. Column chromatography was performed with silica gel 200–300 mesh (Qingdao Marine Chemical Inc. Qingdao, China). TLC was performed on precoated silica gel GF₂₅₄ plates.

41.2.2 Biotransformation

Each bacterial strain was taken from the agar slant to shaking seed bottle culture (250 mL containing 100 mL of medium) by a platinum ring and was cultivated at 30 °C on rotary shakers (120 rpm) for 48 h.

The above seedlings were transferred to fresh medium in an amount of 1 %, and incubated at 30 °C for 24 h. Each flask was fed with 10 mg of compounds 1 and 2 (dissolved in DMSO, 100 g L^{-1}) and incubated for another 7 days. Thus, the total culture volume was 10 L, and the total of 400 mg of compounds 1 and 2 were fed as substrates.

41.2.3 Extraction and Isolation of Transformed Products

At the end of the biotransformation, the cultures were filtered through cheesecloth and centrifuged to separate the mycelium. The supernatant was condensed at 40 °C to approximately 3 L and extracted with same volume of EtOAc [8]. The mycelium was immersed and extracted with EtOAc (50 mL \times 3). The combined organic phases were dried over Na₂SO₄ anhydrous, and concentrated in vacuo as well as tested using TLC monitoring at 254 nm and visible light. The crude transformation products were purified by chromatography (petroleum ether/EtOAc = 100:0, 20:1, 10:1, 5:1, 3:1, 2:1, 1:1, v/v). The products were recrystallized with CHCl₃ and hexane to give the yellow crystals. All products (3–4) were identified on the basis of spectroscopic data. The 1 H, 13 C NMR and ESI–MS data are presented in Table 41.2.

41.2.4 Cytotoxic Assay

Cells (100 μ L) were cultured in 96-well plates at a density of 5 × 10⁴ cells/mL for 2 h (K562) or overnight (HepG2 and HT-29). Compounds (DMSO solution of 0.5 μ L) were added to each well to culture for another 48 h. MTT assay was performed using thermo microplate reader. The DMSO-treated controls were calculated as a cell viability value of 100 %. The IC₅₀ values were obtained by nonlinear regression using GraphPad Prism 4.0. IC₅₀ measurements for each compound were done three times.

394 X. Peng et al.

41.3 Results and Discussion

Six fungi were screened for their ability to transform compounds 1 and 2 in liquid potato-dextrose medium (*T. koningii* AS 3.4290 and *T. viride* AS 3.2942), and liquid Czapek-Dox medium (*A. flavus* AS 3.3950, *A. sydowii* AS 3.4258, *Aspergillus* AS 3.3885, and *A. niger* AS 3.3928). The secondary screening results are shown in Table 41.1. The compounds 1 and 2 could be converted into at least one product by *Aspergillus* AS 3.3885. Especially, an obvious product could be observed from the cultures of compound 1 with *Aspergillus* AS 3.3885. In addition, *T. koningii* AS 3.4290 and *A. niger* AS 3.3928 exhibited biotransform ability to compound 1. Therefore, the fungus *Aspergillus* AS 3.3885 cultivated in the liquid Czapek-Dox medium was found to be the most effective and was selected for the preparative-scale biotransformation of both compounds 1 and 2.

The obtained products were purified and identified by the nuclear magnetic resonance hydrogen spectrum (1 H-NMR), carbon spectrum (13 C-NMR), and Mass spectrometry (LC-MS). The spectroscopic data are shown in Table 41.2. Product **3** was obtained as a yellow crystal. The molecular weight was determined by ESI-MS from the [M+H]⁺ peak at m/z 308.4. The MS spectrum indicated the absence of methyl group (CH₂) comparing with compound **1** (molecular weight 321.3). The methyl peaks of acrylic acid methyl ester disappeared in both 1 H-NMR and 13 C-NMR. Therefore, the structure of product **3** was speculated into (E)-3-(1-benzyl-2,3-dioxoindolin-5-yl) acrylic acid. The product **4** was a yellow crystal and [M+H]⁺ peak at m/z 338.2. Comparing with compound **2** (molecular weight 351.4), the structure of product **4** could be (E)-3-(1-(4-methoxybenzyl)-2,3-dioxoindolin-5-yl) acrylic acid. The 1 H-NMR and 13 C-NMR analysis verified the correctness of the structure of compound **4**. The biotransforming scheme is shown in Fig. 41.2.

The anticancer activity of one of the representative products **4**, against three human tumor cell lines (K562, HepG2, and HT-29), was evaluated by MTT assay in vitro. Comparing with starting material **2**, the cytotoxicity of hydrolysis compound **4** was decreased (>10 μ M) (Table 41.3). This observation suggested that the methyl ester is a necessary structure unit for the activity of **2**. However, the hydrolysis products provide the free carboxyl group, which can be modificated into other ester, amide, aldehyde, hydroxyl, etc. In our previous work [6], the compounds **3** and **4** could be obtained using basic hydrolysis. But the reaction

Table 41.1 The screening results of isatin derivatives biotransformed by 6 strains

Strain	HKL-2c	HKL-2h
Trichoderma koningii AS 3.4290	+	_
Trichoderma viride AS 3.2942	-	_
Aspergillus flavus AS 3.3950	-	_
Aspergillus sydowii AS 3.4258	_	_
Aspergillus AS 3.3885	++	+
Aspergillus niger AS 3.3928	+	_

Note (-) No products; (+) products; (++) obvious products

Compounds	Spectroscopic data				
	¹ H-NMR	¹³ C-NMR	ESI-MS		
3	4.94 (s, 2H), 6.52 (d, 1H, J = 16.0 Hz), 7.00 (d, 1H, J = 8.0 Hz), 7.28–7.36 (m, 5H), 7.43 (d, 1H, J = 4.0 Hz), 7.55–7.59 (d, 1H, J = 16.0 Hz), 7.90 (d, 1H, J = 1.6 Hz), 7.93 (d, 1H, J = 4.0 Hz), 12.34 (s, 1H)	43.5, 111.8, 118.8, 119.1, 124.4, 127.9, 128.0, 129.1, 130.2, 135.9, 138.0, 143.0, 151.7, 159.0, 168.0, 183.0	308.4 [M+H] ⁺		
4	3.72 (s, 3H), 4. 86 (s, 2H), 6.52 (d, 1H, <i>J</i> = 16.0 Hz), 6.89 (d, 2H, <i>J</i> = 8.8 Hz), 7.02 (d, 1H, <i>J</i> = 8.8 Hz), 7.37 (d, 2H, <i>J</i> = 8.8 Hz), 7.57 (d, 1H, <i>J</i> = 16.0 Hz), 7.90–7.92 (m, 2H), 12.34 (s, 1H)	42.9, 55.5, 111.9, 114.51, 114.51, 118.7, 119.1, 124.3, 127.6, 129.3, 129.3, 130.1, 137.9, 143.0, 151.6, 158.9, 159.2, 168.0, 183.1	338.2 [M+H] ⁺		

Table 41.2 The ¹H, ¹³C NMR, and ESI-MS data of compounds 3 and 4

Fig. 41.2 The scheme of compounds 1 and 2 were transformed by Aspergillus AS 3.3885

Table 41.3 In vitro antitumor activities of compounds 2 and 4

Compounds	IC ₅₀ (μM)			
	K562	HepG2	HT-29	
2	0.0030 ± 0.001	0.030 ± 0.02	0.030 ± 0.01	
4	>10	>10	>10	
Camptothecin ^a	0.040 ± 0.02	0.050 ± 0.02	0.060 ± 0.01	

Note a Positive control

conditions were vigorous and polluting. By contrast, the condition of microbial transformation was mild and green.

Therefore, *Aspergillus* AS 3.3885 could be useful for mild, selective hydrolysis of an ester of isatin substrates. The further microbial transformation study of antitumor isatin derivatives is ongoing in our laboratory.

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396 X. Peng et al.

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Chapter 42 Effect of Ultrasound on Lysine Muriate Crystallization

Aijun Hu, Zili Chen, Shuting Jiao, Huanqin Peng, Yanshu Fan, Lin Chen, Meiling Liu and Jie Zheng

Abstract Ultrasound was applied to the crystallization process of lysine muriate solution. The effects on crystal yield of lysine muriate were investigated, including ultrasonic time, ultrasonic power, and crystallization time. The experimental results showed that the crystal yield reached 84.08 % with the adequate conditions of ultrasonic power 112 W, ultrasonic time 50 min, and crystallization time 5 h. And the effects of ultrasonic power on the nucleation induction period and crystal appearance of lysine muriate were also researched, the results indicated that with ultrasound, induction period was shortened significantly compared to that without it at the same supersaturation ratio. The higher the ultrasonic power, the shorter the induction period, and the smaller the crystals. With the increase of ultrasonic power, the crystals became smaller and more uniform. It can be concluded that using ultrasound-assisted crystallization cannot only improve the crystallization rate and crystal yield, but also obtain crystalline product with the characteristics of small particles and uniform distribution by controlling ultrasonic power.

Keywords Lysine muriate • Ultrasound • Crystallization

42.1 Introduction

Lysine is not only one of the most important amino acids, but also one of the essential amino acids for people's health [1]. Except for glutamic acid, the yield of lysine is the highest. Lots of foodstuffs and feedstuffs have few of lysine, so that lysine is needed to add to both human food and animal feed in order to improve their nutritional value [2]. In recent years, the market demand for lysine has become

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398 A. Hu et al.

more great because of the rapid development of feed and food industries, as well as its applications in medical area [1].

Generally, lysine exists in the form of muriate (i.e., lysine muriate) in both production and application, and lysine muriate is transferred to lysine after it is being taken by people. The higher its purity is, the more expensive it is. Among all methods to improve its purity, crystallization is an extremely important process for obtaining highly pure lysine muriate.

Conventional crystallization methods include cooling crystallization, vacuum cooling crystallization, evaporative crystallization, and reaction crystallization [3]. Ultrasonic crystallization as a novel method is received much attention in recent years. It has been studied with various crystallization systems and its advantages in several crystallization applications [4, 5] are disputed. As a size reduction method, ultrasonic crystallization can be considered as a more attractive size reduction method compared to grinding since under ultrasound conditions the crystallinity of the crystals does not decrease in most cases. Ultrasound narrows the metastable zone which can be also concluded from shortened induction period when the nucleation rate is determined empirically under isothermal conditions, i.e., ultrasound promotes nucleation dominated by a heterogeneous primary nucleation mechanism [6]. Li et al. [7] employed ultrasound in salting out crystallization and analyzed its influence on the crystal size distribution and shape. Chow et al. [8] applied ultrasound to a melt crystallization system for ice crystallization from sugar solutions. They also studied the influence of ultrasound on the nucleation rates and cavitation effects on ice fragmentation. According to Devarakonda et al. [9], ultrasound impacts crucially on the seed size, breakage of the solute lumps, and crystallization kinetics. Amara et al. [10] found out that ultrasound increased the crystal growth rate of potash alum, while crystals grown under ultrasound were smaller than those produced in stirred crystallizer without ultrasound. Marjatta et al. [11] examined crystallization of glycine with ultrasound. Martini et al. [12] investigated the effect of high intensity ultrasound (HIU) on crystallization behavior of anhydrous milk fat, the results showed that HIU application not only decreased the induction time of crystallization (faster crystallization) at a constant crystallization temperature, but also generated smaller crystals. Patel et al. [13] studied the recovery of lactose from reconstituted lactose solutions with the aid of ultrasound.

Ultrasound is a kind of mechanical wave in elastic medium. Acoustically induced cavitation is one of the primary effects of power ultrasound in a continuum fluid. Ultrasound can impose an oscillatory pressure on the fluid. At low intensity, this pressure wave induces motion and mixing within the fluid, a process known as acoustic streaming. Mixing can be increased by above process. At higher intensity, in addition to acoustic streaming and mixing effects, the local pressure in the expansion phase of the cycle may fall below the vapor pressure of the fluid, which causes any minute bubbles or cavities to grow. If the ultrasonic intensity is increased further, negative transient pressures can be produced in the fluid medium. This pressure not only enhances the bubble growth process, but also produces new cavities by the tensioning effect on the fluid [14]. Ultrasound also has mechanical effects on crystallization systems. The shock wave which is produced by the

implosion of cavitation bubbles is responsible for the effects of ultrasound on particles in suspension. At vicinity of the implosion, the pressure can reach as high as 300 MPa. According to Doktycz and Suslick [15], the shock wave can promote the mixing process as well. If the implosion of a cavitation bubble is near an interface, besides the shock wave, it can generate a liquid jet [16]. Andrew Van Hook stated further that sound wave irradiation is helpful to supplement and strengthen the wave action needed in forming critical crystal nucleus, thus, it is able to accelerate crystallization [17]. In addition, in our previous research [18], it was found that ultrasound has important effect on enhancing crystal velocity, shortening crystal time, controlling the distribution of crystalline grain diameter, improving product quality, and so on. So the application of ultrasonic crystallization of lysine muriate would have bright prospects.

The aim of this study was to extract lysine muriate crystallization with ultrasound and evaluate the influence of some critical factors on the process. The employed crystallization method was cooling crystallization enhanced with ultrasound.

42.2 Experiment

42.2.1 Materials and Instruments

Lysine muriate was crystallized from aqueous solutions. For cooling crystallization experiments, the solutions were prepared by weighing certain amounts of solid lysine muriate (purity >99.5 %), bought from Sugar Hill East King Co. Ltd. in deionized water. Ultrasonic processor (Model KQ5200DB NC) from Kunshan Ultrasonic Instrument Co. Ltd.; Electric constant temperature water bath pot from Beijing Guangming Medical Instrument Factory Wing; Electric Blast Oven from Experiment Instrument Co. Ltd. Tianjin Tianyu; Electronic Balance from Dragon Electronics Co. Ltd. Shenyang; Microscope from Shanghai Boaimei Light Instruments Manufacturing Co. Ltd.; Digital camera from Powershot 480, Nikon Instruments Sales Co. Ltd (China).

42.2.2 Ultrasonic Processor

Ultrasonic processor, Model KQ5200DB NC with a frequency of 40 kHz, was used as a ultrasonic power generator. The ultrasonic processor was equipped with a wattmeter. The ultrasonic power was changed by varying the amplitude of the ultrasound.

400 A. Hu et al.

42.2.3 Determination and Calculation of Crystal Yield

Crystal yield =
$$N \times 100 \%/M$$
 (42.1)

where N is the mass of lysine muriate crystals obtained by ultrasound, g; M is the mass of lysine muriate powder used to make the solutions of certain concentrations, g.

42.2.4 The Effect of Ultrasonic Time on Crystal Yield of Lysine Muriate

25 g lysine muriate powder was dissolved with 40 mL deionized water to prepare a solution with the concentration of 38.46 %, which was used for ultrasonic treatment and crystallization. The ultrasonic conditions were: ultrasonic power 160 W, temperature 35 °C, ultrasonic time was 0, 10, 30, 50, 70 and 90 min, respectively. After being treated by ultrasound each time, the solutions were kept for 3 h at 25 °C for crystallization. Crystals were filtered by using a Büchner funnel under vacuum and the crystals were collected, then dried for 3 h at 100 °C. The crystal yields were determined and calculated according to the method described in Sect. 42.2.3.

42.2.5 The Effect of Ultrasonic Power on Crystal Yield of Lysine Muriate

Ultrasonic time was 50 min, ultrasonic power was 64, 96, 112, 128 and 160 W, separately. The other ultrasonic conditions and operations were the same as those mentioned in Sect. 42.2.4.

42.2.6 The Effect of Crystal Time on Crystals Yield of Lysine Muriate

Ultrasonic time was 50 min, ultrasonic power was 112 W, and crystallization time was 3, 4, 5, 6 and 7 h, respectively. The other ultrasonic conditions and operations were the same as those mentioned in Sect. 42.2.4.

42.2.7 The Preparation of Supersaturation Solution

Rapid cooling method is used for the preparation of supersaturated solution. This method [19] shows that the supersaturated solution is obtained through rapid

cooling of a certain concentration of saturated solution. For instance, when the solution is confected, which saturation solubility and supersaturation ratio S is, respectively, 85 g/100 mL and 1.4 at 35 °C, according to the definition of supersaturation ratio [20]: $S = C/C_0$ (where S is supersaturation ratio; C is solubility of supersaturation solution; C_0 is solubility of saturated solution), the solubility of supersaturation solution $C = 85 \text{ g} \times 1.4/100 \text{ mL} = 119 \text{ g/}100 \text{ mL}$. So, firstly, the solution with the saturation solubility of 119 g/100 mL is needed to be confected, then rapidly cool to 35 °C and obtain the supersaturation solution needed.

According to the method mentioned above, the supersaturation ratios of the solutions used in our study, which were 1.45, 1.50, 1.55, 1.60, 1.65, 1.70, 1.75, 1.80, 1.85, 1.90, 1.95, and 2.0, can be confected, respectively.

42.2.8 The Measurement of Induction Period of Traditional Crystallization

Supersaturation solution of lysine muriate obtained by the above method (Sect. 42.2.7) was placed into 35 °C ultrasonic processor, however, without ultrasound. Then, the time was recorded and the record was kept until the crystals can be observed.

42.2.9 The Measurement of Induction Period of Ultrasonic Crystallization

Supersaturation solutions of lysine muriate obtained by the above method (Sect. 42.2.7) were placed into 35 °C ultrasonic processor. Then, we introduced ultrasound and started to record ultrasonic time until the crystals can be observed.

42.2.10 The Effect of Ultrasonic Power on Induction Period of Lysine Marinate Crystallization

A series of supersaturated solutions with supersaturation ratio of 1.45, 1.50, 1.55, 1.60, 1.65, 1.70, 1.75, 1.80, 1.85, 1.90, 1.95, and 2.0 were prepared, respectively, according to above method (Sect. 42.2.7). They were placed, respectively, into 35 °C ultrasonic cleaning tank, the introduced ultrasonic power was 0, 96, 112, and 128 W, respectively. Each induction period corresponding to supersaturation ratio was measured, respectively.

402 A. Hu et al.

42.2.11 The Effect of Ultrasonic Power on Crystal Appearance of Lysine Muriate

25 g lysine muriate powder was dissolved with 40 mL deionized water to prepare a solution with the concentration of 38.46 %, which was used for ultrasonic crystallization. Ultrasonic power was 64, 96, 112, 128 and 160 W, separately, temperature 35 °C, ultrasonic time was 50 min. After being treated by ultrasound each time, the solutions were kept for 5 h at 25 °C for crystallization. Crystals were filtered by using a Büchner funnel under vacuum and the crystals were collected, then dried for 3 h at 100 °C. The crystal yields were determined and calculated according to the method described in Sect. 42.2.3.

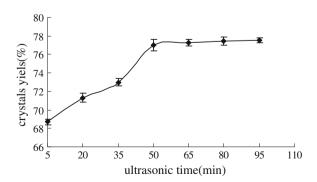
42.3 Results and Discussion

42.3.1 Effect of Ultrasonic Time on Crystal Yield

It can be seen from Fig. 42.1 that without ultrasound, namely, ultrasonic time was 0 min, crystal yield was less than that with ultrasound. Between 0 and 50 min, crystal yield increased rapidly with the extension of ultrasonic time. Between 50 and 95 min, crystal yield kept almost the same. So, the reasonable ultrasonic time of the experimental process was 50 min.

There are several possible reasons for the above phenomenon. Cavitation bubbles may be formed because of introducing ultrasound in the crystallization of lysine muriate. The burst of cavitation bubbles will produce high temperature, pressure, and a certain degree of microjet which can shatter crystal particles and generate a large number of crystals which secondary nucleations grow into. Furthermore, the high temperature generated in the process of cavitation bubbles burst, which can accelerate the diffusion of solutes and break crystal particles, is increasing with the increase of ultrasonic power. However, the stabilization of bubbles is needed for a certain time, when ultrasonic time is shorter, ultrasound fails

Fig. 42.1 Effect of ultrasonic time on crystal yield



to promote nucleation, so that crystal yield will be lower; when ultrasonic time is longer, ultrasound can promote nucleation and the second nucleation will appear rapidly after the ultrasonic induction, so that more crystal nucleus will be formed. In addition, ultrasonic crushing effect will also produce more crystal nucleus and increase the crystal yield. On the other hand, for a certain volume and supersaturation ratio of lysine muriate solution, the maximal crystal yield is certain since it depends on them. As a result, the increase of crystal yield becomes slow with the extension of ultrasonic time.

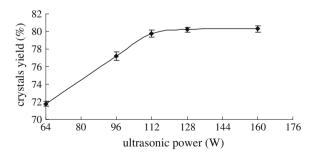
42.3.2 Effect of Ultrasonic Power on Crystal Yield

As clearly shown in Fig. 42.2 that the crystal yield was increasing with the raising of ultrasonic power, but the increasing became not obvious when ultrasonic power was over 112 W. This may be because ultrasonic cavitation leads lots of bubbles to produce. And those bubbles will crack suddenly when they grow to certain extent, which will supply especial space and energy for the form of crystal nucleus, then the crystal yield is enhanced. However, whether and how much ultrasonic cavitation takes place depends on ultrasonic power, in a certain system, ultrasonic cavitation will be restrained when ultrasonic power is increased to a certain extent and then affect crystal yield. Taking all factors into consideration, ultrasonic power 112 W was relatively reasonable in the experiment.

42.3.3 The Effect of Crystallization Time on Crystal Yield

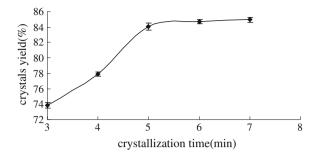
As shown in Fig. 42.3 that the crystal yield was increasing with the raising of crystallization time, but the increasing became not obvious when crystallization time was over 5 h. This phenomenon can be demonstrated as follows: solution is divided into solid–liquid two-phase once crystal nucleus is formed, crystal nucleus is enfolded with a layer of liquid film and solution in the film converts saturation from supersaturation because some of its solutes have been absorbed to the surface of crystal nucleus. However, solution out of the film is still supersaturated and its solutes

Fig. 42.2 Effect of ultrasonic power of crystal yield



A. Hu et al.

Fig. 42.3 Effect of crystallization time on crystal yield



will continually diffuse into the liquid film and also be absorbed to the surface of crystal nucleus because of the impetus of concentration difference. The solutes are arrayed layer by layer on the surface of crystal nucleus, which prompt continually the crystal growth until the concentration of solution falls to saturation [20]. So, the quality of crystal is increasing with the extension of time, as a result, crystal yield is enhanced continually until the concentration of solution falls to that solutes can no longer precipitate from the solution. Thus, in the period of 3–5 h, crystal yield increased rapidly, after 5 h it had only little increase. In the experiment, crystallization time 5 h was relatively reasonable and the crystal's yield reached 84.08 %.

42.3.4 Effect of Ultrasonic Power on Induction Period of Lysine Muriate Crystal Nucleation

It can be seen from Fig. 42.4 that under the circumstance of the same supersaturation ratio, the crystal induction period of lysine muriate with ultrasound was shorter than that without ultrasound. And the higher the ultrasonic power, the shorter the induction period. The shortening amplitude decreased with the increase of supersaturation ratio, and it can be concluded that the effect of ultrasound on crystal nucleation of lysine muriate in the lower supersaturation ratio solution was more significant than that in the higher.

42.3.5 Effect of Ultrasound on Crystal Appearance of Lysine Muriate

In the crystal growth process, the growth rates of different crystal surfaces were different. Crystal surfaces with faster growing rates eventually disappear, the final exposed surface decided the crystal shape. Introducing ultrasound can accelerate the transfer of material and heat, then strengthen crystallization [21]. The effect of ultrasonic power on crystal appearance was investigated by analyzing the pictures of lysine muriate crystals.

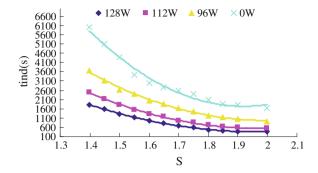


Fig. 42.4 Effect of ultrasonic power on induction period

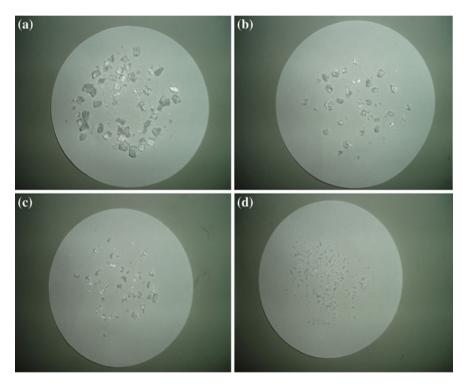


Fig. 42.5 a Ultrasonic power 0 W. **b** Ultrasonic power 64 W. **c** Ultrasonic power 96 W. **d** Ultrasonic power 160 W

From Fig. 42.5, the results were observed as follows: Without ultrasound, the crystals were irregular, larger, and the size distribution was broader, however, with adequate ultrasound, they were regular, smaller, and uniform, the particle size

406 A. Hu et al.

decreased with the increase of ultrasonic power from 64 to 160 W. When crystal particles were larger, the coalescence phenomenon was obvious, the dispersion of crystals was getting better and particles became more uniform.

42.4 Conclusions

In the study, the process of ultrasound-assisted crystallization and the effect of some factors including ultrasonic time, ultrasonic power, and crystallization time on crystals yield were investigated. The results showed that the crystal yield can reach 84.08 % with the adequate conditions of ultrasonic power 112 W, ultrasonic time 50 min, and crystallization time 5 h. Under the same supersaturation ratio, when ultrasound was introduced, the induction period was significantly shortened, the higher the ultrasonic power, the shorter the induction period. The crystals obtained by ultrasound were small and uniform, and become smaller with the increase of ultrasonic power. It can be concluded that using ultrasound-assisted crystallization cannot only improve the crystallization rate and crystal yield, but also obtain crystalline product with the characteristics of small particles and uniform distribution by controlling ultrasonic power.

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Chapter 43 Isolation and Identification of a *Bacillus*amyloliquefaciens Strain Against Grape Downy Mildew and Optimization for Its Liquid Fermentation Medium

Jiping Guo, Guang Ma, Huixia Zhu and Junfan Fu

Abstract An antagonistic bacterial strain named J12 against the grape downy mildew was isolated from 92 bacteria on grape leaves. Antagonistic strains were isolated by exploiting streak plate method. J12 was identified based on morphology observation, physiological and antibacterial characterizations, 16S rDNA sequence analysis, and the phylogenetic tree. The controlling effect of J12 on grape downy mildew was 78.6 %. The liquid fermentation medium of J12 was optimized with the response surface methodology to improve the number of live bacteria. The results showed that the strain J12 was classified as *Bacillus amyloliquefaciens*. For its liquid fermentation medium, the optimum carbon source was sucrose, the best nitrogen source was soybean meal, and the best C/N ratio was 3:1. By Plackett–Burman design and response surface methodology, the best culture medium formula is 31.18 g/L sucrose, 10.57 g/L soybean meal, 2.206 g/L K₂HPO₄, 2.5 g/L yeast, 0.5 g/L MgSO₄·7H₂O. Theoretic optimum cell number was 90.87 × 10⁸ cfu/mL, indicating that the equation fits well to the actual production.

Keywords Bacillus amyloliquefaciens • 16S rDNA • Phylogenetic tree • Response surface methodology

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J. Guo et al.

43.1 Introduction

Bacillus amyloliquefaciens belongs to the aerobic or facultative anaerobic bacteria, which can produce endospores with resistance under certain conditions. It can be used as biological antagonistic bacteria [1, 2]. According to the statistics by the United Nations Food and Agriculture Organization, because the world suffered from plant diseases, 10 % of total production was lost every year [3]. 4 % of the above plant diseases are caused by fungi. At present, the main means to control fungi is chemical pesticide [4]. But environmental pollutants caused by the chemical pesticides have been threatening the health of humans. With the development of green agriculture, the use of biological means to control plant diseases are more and more concerned [5]. For example, Bacillus subtilis B-FS01 was used to control grape downy mildew, which effect was 88.25 %. However, the agent has not been used in factory production.

The new type of microecological preparation of *B. amyloliquefaciens* was used in agricultural production in our country from 1990s. Grape downy mildew is one of the most damaging fungal diseases of grapevine (*Vitis* spp.) worldwide [6–8]. It is caused by the heterothallic diploid *Plasmopara viticola*, a biotrophic Oomycetes native of North America. In this study, a strain named J12 was separated from the grape leaves, which has a good control effect on downy mildew of grape. After identification, the bacteria were classified as *B. amyloliquefaciens*. The nutrient requirements of this strain are relatively simple, which has a strong adaptive capacity to environment. It also has a good future as biological pesticides.

43.2 Materials and Methods

43.2.1 Separation of Bacteria on Grape Leaves

Under sterile conditions, grape leaves were cut into Erlenmeyer flask containing 50 mL of sterile water and 3 drops of Tween 80. After oscillating at 150 r/min for 30 min, the water containing bacteria was diluted to 10^{-3} and 10^{-4} with sterile water. 0.1 mL dilution was put into petri dish, and then the NA medium with 45 °C was mixed. Then the bacteria in NA medium was cultured at 28 °C for 3 days. A single colony with different forms was selected for slant culture for preservation.

43.2.2 Screening of Antagonistic Bacteria

Grape leaves infected by grape downy mildew fungus and the new shoots 4–6 wheel without downy mildew fungus were collected from the Hengshui grape garden. The grape downy mildew fungus on infected leaves was brushed away. After cultured at 22 °C for 24 h, the new grape downy mildew fungus was brushed into sterile water. The leaves without fungus were used to make leaf discs of

diameter 1 cm. Leaf discs were placed in different antagonistic bacteria liquid (cell concentration = 1×10^7 /mL) for 1 min. After drying, the leaf discs were placed on wet absorbent paper with back upward. The spore suspensions containing grape downy mildew fungus were sprayed on the leaf discs evenly. In each treatment, 15 leaf discs were put in a culture dish. Each treatment was replicated three times. The disease index was calculated after the leaf discs were cultured at 22 °C for 7 days.

43.2.3 Identification of Antagonistic Bacteria

Using genomic DNA as a template, the 16S rDNA was amplified. The upstream primer was 27F and the downstream primer was 1492R. The sequence of 27F is 5'-AGAGTTTGATCCTGGCTCAG-3'. And the sequence of 1492R is 5'-GGTTA CCTTGTTACGACTT-3'. PCR reaction conditions are as follows: predegeneration at 94 °C for 5 min, degeneration at 94 °C for 1 min, annealing at 55 °C for 1 min, extension at 72 °C for 1.5 min, and after 36 cycles, extension at 72 °C for 10 min, then preservation at 4 °C. The sequenced 16S rDNA gene of strain J12 was compared and analyzed by blastn software of NCBI to find the homologous sequences and determine the level of region of homology. A phylogenetic tree was constructed by the software of ClustalW and MEGA 5.05.

43.2.4 Determination of the Optimal Carbon Source, Nitrogen Source, and C/N

To determinate the optimal carbon source, the glucose, sucrose, maltose, and soluble starch were added as the only carbon sources, respectively. Then fermentation medium was cultured for 72 h with the concentration of 20 g/L. After determining the carbon source, nitrogen source was prepared with soybean meal, corn flour, peptone, fish meal, and urea (20 g/L), respectively. The determined best carbon source and nitrogen source were mixed with different C/N ratio of by 1:3, 1:2, 1:1, 2:1, 3:1, 4:1, 5:1, and 6:1, respectively. The above mediums were used to culture the J12 strain for 72 h to determine the number of live bacteria in fermentation liquid.

43.2.5 Screening the Critical Factor by Plackett–Burman Design Method

Using nine factors and three levels of Plackett–Burman (PB) design, X1, X4, X7, and X9 were void items to estimate experimental error. X2, X3, X5, X6, and X8 were sugar, soybean meal, yeast paste, K₂HPO₄·MgSO₄·7H₂O, respectively. Each

J. Guo et al.

factor has two different level of (-1, 1), which was 30.000 and 37.500 g/L, 10.000 and 12.500 g/L, 2.500 and 3.125 g/L, 2.000 and 2.500 g/L, and 0.500 and 0.625 g/L, respectively.

43.2.6 Using the Steepest Ascent Search Experiment to Define the Optimal Concentration Range of Important Factor

In steepest ascent search experiment, the concentrations of sucrose, soybean meal, and K₂HPO₄ were changed by certain gradient and the other two factors was initial concentration. By detection of number of live bacteria in each fermentation liquid, the optimal concentration range of important factor can be determined.

43.2.7 Determine the Optimal Formula of Fermentation Culture Medium by Response Surface Method

In response surface experiments, the Box–Behnken central composite design method was selected. Three important factors were tested by three concentration levels of -1, 0, and 1, respectively. Then the data were processed by quadratic regression fitting, which can get 1° polynomial equation. In a certain range of concentration, the best culture medium formula was obtained. The equation was solved within a certain range of concentrations to get the best medium formula.

43.3 Results and Analysis

43.3.1 Separation of Bacteria on Grape Leaves

Using dilution plate method, according to colony morphology, size, color, and growth characteristics, 92 strains of bacteria were isolated from the grape leaves.

43.3.2 Screening Result of Antagonistic Bacteria

Using the leaf disc method, the 92 strains of bacteria were tested for their control effect of grape downy mildew. The results showed that there are 12 strains had the control effect of >20 %. The strain that had the best control effect is J12, which reached 78.6 %. And other strains had the control effect between 21.2–65.8 % (Table 43.1). The J12 strain was cultured 5 times for passage and its antibacterial effect was found to be stable.

Strains	J1	J2	J3	J4	J5	J6	J7	Ј8	J9	J10	J11	J12
Disease index	90.12	52.63	30.82	57.23	59.21	71.01	39.56	46.59	65.25	59.93	37.22	19.29
Control effect (%)		41.6	65.8	36.5	34.3	21.2	56.1	48.3	27.6	33.5	58.7	78.6

Table 43.1 Control effects of different strains on grape downy mildew

43.3.3 Sequencing and Phylogenetic Tree Analysis of 16S rDNA

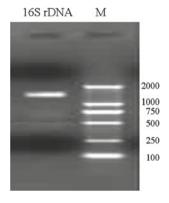
The total DNA of strain J12 was amplified with universal primers of 27F and 1492R. The PCR product length is about 1.6 KB (Fig. 43.1). After sequencing, 16S rDNA length of J12 strain was 1512 bp.

By blastn software of NCBI, the 16S rDNA sequence of strain J12 and its related species were compared. Nine strains were selected to construct the phylogenetic tree with J12. It can be seen from Fig. 43.2, strain J12 and *B. amyloliquefaciens* were clustered into one branch. They have the closest genetic relationship. The morphology, growth characteristics, physiological, and biochemical characteristics of J12 strain are also similar to *B. amyloliquefaciens*. So J12 strain was finally identified as *B. amyloliquefaciens*

43.3.4 Determination of the Optimal Carbon Source, Nitrogen Source, and C/N

To get the best carbon source, 20 g/L of glucose, sucrose, maltose, and soluble starch were used as carbon source, respectively. The result showed that when sucrose was a sole carbon source, the number of live bacteria of 50.95×10^8 cfu/mL

Fig. 43.1 The PCR amplification results of strain J12 16S rDNA. *M* DL2000 (DNA Maker), 16S rDNA: PCR product



J. Guo et al.

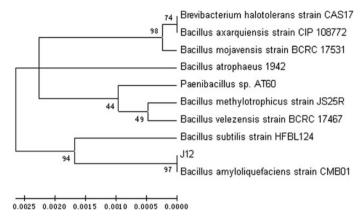


Fig. 43.2 The 16S rDNA phylogenetic tree of J12 and related strains

was highest. So the best carbon source is sucrose. 20 g/L of soybean meal, corn flour, peptone, fish meal, and urea were used to screen the best nitrogen source. The best nitrogen source is soybean meal, which number of live bacteria was 76.01×10^8 cfu/mL. Sucrose and soybean meal were mixed by different ratios of 1:3, 1:2, 1:1, 2:1, 3:1, 4:1, 5:1, 6:1, respectively to prepare different fermentation medium. The number of live bacteria was detected after fermentation for 72 h. When the C/N ratio is 3:1, the number of live bacteria was higher than other groups during which the number of live bacteria reached 77.81×10^8 cfu/mL.

43.3.5 Results of Plackett-Burman Design

Plackett–Burman design process and results were shown in Table 43.2. X2, X3, X5, X6, and X8 were sucrose, soybean meal, yeast extract, K₂HPO₄, and MgSO₄, respectively. X1, X4, X7, and X9 were four void items to estimate the test error.

By analysis of the obtained results (Table 43.3), it can be concluded that sugar, soybean meal, and K₂HPO₄ were the three most important influence factors. Effects of sucrose and K₂HPO₄ reached significant levels. Soybean meal reached extremely significant level.

43.3.6 The Results of Steepest Ascent Experiment

The steepest ascent experiment was designed with factors of sucrose, soybean meal, and K₂HPO₄, with initial level of yeast extract and MgSO₄. The ratio of sucrose and soybean meal was 3:1. The results were shown in Table 43.4. The number of live

No. Factors Number of live bacteria X1 X2 X3 X4 X5 X6 X7 X8 X9 (10^8 cfu/mL) 22.70 1 -1-1-11 -1-11 1 2 1 1 -1-1-11 -1-11 32.56 3 24.46 1 1 -1-1-11 -1-14 1 1 1 1 -1-1-11 -122.44 5 -1 1 1 1 1 -1-1-11 62.04 6 1 -11 1 1 -1-1-126.84 1 7 -11 -11 1 1 1 -1-132.56 8 1 -11 -11 1 1 1 -150.42 9 1 1 -11 -11 1 1 1 26.05 -1-11 10 1 1 -11 1 1 63.80 11 -1-11 1 -11 -1 1 92.14 1 12 1 -1-11 1 -11 -11 34.76 -135.64 13 1 -1-11 1 -11 -1-1-1-11 -114 1 -11 1 88.56 -115 -1-11 -1-11 1 -140.92 16 -1-1-1-1-1 -1-1 -134.32

Table 43.2 Design and results of Plackett-Burman design

Table 43.3 Experimental factors, level and analysis of results

Factors	Levels		t	Pr > t	Order of importance
	1	-1			
X2	37.500	30.000	-2.63	0.031	2
X3	12.500	10.000	3.59	0.008	1
X5	3.125	2.500	-1.45	0.17	4
X6	2.500	2.000	2.52	0.042	3
X8	0.625	0.500	0.91	0.525	5

Table 43.4 Design and results of steepest ascent experiment

No.	Sucrose	Soybean meal	K ₂ HPO ₄	Number of live bacteria (10 ⁸ cfu/mL)
1	27.00	8.50	1.25	81.17
2	28.00	9.00	1.50	75.21
3	29.00	9.50	1.75	83.13
4	30.00	10.00	2.00	89.45
5	31.00	10.50	2.25	90.51
6	32.00	11.00	2.50	57.14
7	33.00	11.50	2.75	72.54

J. Guo et al.

bacteria in No. 5 was highest, which was 90.51×10^8 cfu/mL. So, sucrose, soybean meal, and K_2HPO_4 content were set at 31, 10.50, and 2.25 g/L as central points of response surface experiment.

43.3.7 Optimization Results by Response Surface Methodology

Using Box–Behnken central composite experimental design, the amount of steepest ascent experiment result was optimized further (Table 43.5). Independent variables were sucrose = (D1 - 31)/3, soybean meal = (D2 - 10.5)/1, and $K_2HPO_4 = (D3 - 2.25)/0.25$. And response value was number of live bacteria. The actual value of each independent variable was shown in Table 43.5, and the experimental design and results were shown in Table 43.6.

Table 43.5 The factors and levels of response surface design

Factors	Levels				
	-1	0	1		
Sucrose	28.00	31.00	34.00		
Soybean meal	9.50	10.50	11.50		
K ₂ HPO ₄	2.00	2.25	2.50		

Table 43.6 Design and the results of response surface

No.	Factors		Number of live bacteria	
	Sucrose (g/L)	Soybean meal (g/L)	K ₂ HPO ₄ (g/L)	(10^8 cfu/mL)
1	28.0	9.5	2.25	78.44
2	28.0	11.5	2.25	77.53
3	34.0	9.5	2.25	76.44
4	34.0	11.5	2.25	82.99
5	31.0	9.5	2.0	85.45
6	31.0	9.5	2.5	85.36
7	31.0	11.5	2.0	83.08
8	31.0	11.5	2.5	82.08
9	28.0	10.5	2.0	77.53
10	34.0	10.5	2.0	79.81
11	28.0	10.5	2.5	75.08
12	34.0	10.5	2.5	75.71
13	31.0	10.5	2.25	88.09
14	31.0	10.5	2.25	88.54
15	31.0	10.5	2.25	90.82

By response surface experiment carried out as given in Table 43.6, the experience model was obtained as Y1 = 89.15 + 0.79625X2 - 0.00125X3 - 0.955X6 - 8.63X2X2 + 1.865X2X3 - 0.4125X2X6 - 1.67X3X3 - 0.2275X3X6 - 3.4875X6X6. Correlation coefficient of the regression model is R. For R = 93.30 %, fitting degree of regression equation is good. By SAS analysis, the maximum response (Y) can be obtained when the coded value of sucrose = 0.06296, soybean meal = 0.07477, and $K_2HPO_4 = -0.17448$. The corresponding best culture medium formula was sucrose 31.18 g/L, soybean meal 10.57 g/L, K_2HPO_4 2.206 g/L, yeast extract 2.5 g/L, and MgSO₄.7H₂O 0.5 g/L. And the best theoretical cell number is 90.87×10^8 cfu/mL. In order to test the feasibility of the used response surface, three parallel experiments were carried out. The number of live bacteria was 92.54×10^8 cfu/mL in the three experiments, which suggested that the fitting degree of the equation and the actual situation is very good. The model was correct.

43.4 Discussion and Conclusion

In the early classification systems, *B. amyloliquefaciens* was classified as a subspecies of *B. subtilis*. They were divided in the later classification systems. A lot of phenotypic characteristics of *B. amyloliquefaciens* and *B. subtilis* are very similar. It is difficult to distinguish them only by morphology, culture, biochemical characteristics, and physiological characteristics. So, further gene sequence comparison such as 16S rDNA is necessary. Physiological and biochemical characteristics are one of the important bases of microbial classification. Theoretically, 16S rDNA sequence analysis results can be verified by physiological and biochemical experiments. However, identification of microorganisms by physiological and biochemical characteristics is difficult, because of small differences in physiological and biochemical characteristics between the species, and some changes in the physiological and biochemical characteristics by the growth environment. So, the physiological and biochemical characteristics of identification are not suitable as verification by 16S rDNA sequence analysis results.

B. amyloliquefaciens had been researched and explored more and more widely with the ability to inhibit fungi and bacteria. It becomes a potential microbial biopesticide in recent years. Now, many papers on the *B. amyloliquefaciens* have been published about new discovery and applications, gene function, and molecular research methods of exploration, fermentation conditions, and medium formulation.

In this study, a strain against grape downy mildew was isolated. The strain belongs to *B. amyloliquefaciens*. The important factors for its liquid fermentation medium were screened by Plackett–Burman design and optimal fermentation medium formula was obtained by response surface methodology. The optimal liquid fermentation medium formula is 31.18 g/L sucrose, 10.57 g/L soybean meal, 2.206 g/L K₂HPO₄, 2.5 g/L yeast extract, and 0.5 g/L MgSO₄·7H₂O.

J. Guo et al.

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Chapter 44 The Adsorption Properties of Macroporous Resin for Fusel Oil of Luzhou—Flavor Liquor

Jinxu Sun

Abstract The D3520 resin was suited for adsorption and desorption of fusel oil by the comparison study of different types of resin. The study confirmed that macroporous resin D3520 had favorable adsorption and desorption capability by comparing different resins based on adsorption quantity, adsorption rate, and desorption rate. Temperature affected the fusel oil adsorption rate of D3520 and with more impact the temperature was high. There was little difference in the adsorption rate at 25 and 30 °C and the rate was up to 58.85 %. Among all strippant, the effect of acetone was higher than all concentration ethyl alcohol, up to 65.80 %. After treatment by D3520, there was little difference in physical, chemical, and sensory indexes of Luzhou flavor liquor and was not a bad influence on Luzhou flavor liquor. The Luzhou flavor liquor was soft and well balanced, with more rich ester aroma, and the wine quality was better by the D3520 resin treatment.

Keywords Macroporous resin • Fusel oil • D3520

44.1 Introduction

The monohydric alcohol of above of three carbons atom calls fusel oil and is brown or yellowish oily liquid in liquor of low alcohol content [1]. It is one of the main aroma substances in Chinese liquor, but more fusel oil will lead the drinker to headache and dizziness, then to be poisoned. High levels of fusel oil could damage the good qualities of Chinese liquor, like white precipitate and foreign flavor. So the level of fusel oil is one of the decisive factors in Chinese liquor quality control.

Macroporous resin is a kind of organic polymer adsorbent and has several advantages such as fast switching speed, fine heat stabilization, little pollute to extract, high mechanical strength, and convenient regeneration [2, 3]. Meanwhile,

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⁴¹⁹

420 J. Sun

macroporous resin is the optimum adsorbing material which can be selective adsorption and easy desorption [4, 5]. In this paper, the different types of resin were suited for adsorption and desorption of fusel oil and the research viewed to control fusel oil content in Luzhou flavor liquor.

44.2 Materials and Methods

44.2.1 Materials

The different types (S-8, D3520, D4006, D101, NKA-9, HPD600, AB-8) of resin were purchased from Nanda synthetic resin co. (Tianjin). Butanol, *n*-propanol, isobutanol, *sec*-Butanol, *tert*-Butanol, and isopentanol were obtained from National standard network. Gas chromatograph is from Aglient HP 6890N (Aglient, USA) with FFAP column.

44.2.2 Determination of Fusel Oil [6]

44.2.2.1 Qualitative Analysis

Every alcohol of chromatographic grade standard and internal standard (*tert*-amyl alcohol) were weighed a certain amount and mixed, then dissolved in 60 % ethanol. Solution (as fusel oil in this paper) was filtered through microporous filter. Every alcohol and internal standard was determined by gas chromatograph.

44.2.2.2 Quantitative Analysis

Chromatographic grade standard and internal standard of every alcohol (*tert*-amyl alcohol) were weighed a certain amount and mixed in 100 mL volumetric flask as constant volume in 60 % ethanol. Mixed solution of 1, 2, 4, 8 and 10 mL were placed in 10 mL volumetric flask, respectively, and constant volume in 60 % ethanol. All samples of constant volume were mixed alcohol standard.

44.2.2.3 Determination of Gas Chromatography

The samples were filtered and analyzed without derivatization. FID detector; Capillary column: HP-FFAP (30.0 m \times 1.00 μ m \times 530.00 μ m); Column temperature: initial temperature 35 °C for 9 min, 4 °C/min risen to 45 °C, 10 °C/min risen to 100 °C, retain 4 min, 22 °C/min risen to 210 °C, retain 4 min; injector

temperature: 200 °C; detector temperature: 240 °C; carrier gas (N_2): 0.085 MPa (1.08 mL/min); make-up gas: 35.0 mL/min; H₂ gas: 30 mL/min; atmosphere: 350 mL/min.

44.2.3 Treatment of Macroporous Resin

44.2.3.1 Pretreatment of Resin

Every resin (S-8, D3520, D4006, D101, NKA-9, HPD600 and AB-8) was soaked for 24 h in 95 % ethyl alcohol and washed by 95 % ethyl alcohol until distilled water added no muddy in washing liquor. Then the resin was washed by distilled water until no ethyl alcohol was added and dipped in HCl (5 % v/v) for a night. Distilled water was added to bring it to neutral pH value and was dipped in NaOH (5 % v/v) for a night. The resin was cleaned by distilled water to neutral pH value and dried at 60 °C.

44.2.3.2 Selection of Resin and Calculation of Adsorption Quantity and Adsorption Rate

For 1 g of every resin, 50 mL of 0.20 mg/mL fusel oil (mixed alcohol solution) was added and then vibrated for 24 h at 30 °C. Fusel oil was determined by gas chromatography and adsorption quantity and adsorption rate were calculated.

44.2.3.3 Desorption

All kinds of resin adsorbed fusel oil were stripped by 70 % ethyl alcohol and vibrated for 12 h at 30 °C. Fusel oil was determined by gas chromatography and desorption rate was calculated.

44.2.3.4 Temperature Effect of Adsorption Quantity

All kinds of resin were shocked and adsorbed fusel oil at 25, 30, 35, 40, 45, 50, and 55 °C, respectively. After adsorption, fusel oil solution was diluted, concentration was determined, and adsorption quantity was also calculated.

44.2.3.5 Comparison of Every Strippant Efficiency

The resin with adsorbed fusel oil was vibrated for 24 h at 30 °C and desorbed by absolute ethyl alcohol, 95 % ethyl alcohol, 75 % ethyl alcohol, 55 % ethyl alcohol,

422 J. Sun

and acetone, respectively. Fusel oil was determined by gas chromatography and desorption rate was also calculated.

$$X = \frac{C_0 - C_1}{C_0} \times 100 \tag{44.1}$$

$$L = \frac{(C_0 - C_1) \times V}{W} \tag{44.2}$$

$$J = \frac{C_2}{C_0 - C_1} \times 100 \tag{44.3}$$

 C_0 was the fusel oil concentration of initial solution (mg/mL); C_1 was the fusel oil concentration of after adsorbed solution (mg/mL); C_2 was fusel oil concentration of desorbed solution (mg/mL); V was solution volume, mL; W was the weight of resin, g; J was the desorption rate (%); L was the adsorption quantity (mg/g); and X was the adsorption rate (%).

44.3 Results and Discussion

44.3.1 Comparative Study of Different Resins on Adsorption Quantity, Adsorption Rate, Desorption Rate

Fusel oil had different adsorption quantity, adsorption rate, and the desorption rate in different resins (Figs. 44.1 and 44.2). The highest adsorption quantity was obtained in D3520 among all resins and up to 0.1662 mg/g, followed by HPD600 and D4006. Other resins had less adsorption quantity of fusel oil. Adsorption rate in different resins had the same trend with the adsorption quantity. D3520 had the

Fig. 44.1 The absorption of different resin to fusel oil

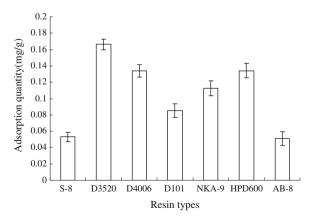
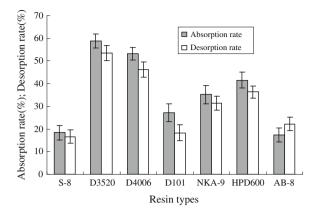


Fig. 44.2 The desorption rate and desorption rate comparison of different types of resin

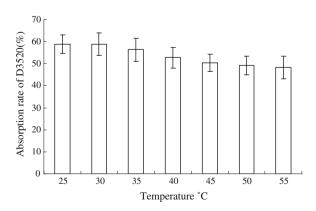


highest desorption rate up to 53.56 % than others resin. By contrast, adsorption and desorption capability of D3520 was better than others. So D3520 was selected for adsorption and desorption of fusel oil.

44.3.2 Temperature Effect of Adsorption and Desorption Capability of D3520

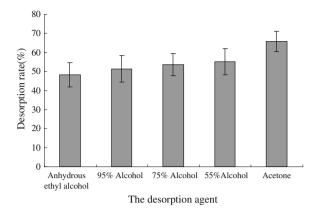
Temperature affected the fusel oil adsorption rate of D3520 and with more impact when the temperature was high (Fig. 44.3). There was little difference in the adsorption rate between 25 and 30 °C and the rate reduced gradually as the temperature increased. All of these had the same trend with others researches. The reason for this trend was that a volatile liquid such as fusel oil was volatile at high temperature; the higher the temperature is, the more volatilization quantity it has. So temperature affected the fusel oil adsorption rate of D3520 (Fig. 44.4).

Fig. 44.3 The temperature effect to desorption rate of fusel oil in D3520



424 J. Sun

Fig. 44.4 The desorption agent effect to desorption rate of fusel oil in D3520



44.3.3 Desorption Efficiency of Different Strippant

Among all strippants, the effect of acetone was higher than all concentration of ethyl alcohol, up to 65.80 %. The result had the same trend with others researches. If ethyl alcohol was selected as a strippant, more desorption efficiency was obtained with less concentration of alcohol.

44.3.4 Fusel Oil Adsorption Quantity Study of D3520 for Lu Zhou Flavor Liquor

Fusel oil quantity of Luzhou flavor liquor was reduced by 45.81 % after treated by D3520 (Table 44.1). The data showed that D3520 resin could effectively reduce fusel oil quantity in Luzhou flavor liquor. In all ingredient of fusel oil, butanol, 1-propanol, isobutanol, and isopentanol were depressed generally when treated by D3520 while 2-butanol had no variation.

44.3.5 Physical, Chemical, and Sensory Index Effects of D3520 for Luzhou Flavor Liquor

There was little difference in the physical, chemical and sensory indexes of Luzhou flavor liquor before and after the treatment by D3520 (Table 44.2) and was not a bad influence on Luzhou flavor liquor. Even fusel oil quantity of Luzhou flavor liquor decreased dramatically, the total acid and ethyl alcohol decreased slightly and total ester was increased slightly. The probability of collisions between

	Butanol	1-Propanol	Isobutanol	2-Butanol	Isopentanol	Total amount of fusel oil
Before treatment (g/L)	0.005	0.136	0.225	0.002	0.265	0.633
After treatment (g/L)	0.003	0.105	0.111	0.002	0.122	0.343

Table 44.1 The content changes comparison of fusel oil for Luzhou flavor liquor

Table 44.2 The mainly liquor sensory index changes for Luzhou flavor liquor before and after treatment

	Physical and	chemical inde	Sensory index		
	Total acid (g/100 mL)	Total ester (g/100 mL)	Fusel oil (g/100 mL)	Ethyl alcohol (v/v)	
Before treatment	0.0289	0.168	0.0633	42	Liquor body clear and transparent; pure and aromatic; stronger pit-flavor; penetrating odor; lingering bitterness; not clear ending wine
After treatment	0.0252	0.179	0.0343	41.8	Liquor body clear and transparent; pure and aromatic; stronger pit—flavor; mealiness; dry and refreshing aftertaste; no lingering bitterness

molecules for wine was increased with the wine through D3520 resin and the esterification reaction was accelerated and the penetrating odor was reduced, this actuated wine softer and improved wine quality.

44.4 Conclusion

The study confirmed that macroporous resin D3520 had favorable adsorption and desorption capability by comparing with other resin. Temperature affected the fusel oil adsorption rate of D3520 and with more impact the temperature was high. There was little difference in the adsorption rate 25 and 30 °C and the rate was up to 58.85 %. The effect of acetone was higher than ethyl alcohol, and the desorption rate arrived up to 65.80 %. There was little difference before and after treatment for physical, chemical, and sensory indexes of Luzhou flavor liquor and was not a bad

426 J. Sun

influence on Luzhou flavor liquor. The Luzhou flavor liquor was soft and well balanced, and had more rich ester aroma. The quality of the wine was better after the treatment of D3520 resin.

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Chapter 45 Low Labeling ¹³C Metabolic Flux Analysis of Saccharomyces cerevisiae Using Gas Chromatography–Combustion–Isotope Ratio Mass Spectrometry

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Abstract The applicability of gas chromatography–combustion–isotope ratio mass spectrometry (GC–C–IRMS) for the quantification of ¹³C enrichment of proteinogenic amino acids in metabolic tracer experiments was evaluated in this paper. We measured the ¹³C enrichment of proteinogenic amino acids in hydrolyzates using GC–C–IRMS from a series of parallel batch cultivations of *Saccharomyces cerevisiae*, which was cultured by mixtures of natural glucose and [1–¹³C] glucose, containing 0, 0.5, 1, and 2 % [1–¹³C] glucose, respectively. By decreasing the [1–¹³C] glucose content, kinetic isotope effects played an increasing role but could be corrected. The ¹³C metabolic algorithm and matrix algorithms were optimized in this study. The central metabolism of vivo fluxes were determined by the calculation method optimization. The obtained flux distribution was similar to published results, which obtained from GC–MS method using conventional high labeling (99 %). The GC–C–IRMS-based method involves low labeling (0.5 %) degree of expensive tracer substrate, and suits well for larger laboratory and industrial pilot-scale fermentations.

Keywords *Saccharomyces cerevisiae* • Metabolic flux analysis • ¹³C-labeling • GC-C-IRMS • Proteinogenic amino acid

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45.1 Introduction

The quantitative analysis of in vivo carbon fluxes in a metabolic network is regarded a powerful tool for the optimization of industrial organisms [2, 19]. Developed from the previous methods using only metabolite balancing, flux estimation has been far advanced by supplementing the previous stoichiometry-based approach with isotopomer balances to overcome shortcomings of the previous method in many realcase situations and the ambiguity of black box model [1, 11, 17, 18, 21, 26]. Theoretically, the most complete information would be obtained by the analysis of all extracellular and intracellular metabolite isotopomers. Unfortunately, it is highly challenging to analyze all positional isotopomers in practice, e.g., using nuclear magnetic resonance spectroscopy (NMR) techniques, and also the information content finally depends on measurement sensitivity. For ¹³C-based metabolic flux analysis (13CMFA), the mass isotopomers of intracellular and extracellular metabolites and biomass constituents can be used. The analysis of intracellular metabolites is often challenging, e.g., due to low concentrations close to or below detection limits as well as often limited stability, and short time constants of certain compounds such as glycolytic metabolites. Amino acids contained in cellular proteins reflect the labeling patterns of various central key metabolites at steady state. They can be easily analyzed by NMR [13, 14, 19], gas chromatography-mass spectrometry (GC-MS) [3, 4, 6, 7, 9, 22], or matrix-assisted laser desorption time-offlight mass spectrometry (MALDI-TOFMS) [25]. Methods have been developed to take into account natural isotope abundances of the analytes [24, 27]. In recent years, the combination of GC-MS and parameter estimation methods using network simulation techniques has been well developed and most frequently applied in metabolic flux analysis [1, 7, 16, 18, 22, 28, 29].

Another challenge in ¹³C-based metabolic flux analysis is that experiments are limited to small scales. So far labeling experiments have been restricted to laboratory-scale fermentation, shake flasks, and microtiter plates [20, 23] due to the requirement for high degree of ¹³C-labeled substrate (generally 100 % tracer substrate) and the resulting high cost. Experiments on a laboratory scale, however, cannot usually represent conditions on much larger scales that are industrially much more relevant. In this regard, a method of ¹³C MFA applicable to large-scale biotechnological processes is highly desired. Here to applying a small amount of isotopic tracer substrates for labeling experiments is a possible solution, yet it leads to extremely low ¹³C enrichments of labeling in proteinogenic amino acids and, consequently, a poor flux resolution. Since ¹³C isotopic enrichments in amino acids present in proteins is usually low (0.001-0.05, APE) as determined in protein turnover experiments [8], the labeling patterns of metabolites cannot reliably be measured by GC-MS, which has a measurable range only higher than 0.5 APE [15]. In contrast, GC–C–IRMS is capable of measuring isotopic composition at low enrichment and even natural abundance level with high precession (0.0002 APE) and accuracy. Thus, GC-C-IRMS has a great potential for the application to cases with extremely low ¹³C enrichments in metabolites. In 2008 and 2010, Elmar Heinzle et al. [10, 30] had established a method to analyze microbial metabolic network using GC-C-IRMS.

In this paper, we implemented ¹³C labeling experiments using different contents of ¹³C glucose mixed with naturally labeled glucose. Cultivations were carried out using *Saccharomyces cerevisia* which is the most important microorganism in industry. The measured values of GC–C–IRMS in low labeling ¹³C glucose were dealt with Matlab and quantum-behaved particle swarm optimization (QPSO). Based on the protocol, the ¹³C fractional enrichment data were corrected from GC–C–IRMS measurements, and applied to ¹³CMFA to verify the potential of GC–C–IRMS for an industrial scale ¹³CMFA. Results were compared with those from conventional ¹³CMFA using 99 % [1–¹³C] glucose and GC–MS as well as with the literature data.

45.2 Materials and Methods

45.2.1 Chemicals, Microorganism, and Medium

Analytical amino acid and organic acid standards were obtained from Sigma–Aldrich. [1^{-13} C] glucose (99 % atom 13 C) was purchased from Omicion Biochemicals. *N*,*N*-Dimethylformamide (DMF, chromatographically pure), *N*-(butyldimethylsilane)-*N*-methyl fluoroacetamide (MTBSTFA, \geq 95 %, Sigma–Aldrich), GC–C–IRMS (Thermo Fisher Scientific), ultraviolet spectrophotometer (UV-1240), biochemical incubator (SPX-250B-Z), refrigerated centrifuge (Sigma), constant temperature shaker, and pressure blowing concentrator (MTN-2800D).

The S. cerevisiae (SCFF-Y-002) was obtained from our lab and the cells were stored at -80 °C with 25 % (v/v) glycerin.

Preculture was grown in complex medium containing 20 g peptone, 10 g yeast extract, and 20 g glucose per liter. The fermentation medium contained 20 g glucose, 0.5 g (NH₄)₂HPO₄, 1.0 g (NH₄)₂SO₄, 0.05 g MgSO₄·7H₂O, 0.5 g KCl, 0.03 g CaCl₂·2H₂O, 3 mg FeCl₃·6H₂O, 2.1 mg MnSO₄·H₂O, 1.8 mg ZnSO₄·7H₂O, 0.5 mg CuSO₄·5H₂O, 60.3 mg myoinositol, 30 mg Ca-panthotenate, 6 mg thiamin·HCl, 1.5 mg pyridoxine·HCl, 0.03 mg biotin, and 50 mmol phosphate buffer per liter.

45.2.2 Cultivation Conditions

At first, a single colony of the yeast was inoculated into 50 mL of precultivation medium in a 250 mL baffled shake flask, and incubated on a rotary shaker at 200 rpm and 30 °C for 20 h. Second, cells were transferred into fermentation medium with a starting cell concentration (OD $_{660}$) of 0.05 and culture for 20 h. Subsequently, cells were washed twice with fermentation medium by centrifugation at 8,500 rpm for 10 min. Then the cells were used as strain. Tracer experiments

were carried out in 50 mL fermentation medium of 250 mL baffled shake flasks. The fermentation medium contained natural glucose and 0, 0.5, 1, and 2 % [1^{-13} C] glucose, respectively.

45.2.3 Determination of Physiological Parameters

Cell growth was monitored at OD_{660} during cultivation. Glucose and organic acids were analyzed by HPLC while ethanol and glycerol were determined by GC with a wax column.

45.2.4 Analysis of the ¹³C-Labeling Patterns of Amino Acids

The isotopic labeling patterns of proteinogenic amino acids were measured by GC–C–IRMS instrument comprising an gas chromatograph, a standard GC/C III interface with a Ni/CuO/Pt combustion reactor set at 940 °C, and a MAT 253 gas isotope mass spectrometer. Biomass (100 mg) was harvested at mid-exponential period (20 h). The cells were washed twice with deionized water, mixed with 4 mL of 6 M HCl, and incubated for 24 h at 110 °C for hydrolysis. The hydrolysis was filtered through a 0.22 μm membrane and then dried with a steady stream of nitrogen at the temperature of 45 °C. Those proteinogenic amino acids were derived by the addition of 100 μL of MTBSTFA and 100 μL of DMF and then directly used for analysis.

The GC was equipped with a split/splitless injector and a HP-5 column, helium was used as a carrier gas with a constant flow of 1.0 mL/min. The injector was set to 250 °C. For analysis, 1 μL of sample was injected in split mode (split 1:10). The GC temperature parameters were set as follows: The initial temperature of 120 °C was kept for 2 min. Subsequently, the temperature was increased by 3 °C/min to 250 °C and then by 10 °C/min to 270 °C and kept for 10 min. The GC–C–IRMS was equipped with a combustion reactor set at 1,000 °C and three Faraday cup collectors for *m*/*z* 44, 45, and 46 were used for detection of CO₂. Isotope ratios were calibrated against reference CO₂ of known isotopic composition introduced directly into the ion source three times at the beginning and at the end of every run. Each measurement was performed at least in triplicate.

45.2.5 Metabolic Flux Estimation

Different from GC-MS method, during GC-C-IRMS measurement all derivatized amino acids were combusted into CO₂ which had only one carbon atom. For this case, the problem became simply the first level of the elementary metabolite units,

which also corresponded to the atomic level problem. Therefore, atom mapping matrices (AMM) were used for the analysis of modeling isotope distributions.

Metabolic network simulations of yeast were performed using the MATLAB. An improved QPSO was used for the network parametrization and numerical flux estimation in this article. In short, fluxes were estimated by minimizing the variance weighted sum of squared residuals between the simulated and the experimental results of each proteinogenic amino acids.

45.3 Results and Discussion

45.3.1 ¹³C Labeling Analysis of Amino Acids by GC-C-IRMS

GC–C–IRMS chromatograms of all proteinogenic amino acids were shown in Fig. 45.1. The 13 C fraction of 15 proteinogenic amino acids could be determined, the order of the peaks were alanine (Ala), glycine (Gly), valine (Val), leucine (Leu), isoleucine (Ile), proline (Pro), serine (Ser), threonine (Thr), phenylalanine (Phe), aspartic acid (Asp), glutamic acid (Glu), lysine (Lys), arginine (Arg), histidine (His), and tyrosine (Tyr). All proteinogenic amino acids present in chromatogram were found satisfactorily separated without any overlapping peaks. The δ^{13} C value for each proteinogenic amino acids were shown in Table 45.1.

45.3.2 Linearity of ¹³C Fractional Labeling

According to the theory, the fraction of carbon atom should always be proportional to the input labeling. A series of experiments were designed to prove this theory

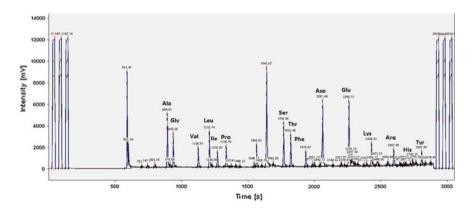


Fig. 45.1 GC-C-IRMS chromatograms of all proteinogenic amino acids for yeast

Table 45.1 δ^{13} C value for each proteinogenic amino acids

Amino acids	δ^{13} C (‰)
Ala	-0.584
Gly	-10.057
Val	22.102
Leu	27.271
Ile	30.014
Pro	25.447
Ser	-6.239
Thr	2.309
Phe	6.679
Asp	0.659
Glu	11.955
Arg	20.704
Lys	-6.266
His	-8.709
Tyr	-2.089

employing a wide range of substrate labeling. The lower fraction of 13 C-substrates containing 0, 0.5, 1, and 2 % [1^{-13} C] glucose, samples from the lower fraction group were measured by GC–C–IRMS due to their low 13 C enrichment in amino acids (<0.5 APE in derivatized analytes). The obtained δ values of individual amino acids were then converted to 13 C enrichment and corrected for the kinetic isotope effect as described previously. And then obtained mass isotopomer distributions of individual amino acids were converted into the fractional 13 C enrichment or fractional labeling (FL) using the following equation:

$$FL = \frac{\sum_{i=0}^{x} i |y_{m+i}|}{X}$$
 (45.1)

Here, y is the carbon mass isotopomer fraction of an amino acid, i is the mass shift by 13 C incorporation, and x is the number of carbon atoms of the amino acid. All experimental results were listed in Table 45.2. Except glycine, the squared correlation coefficients (R^2) from the listed amino acids were close to 1 (>0.9999) at the range from 0 to 2 % of $[1^{-13}\text{C}]$ glucose fraction obtained from GC–C–IRMS data. For the whole range from 0 to 2 %, R^2 resulting from the amino acids, again except glycine, were also quite close to 1 (>0.9999). The high linear correlation over the whole range testing allows us to predict ^{13}C enrichment of unknown labeling (e.g., 100 % $[1^{-13}\text{C}]$ glucose fraction) using extrapolation.

Designation	δ ¹³ C (‰)	δ^{13} C (‰)						
	0 %	0.50 %	1 %	2 %				
Ala	-27.338	-13.143	-0.584	26.163	0.9997			
Gly	-29.773	-19.981	-10.057	11.872	0.9991			
Val	-25.503	-4.805	22.102	59.418	0.9948			
Leu	-25.482	1.417	27.271	84.472	0.9995			
Ile	-25.049	2.832	30.014	93.591	0.9984			
Pro	-25.705	0.549	25.447	84.632	0.9984			
Ser	-28.889	-17.298	-6.239	14.661	0.9994			
Thr	-28.922	-13.007	2.309	36.4025	0.9994			
Phe	-25.890	-9.156	6.679	39.848	0.9990			
Asp	-28.696	-13.437	0.659	32.2035	0.9995			
Glu	-27.218	-7.359	11.955	46.7465	0.9988			
Arg	-25.785	-1.205	20.704	70.2065	0.9531			
Lys	-28.657	-14.928	-6.266	5.075	0.9995			
His	-25.909	-18.185	-8.709	12.901	0.9949			
Tyr	-26.526	-13.846	-2.089	22.285	0.9999			

Table 45.2 13 C fractional labeling of proteinogenic amino acids from 0 to 2 % input [1 $^{-13}$ C] labeled glucose

45.3.3 Biomass Yield and Extracellular Metabolites

The yield coefficients for the formation of biomass and extracellular products on glucose during the exponential phase are listed in Table 45.3 and the precursor demands used for biomass formation in yeast are listed in Table 45.4.

Table 45.3 Anabolic demand of yeast during aerobic glucose-limited culture^a

Compound	Demand μmol/(g dry cell mass)
	<u> </u>
Glucose 6-phosphate	2,089
Erythrose 4-phosphate	243
Ribose 5-phosphate	115
Glyceraldehyde	77
3-phosphate	
Phosphoglycerate	623
Oxaloacetate	683
Phosphoenolpyruvate	457
Pyruvate	1,285
Acetyl-CoA	2,324

^a The values were obtained from the literature [28]

Biomass and metabolites	Yield (g/mmol (GLC), mmol/mmol (GLC))
Biomass	0.115
Ethanol	0.499
Acetic acid	0.0047
Lactate	0.0029
Glycerin	0.086

Table 45.4 Biomass and metabolites yield coefficients of yeast

Biomass and metabolites yields of *Saccharomyces cerevisiae* SCFF-Y-002 cultured on glucose. All yields are given in (C-mol product)/(C-mol glucose), except the yield for biomass that is given in (g dry cellmass)/(mmol glucose)

45.3.4 Central Metabolic Network

The central metabolic network, like glycolysis, tricarboxylic acid cycle (TCA), and pentose phosphate pathway (PPP) of yeast are analyzed. The main reactions in the central metabolic network are shown as follows:

V1: G6P = F6P	V2: F6P + ATP = GA3P + DAHP + ADP
V3: DAHP = GA3P	V4: GA3P + NAD + ADP = 3-
	PG + NADPH + ATP
V5: 3-PG + ADP = PYR + ATP	V6: PYR + H-COA + NAD = AC-
	COA + NADH + CO ₂
V7: $G6P + NADP = RIB-5P + CO_2 + NADPH$	V8: 2RIB-5P = S7P + GA3P
V9: $S7P + GA3P = E-4P + F6P$	V10: RIB-5P + E-4P = F6P + GA3P
V11: AC-CoA + OAA = ICI + H-CoA	V12: ICI + NADP = 2 -OXO + CO_2 + NADPH
V14: SUCC + FAD + NAD = OAA	V15:
+ FADH + NADH	$PYR + ATP + PEP + 2CO_2 = 2OAA + ADP$
V13: 2-OXO + NAD + H-	
$COA + ADP = SUCC + NADH + CO_2 + ATP$	

Back reactions are represented as Vir.

The atom transitions and transitions of reactants and products are shown in Tables 45.5 and 45.6, respectively.

45.3.5 Flux Estimation in Central Metabolic Pathways of Yeast

The central metabolic fluxes of *S. cerevisiae* are estimated with the measurement information, and the results are shown in Table 45.7.

The aim of this study was to develop a new straight forward and cost-efficient approach for the quantification of the in vivo flux distribution in the central

Table 45.5 Atom transitions between reactants and products

Reactants	Products	Carbon atom transitions
G6P	F6P	abcdef
F6P	G6P	abcdef
F6P	GAP	def
F6P	DHP	abc
DHP	GAP	cba
GAP	3PG	abc
3PG	PYR	abc
PYR	ACA	bc
G6P	P5P	bcdef
P5P	GAP	cde
P5P	S7P	00abcde
P5P	S7P	ab00000
S7P	P5P	cdefg
GAP	P5P	00abc
S7P	P5P	ab000
GAP	F6P	000abc
S7P	F6P	abc000
S7P	E4P	defg
F6P	S7P	abc0000
E4P	S7P	000abcd
P5P	F6P	ab0000
E4P	F6P	00abcd
P5P	GAP	cde
F6P	P5P	ab0000
GAP	P5P	00abc
F6P	E4P	cdef
OAA	ICT	dcb00a
ACA	ICT	000ba0
ICT	aKG	abcde
aKG	SUC	bcde
SUC	OAA	abcd
SUC	OAA	deba
PYR	OAA	abc0
CO ₂	OAA	000a
OAA	PYR	abc
GLC	G6P	abcdef
DHP	GLR	abc
3PG	GLY	ab
PYR	LAC	abc
F6P	GAP	def

Table 45.6 Atom transitions between intermediate metabolites and amino acids

Reactants	Products	Carbon atom transitions
PYR	ALA	abc
PYR	VAL	ab00c
PYR	VAL	00bc0
3PG	SER	abc
OAA	THR	abcd
3PG	GLY	ab
OAA	ASP	abcd
OAA	ILE	b0cd0
PYR	ILE	0b00c
PYR	aKI	00bc0
PYR	aKI	ab00c
aKI	LEU	Obcde
aKG	GLU	abcde
ACA	LEU	b0000
PYR	S5P	b0000ca
E4P	S5P	0abcd00
PYR	PHE	000000cba
S5P	PHE	abcdef000
PYR	TYR	000000cba
S5P	TYR	abcdef000
LYS	LYS	abcdef
aKG	ARG	abcde0
CO ₂	ARG	00000a
P5P	HIS	Obacde
CO ₂	HIS	a00000
		-

Table 45.7 Metabolic fluxes of *Saccharomyces cerevisiae*

Reactions	Fluxs (%)
V1	80.5
V2	85.5
V3	78
V4	159.5
V5	151.5
V6	73.3
V7	17.5
V8	4.9
V9	3
V10	2
V11	18.6
V12	18.6
V13	7.4
V14	7.5
V15	25.5

metabolism of microorganisms. Based on the model-based experimental design [12], [1-13C] glucose was selected as the tracer substrate because of its commercial availability and low cost. Due to the ability for the detection of extremely low isotopic enrichment and high precision, GC-C-IRMS measurement of isotope enrichment was chosen for this study. To our knowledge, GC-C-IRMS had not been applied on isotope-based in vivo flux analysis yet. Our study opens up the door for the application of GC-C-IRMS to in vivo ¹³CMFA even at industrially relevant fermentation scales in situ. This research increased in importance with increasing competition, e.g., in the production of biopharmaceuticals [5].

The theoretically expected linear relationship between input ¹³C labeling content and the ¹³C enrichment of metabolic products were also observed experimentally, which suggested the GC–C–IRMS applicability for ¹³CMFA. As verified, for the proteinogenic amino acids investigated here in, other metabolic products could also be applied to ¹³CMFA using this novel approach, e.g., some intracellular metabolites. With this method, it is possible to calculate ¹³C enrichment at any input ¹³C substrate content by extrapolation with at least two or more different amounts of ¹³C substrate in mixtures with unlabeled ones.

Typically, commercially available 99 % [1-13C] glucose applied in ¹³C labeling experiment contained 1 % ¹²C in the first carbon, and natural abundance of isotopes of carbon and other elements at other positions. Additionally, elements with natural abundance of isotopes were introduced during derivatization, for example, silicon and nitrogen. In the case of GC-MS-based estimation, the effect of all these naturally labeled atoms could be eliminated by matrix correction [27]. The so-called correction matrix represents the abundance of naturally occurring isotopes other than the carbon atoms in the skeleton of a metabolite. Therefore, liability of carbon mass isotopomer distributions computed from measurements could be subject to the reliability of the data given for the natural abundance of isotopes.

In contrast to, all the separated derivatized analytes were combusted to CO₂ in GC-C-IRMS measurement and only CO₂ was analyzed afterwards. Therefore, only carbon and oxygen need to be considered for the correction. In this regard, the corrected GC-C-IRMS data should have a higher accuracy than the corrected GC-MS data. The new approach with the use of low degree ¹³C substrate and GC-C-IRMS measurement allows in vivo flux estimation in central metabolism, and the final estimated fluxes showed a slightly lower precision only at very low labeling input of 0.5% [1- 13 C] glucose compared with those from high degree labeling and GC-MS measurement of literature date [31, 32]. The obtained flux distribution was very similar (Fig. 45.2). Contrary to the GC-MS method, many reactions in the central metabolism which were regarded as reversible, the transaldolase and transketolase reactions in the PPP; phosphoenolpyruvate carboxylase (oxaloacetate to phosphoenolpyruvate), malic enzyme (malate to phosphoenolpyruvate), and oxaloacetate decarboxylase (oxaloacetate to pyruvate) could not be determined by the new approach with $[1-^{13}C]$ glucose as only tracer substrate. However, the net fluxes were determined with high precision and accuracy using GC-C-IRMS data in this study. The above limitation could be overcome. In terms of information content, GC-MS measurement was superior to GC-C-IRMS, since it allows the determination of mass

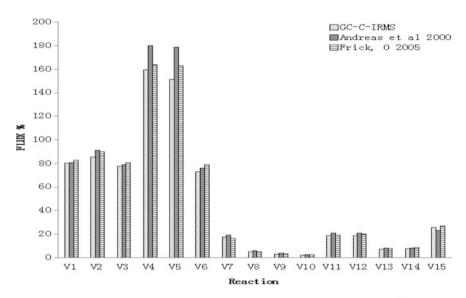


Fig. 45.2 Comparison of estimated fluxes. GC/MS-flux analysis using 99 % $[1^{-13}C]$ glucose; literature data [31, 32]; GC-C-IRMS from single labeling experiment using 0.5 % $[1^{-13}C]$ glucose corrected for natural isotope abundance in culture with naturally labeled glucose

isotopomer distribution for each analyte, whereas GC-C-IRMS determined only average carbon labeling. The present approach had a great potential for ¹³CMFA at low labeling degree of substrates and, especially, was the most promising one for investigating in vivo fluxes in a large-scale bioprocess in situ.

45.4 Conclusion

Currently, ¹³C metabolic flux analysis is the most reliable technique for measuring in vivo metabolic fluxes. The high cost of labeling is the main bottleneck of isotope tracer. This research introduced a simple and economical method to estimate metabolic fluxes of yeast using GC–C–IRMS with low labeling degree of 0.5 % (w/w) [1–¹³C] glucose mixed with 99.5 % (w/w) natural glucose. The central metabolism flux of *S. cerevisiae* was determined by the calculation method optimization. The obtained flux distribution was similar to published results, which obtained from GC–MS method using conventional high labeling (99 %). The GC–C–IRMS-based method reduces the experiment cost greatly, and well suits for larger laboratory and industrial pilot-scale fermentations.

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Chapter 46

A Comparative Study on the Antioxidant Activity of Two Polysaccharides from *Ganoderma lucidum*

Ruyu Tao, Limin Hao, Shiru Jia, Xin Zheng, Jianyong Yu and Qingwu Jiang

Abstract Two polysaccharides from *Ganoderma lucidum* (GLP1 from fermentation broth and GLP2 from fruiting body) were obtained and investigated for their in vitro antioxidant properties. The two polysaccharides showed strong antioxidant activities in a dose-dependent manner. At 5.0 mg/mL, the reducing power and the scavenging rate on DPPH radical and hydroxyl radical of GLP1 were 1.25, 50 and 80 %, while that of GLP2 were 1.46, 85, and 78 %, respectively. GLP1 and GLP2 also showed strong protective effects on yeast cells from UV and H₂O₂ damage. At 20 mg/mL, the protective effects on yeast cells from UV and H₂O₂ of GLP1 were 57 and 49 %, and that of GLP2 were 84 and 79 %, respectively. Thus, both GLP1 and GLP2 could be considered as antioxidants for food and pharmaceutical industries.

Keywords Ganoderma lucidum • Polysaccharides • Antioxidant activity • Protective effect

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46.1 Introduction

Oxidation, which provides energy to biological process, is an essential biological process to many organisms [1]. However, overproduced oxygen free radicals under pathological conditions could generate damage to cells, such as interference and manipulation of protein, tissue loosening, genetic damage, promotion of disease, and aging [2–5]. However, recent researches suggested that using antioxidants is an effective method to reduce oxidation damage. Thus, it is essential to develop and utilize more natural antioxidants to protect human body from free radicals damage [6–8].

Ganoderma lucidum is a mushroom being widely used in China and other Asian countries [9]. It is regarded as a tonic for promoting longevity and health, which has been used to treat a variety of diseases, such as atherosis, hepatopathy, chronic hepatitis, neurasthenia, insomnia, bronchitis, nephritis, hypertension, asthma, and gastric ulcers [10–12]. Modern medical studies have shown that these functions are associated with polysaccharides, which are mainly active components of *G. lucidum*. These polysaccharides are capable of antioxidation, immunity improvement, and antitumor, etc. [13].

Ganoderma lucidum polysaccharides (GLP) are mainly extracted from fruit bodies and spores of *G. lucidum*. However, the incubation time is more than 60 days, and the yield of polysaccharides is below 100 mg/g [14]. The submerged fermentation of *G. lucidum* is another way to gain a large amount of extracellular polysaccharides [15, 16]. Compared with the artificial cultivation of *G. lucidum*, the submerged fermentation of mycelium has a short cycle, high yield, and stable product quality. Therefore, more attention has been paid to the submerged fermentation of *G. lucidum* in recent years. Although the nutritional value and pharmacologic action of the products from the fermentation broth are not lower than that of fruiting body, the contrasts of GLP from fermentation broth and fruiting body have been little studied and need further research.

In this paper, one fermentation broth polysaccharide (GLP1) and one fruiting body polysaccharide (GLP2) were extracted from *G. Lucidum*. Their antioxidant activities (DPPH, hydroxyl radical scavenging assay and reducing power) and protective effects on yeast cells with physical and chemical oxidative damage were investigated.

46.2 Materials and Methods

46.2.1 Materials

Both the strain and fruiting body of *G. lucidum* were provided by the Quartermaster Equipment Institute of General Logistics Department of People's Liberation Army. Trifluoroacetic acid solution (TFA), hydrogen peroxide (H₂O₂), ascorbic acid,

ferrous sulfate, trichloroacetic acid, salicylic acid, ferric chloride, and potassium ferricyanide were purchased from Tianjin chemical factory (Tianjin, China) and of analytical grade. 1-Diphenyl-2-picrylhydrazyl (DPPH) was purchased from Sigma-Aldrich (St. Louis, MO, USA). All other chemicals used were of analytical grade.

46.2.2 Preparation of Crude Polysaccharides

The strain of *G. lucidum* was grown in a 500 mL flask containing 150 mL medium at 28 °C for 5 days with shaking at 180 rpm. After cultivation, the cultured media were filtered to remove the mycelia and the aqueous fraction obtained from filtration was concentrated with a rotary evaporator (RE3000) under reduced pressure at 50 °C and precipitation at 4 °C for 12 h with fourfold volumes of ethanol. The crude polysaccharide GLP1 was collected by centrifugation and then freeze-dried.

The fruiting body of *G. lucidum* was dried in a drying oven (DGG) at 60 °C and ground to powder. The impurities in crushed powder were removed with 80 % ethanol for 24 h at room temperature. The crude polysaccharide GLP2 was extracted from the crushed powder by hot water at 90 °C in a 1:25 (w/v) ratio for 4 h. Then, the extract was filtered and centrifuged at 10,000 rpm for 20 min at room temperature. The supernatant was concentrated, precipitated, and freeze-dried according to the method described above. So the crude polysaccharide GLP2 was obtained. The total sugar content of GLP1 and GLP2 was determined by the phenol–sulfuric acid method [17]. And the polysaccharide yield of GLP1 and GLP2 were 95.6 and 94.5 %, respectively.

46.2.3 In Vitro Antioxidant Activity of GLP1 and GLP2

46.2.3.1 DPPH Radical Scavenging Activity

The DPPH radical scavenging activities of GLP1 and GLP2 were tested according to the method of Shimada et al. [15] with some modifications. Briefly, 3 mL DPPH solution (0.4 mmol/L DPPH in ethanol) was mixed with 1 mL polysaccharide solution or ascorbic acid solution (1.0, 2.0, 3.0, 4.0, and 5.0 mg/mL) and reacted at room temperature for 30 min in dark. Then, the absorbance was measured at 517 nm and ascorbic acid was included as the positive control. The percent DPPH radical scavenging activity was calculated using the following equation:

$$\text{DPPH radical scavenging activity } (\%) = \left\lceil \frac{A_0 - (A_1 - A_2)}{A_0} \right\rceil \ \times \ 100$$

where A_0 was the absorbance of the control (without sample) and A_1 was the absorbance of the sample, A_2 was the absorbance without DPPH.

R. Tao et al.

46.2.3.2 Hydroxyl Radical Scavenging Activity

Hydroxyl radical scavenging activity was carried out according to a literature method [16] with some modifications. Hydroxyl radical was generated from FeSO₄ and $\rm H_2O_2$ and detected by their ability to hydroxylate salicylate. The polysaccharides solution at the concentrations of 1.0, 2.0, 3.0, 4.0, and 5.0 mg/mL (1 mL) was mixed with 5 mmol/L FeSO₄ (1 mL), 5 mmol/L salicylic acid–ethanol solution (1 mL), 5 mmol/L $\rm H_2O_2$ (1 mL), and incubated in water bath at 37 °C for 30 min. The absorbance was measured at 510 nm and ascorbic acid was used as the positive control. The hydroxyl radical scavenging effect was calculated as:

hydroxyl radical scavenging activity (%) =
$$\left[\frac{A_0 - (A_1 - A_2)}{A_0}\right] \times 100$$

where A_1 was the absorbance of the sample or ascorbic acid, and A_0 was the absorbance of the solvent control (without sample or ascorbic acid), A_2 was the absorbance of the reagent blank without salicylic acid.

46.2.3.3 Reducing Power

The reducing power was measured according to the method of Oyaizu [18] with a minor modification. Briefly, 2.5 mL various concentrations of polysaccharide samples (GLP1 and GLP2, 1.0, 2.0, 3.0, 4.0, and 5.0 mg/mL) were mixed with 2.5 mL potassium ferricyanide (1.0 %, w/v) and were incubated at 50 °C for 20 min. Following, 2.5 mL trichloroacetic acid (10 %, w/v) was added, and the mixture was centrifuged at 5,000 rpm for 10 min. The supernatant (5 mL) was mixed with ferric chloride (0.50 mL, 0.1 %, w/v) and measured at 700 nm and ascorbic acid was used as the positive control. The reducing power was calculated as:

Reducing Power =
$$A_1 - A_2$$

where A_1 was the absorbance of the sample or ascorbic acid, and A_2 was the absorbance of the reagent blank without ferric chloride.

46.2.3.4 Protective Effect on Yeast Cells from UV

Firstly, the yeast cells were cultivated to the early stage of the logarithmic phase in yeast extract peptone dextrose medium, and then 20 mL fermentation broth was centrifuged at 4,000 rpm for 5 min. Following, the precipitation was washed with phosphate buffer solution (pH 6.9) twice and then mixed with 20 mL phosphate buffer solution. A total of 1 mL yeast suspension was mixed with 4 mL various concentrations of GLP1 or GLP2 sample (4, 8, 12, 16, 20 mg/mL) and disposed

with UV radiation for lethal time (1 min), respectively. The control group was added with phosphate buffer solution. Finally, the mixture was diluted to a suitable concentration and spread on a plate medium, cultured at 28 °C for 48 h. Ascorbic acid was used as the positive control. The survival rate was calculated as follows:

The survival rate (%) =
$$\left[\frac{A_1 - A_2}{A_0 - A_2}\right] \times 100$$

where A_0 was the amount of cells which had not been disposed by UV radiation, A_1 was the amount of polysaccharide sample, and A_2 was the amount of the blank control group.

46.2.3.5 Protective Effect on Yeast Cells from H₂O₂

Similar to the oxidative injury of UV irradiation above, the yeast suspension was prepared according to the method described in Sect. 46.2.3.4. A total of 1 mL yeast suspension was mixed with 4 mL polysaccharide samples (GLP1 or GLP2 of 4, 8, 12, 16, 20 mg/mL), and then disposed by 1 mL H₂O₂ solution (lethal concentration 2 %) for 30 min, respectively. The control group was mixed with 1 mL phosphate buffer solution (instead of the polysaccharide samples). The survival rate was calculated using the following equation:

The survival rate
$$(\%) = \left[\frac{A_1 - A_2}{A_0 - A_2}\right] \times 100$$

where A_0 was the amount of cells which had not been disposed by H_2O_2 solution, A_1 was the amount of polysaccharide sample, and A_2 was the amount of the blank control group.

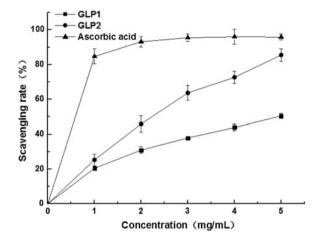
46.3 Results and Discussions

46.3.1 Scavenging Ability on DPPH Radicals

The method of scavenging the stable DPPH radicals is a widely used assay to evaluate the antioxidant activities of different compounds. The effects of scavenging DPPH radicals of GLP1 and GLP2 were measured and shown in Fig. 46.1. The results indicated that both GLP1 and GLP2 showed obvious scavenging activity on DPPH radicals in a concentration-dependent fashion. Compared with GLP2, GLP1 showed a weaker scavenging activity on DPPH radicals. The scavenging abilities at high concentration of GLP1, GLP2, and ascorbic acid were 50, 85, and 95 %, respectively. It has been reported that *G. lucidum* fruit body polysaccharides showed 46–65 % (4.0 mg/mL) of DPPH radical scavenging activities [19], similar

R. Tao et al.

Fig. 46.1 Scavenging effects of GLP1 and GLP2 on DPPH free radicals

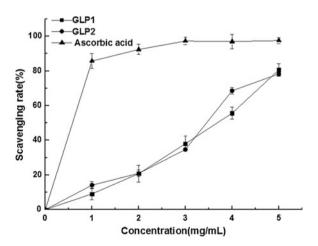


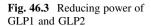
to the polysaccharides in this study. The difference between GLP1 and GLP2 may be related to monosaccharide component and molecular size.

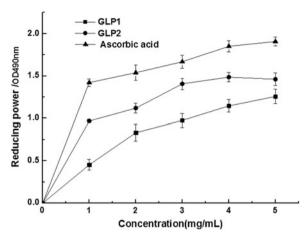
46.3.2 Scavenging Ability on Hydroxyl Radicals

Hydroxyl radical reacts with most biomolecules including carbohydrates, proteins, lipids and DNA in cells, causing damage to the heterocyclic DNA bases or cell death by a variety of mechanisms [20]. Thus, scavenging hydroxyl radicals are important for the protection of living systems. The antioxidant mechanism may be related to hydrogen supplied by polysaccharides, which combines with radicals and forms a stable radical to terminate the radical chain reaction [14]. Figure 46.2

Fig. 46.2 Scavenging effects of GLP1 and GLP2 on hydroxyl free radicals







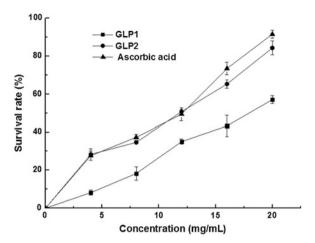
showed the results of the scavenging effects on hydroxyl radicals of GLP1, GLP2, and ascorbic acid. At the test concentrations, the scavenging activities of various samples increased with the increase of sample concentration ranging from 1.0 to 5.0 mg/mL. At 5.0 mg/mL, the scavenging activities of GLP1 and GLP2 were 80 and 78 %, respectively. The results were agreed with the previous report by Fan et al. [21]. They found that the scavenging ability on hydroxyl radicals of *G. lucidum* polysaccharides were 33.6–63.6 % at 2.5 mg/mL. Our data on the activities of scavenging hydroxyl radical of two polysaccharides suggested that they were potential antioxidants.

46.3.3 Reducing Power

In the reducing power assay, the antioxidants in the samples would result in the reduction of Fe³⁺ to Fe²⁺ by donating an electron. The antioxidant activity was concomitant with the reducing power. The reducing capacity of GLP1, GLP2, and ascorbic acid was tested and the results were shown in Fig. 46.3. The results showed that the reductive potential of GLP1 and GLP2 increased to 1.25 and 1.46 as the concentrations increased to 5 mg/mL, respectively. While ascorbic acid showed a stronger reducing power (1.91 at 5 mg/mL) than the polysaccharides samples. The results were much higher than the previous report [14], in which the reducing power of *G. lucidum* fruit body polysaccharides were 0.68–1.48 at 10.0 mg/mL. The reducing capacity of a compound may be considered as a significant marker of its potential antioxidant activity.

448 R. Tao et al.

Fig. 46.4 Protective effects of GLP1 and on yeast cells from UV oxidative damage



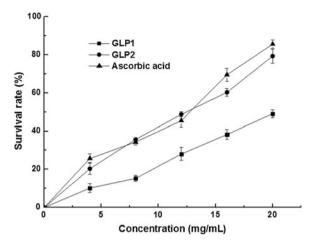
46.3.4 Protective Effect on Yeast Cells from UV

Studies have proved that UV radiation may cause oxidative-induced DNA damage or cell death, which is induced by the generation of oxygen-derived species including free radicals, such as O_2 . O_2 , O_2 , O_2 , O_3 , O_4 , O_4 , O_5 . O_7 . The presence of antioxidants could effectively reduce the DNA damage caused by UV radiation. The yeast cells were used in the assay to measure the protective effects of polysaccharides samples (at the dose of 4, 8, 12, 16, 20 mg/mL) on yeast cells from UV. The antioxidant activity was related to the survival rate of the yeast cells. As shown in Fig. 46.4, GLP1 and GLP2 demonstrated protective effects on yeast cells in a dose-dependent tendency at the concentrations examined. At 20 mg/mL, the protective effects on yeast cells of GLP1 and GLP2 were 57 and 84 %, respectively. However, the protective effect of ascorbic acid was a little stronger (91 % at 20 mg/mL) than GLP1 and GLP2. The protective effect on yeast cells of two polysaccharides were closely correlated with their antioxidant activities.

46.3.5 Protective Effect on Yeast Cells from H₂O₂

 H_2O_2 is an interesting class of cytotoxic agents that tends to arise naturally from a wide variety of chemical and biochemical oxidations. Hydrogen peroxide is an essential component for toxicity and it can interact with macromolecules (e.g., DNA), cells and tissues to produce damage [22]. Thus, in this assay, the protective effect on yeast cells from H_2O_2 was used to determine the antioxidant activities of GLP1 and GLP2, and the survival rate of yeast cells is shown in Fig. 46.5. At the test concentrations, GLP1 and GLP2 exhibited protective effects on yeast cells from H_2O_2 in a dose-dependent manner. At 20 mg/mL, the protective effects on yeast

Fig. 46.5 Protective effects of GLP1 and GLP2 on yeast cells from H₂O₂ oxidative damage



cells of GLP1 and GLP2 were 49 and 79 %, respectively. However, the results were a little less than that of ascorbic acid (85 % at 20 mg/mL).

From the above results, both GLP1 and GLP2 exhibited potential antioxidant capacities and could be explored as a source of bioactive compounds.

46.4 Conclusion

In this study, two polysaccharides from the fermentation broth and fruiting body of *G. lucidum* were isolated. Moreover, the antioxidant activities of GLP1 and GLP2 indicated that they had significant scavenging effects on free radicals (DPPH, hydroxyl radicals), reducing power and protective effect on yeast cells from UV and H₂O₂. The results showed that both the two polysaccharides exhibited antioxidant activities in a concentration-dependent manner and the antioxidant activities of GLP2 were superior to GLP1. Above all, the polysaccharides of *G. lucidum* could be utilized as antioxidant for food and pharmaceutical industries. Further research is needed to elucidate the mechanism of their antioxidant activities.

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450 R. Tao et al.

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Chapter 47 Biofortification Using Bacteria Containing an Atrazine-Degrading Gene and Its Effects on Reactor Operating Efficiency

Yue Wang and Jinxu Sun

Abstract To investigate the effects of biofortification of activated sludge reactors using genetically engineered microbes containing a plasmid encoding atrazine chlorohydrolase, we compared biodegradation efficiency and sludge properties during three operating stages in a conventional activated sludge (CAS) reactor and a MBR. Our results show that with the addition of genetically engineered bacteria and the selection for indigenous biodegrading microbes, membrane fouling was reduced. Atrazine has a certain level of biotoxicity toward activated sludge and shows an inhibitory effect toward pollutant removal. After biofortification, atrazine degradation was superior in MBR than that in CAS reactors. The atrazine removal rate was 92.6 % in the MBR, while it was 82.6 % in the CAS reactor. Following biofortification, the sludge concentration in the MBR was maintained at 7.3 g/L, while the sludge concentration in the CAS reactor was 2.3 g/L.

Keywords Biofortification • Genetically engineered bacteria atrazine degradation • Conventional activated sludge reactor • Membrane bioreactor

47.1 Introduction

Biodegradation-resistant organic pollutants, particularly synthetic compounds, can only be broken down through the actions of complex series of cooperative metabolic mechanisms found in different species of biodegradative bacteria [1]. Superbiodegrading bacteria overcome this need for interspecies collaboration, as they are genetically engineered to contain recombinant genes coding for degradative enzymes originating from different bacterial species. This type of genetic

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⁴⁵¹

452 Y. Wang and J. Sun

engineering can effectively improve the biodegradative activity of the superbiodegrading bacteria and increase pollutant removal.

Biofortification is a process that can help improve the removal efficiency of biodegradation-resistant organic pollutants. Beginning in the late 1980s, researchers studied the use of genetically engineered bacteria for the biofortification of wastewater treatment reactors [2]. Since then, there has been a growing interest in using genetically engineered bacteria to assist in the removal of organic pollutants through biofortification. Experimental evidence shows that the use of genetically engineered bacteria can effectively improve the removal efficiency of biodegradation-resistant pollutants, greatly improving both the rate of degradation of pollutants and the ability of the reactors to handle a greater pollutant load [3–8].

47.2 Materials and Methods

47.2.1 Reactor Parameters

Our experimental setup consisted of a CAS reactor and an MBR system. The main portion of the reactor was made of plexiglass.

The CAS reactor is shown in Fig. 47.1 and was constructed using aeration and sedimentation tanks. Artificially prepared wastewater used in this study was fed into the reactor via a peristaltic pump. The reactor was made of plexiglass and its dimensions are $150 \text{ mm} \times 300 \text{ mm} \times 350 \text{ mm}$, with an effective volume of 9 L.

The MBR system is shown in Fig. 47.2. Artificially prepared wastewater was fed into the reactor via the intake pump. The dimensions of the reactor are $140 \text{ mm} \times 250 \text{ mm} \times 660 \text{ mm}$, with an effective volume of 15 L. The membrane was a hollow fiber PVDF ultrafiltration membrane.

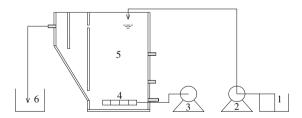
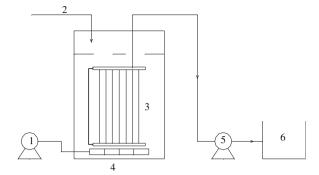


Fig. 47.1 Conventional activated sludge system schematic diagram. *1* Water supply tank. 2 Peristaltic pump. *3* Air pump. *4* Revealed the trachea. *5* The reactor. *6* Out of the water tank

Fig. 47.2 The MBR system schematic diagram. *I* Air pump. 2 The water. *3* Membrane module. *4* Reactor. *5* Out of the water pump. *6* Out of the water tank



47.2.2 Source of Sludge and Artificially Prepared Wastewater

In this study, secondary settling tank sludge reflux was collected from the sewage treatment plant at Shijiazhuang Bridge West. The sludge concentration was 4 g/L, with high floc density and good settling characteristics.

The experimental wastewater was artificially prepared [9]. After initiating the reactor with the appropriate startup operations, atrazine was added to the influent water at a concentration maintained between 10 and 20 mg/L. The reactor's operating temperature was maintained at approximately 25 °C.

Using the same hydraulic retention time (HRT), the reactor operation was divided into two major stages:

- (1) Sludge culture stage (Run-0);
- (2) Sludge operation stage, which was further divided into three shorter stages, as shown in Table 47.1.

47.2.3 Preparation of Bacterial Strains and Suspensions

The genetically engineered bacteria used in this study were Escherichia coli DH5 α cells containing the pACYC184 plasmid encoding atrazine chlorohydrolase and the

Table 47.1	Table experiment	contents and	operating	conditions
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Table experiment contents	Operating conditions
Blank operation phase (Run-1)	COD concentration in the influent water was 300–500 mg/L; ammonia was 40–50 mg/L; HRT was 24 h; no atrazine
Abiotic strengthen the operating period (Run-2)	Atrazine concentration was 25–30 mg/L; other conditions the same as Run-1
Biological strengthen the operating period (Run-3)	Genetic engineering bacteria in the influent water was 2.16×107 CFU/mL; other conditions the same as Run-2

454 Y. Wang and J. Sun

pUC18-gfp plasmid encoding green fluorescent protein [10]. As a result, the cells are resistant to ampicillin and chloramphenicol.

A single colony was picked and grown overnight in the LB medium (containing 25 μ g/mL of chloramphenicol) at 37 °C with shaking at 120–140 rpm. The culture was centrifuged and washed with phosphate buffer (pH 7.0). Cells were harvested to prepare bacterial suspensions for subsequent experiments.

47.2.4 Detection of Pollutants

COD and ammonia were measured using national standardized methods.

Atrazine concentrations were measured by high-performance liquid chromatography (Shimadzu) analysis of water samples filtered through a 0.45 μ m membrane. A C18 reverse phase column and UV detector were used and the methanol:water ratio was 70:30. The detection wavelength was 223 nm.

47.2.5 Sludge Performance Testing

Measurements were performed in accordance with national standards.

Sludge floc particle sizes were measured using a laser particle size distribution analyzer (BT-9300H).

Sludge activity was measured using the total oxygen uptake rate (OUR).

47.3 Results and Analyses

47.3.1 Membrane Pressure Changes in the MBR

Changes in membrane pressure during the three stages of reactor operation are shown in Fig. 47.3.

The membrane must be cleaned when the pressure reaches 180 kPa. The regular rapid decreases in pressure shown in Fig. 47.3 reflect the times when the membrane was cleaned during reactor operation. During the different stages, membrane fouling took place quickly during Run-1, resulting in membrane cleaning every 2 months. During Run-2, the addition of atrazine disrupted the activity of the microbes in the system, resulting in rapid increases of membrane pressure due to increased membrane fouling. The frequency of membrane cleaning during this stage was once a month. During Run-3, when the genetically engineered bacteria were added, atrazine was degraded, and it took longer for the membrane pressure to increase to the maximum tolerable value of 180 kPa than during Run-2. On average, the membrane was cleaned once in every 2 months.

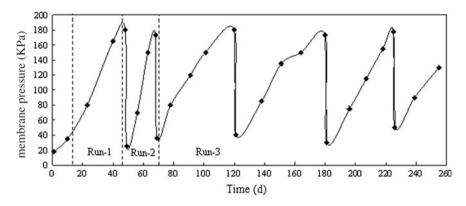


Fig. 47.3 Membrane pressure difference along with the change of running time

Bacterial growth, metabolites, organic molecules from wastewater, dissolved substances, and solid particles all contribute to the film formation on the surface of the membrane or to membrane obstruction. We show that atrazine toxicity toward sludge microbes results in accelerated membrane fouling, likely due to increased secretion of extracellular polymers and the accumulation of the secretions. However, the addition of genetically engineered bacteria improved the sludge properties, resulting in decreased membrane fouling.

47.3.2 Analysis of Pollutant Removal

47.3.2.1 Conventional Pollutants

The COD removal efficiency of the MBR and CAS reactors during all four operating stages is shown in Fig. 47.4.

As shown in Fig. 47.4, at each of the four operating stages, the MBR showed an average COD removal efficiency of 80.6, 95.6, 75.6, and 97.3 %. The average COD removal efficiency of the CAS reactor was 73.8, 83.8, 77.9, and 88.4 % during the same four stages of operation.

Clearly, the MBR has significantly higher COD removal efficiency than the CAS reactor, due to more efficient sludge and microbe retention by its membrane. Sludge is easily lost in the CAS reactor, and the effluent water contains suspended solids, indicating lower effluent water quality.

Meanwhile, in the two reactors, the trends of the rate of COD removal were roughly the same. The addition of atrazine during Run-2 led to a significant decrease in the rate of COD removal, which is reflective of the negative effect of atrazine on the biological activity of the activated sludge. During Run-3, the efficiency of COD removal returned to the levels similar to Run-1, indicating that

456 Y. Wang and J. Sun

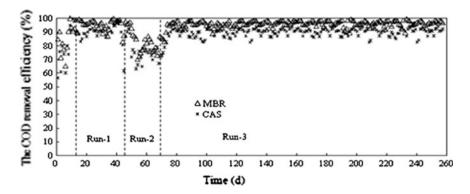


Fig. 47.4 The COD removal efficiency of the MBR and CAS reactors

biofortification using our genetically engineered bacteria helped remove atrazine in the system and reduced the adverse effects of atrazine on sludge bioactivity.

The ammonia removal efficiency of the MBR and CAS reactors during the four operating stages is shown in Fig. 47.5.

Figure 47.5 shows that the average concentration of ammonia in the influent water in both the reactors was 52.3 mg/L. The average rates of ammonia removal by the MBR during the four stages of operation were 65.4, 83.4, 71.9, and 87 %. The rates were 52.0, 78.3, 63.4, and 80 % in the CAS reactor.

These results show that the MBR is significantly better than the CAS reactor at removing ammonia. As the membrane in the MBR traps a large quantity of nitrifying bacteria, ammonia removal efficiency was improved.

Meanwhile, the ammonia removal efficiency shows the same trend in both the reactors. Addition of atrazine during the Run-2 stage significantly decreases the ammonia removal efficiency in the MBR, from 83.4 to 71.9 %. This suggests that

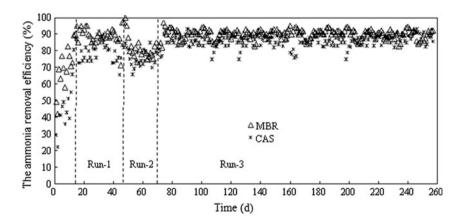


Fig. 47.5 The ammonia removal efficiency of the MBR and CAS reactors

atrazine has an inhibitory effect on nitrifying bacteria and their ability to remove ammonia. However, the addition of genetically engineered bacteria during Run-3 led to the increase of ammonia removal efficiency to 87 %, indicating that biofortification using genetically engineered bacteria reduced the quantity of atrazine in the system and further reduced the adverse effects of toxic compounds on the bioactivity of nitrifying bacteria.

Our genetically engineered bacteria showed high biodegradation activity toward atrazine. Their effect on reactor efficiency following biofortification is well demonstrated through the efficiency of atrazine removal.

The efficiency of atrazine removal by both the MBR and CAS reactors during the Run-2 and Run-3 operation stages is shown in Fig. 47.6.

During the Run-2 stage, reactors with no biofortification were not able to biodegrade atrazine.

During the Run-3 stage, the average rate of atrazine removal was 92.6 and 82.6 %, respectively. Initially after biofortification, the atrazine removal rate of the MBR increased rapidly to over 85 %. Soon after, the efficiency of atrazine biodegradation decreased due to indigenous bacteria preying on and competing with the genetically engineered bacteria, but atrazine removal remained at a steady rate during the later stage of operation. Atrazine removal by the CAS reactor gradually increased from 30 % to approximately 50 % during the early stage. However, due to the loss of genetically engineered bacteria from lack of retention or predation and competition by indigenous microbes, there was a slight decrease in atrazine removal during the subsequent stage of operation. When the efficiency of atrazine removal was compared between both the reactors, the MBR showed superior removal efficiency under the same operating conditions, likely due to microbial retention by the membrane in the MBR. This phenomenon may also be explained by the selection for indigenous atrazine biodegrading microbes.

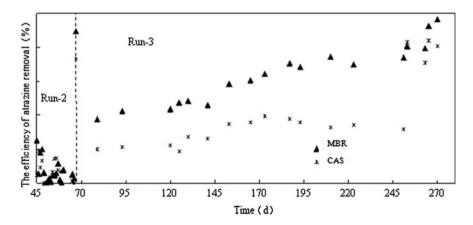


Fig. 47.6 The efficiency of atrazine removal by both the MBR and CAS reactors

458 Y. Wang and J. Sun

47.3.3 Sludge Characteristics

We analyzed the sludge particle size distribution, which indirectly reflects the operating conditions of the reactor. The distribution of sludge particle sizes is shown in Fig. 47.7.

As shown in the figure, during the early stages of Run-1, when the sludge was first added, the particles had rather large diameters in both the MBR and the CAS reactors. After a certain time of operation, the sludge particle diameters decreased and finally stabilized at approximately 50 μ . During Run-2, the sludge particle diameters gradually decreased in both the MBR and the CAS reactors, indicating deterioration in sludge settling and reactor performance. In Run-3, sludge particle sizes grew larger in both the reactors. Overall, sludge particles in the MBR were larger than those in the CAS reactor, and showed good settling properties.

The activity of the sludge on organic matter degradation and nitrification are reflected by the reactor's treatment of COD and ammonia. We measured the rate of oxygen consumption during all three operational stages: Run-1, Run-2, and Run-3 to calculate the sludge activity in both the reactors. The results are shown in Figs. 47.8, 47.9 and 47.10.

Based on our results, the activity of the MBR was higher than that of the CAS reactor in Run-1, Run-2, and Run-3. We believe that this is due to microbial retention by the membrane in the MBR, leading to higher quantities of retained microorganisms and higher sludge activity. For this reason, the MBR showed superior COD and ammonia removal efficiencies to the CAS reactor. Meanwhile, both the MBR and the CAS reactors showed decreased activity in Run-2, which we believe to be due to the biotoxicity of the added atrazine toward the reactor microorganisms, which consequently negatively affected sludge activity, ultimately resulting in a decrease in the removal efficiency of COD and ammonia. Both the reactors recovered their initial sludge activity in Run-3.

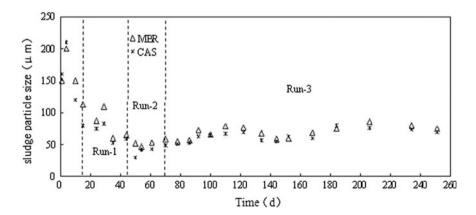


Fig. 47.7 The distribution of sludge particle sizes

Fig. 47.8 The sludge activity of Run-1 operational stage

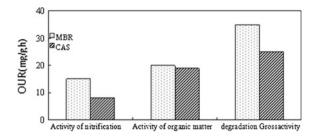


Fig. 47.9 The sludge activity of Run-2 operational stage

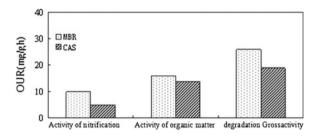
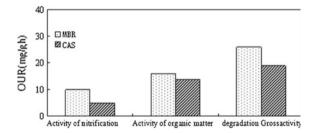


Fig. 47.10 The sludge activity of Run-3 operational stage



47.4 Conclusion

Atrazine is biotoxic and can negatively affect sludge settling, cause sludge floc dispersion, and increase extracellular secretions, thereby leading to accelerated membrane fouling in the MBR. With the use of genetically engineered bacteria and the colonization of indigenous biodegrading bacteria, membrane fouling was alleviated.

After biofortification, the MBR showed higher atrazine removal efficiency than the CAS reactor. Physical adsorption of atrazine by the activated sludge leads to lower effluent atrazine concentrations immediately after its addition to the influent water. After adsorption saturation, the effluent atrazine concentration increases. Loss of genetically engineered bacteria due to predation by the indigenous microbes in the activated sludge, as well as the loss of sludge, can lead to a decrease in the density of genetically engineered bacteria in the sludge. With bacterial growth and selection for indigenous atrazine-degrading microbial species, atrazine removal

460 Y. Wang and J. Sun

efficiency grows progressively higher. Ultimately, the efficiency of atrazine removal by the MBR was 92.6 %, while the efficiency of the CAS reactor was 82.6 %.

Under the same conditions of HRT, influent water load, and addition of the same concentration of genetically engineered bacteria, both the reactors showed similar changes in the removal of conventional pollutants. During the Run-2 stage, the addition of atrazine to the influent water decreased the removal efficiency of conventional pollutants, indicating that atrazine has some biotoxicities toward activated sludge as well as inhibitory activity toward pollutant removal. During the Run-3 stage, the efficiency of pollutant removal was restored to pre-atrazine levels, indicating that biofortification using genetically engineered bacteria reduces the adverse biological effects of atrazine on sludge activity, which is reflected by the restoration of removal efficiency of conventional pollutants by biofortified activated sludge.

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Chapter 48 A Novel One-Pot Five-Component Synthesis of Tetrahydro-pyrrolo[3,4-b] pyridine-5-one via Ugi/Aza-Diels-Alder

Yan Liu, Tianyi Shang, Chuanming Xu, Hui Yang, Peng Yu and Kui Lu

Tandem Reaction

Abstract A novel five-component domino processes to tetrahydro-pyrrolo[3,4-b] pyridine-5-one (**8**) were developed. Reaction of 5-phenyloxazole-2-carbaldehyde (**1**), *p*-toluidine (**2**), (Z)-4-methoxy-4-oxobut-2-enoic acid (**3**), and *t*-butyl isocyanide (**4**) in methanol at 60 °C provides an efficient access to the drug-like bicycle compound (**8**) in a good yield. In this one-pot process, a traditional oxo-bridged tricycle intermediate was formed via Ugi/ara-Diels-Alder tandem reaction and then the solvent, methanol, sequentially attacks and cuts off the oxo bridge under mild conditions via SN2 reaction. Formation of one C-N, two C-O, and three C-C bond with the triple domino sequence (Ugi/IMDA/SN2) is involved in this new scaffold generating reaction. The operational simplicity and maximizing the buildup of structural complexity makes this novel multicomponent reaction valuable in diversity-oriented synthesis of tetrahydro-pyrrolo[3,4-b]pyridine-5-one.

Keywords MCR · Ugi/aza-Diels-Alder tandem reaction · SN2 reaction · Domino process · Diversity-oriented synthesis

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⁴⁶¹

462 Y. Liu et al.

48.1 Introduction

Multicomponent reaction (MCR) is being recognized as one of the most promising approaches in diversity-oriented synthesis and is undoubtedly well suited for the drug discovery program [1]. Among named MCRs, the Ugi four-component reaction (Ugi-4CR) is without doubt one of the most powerful transformations [2]. By saving synthetic operations while maximizing the buildup of structural and functional complexity, this highly step economic reaction is particularly appealing in application of diversity-oriented synthesis [3]. Clearly, a proper union of Ugi with other reactions in a sequential or a domino process will further expand its potential [4, 5].

Indeed, pyrrolopyridine has attracted increased attention as biologically active compounds such as central nervous system agents [6], as herbicides [7] and as antidiabetic agents [8]. This skeleton has also been found in medically important natural product such as camptothecine [9] and meppicine ketone [10] (Fig. 48.1). As part of our research program directed at synthesizing drug-like heterocycles, we were interested in the development of a novel multicomponent synthesis to generate new scaffolds such as polysubstituted pyrrolo[3,4-b]pyridine-5-one.

One possible new scaffold generating reaction was illustrated by Ugi/aza-Diels–Alder tandem reaction with Lewis acid to afford the pyrrolopyridine (Scheme 48.1) [11].

Janvier et al. [12] have already shown that MCR/IMDA reaction cascades of 5-amideoxazole with a dienophile was possible leading to [4+2] cycloaddition product (Scheme 48.2).

To further apply this strategy to diversity-oriented synthesis, we set out to examine the previously unexplored Ugi reaction subsequentially combined with intramolecular Diels-Alder cycloaddition reaction of 5-phenyloxazole. According to the references, the reaction sequence we envisaged is highlighted in Scheme 48.3.

A reaction of 5-phenyloxazole-2-carbaldehyde (1), *p*-toluidine (2), (Z)-4-methoxy-4-oxobut-2-enoic acid (3), and *t*-butyl isocyanide (4) in methanol with Lewis acid in one pot was designed to provide a key pyrrolo[3,4-b]pyridine-5-one (9), which could then be used as a point to produce diverse bioactive libraries.

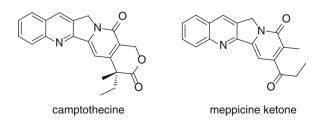


Fig. 48.1 Pharmaceutically important natural products camptothecine and meppicine ketone

Scheme 48.1 Construction of heterocycle via sequential Ugi reaction/Diels-Alder cycloaddition reaction

Scheme 48.2 Construction of pyrrolo[3,4-b]pyridine-5-one via MCR/IMDA reaction cascades of 5-amideoxazole

Scheme 48.3 Synthesis of tetrahydro-pyrrolo[3,4-b]pyridine-5-one

464 Y. Liu et al.

However, to our surprise, there is no 9 producing in the reaction instead of 8. From the results of the experiment, a sequence of reaction involving intramolecular Diels-Alder cycloaddition and SN2 reaction was proposed to explain the formation of adduct 8. In this one-pot process, a traditional oxo-bridged tricycle intermediate 7 was formed via Ugi/Aza-Diels-Alder tandem reaction.

48.2 Results and Discussions

Using 5-phenyloxazole-2-carbaldehyde (1), *p*-toluidine (2), (Z)-4-methoxy-4-oxobut-2-enoic acid (3), and *t*-butyl isocyanide (4) as substrates, we got the adduct 8 in 62 % yield. In this one-pot process, a traditional oxo-bridged tricycle intermediate 6 was formed via Ugi/aza-Diels-Alder tandem reaction, and then the solvent, methanol, sequentially attacks α-atom of -C=N and cuts off the oxo bridge in mild conditions via SN2 reaction. In order to get the tetracyclic intermediate 7, we performed a survey of reaction conditions varying the solvents (MeOH, THF, H₂O, DCM, toluene, MeCN). The experimental results are summarized in Table 48.1. It is proved that the presence of MeOH was of utmost importance for the success of this reaction. Indeed, in its absence, there is no product 8 was obtained in other solvents. So MeOH may act as solvent as well as an activator for the domino sequence of Ugi reaction, intramolecular Diels-Alder reaction, and subsequent SN2 reaction.

In summery, we have developed an efficient five-component synthesis of tetrahydro-pyrrolo[3,4-b]pyridine-5-one. The optimized conditions we found for this five-component reaction consisted of heating a MeOH solution of 1, 2, 3, and 4 at 60 °C for 10–12 h. Taking advantage of the chemical reactivity of 5-phenyloxazole and by fine-tuning the structure into the starting materials, we are expecting that this new five-component reactions can be devised for the synthesis of a number of tetrahydro-pyrrolo[3,4-b]pyridine-5-one. It is worth noting that one C–N, two C–O, and three C–C bonds were formed with the triple domino sequence (Ugi/IMDA/SN2) in this one-pot process. The efficiency of this reaction was thus truly remarkable if one looks at the yield per bond formation. At the same time, the

Table 48.1 Reaction conditions varying the solvents^a

Entry	Sovelution	[1] (M)	Temp (°C)	Time (h)	Yield (%) ^b
1	МеОН	1.0	60	12	62
2	THF	1.0	60	12	n.r.
3	H ₂ O	1.0	60	12	n.r.
4	DCM	1.0	60	12	n.r.
5	Toluene	1.0	60	12	n.r.
6	MeCN	1.0	60	12	n.r.

^aReaction conditions: **1** (1.0 mmol), **2** (1.0 mmol), **3** (1.0 mmol),

^{4 (1.0} mmol) in 2 mL MeOH

^bIsolated yield

significant feature of this procedure is the exploitation of methanol to cut off oxo bridge under mild conditions. It was the first time to report herein full details in development of this five-component reaction via the triple domino sequence (Ugi/IMDA/SN2) to successfully synthesize tetrahydro-pyrrolo[3,4-b]pyridine-5-one.

48.3 Experimental

48.3.1 General Procedure for the Three-Component Synthesis of Methyl 7-(tert-butylcarbamoyl)-7a-hydroxy-3-methoxy-5-oxo-3-phenyl-6-(p-tolyl)-4, 4a,5,6,7,7a-hexahydro-3H-pyrrolo[3,4-b] pyridine-4-carboxylate (8)

In a 10 mL vial equipped with a magnetic stir bar, 5-phenyloxazole-2-carbaldehyde (1) (173 mg, 1.0 mmol), *p*-toluidine (2) (107 mg, 1.0 mmol), (Z)-4-methoxy-4-oxobut-2-enoic acid (3) (130 mg, 1.0 mmol) in MeOH (2.0 mL) was added *t*-butyl isocyanide (4) (83 mg, 1.0 mmol). The mixture was stirred at 60 °C for 10–12 h. Once the product formation deemed completed by TLC analysis, stopped the reaction. The mixture was cooled to room temperature and concentrated under reduced pressure to give a residue which was purified by silica gel column chromatography with hexane/ethyl acetate/CH₂Cl₂.

48.3.2 Characterization of Compound 8

Compound 8 (314 mg, 62 %) was obtained as white solid. 1 H NMR (400 MHz, d_6 -DMSO): δ 7.68 (s, 1H), 7.61 (d, J = 3.2 Hz), 7.48–7.44 (m, 4H), 7.25–7.22 (m, 3H), 7.07 (d, J = 8.4 Hz, 2H), 5.27 (s, 1H), 5.16 (s, 1H), 4.47 (d, J = 2.8 Hz, 1H), 3.55 (s, 1H), 3.53 (s, 3H), 3.30 (s, 3H), 2.23 (s, 3H), 1.15 (s, 9H). 13C NMR (100 MHz, d_6 -DMSO): δ 20.8, 28.4, 45.7, 51.3, 52.1, 55.4, 56.7, 62.9, 73.6, 88.7, 98.7, 117.8, 125.9, 127.7, 128.1, 129.4, 131.1, 137.7, 144.1, 153.7, 164.3, 169.9, 171.1. LRMS (ESI) m/z 508.4 [M+H]⁺.

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Chapter 49 Study on Preparation and Application of High Esterification Green-Liquor Daqu

Xiao-Dan Wang, Bei Xiao, Shi-Dong Ban, Si-Xia Xu, Xi-Cui Shen and Shu-Yi Qiu

Abstract The optimal conditions of esterification Daqu production requires medium temperature to be used in the Daqu making process. The ratio of raw materials is 6:3:1 (wheat: barley: pea), the inoculation of *Monascus (Monascus purpureus*) FBKL3.0018 is 10 g/kg, and the water of Daqu is controlled at 40 %. The highest temperature is more suitable, which is controlled at 55 °C, and the culture time is 25–28 days. The best storage period of Daqu esterification is 2–3 months. This study could improve the quality of Daqu, the development, and research of fortified Daqu, which will become a hot point in increasing the quality rate of Luzhou-flavor liquor.

Keywords Esterification · Vinosity · Daqu

The yield, quality, and style of Luzhou-flavor liquor are directly affected by the quality of Luzhou-flavor Daqu. The bacteria, enzyme, and composite carrier of Daqu are made by environmental microorganisms. Fortified Daqu is used by the isolation technique of microorganisms, screening of beneficial bacteria, which could fully improve the biochemical properties, Then adding the beneficial bacteria into the Daqu improves the fortified Daqu [1, 2]. The combination of traditional brewing technology and modern biological engineering technology is used for screening and breeding of beneficial bacteria. The development and research of

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⁴⁶⁷

468 X.-D. Wang et al.

fortified Daqu became a hot point to increase the quality rate of Luzhou-flavor liquor [3–7].

This study used the strong esterifying power of *Monascus purpureus* (FBKL3.0018) to make Daqu. In order to improve the quality of Daqu and yeast wine, the conditions and parameters of starter-making technology should be explored.

49.1 Materials and Methods

49.1.1 Raw Materials

Monascus purpureus (FBKL3.0018), wheat, barley, and pea.

49.1.2 Equipment

Agilent GC 6890 FID (NYSE:A); U-3010 ultraviolet-visible light detector (Hitachi).

49.1.3 Medium

Rose Bengal agar, malt agar medium, NA.

49.1.4 Adding the Proportion of Monascus purpureus

Starter-making technology:

Raw materials (wheat, barley, pea) \rightarrow Crushing \rightarrow Water mixing \rightarrow Strain breeding \rightarrow Strain culturing \rightarrow Then inoculate \rightarrow Bacterium suspension \rightarrow Artificial on Daqu \rightarrow Entry \rightarrow Arrangement \rightarrow Mildew-growing \rightarrow Tide phase \rightarrow Fire stage \rightarrow Insulating \rightarrow Out of the house \rightarrow Storage \rightarrow Product Daqu.

Before the production of esterification Daqu, the room must be cleaned up. The ground is covered with a layer of fresh rice husk of about 5 cm.

The medium temperature is used in the Daqu making process. The ratio of raw materials is 6:3:1 (wheat: barley: pea), 1 kg material with *Monascus* and respective quantities of 2, 5, 8, 10 and 15 g. Through the evaluation of physical and chemical indexes and aesthetic quality, the proportion of *Monascus* is determined.

Test number	Wheat (mass fraction %)	Barley (mass fraction %)	Pea (mass fraction %)
1	100	0	0
2	80	10	10
3	70	20	10
4	60	30	10
5	50	40	10

Table 49.1 Different ratio of raw material

49.1.5 The Different Ratios of Raw Materials

The medium temperature is used in the Daqu making process. Daqu are made by the different ratios of raw materials through the evaluation of physical and chemical indexes and aesthetic quality to determine the proportion of raw materials (Table 49.1).

49.1.6 Effect of Water Addition on the Quality of Daqu

Water addition is key during the experiments; excess water can lead to difficulty in molding of grains, and it is also not conducive to make beneficial bacteria grow in Qu Xin. Excess heat causes rancidity; however, if the water is too little, the grains do not bond easily and this would also prevent the microbial from growing. The mass fraction of water, respectively, is controlled at 37, 40, 43 % of the raw materials through the evaluation of physical and chemical indexes and aesthetic quality to determine the best temperature.

49.1.7 Effect of Cultivation Time on Daqu Quality

During the cultivation process of Daqu, take samples which have been cultivated at 15, 20, 25, 30 days, respectively, for analysis. Through the evaluation of physical and chemical indexes and aesthetic quality the best time of starter-making is determined.

49.1.8 Determination of Moisture Content [8]

49.1.9 Determination of Acidity [8]

470 X.-D. Wang et al.

49.1.10 The Determination of Fermenting Power [8]

49.1.11 Determination of Starch Content [8]

49.1.12 The Determination of Enzyme Activity

- 49.1.12.1 Determination of the Liquefying Power: Iodine Color-Fading-Photometric Method at 30 °C [9]
- 49.1.12.2 Determination of the Proteolytic Power: Precipitation [10]
- 49.1.12.3 Determination of the Saccharifying Power: Lam Method [10]
- 49.1.12.4 Determination of the Esterifying Power [9]

49.1.13 Determination of Microorganisms

Colony counting method.

49.2 Results and Analysis

49.2.1 Addition of Monascus

Before starter-making, the bacteria suspension is made by the *Monascus* and water (1:20), then 2, 5, 8, 10, 15 g in 1 kg raw materials, respectively, is added while mixing in water to make the Daqu.

We can see from Table 49.2 that there is a big difference between esterification and medium temperature Daqu in acidity, which indicates that the acidity of the medium temperature Daqu changes to bigger than esterification Daqu.

We can see the following from Table 49.3:

- (1) Compared with ordinary Daqu, the esterifying power of the esterification Daqu is much higher.
- (2) With the increasing addition of *Monascus*, the esterifying power, diastatic power, fermenting power, and liquefaction power of the esterification Daqu are all improved and the esterifying power is most appreciated. The highest increase can be by 84.15 %, while the minimum increase can be by 50 %.

Different Daqu	Number	Moisture fraction %	*	Acidity		Addition of Monascus (g/kg)
		Into the room	Out of the room	Into the room	Out of the room	
Esterification	1	37.0	15.5	0.20	1.04	2
Daqu	2	37.1	15.9	0.20	1.07	5
	3	37.4	16.0	0.21	1.05	8
	4	37.1	16.4	0.23	1.08	10
	5	37.0	15.5	0.21	1.09	15
Ordinary Daqu	1	37.0	15.5	0.19	1.35	0

Table 49.2 The physiochemical indexes comparison of esterification Daqu and ordinary Daqu

The above data are means of three samples

Table 49.3 The diastatic power, fermenting power, liquefaction power and esterifying power comparison of esterification Daqu and ordinary Daqu

Different Daqu	Number	Diastatic power (mg/g h)	Fermenting power (g/kg)	Liquefaction power (g/g h)	Esterifying power (mg/g 100 h)	Esterifying growth (%)
Esterification	1	320	30.7	6.35	110.8	50.14
Daqu	2	300	32.8	6.35	114.5	55.15
	3	345	33.1	6.36	129.8	75.88
	4	360	32.4	6.33	133.1	80.35
	5	330	33.4	6.45	135.9	84.15
Ordinary Daqu		720	31.2	6.35	73.8	0.00

The above data are means of three samples

(3) There is a slight difference in esterifying power between tests 4 and 5. In order to reduce the costs of production, it is more suitable that the inoculum concentration be 10 g/kg.

49.2.2 Different Ratios of Raw Materials

From Table 49.4 we can see that when the esterification Daqu is produced with pure wheat, the esterifying power is lower, which may be due to the strength adhesion of wheat. When the Daqu is stepped too tight, due to poor permeability in Qu Xin, the growth of *Monascus* is restrained. With the decreasing of wheat and adding other materials, the esterifying power of Daqu increases gradually. Through the comprehensive analysis of the above data, test no. 4 is chosen (wheat: barley: pea = 6:3:1) to make the esterification Daqu.

472 X.-D. Wang et al.

Item	Test no. 1	Test no. 2	Test no. 3	Test no. 4	Test no. 5
Moisture (%)	16	15.3	16.5	16.4	15.8
Diastatic power (mg glucose/ g h)	387	314	380	385	337
Fermenting power (g/kg)	31.5	32.1	34.5	32.4	32.1
Esterifying power (mg/g 100 h)	94.5	109.8	123.6	130.2	115.5
Liquefaction power (g/g h)	6.32	6.50	6.41	6.42	6.57
Acidity	1.35	1.28	1.30	1.34	1.36

Table 49.4 The test results of different ratio of raw materials

The test no. 1 is making by pure wheat, the barley and peas are being added respectively

49.2.3 Effect of Water Addition on the Quality of Daqu

We can see the following from Table 49.5:

- (1) The more water is added into the room, the more it is when it is taken out of the room. At the same time the acidity also improves, thus we can see that when the Daqu has taken in too much water, it will improve the acidity.
- (2) More or less water can have a measure of influence on the diastatic, fermenting, and esterifying powers, indicating that water can prevent the growth of microorganisms. So we choose the water of Daqu to be controlled at 40 %.

49.2.4 Effect of Temperature on the Quality of Daqu

We can see from Table 49.6 that esterification Daqu is made at medium temperature. The highest temperature is more suitable which is controlled at 55 °C.

49.2.5 Effect of Cultivation Time on Daqu Quality

We can see from Table 49.7 that with extension of time, the physical and chemical indexes are increased. Considering the concrete operation, the best storage period of esterification Daqu is 2–3 months.

Table 49.5 The Daqu physiochemical indexes comparison of different wat	er
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Added water (%)	Moisture (%)	Acidity	Diastatic power (mg glucose/g h)	Fermenting power (g/kg)	Esterifying power (mg/g 100 h)	Liquefaction power (g/g h)
37	16.0	1.04	341	31.4	100.3	6.41
40	16.5	1.07	359	32.5	124.5	6.46
43	17.5	1.09	336	30.1	98.8	6.16

Temperature	Moisture	Acidity	Diastatic power	Fermenting	Esterifying	Liquefaction
(°C)	(%)		(mg glucose/g h)	power	power	power (g/g h)
				(g/kg)	(mg/g 100 h)	
60	15.5	1.32	335	32.5	114.5	6.42
55	15.4	1.29	356	32.1	124.8	6.56
50	15.4	1.24	308	31.8	120.9	6.50

Table 49.6 The quality comparison of different culture temperature

Table 49.7 The comparison of cultivation time on Dagu quality

Time (day)	Moisture (%)	Acidity	Diastatic power (mg glucose/g h)	Fermenting power (g/kg)	Esterifying power (mg/g 100 h)	Liquefaction power (g/g h)
15	25.0	1.75	105	21.0	46.5	4.54
20	21.5	1.58	268	25.2	84.5	4.83
25	17.4	1.45	325	30.1	105.8	5.76
30	15.4	1.29	343	32.0	117.3	6.41

Through the production conditions of high esterification Daqu, the suitable parameters of production are selected and management operation is also developed:

- 1. The Daqu should be taken into the room quickly when the starter-making is completed because the Daqu heats up quickly and the heat dissipating capacity is big. In order to control temperature and prevent the moisture to evaporate from the surface, the length is increased by 2 cm and the total Daqu is decreased by 20 % than ordinary Daqu. When being taken into the room, the Daqu are arranged in two layers. The four sides of Daqu are enclosed well with straw and covered with a layer of thin straw on top, at the same time, choosing a room of ordinary Daqu for reference.
- 2. After 24 h, the temperature can be up to 40 °C, 12–15 h faster than ordinary Daqu. So after about 33 h the Daqu must be turned, and the layer increased by one. The surface of Daqu is covered with white or milky white fungus colony. The room temperature and humidity are strictly controlled to prevent moisture evaporating from the surface. When the temperature of the room is raised, it is cooled down. The other management methods are the same as with ordinary Daqu, where the layers are increased from three to four, the starter-making time is 28 days, 2 days shorter than ordinary Daqu, with the highest temperature controlled at 55 °C.

49.2.6 Physicochemical Indexes Compared with Esterification and Ordinary Daqu

We can see from Table 49.8 that compared with ordinary Daqu, the fermenting and esterifying powers are improved, and esterifying power is increased more than

474 X.-D. Wang et al.

Physicochemical indexes	Esterification of Daqu	Ordinary Daqu
Moisture (%)	16.0	15.8
Acidity	1.29	1.35
Diastatic power (mg glucose/g h)	349	720
Fermenting power (g/kg)	39.0	35.0
Esterifying power (mg/g 100 h)	118.3	67.5
Liquefaction power (g/g h)	6.47	6.35

Table 49.8 The physiochemical indexes comparison of esterification Daqu and ordinary Daqu

once. The esterification Daqu is covered with a milk-white covering significantly, the mycelium is more thickset and consistent, the cross-section is thin and yellow, the Daqu gives out a sweet aroma, and no spoilage or other flavors. The ordinary Daqu has little surface colony and clothing is not obvious, the section is slightly thick, the mycelium is white and red with yellow spots, and the Daqu has strong atmosphere with no spoilage or other taste.

49.2.7 Effect of Cultivation Time on Daqu Quality

We can see the following from Table 49.9:

(1) The water is dropped quickly after the Daqu out of the room, but a few months later the water is increased again.

physiochemical	indexes	comparison	ot	cultivation time
į	hysiochemical	hysiochemical indexes	hysiochemical indexes comparison	hysiochemical indexes comparison of

Index	Daqu	January	February	March	April	May	June	July
Moisture (%)	16.0	13.2	12.4	12.2	13.0	14.5	14.0	14.7
Acidity	1.29	1.30	1.42	1.32	1.38	1.40	1.40	1.28
Diastatic power (mg glucose/g h)	349	398	402	385	365	342	336	339
Fermenting power (g/kg)	39.0	40.0	39.0	38.0	35.0	35.0	33.0	31.0
Liquefaction power (g/g h)	6.47	6.67	6.52	6.47	6.42	6.40	6.00	5.32
Esterifying power (mg/g 100 h)	118.3	125.5	134.4	142.3	135.7	130.2	121.5	115.7
Total plate count (104/g)	897.5	944.9	1165	1055	984	871	800	610
Total number of yeast (104/g)	45.3	47.2	45.5	42.1	40.5	39.7	36.2	30.6
Total counts of molds (104/g)	146.7	192.1	201.8	208.1	155.2	125.7	118.4	131.5

- (2) After being stored for 2 months, the acidity reaches a maximum, then it is reduced gradually. When stored for 7 months, the acidity is equal to the Daqu out of room.
- (3) After being stored for 2 months, the liquefaction, fermenting, and diastatic powers reaches a maximum. Compared with the Daqu stored for 2 months, the liquefaction, fermenting, and diastatic powers all decrease by the Daqu stored for 7 months. That is to say, when Daqu is stored for 7 months, the index of saccharification and fermentation decreases.
- (4) In the first 3 months, the esterification power of Daqu rises from month to month, but from the fourth month, the esterification decreases every month; when the Daqu is stored at 7 months, esterification is decreased by 2.2 %.
- (5) From the results of microbial test, yeast, bacteria, molds increase month by month, then reduce gradually. That is, the microorganisms do not stop their growth, so management at this time is particularly important for all. When stored at 7 months, the total bacterial decreases by 32.03 %, yeast decreases by 32.45 %, and molds decrease by 10.36 %.

According to the results, the best storage period of esterification Daqu is 2–3 months.

49.3 Conclusions

The optimal conditions of esterification Daqu production is the ratio of raw materials at 6:3:1 (wheat: barley: pea), the inoculation of *Monascus* (*M. purpureus*) FBKL3.0018 is 10 g/kg, and the water of Daqu is controlled at 40 %. The highest temperature is more suitable which controlled at 55 °C, and the time of cultivation is always between 25 and 28 days. Fermenting, liquefaction, and esterifying powers are higher than the ordinary Daqu, and esterifying power increases more than once. The sensory quality of esterification Daqu is better than ordinary Daqu. During storage, the esterification power of Daqu rises from month to month, but from the fourth month, the esterification power decreases every month. Based on the analysis of other factors, the best storage period of esterification Daqu is 2–3 months.

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Chapter 50 Metabolomics Analysis Between Wild-Type and Industrial Strains of Streptomyces avermitilis Based on Gas Chromatography–Mass Spectrometry Strategy

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Abstract *Objective* To conduct a metabolic analysis of the intracellular metabolites in *Streptomyces avermitilis*, so as to search for possible biomarkers and to discover the mechanism of higher avermectin production. *Methods* GC–MS analysis was used to obtain the fingerprint of wild-type and industrial *S. avermitilis* 9-39 strains. *Results and conclusion* multivariate statistical analysis demonstrated that D-cellobiose, D-galactose, D-glucopyranose, D-mannose, D-turanose, glutamine, L-serine, and maltose were mainly responsible for distinguishing the wild-type *S. avermitilis* and industrial strain 9-39, and most of them belong to glycometabolism. Thus, strengthened glycometabolism is the reason for high yield of avermectin in *S. avermitilis*.

Keywords *Streptomyces avermitilis* \cdot Metabolomics \cdot Gas chromatography–mass spectrometry \cdot Multivariate

50.1 Introduction

Streptomyces avermitilis is a Gram-positive bacterium in the genus Streptomyces (family Streptomycetaceae, class Actinobacteria) [1, 2]. The major interest in this genus Streptomyces is the diversity of its production of secondary metabolites as an industrial microorganism. Streptomyces avermitilis carries out not only a complex morphological differentiation but also production of various secondary metabolites

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478 G. Guo et al.

[3, 4], one of which, avermectin, is used as a major commercial antiparasitic agent in the fields of animal health, agriculture, and human infections. Although much progress has been made, the detailed avermectin production mechanism by *S. avermitilis* is unclear up to now [4, 5].

Metabolomics aims to systematically analyze the entire small molecule metabolite composition and dynamic response in microorganisms [6–8]. Metabolomics analysis also assesses the last step in the series of changes under an external stress or a pathological insult [9, 10], and the changes of composition of metabolites directly reveals the phenotypic changes in a living system [11, 12]. By this way, we can have further insight into the difference of *S. avermitilis* industrial mutant 9-39 and its wild-type on microbial metabolism via the study of variation of intracellular metabolites, which is important for further understanding of higher production mechanism of avermectin.

In the present study, we applied a GC-MS-based metabolomics approach [13–15] and employed a multivariate analysis to explore the global metabolite profiles of industrial strain 9-39 and wild-type *S. avermitilis* at different fermentation time point. The aim is to reveal their difference in metabolomics, and the intermediate metabolites related with higher production, and find out the correlated pathway of avermectin in *S. avermitilis*.

50.2 Materials and Methods

50.2.1 Strains, Media, and Culture Conditions

A wild-type (WT) strain of *S. avermitilis* and one of its industrial mutants, strain 9-39, were offered by Microbiology Institute of CAS, China used for the metabolosome comparison in this study. All strains were maintained as spore stocks on agar slant at 4 °C. About 1 cm² of frozen spore culture was scalped from the agar surface and transferred into 250-mL flasks containing 40-mL seed medium [16] and incubated at 28 °C on a rotary shaker at 220 r/min for 40 h. Fermentation medium [16] were then inoculated with 5 % seed cultures and incubated on a rotary shaker at 220 r/min at 28 °C for 10 days.

50.2.2 Preparation of Metabolome Samples

Cell samples were, respectively, collected on the 6th, 8th, and the 10th day's cultivation for both 9-39 and WT. Metabolome samples were quenched and extracted in 4 replicates for each sample as described below. Cell suspensions were immediately quenched at -40 °C for 5 min with 40 % (v/v) methanol in water. The quenching solution was prechilled at -80 °C and used at a ratio of 4:1 (v/v) of quenching solution to cell suspension. Then cells were harvested by centrifugation

at $8,000 \times g$ for 5 min with 4 °C and washed twice with 0.6 % NaCl to remove the residual culture medium [17]. Then the cells (100 mg) were disrupted with bead mill by 3 cycles for 48 s each time which contain 1 mL of extraction buffer (chloroform: ethanol: water = 2:2:1, v/v/v). The mixture was frozen in liquid nitrogen for 2 min and then was thawed, this freeze-thaw process was repeated five times. After centrifugation at $1,000 \times g$ for 5 min, 200 μ L supernatant was mixed with 10 μ L of internal standard solution and dried in a vacuum centrifuge dryer.

50.2.3 Sample Derivatization

Before GC–MS analysis, a two-stage chemical derivatization was carried out. First, methoximation was performed by dissolving the dried metabolite samples in 50 μL methoxamine hydrochloride (20 mg/mL in pyridine, Sigma), and the solution was incubated at 40 °C for 80 min. Then, 80 μL *N*-methyl-*N*-(trimethylsilyl)trifluoro-acetamide (MSTFA, Sigma) were added to the samples, which were subsequently incubated at 40 °C for 80 min for trimethylsilylation.

50.2.4 GC-MS Analysis

Gas chromatography analysis was performed on a GC-MS system (Agilent Technologies, Palo Alto, CA, USA) equipped with an HP-5 capillary column (60 m × 320 μm i.d., 0.25 μm film thickness; Agilent J&W Scientific, Folsom, CA, USA). Samples (1 µL each) were injected into HP-5 capillary column by split injection mode with a split ratio of 8:1 using an autoinjector. Helium was used as the carrier gas at a constant flow of 1 mL/min. The injection, ion source, and ion source surface temperatures were set to 280, 250 and 280 °C, respectively. The GC oven temperature was set to 70 °C for 2 min, then raised to 290 °C at a rate of 5 °C/min, and maintained at 290 °C for 3 min. The electron impact ionization (70 eV) was set at a full scan mode (m/z 50-800). MSD Productivity ChemStation software (version E.02 01.1177, Agilent Technologies) was used to acquire mass spectrometric data. Compound identification was performed by comparing the mass spectra with a commercially available standard library, the National Institute of Standards and Technology mass spectral library 2008. The compounds were also identified by comparing their mass spectra and retention times with those of commercially available reference compounds.

50.2.5 Data Analysis

The relative intensity of each metabolite peak was expressed as the ratio of its peak area to that of the internal standard on the same chromatograph. The generated

480 G. Guo et al.

normalized peak areas (variables) were imported into SIMCA package (ver. 11.5) (Umetrics, Umea, Sweden) for the multivariate statistical analysis. Principal component analysis (PCA) and partial least-squares-discriminant analysis (PLS-DA) were applied to analysis the data after mean-centering and UV-scaling.

50.3 Results and Discussion

50.3.1 Typical GC-MS TIC Chromatograms of 9-39 and WT

The wild-type and industrial strains of *S. avermitilis* were cultured in 250-mL flasks with 5 replicates. After extraction and derivatization, the typical GC–MS TIC chromatograms obtained by GC–MS analysis system were shown in Fig. 50.1.

50.3.2 Metabolites Analyzed by PCA and PLS-DA

The generated normalized peak areas were imported into SIMCA for the multivariate statistical analysis. Firstly, PCA was used to investigate the metabolic differences between 9-39 and WT at different sampling time point, and the PC1/PC2 scores plots ($R^2X_{\text{cum}} = 0.977$, $Q_{\text{cum}}^2 = 0.918$) were performed to represent the sample distribution in the new multivariate space (Fig. 50.2a). Distinct clustering was observed among WT at each time point and 9-39 on the 6th day. There was no clear difference between 9-39 on the 8th day and on the 10th day. To confirm the

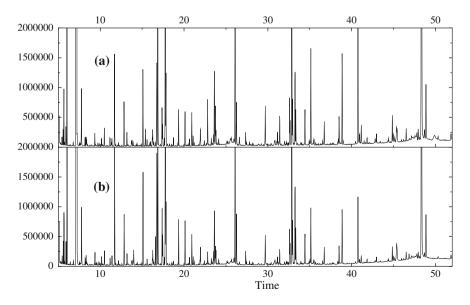


Fig. 50.1 Typical GC-MS TIC chromatograms of 9-39 (a) and WT (b)

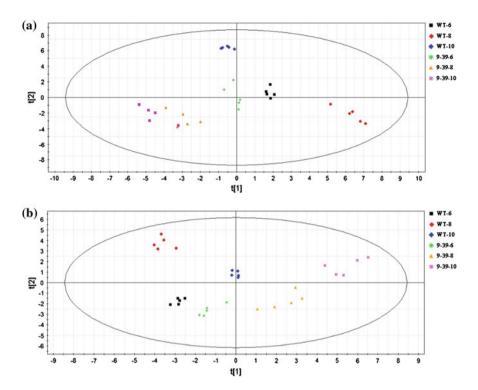


Fig. 50.2 Scores plot for 9-39 and WT. **a** PCA-derived metabolites profiles and **b** PLS-DA-derived metabolite profiles. In the scores plot, the confidence interval is defined by the Hotellin's T2 ellipse (95 % confidence interval), and observations outside the confidence ellipse are considered outliers

difference of every sampling points in metabolite variations, the class separation was further optimized using the PLS-DA model ($R^2X_{\text{cum}} = 0.965$, $Q_{\text{cum}}^2 = 0.946$), and good discrepancy was observed (Fig. 50.2b).

To further verify the differences, multivariate data analysis was also performed by HCA. The HCA plot of the 52 differential metabolites reflected a clustering pattern that was similar to the result of the PLS-DA middle, and good discriminability was observed (Fig. 50.3).

50.3.3 Changes in Intracellular Metabolites Between 9-39 and WT at Different Time Points

The intracellular metabolites of 9-39 and data were analyzed with PLS model at different culture time. As shown in scores plots (Fig. 50.4a, c, e), good discrepancy was observed. It indicated that 9-39 were metabolically distinct from WT in all

482 G. Guo et al.

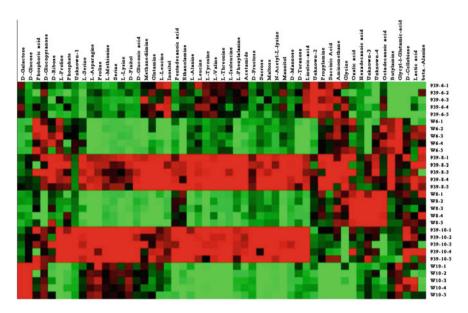


Fig. 50.3 Hierarchical cluster analysis of the 52 identified differential metabolites

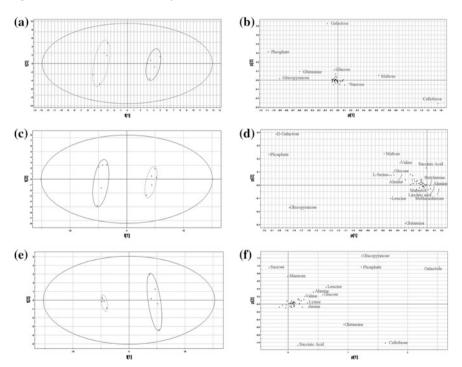


Fig. 50.4 Scores plot and loading plot generated by PLS: **a**, **b** sampling time on the 6th day $(R^2X_{\text{cum}}=0.980, Q_{\text{cum}}^2=0.949)$; **c**, **d** sampling time at 8th day $(R^2X_{\text{cum}}=0.984, Q_{\text{cum}}^2=0.957)$; **e**, **f** sampling time on the 10th day $(R^2X_{\text{cum}}=0.986, Q_{\text{cum}}^2=0.981)$

those days. The major metabolic perturbations that caused these discriminations were identified from the PLS loading plot (Fig. 50.4b, d, f). In the loading plots, points far away from the cross central means these metabolites have a larger contribution. Also, the points in loading plots have positive effect on the same quadrant in scores plots.

In the PLS models, the VIP coefficient reflects the contribution of each metabolite, the higher VIP value means the metabolite has a larger contribution. A metabolite with a VIP value greater than 1 demonstrates a significant contribution to the separation of groups within PLS model. When *S. avermitilis* fermentation was conducted for 6 days, D-cellobiose, phosphate, D-glucopyranose, maltose, glutamine, succinic acid, D-turanose, and D-galactose were responsible for the differentiation between 9-39 and WT (Fig. 50.5a). On the 8th day, the difference of intracellular metabolites between WT and 9-39 were from phosphate, D-galactose, D-glucopyranose, D-turanose, maltose, and L-serine (Fig. 50.5b). On the 10th day, the difference came from D-turanose, D-galactose, D-glucopyranose, D-mannose, and phosphate (Fig. 50.5c).

On the whole, phosphate, D-galactose, D-glucopyranose, and D-mannose were involved in variable importance of the projection plots of three sampling time

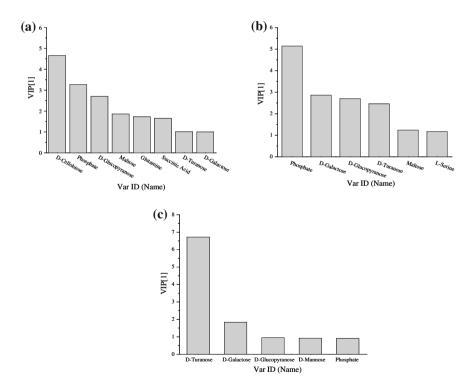


Fig. 50.5 Variable importance of the projection plots for the intracellular metabolites along component 1. **a** WT versus 9-39 on the 6th day using PLS-DA model, **b** WT versus 9-39 on the 6th day using PLS-DA model, and **c** WT versus 9-39 on the 6th day using PLS-DA model

484 G. Guo et al.

points. By comparison of PLS-DA scores plots and loading plots, phosphate expresses higher level in WT than 9-39 on the 6th day. While on the 8th day and the 10th day, 9-39 got higher phosphate level than WT. High levels of phosphate represent vigorous intracellular energy metabolism. Avermectin synthesis process requires a lot of energy and reducing power. In other words, the intracellular energy metabolism was enhanced in high-yielding strain, while the avermectin production reached the highest. The result coincides with the fact, so metabolomics strategy can be used to research into the mechanism of high avermectin yield in *S. avermitilis*. Compared VIP of three sampling time points, these marker substances (VIP > 1) mainly focused on the glycometabolism, and 9-39 got higher level than WT. As macrolide antibiotics, avermectins production requires a large amount of carbon source. Research has shown that all carbon skeleton synthesis of avermectin from sugar [18].

50.4 Conclusion

In this study, the variation of metabolites in wild-type and industrial strains of *S. avermitilis* were investigated. Over 200 intracellular metabolites were detected and a total of 52 among them were identified by GC–MS at different culture time points. The difference between 9-39 and WT were clearly separated by PCA or PLS methods in the scores plot. Furthermore, it was found that D-cellobiose, D-galactose, D-glucopyranose, D-mannose, D-turanose, glutamine, L-serine and maltose were mainly responsible for distinguishing 9-39 and WT, and most of which belong to glycometabolism. It could be inferred the enhancement of glycometabolism is the reason of high yield of avermectin in *S. avermitilis*. Overall, this work is conducive to shed light on the avermectin high-yielding mechanism, and provides a rational strategy of metabolic engineering and synthetic biotechnology for improving avermectin production.

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Chapter 51 Application of Orthogonal Design to Optimize Fermentation Conditions of *Bacillus amyloliquefaciens* BI₂

Yajun Wang, Zhanglei Cao, Miao Yu and Depei Wang

Abstract Bacillus amyloliquefaciens BI₂, isolated from corn straw silage, exhibits to be antagonistic against many kinds of molds, especially Aspergillus flavus. In this work, optimization of the medium and culture conditions was studied by adopting the combining single-factor tests and an orthogonal experiment. On the study of the medium optimization, yeast powder, monosodium glutamate, sodium nitrate, and the proportion of starch and maltose were selected as four factors at three levels each. On the study of the optimizing culture conditions, medium volume, culture temperature, fermentation time, and inoculums size were used as four factors at three levels each. The results demonstrated that yeast powder and monosodium glutamate have significant effects on medium optimization; medium volume, culture temperature, and fermentation time give crucial influences on culture conditions, and the influences of the other factors are not remarkable. The optimized medium results shown by the interaction tests were as following: yeast powder 1.5 %, monosodium glutamate 2 %, sodium nitrate 0.8 %, starch 0.17 %, and maltose 0.33 %; the optimal culture conditions were displayed as following: medium volume 60 mL, culture temperature 30 °C, inoculums size 2 %, and fermentation time 36 h.

Keywords Bacillus amyloliquefaciens BI_2 · Orthogonal experiment · Optimal medium · Optimal culture conditions

51.1 Introduction

Widely distributed in nature, *Aspergillus flavus* is one of the common food spoilage organisms and often appears in the moldy grains such as corn, peanuts, seeds, and nuts. Aflatoxins, the metabolite of *A. flavus*, can induce gene mutation, suppress the

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488 Y. Wang et al.

immune system, and are highly toxic and carcinogenic to human beings and animals [1, 2]. Therefore, there are long-term economic benefits in developing biological control methods which can decrease the toxicity and inhibit the growth of pathogen. Recent research has shown that certain *Bacillus* sp. can inhibit *A. flavus* and its toxin [3].

The antifungal mechanism includes several aspects and the substances mainly inhibit the growth of the pathogenic fungus. *Bacillus* sp., which can produce a variety of metabolites with antifungal activity, including amino acids, peptides, grease peptide, phospholipids, polyene, nucleic acid, have a good inhibitory effect for many plant and animal pathogens [4]. Thus, they are commonly used for biological controlling of crop diseases and expected to be applied in the field of food antiseptics.

Bacillus amyloliquefaciens BI₂ is a beneficial bacterium with strong environmental adaptability and resistance because of its tolerance spores. Its secondary metabolites can also antagonize a variety of pathogens, especially aflatoxin-producing A. flavus [5]. In this research, we investigated the optimal medium and culture conditions by adopting the method of combining the single-factor experiment and the orthogonal design for improving its antifungal effect.

51.2 Materials and Methods

51.2.1 Microorganisms

Bacillus amyloliquefaciens BI₂ was isolated from silage feed by our previous work, and preserved by China General Microbiological Culture Collection Center, preservation number CGMCC No. 3413. Aspergillus flavus was preserved in the Laboratory of Industrial Fermentation Microbiology in College of Biotechnology, Tianjin University of Science & Technology.

51.2.2 Media and Culture Conditions

Strain BI₂ was cultured in nutrient broth (pH 7.2) containing 0.5 % beef extract, 1 % peptone, 0.5 % NaCl, and in addition 2 % agar for solid medium, sterilized at 1.0×10^5 Pa for 20 min. The components of the initial fermentation medium were 0.5 % beef extract, 0.5 % soluble starch, 0.5 % NaCl, 1.2 % NaNO₃, 1.2 % corn pulp, 0.06 % Tween 80, and 2 % sodium glutamate (pH 7.5), sterilized at 1.0×10^5 Pa for 20 min. Potato Dextrose Agar: weigh 200 g potato, dice them, add 1 L distilled water and boil for 30 min, six layers of cotton gauze filter, add the filtrate water to 1,000 mL, add 20 g glucose, pH natural, 1.0×10^5 Pa sterilized for 20 min.

51.2.3 Single-Factor Experiments for Medium Optimization

To determine the optimal factors for the following orthogonal experiment, a series of single experiments were designed. The optional carbon sources were glucose (1 %), sucrose (0.5 %), maltose (0.5 %), starch (0.5 %), and dextrin (0.5 %) to replace starch in the initial fermentation medium. The optional nitrogen sources were ammonium sulfate, ammonium nitrate, sodium nitrate, peptone, corn steep liquor, yeast extract, and beef extract. 1.5 % of sodium nitrate and corn steep liquor were added in the initial fermentation medium to replace beef extract respectively (Sect. 51.2.4).

The method of comparing the inhibition rates of *A. flavus* spore germination was used. First, the concentration of *A. flavus* spore suspension was made to be 5×10^4 cfu/mL. The centrifuged fermentation broth and the unset PDA medium were well mixed in 1:6 proportion, and poured onto the plate, coat 100 μ L of the spore suspension on the medium in plate after solidification, with the blank PDA medium as the control. The colonies of *A. flavus* were reckoned after being cultured at 30 °C for 36 h, and the inhibition rates were calculated. Each group has three parallels.

51.2.4 Method of Calculating Antifungal Rate [6]

$$\alpha(\%) = \frac{N_C - N_E}{N_C} \times 100$$

where α represents antifungal rate, N_C represents the colony number of control group, and N_E represents the colony number of experimental group.

51.2.5 Orthogonal Design of Medium Optimization

For the interaction tests, a group of orthogonal tests on yeast extract (0.5, 1, 1.5%), sodium glutamate (1, 1.5, 2), NaNO₃ (0.4, 0.8, 1.2%), and the ratio of starch and maltose (1:2, 1:1, 2:1) were designed according to the results of single-factor experiments (Table 51.1). In addition, the orthogonal experiment needs to consider three nitrogen sources and the interaction between any two factors. So, 27 experiments were designed according to the orthogonal array (Table 51.2). All the experiments were carried out in triplicate.

490 Y. Wang et al.

Table 51.1 Orthogonal array of medium optimization

Factor	Level		
	1	2	3
A	0.5	1	1.5
В	1	1.5	2
С	0.4	0.8	1.2
D	1:2	1:1	2:1

Note A, B, and C represent yeast extract, sodium glutamate, and NaNO₃, respectively. The data of A, B, and C represents the percentage of the volume of medium, the data of D represents the ratio of starch and maltose content

Table 51.2 The L₉(3⁴) orthogonal test was used for culture condition

Factor	Level		
	1	2	3
A'	50	60	70
B'	30	33	37
C'	1	2	3
D'	32	36	40

Note A' medium volume; B' culture temperature; C' inoculums size: D' fermentation time. Each factor has three levels

51.2.6 Orthogonal Experiments of Culture Conditions

For the combinational effect trial, an $L_9(3^4)$ orthogonal test [7, 8] on medium volume (50, 60, and 70 mL), culture temperature (30, 33, 37 °C), inoculums size (1, 2, 3%), and fermentation time (32, 36, 40 h) was designed according to the results of the single-factor experiments (Table 51.2). All the experiments were carried out in triplicate.

51.3 Results and Discussion

51.3.1 The Results of Single-Factor Experiments for Medium Optimization

From the results of the single-factor experiments for carbon sources in Fig. 51.1, maltose and starch among the tested factors were selected as the optimal carbon sources for fermentation. The results of adding sodium nitrate, yeast extract, and beef extract showed high inhibition rates and there is no significant difference between any two factors. But the ingredient beef exact is instable; accordingly, sodium nitrate and yeast extract are selected to optimize the combination of them in the orthogonal tests (Fig. 51.2).

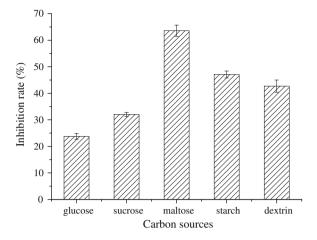


Fig. 51.1 The single-factor experiments of carbon sources for medium optimization

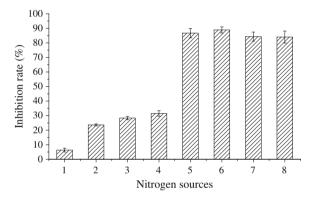


Fig. 51.2 The single-factor experiments of nitrogen sources for medium optimization. No. 1, 2, 3, 4, 5, 6, 7, and 8 represent ammonium sulfate, ammonium nitrate, corn steep liquor, peptone, sodium nitrate, yeast extract, beef extract, the initial nitrogen sources, respectively

51.3.2 The Experimental Data of Orthogonal Array for Medium Optimization

According to the components of the medium, the orthogonal array was designed in Table 51.3. Six columns were picked to show the interactions of the three nitrogen sources: yeast extract, sodium glutamate, and sodium nitrate. Three blank columns were added to reduce the error of orthogonal test. Thus, specific operation in experiments did not involve the interaction columns and blank columns.

By the analysis of SPSS software [9–11], the optimized medium was obtained. As indicated in Table 51.4, A, B and the interaction between A and B had

Table 51.3 The results of medium optimization by orthogonal experiment including interaction

No.	Factor													Inhibition rate (%)
	A	В	$A \times B$	$A \times B$	C	$A \times C$	$A \times C$	$\mathbf{B} \times \mathbf{C}$	D		B × C			
1	1	1	1	1	1	1	1	1	1	1	1	1	1	75.67
2	1	1	1	1	2	2	2	2	2	2	2	2	2	77.94
3	1	1	1	1	3	3	3	3	3	3	3	3	3	76.91
4		2	2	2	1	1		2	2	2	3	3	3	81.65
5	1	2	2	2	2	2	2	3	3	3	1	1	1	75.67
9	1	2	2	2	3	3	3	1	1	1	2	2	2	74.64
7	1	3	3	3	1	1	1	3	3	3	2	2	2	76.70
8	1	3	3	3	2	2	2	1	1	1	3	3	3	70.10
6	1	3	3	3	3	3	3	2	2	2	1	1	1	71.75
10	2	1	2	3	1	2	3	1	2	3	1	2	3	73.81
111	2	1	2	3	2	3	1	2	3	1	2	3	1	73.40
12	2	1	2	3	3	1	2	3	1	2	3	1	2	66.39
13	2	2	3	1	1	2	3	2	3	1	3	1	2	73.81
14	2	2	3	1	2	3	1	3	1	2	1	2	3	76.29
15	2	2	3	1	3	1	2	1	2	3	2	3	1	81.65
16	2	3	1	2	1	2	3	3	1	2	2	3	1	85.77
17	2	3	1	2	2	3	1	1	2	3	3	1	2	84.54
18	2	3	1	2	3	1	2	2	3	-	1	2	3	86.39
19	3	1	3	2	1	3	2	1	3	2	1	3	2	78.97
20	3	1	3	2	2	1	3	2	1	3	2	-	3	79.79

(continued)

No.	Factor													Inhibition rate (%)
	А	В	$A \times B$	$A \times B$	၁	$A \times C$	$A \times C$ $A \times C$	$\mathbf{B} \times \mathbf{C}$	D		$\mathbf{B} \times \mathbf{C}$			
21	3	_	3	2	3	2		3	2		3	2		80.82
22	3	2		3	1	3	2	2	1	3	3	2	1	88.45
23	3	2		3	2	1	3	3	2			3	2	85.36
24	3	2		3	ж	2			3	2	2	1	3	99:88
25	3	3	2			3	2	3	7		2	1	3	92.16
26	3	3	2		2	1	3		3	2	3	2	1	87.22
27	3	3	2		3	2		2	1	3		3	2	89.28

494 Y. Wang et al.

Source	Type III sum of squares	df	Mean square	F	Sig.
Calibration model	1090.211 a	20	54.511	6.852	0.012
Intercept	171807.828	1	171807.828	21597.374	0.000
A	488.835	2	244.417	244.417	0.001
В	212.746	2	106.373	13.372	0.006
С	15.802	2	7.901	0.993	0.424
D	30.167	2	15.084	1.896	0.230
$A \times B$	291.991	4	72.998	9.176	0.010
$A \times C$	19.386	4	4.847	0.609	0.671
$B \times C$	31.283	4	7.821	0.983	0.482
Error	47.730	6	7.955		
Total	172945.769	27			
Total correction	1137.941	26			

Table 51.4 The results of the orthogonal experiment analyzed by SPSS

significant effects on the inhibition rates, that is to say, yeast powder, monosodium glutamate, and the interaction between them had significant influences on the experiment (P < 0.05). Meanwhile, the results of Level 3 of yeast exact had more significant difference comparing with the results of Level 1 and Level 2. Therefore, 1.5 % yeast powder was selected to add into the medium. In a similar way, Level 3 of sodium glutamate and Level 2 of sodium nitrate were chosen to be the optimal levels. Considering the interaction of A and B, the combination A_3B_3 test had the highest inhibition rates. The results of factors C and D did not show significant differences. The level 2 of them was chosen to be the optimal one by comparing the inhibition rates. The final medium formula was $A_3B_3C_2D_1$ combination: yeast powder 1.5 %, monosodium glutamate 2 %, sodium nitrate 0.8 %, the ratio of starch to maltose 1/2, i.e., starch 0.17 %, maltose 0.33 %, and the remaining was Tween-80 0.06 % and NaCl 0.5 %.

51.3.3 The Experimental Data of Orthogonal Array for Culture Conditions Optimization

According to Table 51.5 for optimizing experiment of culture conditions, the analysis was using SPSS in Table 51.6.

From Table 51.6, medium volume, temperature, and fermentation time had remarkable influences on the results of culture conditions, but inoculums size had no significant impact on it. The optimal culture conditions were obtained as $A_2B_1C_2D_2$ combination: medium volume 60 mL, culture temperature 30 °C, inoculums size 2 %, and fermentation time 36 h.

No.	A'	B'	C'	D'	Inhibition rate (%)
1	1	1	1	1	88.39
2	1	2	2	2	89.06
3	1	3	3	3	84.15
4	2	1	2	3	90.18
5	2	2	3	1	89.73
6	2	3	1	2	89.51
7	3	1	3	2	91.07
8	3	2	1	3	87.50
9	3	3	2	1	86.16

Table 51.5 The result of the $L_9(3^4)$ orthogonal test for the culture conditions optimization

Table 51.6 The analysis of the culture conditions optimization by SPSS

Source	Type III sum of squares	df	Mean square	F	Sig.
Calibration model	116.534 a	8	14.567	53.707	0.000
Intercept	211589.357	1	211589.357	780122.165	0.000
A'	33.770	2	16.885	62.254	0.000
B'	46.726	2	23.363	86.138	0.000
C'	0.911	2	0.456	1.680	0.214
D'	35.127	2	17.563	64.756	0.000
Error	4.882	18	0.271		
Total	211710.772	27			
Total correction	121.416	26			

51.4 Conclusions

From the tests and analyses above, the optimal medium formula is determined as: yeast powder 1.5 %, monosodium glutamate 2 %, sodium nitrate 0.8 %, starch 0.17 %, maltose 0.33 %; medium volume 60 mL, and culture temperature 30 °C. Also, inoculums size 2 % and fermentation time 36 h are the optimal culture conditions. According to the optimal medium and culture conditions for the experiments, the inhibition rate of strain BI_2 reaches 92 % in line with the experimental predicted results.

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496 Y. Wang et al.

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Chapter 52 Optimization of Fermentation Medium for Citric Acid Production by Aspergillus niger

Jianhua Zhang, Kun Li, Juan Huang and Depei Wang

Abstract A statistically based method was used to optimize the fermentation medium for enhancing citric acid production by *Aspergillus niger* TCCC 41661. The optimization of single-factor experiments (soy peptone, (NH₄)₂SO₄, dextrin) and an orthogonal experiment were applied to determine the optimal concentration of each significant variable. The optimum values for the critical components were obtained as follows: 0.25 g/L soy peptone, 12.0 g/L dextrin, 2.0 g/L KH₂PO₄, 2.0 g/L (NH₄) SO₄, 0.1 % Tween 80. Under this condition, the citric acid production of 16.11 g/L was obtained.

Keywords Aspergillus niger • Citric acid • Morphology • Medium optimization • Orthogonal experiment

52.1 Introduction

Citric acid is widely used as pharmaceutical additive, cosmetic and detergent additive [1]. Citric acid is an important commercial product of fermentation industry, its annual production is over 700,000 tons [2]. The study found that there was too much protein in coarse corn flour [3], filamentous fungi will overgrow if it takes coarse corn flour as raw material. There are unique advantages in using the corn liquefied clarifying solution as raw material for its low content of fat and protein, it helps to improve oxygen level and the conversion rate, shorten the

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⁴⁹⁷

498 J. Zhang et al.

fermentation period [4]. In this paper, factors affecting the production of citric acid and mycelium pellet shape of *Aspergillus niger* prefermentation were studied using the corn liquefied clarifying solution as raw material and *A. niger* TCCC 41661 as a research strain [5]. It laid a foundation for further optimization of citric acid production process.

52.2 Materials and Methods

52.2.1 Material and Equipment

A high-yield strain (*A. niger* TCCC 41661) is kept at the Microorganism Preservation Center in Tianjin University of Science and Technology. The experimental equipment consists of professional digital microscope OLYMPUS DP7 (Olympus (China) Co., Ltd.), table model high-speed centrifuge (Hunan Xiang Yi Laboratory Instrument Development Co., Ltd.), the citric acid test kit was purchased from Boehringer Mannheim Company, FRG.

The initial liquid seed medium: 10 % corn liquefied serum, 0.1 % soy peptone, 0.15 % (NH₄)₂SO₄, 0.5 % dextrin, 0.270 g/L KH₂PO₄, 0.065 g/L MgSO₄·7H₂O, 0.0792 g/L MnCl₂·4H₂O, 0.016 g/L CoCl·6H₂O, 0.0647 g/L ZnSO₄·7H₂O.

Culture conditions: the optimal fermentation conditions were: 36.5 °C, 30 mL/ 250 mL medium volume, 330 rpm for 24 h.

52.2.2 Analytical Methods

Determination of citric acid: approximately 20 mL fermentation broth at different time points was taken in a 50 mL Eppendorf tube and centrifuged at 10,000×g and 4 °C for 5 min. The supernatant was removed in a 10-mL Eppendorf tube was placed in a water bath at 80 °C for 10 min to stop the possible enzymatic reactions. Then the supernatant was centrifuged again for the assay. Citric acid was measured enzymatically according to the manufacturer's instruction (test kit Cat. No. 10139076035). All the assays in this study were performed in triplicate.

52.2.3 The Effect of Nitrogen on Citric Acid Fermentation

Nitrogen plays an important role in cellular material synthesis and metabolism regulation. To determine the optimal nitrogen content, a series of experiments were performed using a range of values. The effect of the nitrogen was determined by varying the amounts of soy peptone.

52.2.4 The Effect of Carbon on Citric Acid Fermentation

The carbon source for the citric acid fermentation has been the focus of much study. To determine the effect of carbon, different amounts of dextrin were added to the fermentation medium. Single-factor experiments were carried out to determine the optimal carbon content.

52.2.5 Orthogonal Experiment

The best ratio of dextrin, KH_2PO_4 , soy peptone, $(NH_4)_2SO_4$ was elected in L_9 . (3⁴) Orthogonal experiment.

52.2.6 The Effect of Tween 80 on Citric Acid Fermentation

Tween 80, an important surfactant has a crucial impact on the spread of mycelium pellets. There are conflicting reports about the importance and influence of morphology. For example, it has been reported that citric acid titer was higher when the morphology was filamentous and dispersed [6, 7]. To determine the optimum value, a series of experiments were performed using a range of values.

52.2.7 The Effect of Inoculum Size on Citric Acid Fermentation

The effect of inoculum size on citrate production was determined by varying the number of spores added as inoculum to fermentation broth [8].

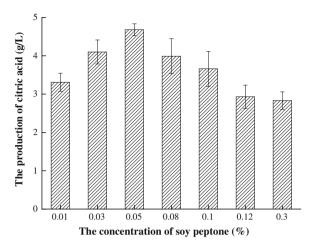
52.3 Results and Discussion

52.3.1 Effect of Nitrogen on Citric Acid Fermentation

The roles of nitrogen are synthesis of cellular material and regulate metabolism [9]. From the perspective of growth, *A. niger* can take advantage of inorganic nitrogen and organic nitrogen [10]. Soy peptone is easy absorbed and used by microorganism; the results (Fig. 52.1) showed an increase in citrate production with the increase of soy peptone concentration. The optimized soy peptone was 0.05 %, when the citric acid production was maximum and mycelium pellet shape was

500 J. Zhang et al.

Fig. 52.1 The influence of soy peptone on citric acid fermentation



relatively homogeneous. The citric acid production decreased with the continuous increase of soy peptone concentration. Mycelium pellets (Fig. 52.2) were larger and more irregular at the soy peptone concentration of 0.01–0.03 % and 0.08–0.30 % compared to that at 0.05 %. At the soy peptone concentration of 0.01–0.03 %, the citric acid production decreased continually, because the nitrogen content was too low for bacterial growth and development. At the soy peptone concentration of 0.08–0.30 %, the rich nitrogen content accelerated the growth of filamentous fungi, which was not conducive to the accumulation of citric acid, although it was advantageous for shortening the fermentation period.

 $(NH_4)_2SO_4$ as inorganic nitrogen directly affects citric acid fermentation. As showed in Fig. 52.3, with the increase of $(NH_4)_2SO_4$, citric acid yields gradually increased and citric acid yield was maximum at 0.24 % $(NH_4)_2SO_4$. Figure 52.4 showed mycelium pellet was large and irregular at the $(NH_4)_2SO_4$ concentrations of 0.16 and 0.18 % compared to that at 0.24 %. At the $(NH_4)_2SO_4$ concentration of 0.24 %, bacteria pellet were uniform and round which was conducive to the citric acid fermentation.

52.3.2 Effect of Dextrin on Citric Acid Fermentation

Carbon nitrogen ratio is a key factor in the citric acid fermentation and directly affects the fermentation period, the rate of acid production. If carbon nitrogen ratio is high, it will cause poor cell growth and slow cell metabolism, if carbon nitrogen ratio is too low, also leads to a large and loose state of mycelium pellet, and low yield of citric acid. Carbon nitrogen ratio can be adjusted, though adding different amounts of dextrin. The carbon source for the citric acid fermentation has been the

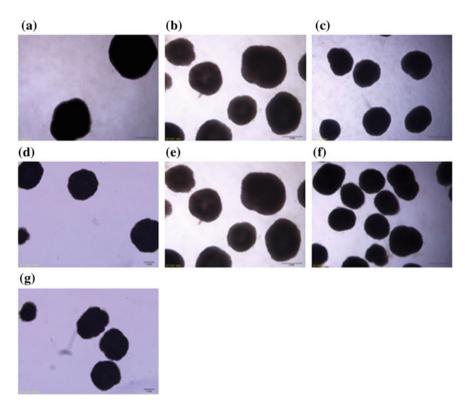
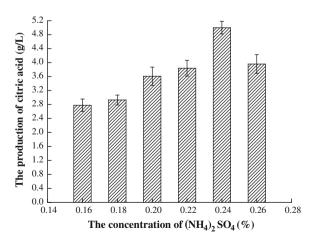


Fig. 52.2 The mycelium pellet morphology of Aspergillus niger in fermentation 24 h. a 0.01~% soy peptone, b 0.03~% soy peptone, c 0.05~% soy peptone, d 0.08~% soy peptone, e 0.10~% soy peptone, f 0.12~% soy peptone, g 0.30~% soy peptone. Magnification $\times 10$, with the edges of the images being 1.0~mm

Fig. 52.3 The influence of different concentration of $(NH_4)_2SO_4$ on citric acid fermentation



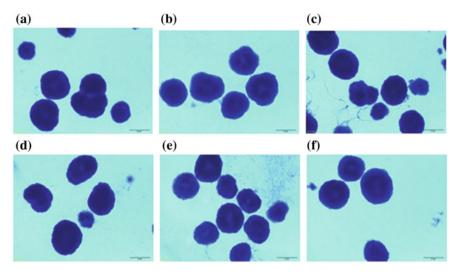
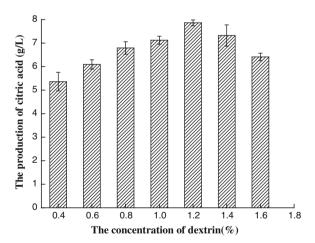


Fig. 52.4 The mycelium pellet morphology of *A. niger* infermentation 24 h. **a** 0.16% (NH₄)₂SO₄, **b** 0.18% (NH₄)₂SO₄, **c** 0.20% (NH₄)₂SO₄, **d** 0.22% (NH₄)₂SO₄, **e** 0.24% (NH₄)₂SO₄, **f** 0.26% (NH₄)₂SO₄. Magnification ×10, with the edges of the images being 2.0 mm

Fig. 52.5 The influence of dextrin on citric acid fermentation



focus of much study [11]. In general, dextrin is rapidly taken up by *A. niger* [12]. Figure 52.5 shows that with the increase of dextrin concentration citric acid production gradually increased, acid production was maximum at 1.2 %. As showed in Fig. 52.6, mycelium pellets were uniform, dense, and had less broken mycelium. As dextrin increased continually, citric acid production declined.

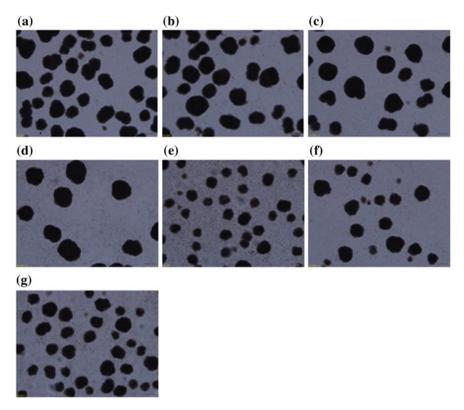


Fig. 52.6 The mycelium pellet morphology of *A. niger* infermentation 24 h. **a** 0.40 % dextrin, **b** 0.60 % dextrin, **c** 0.80 % dextrin, **d** 1.0 % dextrin, **e** 1.2 % dextrin, **f** 1.4 % dextrin, **g** 1.6 % dextrin. Magnification $\times 10$, with the edges of the images being 100 μ m

52.3.3 Orthogonal Experiment

 L_9 (3⁴) orthogonal experiment was used to screen the best ratio of dextrin, KH_2PO_4 , soy peptone, $(NH_4)_2SO_4$. The results were shown in Table 52.1.

As showed in Table 52.1, in different proportions of four media components, the combination with the maximum production of citric acid was: dextrin 12 g/L, KH₂PO₄ 2.0 g/L, soy peptone 0.25 g/L, (NH₄)₂SO₄ 2.0 g/L. The degrees of influence of four factors were showed though range analysis: (NH₄)₂SO₄ > KH₂PO₄ > Soy peptone > Dextrin.

The theoretically optimized medium was coincident with medium of the ninth group in orthogonal table. According to verification experiment, production of citric acid was 10.59 g/L.

504 J. Zhang et al.

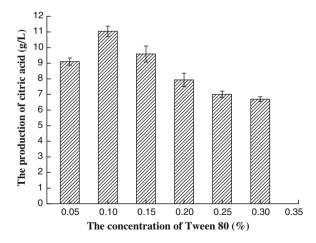
Experiment	Dextrin	KH ₂ PO ₄	Soy peptone	$(NH_4)_2SO_4$	Citric acid
no.	(g/L)	(g/L)	(g/L)	(g/L)	(g/L)
1	5.0	1.0	0.10	2.0	6.50
2	5.0	1.5	0.25	2.5	5.77
3	5.0	2.0	0.50	3.0	4.84
4	8.0	1.0	0.25	3.0	4.69
5	8.0	1.5	0.50	2.0	6.85
6	8.0	2.0	0.10	2.5	5.82
7	12.0	1.0	0.50	2.5	4.95
8	12.0	1.5	0.10	3.0	4.80
9	12.0	2.0	0.25	2.0	8.61
K_1	0.5703	0.5380	0.5707	0.7320	
K ₂	0.5787	0.5807	0.6357	0.5513	
K ₃	0.6120	0.6423	0.5547	0.4777	
R	0.0417	0.01043	0.0810	0.2543	

Table 52.1 Results and variance analysis of orthogonal experiment of medium components

52.3.4 Effect of Tween 80 on Citric Acid Fermentation

As shown in Fig. 52.7, the optimized concentration of Tween 80 was 0.10 %. With the continual increase of Tween 80, citric acid production gradually decreased. Tween 80 had damaging effects on the cell membrane. Too much Tween 80 will inhibit the growth of filamentous fungi, which leads to low citric acid production. From Fig. 52.8, the Tween 80 concentration of 0.10 %, the mycelium pellets were relatively homogeneous and dense, the hyphae was stubby, which was consistent with normal morphology of *A. niger* strains [13]. With the increase of Tween 80 concentration, the large and irregular mycelium pellets were disadvantageous to the production of acid.

Fig. 52.7 The influence of Tween 80 on citric acid fermentation



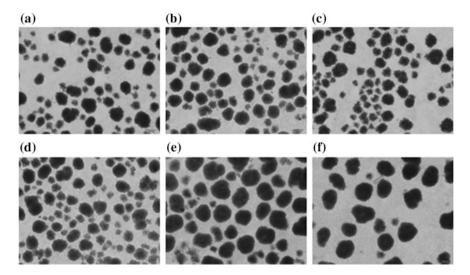
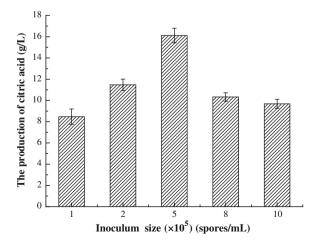


Fig. 52.8 The mycelium pellet morphology of *A. niger* infermentation 24 h. **a** 0.050 % Tween 80, **b** 0.10 % Tween 80, **c** 0.15 % Tween 80, **d** 0.20 % Tween 80, **e** 0.25 % Tween 80, **f** 0.30 % Tween 80. Magnification $\times 10$, with the edges of the images being 100 μ m

52.3.5 Effect of Inoculum Size on Citric Acid Fermentation

Generally speaking, the adequate dissolved oxygen level under small inoculum size brought excess nutrients [14] and led to large mycelium pellets and low acid production rate. In contrast, the excessive inoculum size will lead to the lack of dissolved oxygen and active anaerobic metabolism. As shown in Fig. 52.9, with the increase of inoculum size the citric acid production gradually increased. When the inoculum size was 5×10^5 spores/mL, the citric acid production was the maximum.

Fig. 52.9 The influence of inoculum size on citric acid fermentation



506 J. Zhang et al.

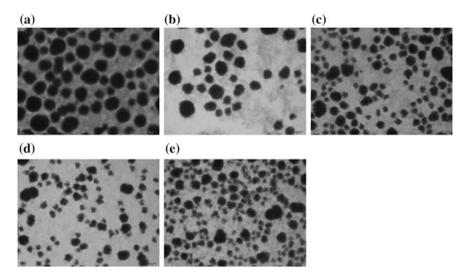


Fig. 52.10 The mycelium pellet morphology of *A. niger* in fermentation 24 h. **a** 10^5 spores/mL, **b** 2×10^5 spores/mL, **c** 5×10^5 spores/mL, **d** 8×10^5 spores/mL, **e** 10^6 spores/mL. Magnification $\times 10$, with the edges of the images being $100 \ \mu m$

From Fig. 52.10, mycelium pellets were too large under the inoculum size of 10^5 spores/mL. The optimum inoculum size was 5×10^5 spores/mL when mycelium pellets were relatively homogeneous and dense. If the inoculum size was increased continually, mycelium pellets were loose relatively, which was unsuitable for citric acid production.

52.4 Discussion

According to the results, the optimized fermentation medium consists of 10 % corn liquefied serum, 0.25 g/L soy peptone, 2.0 g/L (NH₄)₂SO₄, 12.0 g/L dextrin, 2.0 g/L KH₂PO₄, 0.10 % Tween 80. The fermentation conditions were 36.5 °C, 330 rpm for 24 h. The inoculation size and medium volume were 5×10^5 spores/mL and 30 mL/250 mL, respectively. Under the optimized conditions above, a maximized citric acid yield of 16.11 g/L was obtained.

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Chapter 53 Study on the Soxhlet's Extraction of Star Anise Oil and Preliminary Investigation of Its Antibacterial Activity

Lin Tian and Ping Li

Abstract In this study, Soxhlet extraction method using ethyl acetate as a solvent was used to extract the volatile oil from star anise fruits. The best technological extraction conditions were obtained by a series of single-factor experiments. The antibacterial activity of star anise oil obtained by Soxhlet extraction method was preliminary determined. The result showed that the most important factor affecting the extraction rate of oil from star anise fruits was extraction temperature, followed by powder granularity, solid-liquid ratio, and extraction time. The best extraction conditions were as follows: 1:22 (w/v), 40 min, 120 °C, and 60-80 mesh, respectively. Under these conditions, the yield of star anise oil reached 25.9827 %. It was proved that star anise oil extracted by Soxhlet extraction method could reduce the cost of investment equipments, improve production significantly, and keep the high-quality and natural flavor of the essential oil. Antibacterial experiment demonstrates that star anise oil method has a strong inhibitory effect against Escherichia coli and Bacillus subtilis, so that it could be applied as a natural and environment-friendly antibacterial agent in many areas, such as food products and pharmaceutical sectors.

Keywords Star anise oil · Soxhlet extraction method · Inhibitory effect

53.1 Introduction

Star anise is an important "Medicinal and Edible" species, mainly in Guangxi, Guangdong, Yunnan, and other provinces. Star anise oil, which is extracted from fresh leaves or ripe fruits, is colorless to light yellow, odor fragrance, and tastes sweet. Anise fruit contains volatile oil, organic acids, flavonoids, anisole, and some

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510 L. Tian and P. Li

active ingredients such as anise aldehyde which is reported to have certain antibacterial effect [1-3].

In the traditional food industry, the utilization of star anise oil is relatively low compared to star anise fruit. With the development of the food industry, the use of star anise oil is far more than the fruit, because of its small size, health, easy to use, and other advantages [4, 5].

Star anise oil has a good bactericidal ability to *Mycobacterium tuberculosis*, *Bacillus subtilis*, *Staphylococcus aureus*, *Streptococcus pneumoniae*, *Aspergillus flavus*, and *Aspergillus niger*. Star anise extract, extracted with ethanol, has certain inhibition against *S. aureus*, *S. pneumoniae*, *Paratyphoid bacillus*, and some common pathogenic Candida [6, 7].

In this paper, we extracted star anise oil from star anise dried fruit with Soxhlet method. Meanwhile, several major factors that affect the oil yield were investigated, such as extraction time, the degree of material crushing, extraction temperature, solid–liquid ratio, etc. Conditions of star anise oil extracted by Soxhlet method were optimized by orthogonal test on the basis of single-factor experiment. Finally, the inhibitory effect of star anise oil against *Escherichia coli* and *B. subtilis* was preliminary studied to provide some references for anise oil on drugs for antibacterial agents.

53.2 Materials and Methods

53.2.1 Drugs and Instruments

53.2.1.1 Materials

Ingredients: Dried star anise

Reagents: ethyl acetate (AR), cyclohexane, petroleum ether, 95 % ethanol, and

acetone

Drugs: anhydrous sodium sulfate (AR)

Experimental apparatus: thermostatic bath with magnetic stirring mill, flat-bottomed flask (150 mL), silica gel plate, and a fat extractor

53.2.1.2 Antimicrobial Experimental Material

strains: E. coli, B. subtilis medium: agar gel medium

53.2.2 *Methods*

53.2.2.1 Plant Material

The star anise was dried at 60 °C for 48 h in a electric air blowing drying oven and then were ground in different periods in order to obtain different powder granularity.

53.2.2.2 Microbial Strains and Bacteria Suspension Preparation

Escherichia coli (gram negative) and B. subtilis (gram positive) were selected as representative bacteria to determine antibacterial activity of star anise oil. After activated, the bacteria were inoculated into liquid standard nutrient broth medium, and cultured at 37 °C with regularity shaking at 180 rpm for 18 h. The number of bacteria was counted using plate dilution method. The final concentration of colonies was 106 CFU/mL. Solid nutrient broth medium was supplemented with 2 % agar into liquid medium.

53.2.2.3 Preparation of Essential Oil

The extraction process was performed by a Soxhlet extraction method. Star anise powder was put into a fat extractor, extraction solvent was added, and then the experiment was conducted under set conditions. After extraction, the solution was filtered and dried under anhydrous sodium sulfate. After drying, the extract was again filtered and the filtrate was concentrated by vacuum rotary evaporator to recover solvent. Anise oil was calculated using the following (53.1):

$$X(\%) = \frac{Q_1}{Q_2} \times 100 \% \tag{53.1}$$

where Q_1 is the quality of extracted star anise oil; Q_2 is the quality of star anise powder in test.

53.3 Results and Discussion

In the study, extraction conditions of star anise oil by Soxhlet method were optimized by a series of single-factor tests and orthogonal test.

512 L. Tian and P. Li

53.3.1 Single-Factor Tests for the Extraction of Star Anise Oil

53.3.1.1 Extraction Solvent Selection

In the extraction process, ethyl acetate, cyclohexane, petroleum ether (30–60 °C), petroleum ether (60–90 °C), 95 % ethanol, and acetone were applied as extraction solvents to examine their effects on the oily yield. Results were shown in Fig. 53.1.

As can be seen from Fig. 53.1, ethyl acetate was the best extraction solvent with the highest extraction yield; hence, ethyl acetate was selected as the solvent to extract star anise oil in the following tests.

53.3.1.2 Effect of Powder Granularity on the Extraction Yield of Star Anise Oil

In this study, star anise was crushed into powder with the mill and was sieved into five different sizes, i.e., using 10–20, 20–40, 40–60, 60–80 and 80–100 meshes, was extracted at 100 °C, the solid–liquid ratio was 1:20, and the extraction time was 40 min. The result was shown in Fig. 53.2.

As shown in Fig. 53.2, the extraction rate of star anise oil was increased because of improved powder granularity, and reached maximum when the star anise powder was 60–80 mesh. After 60–80 mesh, the rate decreased. The possible reason is that crushing anise into powder could increase the specific surface area and promote the leaching of anise oil. However, excessive grinding could lead to large specific surface area which could cause enhanced adsorption and gradually reduced yield.

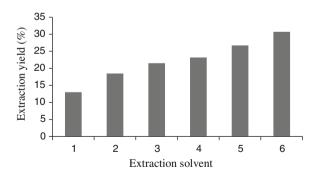


Fig. 53.1 Effect of different solvents on the extraction yield of star anise oil. *I* Ethanol (95 %); 2 petroleum ether (b.p. 30–60 °C); *3* petroleum ether (b.p. 60–90 °C); *4* cyclohexane; *5* acetone; *6* ethyl acetate

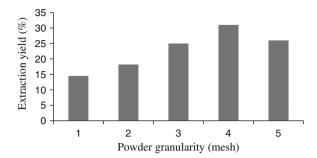


Fig. 53.2 Effect of powder granularity on the extraction yield of star anise oil. *I* 10–20 mesh; 2 20–40 mesh; 3 40–60 mesh; 4 60–80 mesh; 5 80–100 mesh

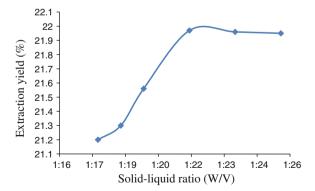
53.3.1.3 Effect of Solid-Liquid Ratio on the Yield of Star Anise Oil

Some experiments were carried out to determine the effect of solid–liquid ratio on the yield of star anise oil (Fig. 53.3). The results showed that with increased amount of ethyl acetate, the yield was relatively gentle rise. When the solid–liquid ratio was 1:22 (w/v), the yield reached 21.97 %. And after that, the yield was little changed with the increase of ethyl acetate.

53.3.1.4 Effect of Extraction Time on the Yield of Star Anise Oil

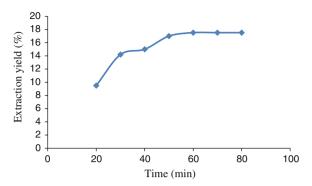
Seven experiments were conducted to study the impact of different extraction time on the yield of star anise oil. As illustrated in Fig. 53.4, with the increase of extraction time, star anise oil yield also improved. The highest rate was achieved at 60 min. After that, there was no significant increase in yield. This showed that the star anise oil had been completely extracted for 60 min.

Fig. 53.3 Effect of solid–liquid ratio on the extraction yield of star anise oil



514 L. Tian and P. Li

Fig. 53.4 Effect of extraction time on the extraction yield of star anise oil



53.3.1.5 Effect of Extraction Temperature on the Yield of Star Anise Oil

It can be seen from Fig. 53.5, with the rising of temperature, the rate of extraction of star anise oil raised, but gradually reduced after 110 °C. It is because the extraction temperature was too high which led to some low-boiling-point components (such as terpenoids) volatilizing.

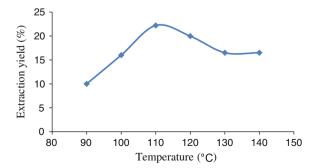
53.3.2 Orthogonal Test for the Extraction of Star Anise Oil

53.3.2.1 Design and Result of the Orthogonal Test

An orthogonal experiment on four factors and four levels was designed and conducted, in order to optimize the extraction conditions of Soxhlet extraction of star anise oil, based on the previous single-factor experiments (listed in Tables 53.1 and 53.2).

As shown in Tables 53.1 and 53.2, extraction temperature had the most significant influence on the yield. But the over-high temperature could result in the

Fig. 53.5 Effect of extraction temperature on the extraction yield of star anise oil



r	A solid–liquid ratio (w/v)	B time (min)	C temperature (°C)	D powder granularity (mesh)
1 1	1:18	40	100	20–40
2 1	1:20	50	110	40–60
3 1	1:22	60	120	60–80
4 1	1:24	70	130	80–100

Table 53.1 The design of experiment factors and levels

Table 53.2 Four-level orthogonal test and experimental results

Number	A solid-liquid	B time	C temperature	D powder	Yield
	ratio (w/v)	(min)	(°C)	granularity (mesh)	(%)
1	1	1	1	1	13.9754
2	1	2	2	2	19.9848
3	1	3	3	3	25.9621
4	1	4	4	4	21.9829
5	2	1	2	3	21.9811
6	2	2	1	4	15.9728
7	2	3	4	1	15.9831
8	2	4	3	2	19.9760
9	3	1	3	4	23.9741
10	3	2	4	3	25.9631
11	3	3	1	2	17.9860
12	3	4	2	1	19.9812
13	4	1	4	2	23.9842
14	4	2	3	1	19.9732
15	4	3	2	4	17.9870
16	4	4	1	3	13.9835
K_1	81.9052	83.9148	61.9187	69.9129	
K_2	73.9140	81.8949	79.9341	81.9310	
<i>K</i> ₃	87.9044	77.9182	89.8854	87.8898	
K_4	75.9279	75.9236	87.9133	79.9178	
k_1	20.4763	20.9787	15.4797	17.4782	
k_2	18.4785	20.4737	19.9835	20.4828	
k_3	21.9761	19.4796	22.4714	21.9725	
k_4	18.9820	18.9809	21.9783	19.9795	
R	3.4976	1.9978	6.9917	4.4943	

reduction of the yield. The possible reason is that the enhancement of the temperature led to some losses of volatile components of essential oil and the decline of the yield. In addition, the powder granularity also had a great effect on the yield, which illustrated smaller powders caused more dissolution of the oil.

516 L. Tian and P. Li

Besides, long extraction time was not conducive to the extraction rate. And the optimum conditions by Soxhlet extraction of star anise oil: 1:22 (w/v), 40 min, 120 °C, 60–80 mesh.

53.3.2.2 Verification Test

Three tests were carried out to verify the repeatability of optimum extraction conditions above, and the results were listed in Table 53.3. The result of verification test demonstrated that the optimum extraction technology had good stability, and under this condition, the oily yield could reach 25.9827 % in average.

53.3.3 Purification of Star Anise Oil

Since the color of the star anise oil extracted through Soxhlet extraction method was very deep (dark green), silica gel column chromatography separation and purification method was used to separate pigments or other impurity out, in order to obtain higher purity of the star anise oil. Before the separation experiments, thin layer chromatography tests were conducted to determine the best ratio of eluent (petroleum ether:ethyl acetate). Finally, the ratio of 1:10 eluent was used for column chromatography separation experiment, so that the high purity of star anise oil was obtained.

53.3.4 Antibacterial Activity of Star Anise Oil Against Escherichia coli and Bacillus subtilis

Disc-diffusion method was conducted in the study of antibacterial activity of star anise oil against *E. coli* and *B. subtilis*. The diameters of inhibition zone and the percentage of growth inhibition were listed in Tables 53.4, 53.5, 53.6, and 53.7.

It can be seen from the results, that star anise oil has a strong antibacterial activity against *E. coli* and *B. subtilis*. This might be due to the presence of its main constituent in star anise oil, such as anisic aldehyde, limonene, and transanethole which were reported to have strong antibacterial activity. However, with the decreased concentration, the inhibitory effect of star anise oil also weakened, and

Table 53.3 Verification test and results

Number	1	2	3
Yield (%)	25.8653	25.9051	26.0112
Average (%)		25.9827	
Relative standard deviation (%)		0.0080	

Table 53.4 Antibacterial activity of star anise oil (without purification) against *Bacillus subtilis*

Concentration (mg/mL)	Diameter of the inhibition zone (mm)	Inhibition percentage (%)
1.00	25.3	76.3
0.50	12.7	52.8
0.25	9.0	33.3
0.17	8.0	25.0
Control	6.0	0

Table 53.5 Antibacterial activity of star anise oil (after purification) against *B. subtilis*

Concentration (mg/mL)	Diameter of the inhibition zone (mm)	Inhibition percentage (%)
1.00	28.3	78.8
0.50	15.2	60.5
0.25	12	50.0
0.17	10	40.0
Control	6.0	0

Table 53.6 Antibacterial activity of star anise oil (without purification) against *Escherichia coli*

Concentration (mg/mL)	Diameter of the inhibition zone (mm)	Inhibition percentage (%)
1	17.7	67.1
0.50	11.7	40
0.25	8.0	25.0
0.17	7.0	14.3
Control	6.0	0

Table 53.7 Antibacterial activity of star anise oil (without purification) against *E. coli*

Concentration (mg/mL)	Diameter of the inhibition zone (mm)	Inhibition percentage (%)
_ (8)	,	1
1	18.3	66.1
0.50	13.0	53.8
0.25	9.0	33.3
0.17	8.0	25.0
Control	6.0	0

this was probably due to the dilution effect of essential oil, which resulted in the content of antibacterial active ingredients declining. Second, inhibitory effect of star anise oil on *B. subtilis* was better than that on *E. coli*. Finally, the oil treated by separation and purification methods displayed stronger inhibitory effect on the two strains than that without purification. This is because the wax, pigment, and other impurities could be removed through the purification experiments which improved the purity of star anise oil.

518 L. Tian and P. Li

53.4 Conclusion

In conclusion, our research showed that the most important factor affecting the extraction rate of oil from star anise fruits was extraction temperature, followed by powder granularity, solid–liquid ratio, and time. The best extraction conditions were as follows: 1:22 (w/v), 40 min, 120 °C, and 60–80 mesh, respectively. Under this condition, the yield of star anise oil extracted from star anise fruits was 25.9827 %. It was proved that star anise oil extracted by Soxhlet method could reduce the cost of investment equipments and improve production significantly, and keep the high-quality and natural flavor of the essential oil. In addition, antibacterial experiment demonstrated that star anise oil obtained by Soxhlet extraction method had a strong inhibitory effect against *E. coli* and *B. subtilis*, so that it could be applied as a natural and environment-friendly antibacterial agent in many areas, such as food products, healthcare products, and pharmaceutical sectors.

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Part III Biological Separation and Biological Purification

Chapter 54 Production of Diosgenin from *Dioscorea*zingiberensis by Mixed Culture of Three Filamentous Fungi

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Abstract The mixed culture of Aspergillus orvzae, Phanerochaete chrysosporium. and Aspergillus niger was explored for the production of diosgenin from Dioscorea zingiberensis. Conditions for biotransformation of saponin using mixed culture of optimized. were The maximum production $(40.06 \pm 0.53 \text{ mg/g})$ was obtained with D. zingiberensis concentration (50.00 g/L), inoculation size (2.0 mL of spore suspension, 1:2:3 ratio of A. oryzae, P. chrysosporium, and A. niger, v/v/v), initial pH (5.0), and cultivation time (6 days). In the optimization conditions, mono and mixed cultures were tested, with mixed culture of three fungi giving better yields of diosgenin and β -glucosidase. The diosgenin yield of optimization biotransformation increased by 50.54 % compared to the traditional acid hydrolysis method (26.61 \pm 0.78 mg/g). The use of mixed culture of these fungi proposes an approach for more efficient and clean diosgenin production by D. zingiberensis.

Keywords Diosgenin · Aspergillus oryzae · Phanerochaete chrysosporium · Aspergillus niger · Dioscorea zingiberensis

54.1 Introduction

Diosgenin (25R-Spriost-5-en-3β-ol) is an important precursor for the synthesis of adrenal cortex hormone, sex hormone, progestational hormone, and anabolic steroid [1, 2]. *Dioscorea zingiberensis* (DZW) tuber is one of the most important resources for diosgenin production, as it has high content of steroidal saponins. DZW tubers contain 1–2 % diosgenin, 30–40 % starch, 10–15 % lignin, and 40–50 % cellulose

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522 H. Xiao et al.

[3, 4]. Diosgenin occurs in plants in the form of saponins attaching glucose or rhamnose to aglycone by glycosidic bonds at C-3 and C-26 [5]. Production of diosgenin from plant tubers depends not only on decomposition starch, lignin, and cellulose which prevent the saponins from releasing, but also on the hydrolysis of these chemical bonds of saponins. It is reported that the glucosyl residue at C-3 sugar chain of steroidal saponins hardly could be hydrolyzed by currently available glycosidases such as β -glucosidase, amylase, and cellulase [6, 7]. The glucose unit at C-26 on furostanol saponins is easily hydrolyzed to form the corresponding spirostanol saponins [8, 9]. The structure of the saponins and diosgenin are shown in Fig. 54.1.

In the traditional method, acids are used to hydrolyze DZW to produce diosgenin. However, the acid hydrolysis causes the structure of diosgenin damaged and generates numerous byproducts, which results in higher chemical oxygen demand (COD, 80,000 mg/L in the waste water) [10] and serious environmental pollution.

Attempts to reduce oxygen consumption during diosgenin production from DZW were reported. For instance, fiber and starch were recovered by physical methods and used to produce ethanol, and then diosgenin was generated by acid hydrolysis. In this way, COD in wastewater could be decreased by at least 50 %, but resulted in the loss of 20 % diosgenin [11] and the process of recovering fiber and starch is complex. Previous studies showed that multiple enzymes (cellulase, β-glucosidase and pectinase) could transform saponins in DZW to diosgenin [11].

Fig. 54.1 The structure of saponins and diosgenin

Trillin

However, the high costs of enzymes are the major drawback for their application and very few reports are available in the literature regarding using enzymes alone to generate diosgenin from saponin.

Diosgenin production through microbial transformation is another potential technique [12]. For example, biotransformation of DZW by *Aspergillus oryzae* [13], *Trichoderma reesei* [4], and *Trichoderma harzianum* [14] have been applied to improve tradition process efficiency. However, the yield of diosgenin is not high. Because of high contents of lignin and cellulose in DZW, the releasing of saponins and the hydrolysis of glycosidic bonds of saponins require the cooperation of many microbial enzymes to accomplish. Mixed culture of fungi thus may produce the optimal combination of enzymes for the most efficient biodegradation process in increasing diosgenin yield. So far, there are few references regarding the use by mixed microbial culture for diosgenin production. With countless possible microbe combinations, the potential for mixed fermentation in natural product drug discovery is huge [15]. Complex mixed cultures have been widely used in biotechnology for many processes, e.g., for the production of antibiotics, enzymes, fermented food, composting, dairy fermentation, bioconversion of apple distillery, and domestic wastewater sludge [12].

The aim of this work was to increase diosgenin yield from DZW and reduce environmental pollution by the mixed culture of three fungi: *A. oryzae*, *Phanero-chaete chrysosporium*, and *Aspergillus niger*. The biotransformation consists of two stages: hydrolysis of lignin and cellulose of DZW to release saponins and hydrolysis of the saponins to diosgenin.

54.2 Materials and Methods

54.2.1 Materials

The dried DZW tubers were supplied by Kangsheng Company, Qingyang, Gansu, China. The materials were ground by a pulverator (DJ-048 Pulverator, Beijing Huanya Tianyuan Co. Ltd., Beijing, China) and the powder was screened through a 40-mesh stainless steel sieve (0.2 cm). Chemical reagents used were of analytical grade commercial products. HPLC-grade acetonitrile was purchased from TEDIA (USA). Diosgenin standard and *p*-Nitrophenyl β-D-glucopyranoside (pNPG) were obtained from Sigma-Aldrich Co. (USA).

54.2.2 Microorganisms and Microbial Transformation

Phanerochaete chrysosporium was preserved in Tianjin University of Science and Technology Microbiological Culture Collection Center. A. niger and A. oryzae were purchased from China Center of Industrial Culture Collection (Beijing,

524 H. Xiao et al.

China). These strains were stored at 4 °C on potato dextrose agar slant temporarily. All culture and biotransformation experiments were performed in 250 mL Erlenmeyer flasks. To prepare the inocula, spores in a 7-day-old agar slant were suspended in 0.01 % Tween 80 solution (10⁷ spores/mL). The optimal inoculation ratio (1:2:3 ratio of *A. oryzae*, *P. chrysosporium*, and *A. niger*) for each fungus was determined in a preliminary experiment (unpublished data). Fermentation media incubated at 30 °C on a rotary shaker at 180 r/min for diosgenin production. All experiments were done in triplicate and data expressed as average values.

54.2.3 Extraction and Analysis Diosgenin

At the end of cultivation, the steroidal saponins extraction was performed according to Zhu et al. [4]. The fermentation culture was centrifuged, precipitated, and dried at 80 °C, extracted with chloroform, and ultrasonicated for 30 min (KQ 3200B ultrasonicator). The samples were centrifuged and analyzed on Agilent 1,100 Multisolvent Delivery System equipped with a Diamonsil C18 column, 250 mm \times 4.6 mm (5 μ m), and detector at 203 nm, the eluent was A (CH₃CN) and B (HCOOH: H₂O = 0.01:300, v/v) with a gradient time program: 0–24 min, 55 % A \rightarrow 60 % A; 24–30 min, 60 % A \rightarrow 80 % A; 30–64 min, 80 % A. The flow rate of the mobile phase was 1.0 mL/min and the temperature of column was 25 °C.

Yield of Diosgenin (mg/g) = diosgenin content (mg)/DZW content (g)

54.2.4 Acid Hydrolysis Conditions

The traditional method uses acids to hydrolyze DZW to produce diosgenin. As a comparison experiment of microbial transformation, the acid hydrolysis conditions: 5.0 g DZW was mixed with 100 mL 1 mol/L $\rm H_2SO_4$ and heated with a reflux condenser at 100 °C for 4 h, then centrifuged and pH was adjusted until it became neutral. The centrifuged residue was dried at 80 °C and diosgenin extraction was the same as mentioned above.

54.2.5 Analysis β-Glucosidase, Glucoamylase, Carboxymethyl Cellulase, and Lignin Peroxidase Activity

A culture in each flask was centrifuged at 4 °C with 8,000 r/min for 10 min, the supernatants were crude enzyme diluted for enzymes activity analysis.

 β -Glucosidase activity was determined at 400 nm by using 1 mol/L pNPG as the substrate as described previously [16]. The reaction mixture containing 0.5 mL pNPG and 0.5 mL enzyme solution was incubated at 50 °C for 10 min. The reaction was terminated by adding 2 mL 1 mol/mL Na₂CO₃. One unit (U) of β -glucosidase activity was defined as the amount of enzyme catalyzing the formation of 1 μ mol p-nitrophenyl per minute.

Carboxymethyl cellulase was determined at 540 nm by measuring the increase in reducing sugar formed by enzymatic hydrolysis of sodium carboxymethyl cellulose [17]. One unit (U) of enzyme activity was expressed as the amount releasing 1 mg of glucose per mL of enzyme in 1 h.

Glucoamylase activity was determined at 540 nm by measuring the increase in reducing sugar formed by enzymatic hydrolysis of starch described previously [6]. One unit of enzyme activity was expressed as the amount releasing 1 μ mol of glucose per milliliter of enzyme per minute.

Lignin peroxidase activity was measured using veratryl alcohol oxidation [18]. The standard reaction mixture (2 mL) contained 0.2 mmol veratryl alcohol, 0.15 mmol $\rm H_2O_2$, and 11.4 nmol LiP in 100 mmol/L sodium tartrate buffer pH 3.0. The reaction was assayed at 310 nm. One unit (U) of enzyme activity was expressed as the amount of generating 1 μ mol of oxidation of veratryl alcohol per minute.

54.3 Results and Discussion

54.3.1 The Optimization of Culture Conditions for DZW

In order to determine the optimum concentrations of DZW, inoculation size, initial pH, and cultivation time for diosgenin yield, mixed spore suspension was inoculated into fermentation medium that consisted of only DZW in tap water at different concentrations (16.67, 33.33, 41.67, 50.00, 58.33, and 66.67 g/L). The results show that diosgenin yield increased with increasing DZW concentrations up to 50.00 g/L and decreased with further increasing DZW concentrations (Fig. 54.2).

The inoculum size is also an important factor for diosgenin production. Inoculation size for DZW fermentation was optimized by inoculating the mixed spore suspension with 1.0, 1.5, 2.0, 2.0, 2.5, 3.0, and 3.5 mL. The results obtained are shown in Fig. 54.3. The highest yield of diosgenin was obtained with 2 mL spore suspension. Using spores for inoculation offered several benefits compared with vegetative cells. For instance, they can serve as a biocatalyst in bioconversion reactions because of their ability to carry out the same reactions as the corresponding mycelia [19, 20].

pH is important for fermentation process as it may affect various metabolic activities through its effect on several components transferred across the cell membrane [21]. The effect of initial pH on diosgenin yield by mixture fermentation was carried out at pH 3.5, 4.0, 4.5, 5.5, and 6.0. The results obtained are shown in

526 H. Xiao et al.

Fig. 54.2 The effect of DZW concentration on diosgenin yield by mixed culture (A. oryzae:P. chrysosporium: A. niger = 1:2:3); inoculum size: 2 mL; culture time: 4 days; unadjusted pH

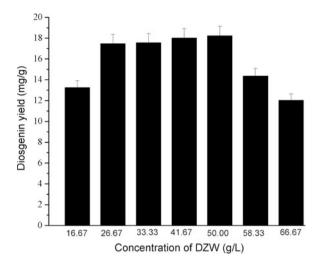


Fig. 54.3 The effect of inoculum size on diosgenin yield by mixed culture (A. oryzae:P. chrysosporium: A. niger = 1:2:3); DZW concentration: 50.00 g/L; culture time: 4 days; unadjusted pH

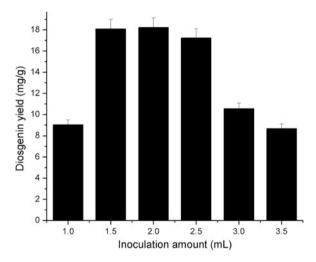


Fig. 54.4. Diosgenin yield gradually increased with increasing pH from 3.5 to 5.0 and reached the optimum level at 5.0.

The effects of culture time on diosgenin yield for mixture culture were carried out at different growth stages. The highest yield of diosgenin was obtained at 6 days, not changed in further increasing culture time (Fig. 54.5). The results of optimization showed that a combination of concentration of DZW 50.00 g/L, inoculation size 2.0 mL, initial pH 5.0, and cultivation time 6 days, the diosgenin yield was 40.06 ± 0.53 mg/g. The final diosgenin yield of optimization process increased by 50.54 % compared to the traditional acid hydrolysis method for DZW production $(26.61 \pm 0.78$ mg/g). The use of mixed culture of these fungi proposes an approach for more efficient and clean diosgenin production from *D. zingiberensis*.

Fig. 54.4 The effect of initial pH on diosgenin yield by mixed culture (*A. oryzae*: *P. chrysosporium:A. niger* = 1:2:3); DZW concentration: 50.00 g/L; inoculum size: 2 mL; culture time: 4 days

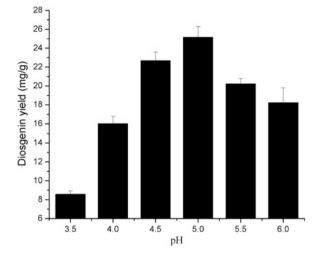
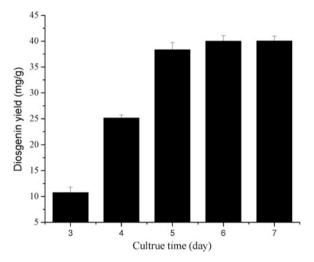


Fig. 54.5 The effect of culture time on diosgenin yield by mixed culture (*A. oryzae:P. chrysosporium: A. niger* = 1:2:3); DZW concentration: 50.00 g/L; inoculum size: 2 mL; initial pH: 5.0



54.3.2 Biotransformation of Monoculture and Mixed Culture

In the optimized conditions, as shown in Fig. 54.6, the diosgenin yield using monoculture of A. oryzae, P. chrysosporium, and A. niger and the mixed culture were 28.44 ± 0.89 , 12.59 ± 0.67 , 30.67 ± 0.23 and 40.06 ± 0.53 mg/g, respectively, after 6 days of fermentation. The final diosgenin yield of optimization process increased by 28.58, 68.57, and 23.44 % compared to that using monoculture of A. oryzae, P. chrysosporium, and A. niger. The results showed that the mixed culture system gave better results than monocultures for diosgenin production. The mixed culture obviously depends more on a multiple enzymatic hydrolysis strategy

528 H. Xiao et al.

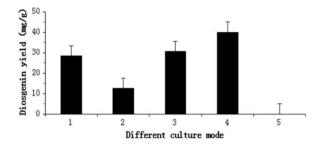
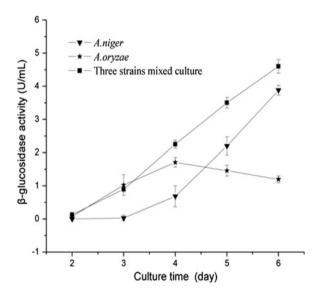


Fig. 54.6 The results of hydrolyzed product by monoculture (1 A. oryzae, 2 P. chrysosporium, 3 A. niger), mixed culture (4 A. oryzae:P. chrysosporium:A. niger = 1:2:3), and control (5 not inoculation fungus). DZW concentration: 50.00 g/L; Inoculum size: 2 mL; initial pH: 5.0 the diosgenin was extracted with chloroform after culturing for 6 days

Fig. 54.7 The β-glucosidase activity of by *A. oryzae* or *A. niger* monoculture and by mixed culture (*A. oryzae*: *P. chrysosporium:A. niger* = 1:2:3). DZW concentration: 50.00 g/L; inoculum size: 2 mL; initial pH: 5.0 the β-glucosidase activity of *P. chrysosporium* was undetectable



for decomposition of DZW cellulose, lignin, and for transformation of saponins. The multiple enzyme system contains β -glucosidase, cellulase, glucoamylase, and lignin peroxidase which has a direct or indirect effect on production of diosgenin.

The β -glucosidase activity by mixed culture was higher than by *A. niger* or *A. oryzae* monoculture (Fig. 54.7). This result was in agreement with that of Raza et al. who reported that mixed cultures of *A. oryzae* and *A. niger* gave better production yield of β -glucosidase than single culture [22]. β -Glucosidase is one of the key glycosidases for conversion of saponin to diosgenin. Lei et al. reported

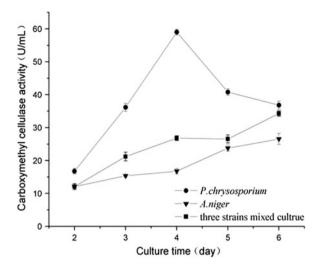


Fig. 54.8 The carboxymethyl cellulase activity of by *A. niger* or *P. chrysosporium* monoculture and by mixed culture (*A. oryzae:P. chrysosporium:A. niger* = 1:2:3). DZW concentration: 50.00 g/L; inoculum size: 2 mL; initial pH: 5.0. the carboxymethyl cellulase activity of *A. oryza* was undetectable

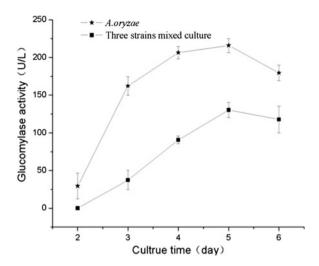
that β -glucosidase from *Aspergillus fumigates* could hydrolyze terminal a-1,2-linked glucopyranosyl of saponins [16]. In addition, Su et al. reported a novel β -glucosidase from *Fusarium proliferatum* ECU 2042 exhibiting almost no activity toward cellobiose, but showing relatively high activity on ginsenoside Rg3 [23].

Filamentous fungi possess powerful abilities to decompose the extensive plant material lignocellulose, and are unique organisms specified to degradation of the xylem cell wall components (cellulose, hemicelluloses, lignins and extractives) [24]. P. chrysosporium, the most widely researched white-rot fungus, produces two families of lignin-degrading peroxidases designated lignin peroxidase and manganese peroxidase [25, 26]. P. chrysosporium not only produced carboxymethyl cellulase production (Fig. 54.8), but also secreted lignin $(188.03 \pm 18.95 \text{ U/L})$ for degrading the lignin of the DZW, thus playing an important role in generating saponins.

Furthermore, as shown in Fig. 54.9, *A. oryzae* secreted both β-glucosidase and glucoamylase. Glucoamylase from *Curvularia lunata* and *A. niger* were shown to be capable of hydrolyzing the terminal 1,2-linked rhamnosyl residues of sugar chains at C-3 position of steroidal saponins [6, 27]. Thus glucoamylase could be used to increase diosgenin yield.

530 H. Xiao et al.

Fig. 54.9 The glucoamylase activity of by *A. oryzae* monoculture and by mixed culture (*A. oryzae*: *P. chrysosporium:A. niger* = 1:2:3). DZW concentration: 50.00 g/L; inoculum size: 2 mL; initial pH: 5.0 the glucoamylase activity of *P. chrysosporium* and *A. niger* was undetectable



54.4 Conclusion

The potential of mixed culture with *A. oryzae*, *P. chrysosporium*, and *A. niger* for optimal diosgenin production was demonstrated. In the optimized conditions, the maximum production of diosgenin was 40.06 ± 0.53 mg/g, an increase of 50.54 % compared to the acid hydrolysis method for DZW production (26.61 ± 0.78 mg/g). The mixed culture system gave better results than monocultures regarding diosgenin production.

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Chapter 55 Characterization of Volatile Constituents of Chinese Hawthorn (*Crataegus* spp.) Fruit Juices

Yuping Zhao, Yangyang Wang, Jiwu Wang, Zhilian Wu, Zuli Sun, Tiantian Tian, Hao Niu, Lili Jing, Zhengyu Fang and Jianrong Yang

Abstract All hawthorn fruits used in this study were collected from four provinces of the China namely Shandong, Shanxi, Hebei, and Liaoning. Hawthorn fruits were prepared for hawthorn fruit juices. Volatile compounds in the juices of 12 cultivars of hawthorn were determined by gas chromatography and mass spectrometry (GC–MS) coupled with headspace solid-phase micro-extraction (HS–SPME) method. Semi-quantitative analysis was employed for the analysis of volatile constituents in hawthorn juice. Among 61 volatile compounds detected in the 12 samples, 21 volatiles were found in all samples. Esters and alcohols were the main volatiles present in the hawthorn fruit. Aldehydes, aromatic compounds, furans, terpenes, acids, and sulfurs also made great contributions to the total volatiles of the hawthorn fruits. Ethyl acetate, 3-methylbutanol, 3-octanone, acetic acid, furfural, dimethyl sulfide, and *p*-menth-1-en-8-ol were suggested to the major volatile compounds in all samples. For most volatile compounds identified in the current research, their content in sample 7 was highest than the others and the sample with lowest contents was sample 6.

Keywords Hawthorn fruit • Volatile constituents • Gas chromatography-mass spectrometry • Headspace solid-phase micro-extraction

55.1 Introduction

Hawthorn fruits, which are well known for their health benefits, are used as raw materials in the field of food and medicine [3, 13]. Hawthorn is considered as a "cardiotonic" herb and has been used in traditional medicine to reduce the risk of

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cardiovascular diseases such as chest pain, irregular heartbeat, and congestive heart failure [1, 14]. The hypolipidemic, free radical scavenging effects and anti-inflammatory potentials of hawthorn have also been reported, together with their vasorelaxing and antioxidative activities [21]. Such wide beneficial effects have been mainly attributed to the presence of phenolic compounds. As one of the commonly cultivated varieties, hawthorn fruits have long been consumed by Chinese and the long history of application supports their health and safety benefits [8].

In China, hawthorn is widely distributed, such as Shandong, Liaoning, Henan, Hunan, and Shanxi provinces. Hawthorn, belonging to the Rosaceae family, involves 18 species in China, of which *Crataegus pinnatifida* Bge. and its variety Shanlihong (*C. pinnatifida*. Bge. var. major N.E.Br.) are the most important [9]. Chinese hawthorn fruit presents intense and unique aroma which influences the acceptability and flavor of formulation of various food products including jam, confect, juice, and canned food [19].

Previous studies have determined the antioxidant activity, phenolic compounds, acids, and sugars of the hawthorn [1–13]. However, to the best of our knowledge, there is to date little research determining the flavor of hawthorn fruits [6, 15]. Hence, our research is focused on the qualitative and quantitative analysis of volatile components in 12 cultivars hawthorn fruits. The analytical technique used to determine aromatic compounds of hawthorn fruits is headspace solid phase microextraction combined with gas chromatography-mass spectrometry (HSSPME–GC–MS). From a practical standpoint, the research was performed to provide sufficient information for further study of volatile components in hawthorn fruits and the development of hawthorn fruit products.

55.2 Materials and Methods

55.2.1 Plant Materials

All hawthorn fruits used in this research were collected in China and picked at commercial maturity during the harvest season. Samples 1–2 were collected in Yantai city and Weifang city of Shandong province. Samples 3–5 were collected in Jincheng city, Yuncheng city and Linfen city of Shanxi province. Samples 6–9 were collected in Chengde city, Baoding city, Zhangjiakou city, and Cangzhou city of Heibei province. Samples 10–12 were collected in Anshan city, Dalian city, and Shenyang city of Liaoning province.

55.2.2 Chemicals

The pectic enzyme preparation was purchased from Novozymes Co. LTD in China (Beijing, China). Sodium chloride was obtained from China National

Pharmaceutical Ground Corporation (Shanghai, China). The internal standard (IS) solution of 3-octanol used in the GC-MS analysis was purchased from Aldrich (Milwaukee, Wis., USA). All other chemicals used were of analytical grade.

55.2.3 Preparation of Hawthorn Juice

The hawthorn juice was prepared according to the method proposed by Zhao et al. [21].

55.2.4 HS-SPME Analysis

A portion of 5 mL hawthorn juice, saturated with 2.0 g of sodium chloride was added into a 20 mL headspace vial spiked with 5 μ L of 3-octanol (101 mg/L standard solution in ethanol), and a 10 mm stirring bar. The vial capped a Teflon-faced silicone septum and was placed in a thermostat bath adjusted to 50 °C with constant agitation in a magnetic stirrer for 10 min.

The fiber of 50/30 μm DVB/CAR/PDMS employed in the analyses was obtained from Supelco (Aldrich, Bellefonte, PA, U.S.A.). Before extraction, the fiber was conditioned by being inserted into the GC injector at the 250 °C for 2 h to prevent contamination. After that, the fiber was exposed to the headspace of the prepared glass vial for 15 min. Finally, the fiber was immediately inserted into the injection port of GC at 250 °C for thermal desorption over 5 min.

55.2.5 GC-MS Analysis

GC–MS analyses were carried out using an Agilent 6890 N gas chromatograph fitted with an Agilent 5975 mass selective detector. The above extraction was analyzed on a DB-Wax column (30 m \times 0.25 mm i.d., 0.25 μ m film thickness, J&W Scientific, Folsom, CA) and the injection technique was splitless. The carrier gas was helium with a constant flow of 2 mL/min. The injector and detector temperature were both 250 °C. The analysis conditions were programmed with an initial temperature of 40 °C for a 2 min hold, then to 230 °C at 4 °C/min with a 5 min hold. Electron ionization mass spectrometric data from m/z 34 to 348 were recorded with an ionization voltage of 70 eV and the temperature of ion source was 230 °C.

536 Y. Zhao et al.

55.2.6 Compound Identification

A majority of compounds were identified by comparison of their mass spectra and retention times with those of standard compounds or by comparison of the mass spectrum with those in the National Institute of Standards and Technology (NIST) 05 MS database or a custom database [20].

55.2.7 Quantitative Analysis

Semi-quantitative analysis was employed for the determination of volatile component in hawthorn juice. The IS solution (3-octanol) was individually prepared in ethanol and stored at 4 °C prior to analysis in GC–MS. All analyses were carried out in triplicate, so the average was taken as the final peak area. The integral for all chromatogram peaks used selective ion method (SIM) [2, 12].

Semi-quantitative concentrations =
$$(peak area/IS peak area) \times IS concentration$$
(55.1)

55.2.8 Statistical Analysis

The results were reported as means of three parallel measurements. Statistical analysis was conducted using the SPSS version 19.0 statistical package for windows (SPSS Inc., Chicago, Illinois).

55.3 Results and Discussion

For a better characterization of the volatile components of the hawthorn juice, a detailed research by GC–MS was conducted. Table 55.1 presents the corresponding volatile composition identified in the hawthorn juice, along with the relative concentrations in the extract. A total of 61 primary volatile compounds were identified and quantified, including 11 esters, 16 alcohols, 12 aromatic compounds, 8 terpenes, 5 furans, and 4 carboxylic acids, 4 ketones, 1 sulfur. According to semi-quantitative concentrations, esters and alcohols had higher concentrations among the volatiles in 12 hawthorn juices, followed by the sulfurs, terpenes, and aromatic compounds.

Table 55.1 Volatile compounds identified in 20 Chinese hawthorn juices by GC-MS analysis

Compounds	Concen	ration (r	ng/L)			
	1	2	3	4	5	6
Alcohols						
Methyl alcohol	1.259	0.754	0.887	1.044	1.065	0.777
2-Methyl-3-buten-2-ol	0.224	0.356	0.154	0.093	0.213	0.329
2-Methyl-1-propanol	0.022	0.049	0.020	0.065	0.052	0.008
1-Butanol	0.018	n.i.	n.i.	0.015	0.073	n.i.
3-Methyl-1-butanol	1.056	1.381	0.890	3.285	1.765	0.288
1-Pentanol	0.223	0.484	0.195	0.078	0.190	0.192
1-Hexanol	0.163	0.229	0.096	0.335	0.291	0.163
3-Hexen-1-ol	0.128	0.153	0.061	0.084	0.119	0.084
1-Heptanol	n.i.	n.i.	0.093	0.528	0.119	n.i.
2-Ethyl-1-hexanol	0.159	0.078	0.040	0.454	0.125	0.071
(E)-2-Hepten-1-ol	0.009	0.015	0.017	n.i.	n.i.	0.016
1-Octanol	0.061	0.114	0.121	0.085	0.167	0.102
(Z)-5-Octen-1-ol	n.i.	n.i.	0.012	0.260	0.006	n.i.
1-Nonanol	0.063	0.026	0.013	0.221	0.047	0.016
(<i>E</i> , <i>Z</i>)-3,6-Nonadien-1-ol	0.019	n.i.	n.i.	0.086	n.i.	n.i.
3-Methyl-2-buten-1-ol	0.217	0.382	0.103	0.216	0.106	0.192
Totals	3.612	4.021	2.702	6.849	4.338	2.238
Esters						
Methyl acetate	1.247	n.i.	2.752	1.290	n.i.	0.813
Ethyl acetate	9.027	3.888	3.641	6.995	6.833	3.070
2-Methylpropyl acetate	n.i.	0.105	0.038	0.520	0.072	n.i.
Butyl acetate	0.022	0.108	0.006	0.017	0.050	0.061
2-Methyl-1-butanyl acetate	0.098	0.881	0.811	0.232	0.172	0.266
3-Methyl-1-butanyl acetate	0.021	n.i.	n.i.	0.010	0.274	0.542
Ethyl hexanoate	0.077	0.093	0.006	0.063	n.i.	0.049
(Z)-3-Hexenyl acetate	n.i.	n.i.	n.i.	0.024	n.i.	n.i.
2-Furanmethanyl acetate	0.066	0.020	0.056	n.i.	n.i.	0.037
Methyl benzoate	0.135	0.056	0.015	0.051	0.061	0.027
1.2-Ethyl-3-hydroxyhexyl	0.073	0.037	0.079	0.054	0.017	0.014
2-methylpropanoate						
Totals	10.766	5.188	7.404	9.256	7.479	4.879
Aldehydes and ketones						
3-Octanone	0.013	0.415	0.055	0.017	0.066	0.192
3-Hydroxy-2-butanone	0.016	0.040	n.i.	0.063	0.036	0.003
6-Ethyl-7-hydroxy-4-octen-3-one	0.100	0.061	0.103	0.028	0.039	n.i.

538 Y. Zhao et al.

Table 55.1 (continued)

Compounds	Concent	ration (r	ng/L)			
	1	2	3	4	5	6
6-Methyl-5-hepten-2-one	0.020	0.083	0.039	0.014	0.075	0.042
Totals	0.149	0.599	0.197	0.122	0.216	0.237
Compounds	Conce	ntration	(mg/L)			
	7	8	9	10	11	12
Alcohols	•					
Methyl alcohol	1.076	0.625	0.832	3.281	0.840	0.683
2-Methyl-3-buten-2-ol	0.114	0.118	0.470	0.151	0.088	0.215
2-Methyl-1-propanol	0.043	0.062	n.i.	0.209	0.008	0.024
1-Butanol	n.i.	0.047	n.i.	0.033	0.067	0.006
3-Methyl-1-butanol	0.517	0.270	0.294	6.040	0.213	0.457
1-Pentanol	0.657	0.606	0.102	0.317	0.223	0.331
1-Hexanol	0.182	0.145	0.127	0.207	0.104	0.289
3-Hexen-1-ol	0.077	0.048	0.110	0.136	0.056	0.237
1-Heptanol	0.157	n.i.	n.i.	0.427	n.i.	n.i.
2-Ethyl-1-hexanol	0.412	0.021	0.030	0.282	0.061	0.135
(E)-2-Hepten-1-ol	0.010	0.039	n.i.	0.033	0.005	0.018
1-Octanol	0.093	0.074	0.072	0.307	0.058	0.102
(Z)-5-Octen-1-ol	n.i.	0.025	n.i.	0.015	n.i.	n.i.
1-Nonanol	n.i.	n.i.	0.005	0.078	n.i.	0.046
(<i>E</i> , <i>Z</i>)-3,6-Nonadien-1-ol	n.i.	n.i.	n.i.	n.i.	n.i.	n.i.
3-Methyl-2-buten-1-ol	0.075	0.270	0.006	0.073	0.163	0.735
Totals	3.413	2.35	2.048	11.589	1.886	3.278
Esters	·	•				
Methyl acetate	1.531	n.i.	0.543	2.732	0.569	0.767
Ethyl acetate	9.001	4.728	4.106	7.897	4.504	2.329
2-Methylpropyl acetate	n.i.	n.i.	n.i.	n.i.	n.i.	n.i.
Butyl acetate	0.662	0.563	0.011	0.038	0.081	0.043
2-Methyl-1-butanyl acetate	3.278	0.820	n.i.	0.435	0.231	0.939
3-Methyl-1-butanyl acetate	3.486	0.078	1.234	0.417	0.961	1.588
Ethyl hexanoate	n.i.	0.098	n.i.	0.105	0.048	0.695
(Z)-3-Hexenyl acetate	n.i.	n.i.	n.i.	n.i.	n.i.	n.i.
2-Furanmethanyl acetate	0.009	0.058	0.033	0.063	0.038	0.104
Methyl benzoate	0.019	0.038	0.026	0.046	0.010	0.112
1.2-Ethyl-3-hydroxyhexyl 2-methylpropanoate	0.274	0.011	0.022	0.128	0.019	0.057
Totals	18.26	6.394	5.975	11.861	6.461	6.634

Table 55.1 (continued)

Note Part	Compounds	Concen	tration (mg/L)			
3-Octanone 0.814 0.066 0.119 0.049 0.223 0.063					10	11	12
3-Hydroxy-2-butanone 0.021 0.023 0.052 0.075 n.i. n.i. 6-Ethyl-7-hydroxy-4-octen-3-one n.i. n.i. n.i. n.i. n.i. n.i. n.i. n.i. 6-Methyl-5-hepten-2-one 0.009 0.098 0.082 0.059 0.088 0.142 Totals 0.844 0.187 0.253 0.183 0.311 0.205 Compounds Concentration (mg/L)	Aldehydes and ketones						
6-Ethyl-7-hydroxy-4-octen-3-one n.i. 0.020 Corneuristion (myll-1 Corneuristic (myll-1 Corneuristic (myll-1 Corneuristic (myll-1 0.001 0.002 0.032	3-Octanone	0.814	0.066	0.119	0.049	0.223	0.063
6-Ethyl-7-hydroxy-4-octen-3-one n.i. 0.020 Corneuristion (myll-1 Corneuristic (myll-1 Corneuristic (myll-1 Corneuristic (myll-1 0.001 0.002 0.032	3-Hydroxy-2-butanone	0.021	0.023	0.052	0.075	n.i.	n.i.
6-Methyl-5-hepten-2-one 0.009 0.098 0.082 0.059 0.088 0.142 Totals 0.844 0.187 0.253 0.183 0.311 0.205 Compounds Concertation (mg/L) 1-Methyl-3-(1-methylethyl)benzene 0.009 0.075 0.062 0.031 n.i. 0.062 Acetic acid phenyl methyl ester 0.010 0.009 0.026 0.032 n.i. n.i. Methoxy-phenyl-oxime 0.037 0.008 n.i. 0.017 0.025 0.031 n.i. n.i. 2-Methyl-3-phenylpropanal 0.146 0.072 0.034 0.330 n.i. n.i. Acetic acid 2-phenylethyl ester 0.089 0.009 0.053 0.021 n.i. 0.038 Acetic acid 2-phenylethyl ester 0.089 0.009 0.053 0.021 n.i. n.i. Acetic acid 2-phenylethyl ester 0.089 0.009 0.053 0.021 n.i. n.i. Benzyal alcohol n.i. n.i. n.i.		n.i.	n.i.	n.i.	n.i.	n.i.	n.i.
Totals		0.009			0.059	0.088	0.142
Concentration (mg/L) 1 2 3 4 5 6 6 Aromatic compounds 1-Methyl-3-(1-methylethyl)benzene 0.009 0.075 0.062 0.031 n.i. 0.062 0.062 0.031 n.i. 0.062 0.062 0.032 n.i. n.i. 0.062 0.064 0.062 0.063 0.013 0.008 0.062 0.063 0.013 0.008 0.062 0.063 0.015 0.008 0.062 0.063 0.015 0.008 0.062 0.063 0.015 0.016 0.062 0.063 0.015 0.016 0.062 0.063 0.015 0.016 0.062 0.063 0.015 0.016 0.062 0.063 0.015 0.016 0.062 0.063 0.015 0.016 0.062 0.063 0.015 0.016 0.063 0.015 0.016 0.0		0.844	0.187	0.253	0.183	0.311	0.205
Aromatic compounds 1-Methyl-3-(1-methylethyl)benzene 0.009 0.075 0.062 0.031 n.i. 0.062 Acetic acid phenyl methyl ester 0.010 0.009 0.026 0.032 n.i. n.i. Methoxy-phenyl-oxime 0.037 0.008 n.i. 0.017 0.025 0.013 0.044 0.072 0.034 0.330 n.i. 0.008 Acetic acid 2-phenylethyl ester 0.089 0.009 0.053 0.021 n.i. 0.044 0.047 0.044 0.047 0.047 0.072 0.078 0.048 0.049 0.059 0.053 0.021 n.i. 0.044 0.047 0.048 0.041 0.072 0.078 0.048 0.049 0.059 0.059 0.059 0.050 0.015 n.i. n.i. n.i. 0.026 n.i. n.i. 0.027 0.030 0.007 0.012 0.036 0.015 n.i. 0.027 0.030 0.007 0.012 0.007 0	Compounds	Conce	ntration				
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1-Methyl-3-(1-methylethyl)benzene 0.009 0.075 0.062 0.031 n.i. 0.062 Acetic acid phenyl methyl ester 0.010 0.009 0.026 0.032 n.i. n.i. Methoxy-phenyl-oxime 0.037 0.008 n.i. 0.017 0.025 0.013 2-Methyl-3-phenylpropanal 0.146 0.072 0.034 0.330 n.i. 0.008 Acetic acid 2-phenylethyl ester 0.089 0.009 0.053 0.021 n.i. 0.044 α,α,4-Trimethyl-benzene methanol 0.118 0.104 0.088 0.111 0.072 0.078 Benzyal alcohol n.i. n.i. n.i. n.i. n.i. n.i. n.i. n.i. Phenyl ethyl alcohol 0.009 0.007 0.012 0.036 0.015 n.i. Benzothiazole 0.009 0.004 0.010 0.028 n.i. n.i. Phenol 0.059 0.006 0.011 0.391 0.007 n.i. 2-Ethyl phenol <td>Aromatic compounds</td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td>	Aromatic compounds						
Acetic acid phenyl methyl ester 0.010 0.009 0.026 0.032 n.i. n.i. Methoxy-phenyl-oxime 0.037 0.008 n.i. 0.017 0.025 0.013 2-Methyl-3-phenylpropanal 0.146 0.072 0.034 0.330 n.i. 0.008 Acetic acid 2-phenylethyl ester 0.089 0.009 0.053 0.021 n.i. 0.044 α,α,4-Trimethyl-benzene methanol 0.118 0.104 0.088 0.111 0.072 0.078 Benzyal alcohol n.i. n.i. n.i. n.i. n.i. 0.026 n.i. n.i. Phenyl ethyl alcohol 0.009 0.007 0.012 0.036 0.015 n.i. Benzothiazole 0.009 0.004 0.010 0.028 n.i. n.i. Phenol 0.009 0.004 0.010 0.028 n.i. n.i. 2-Ethyl phenol 0.025 0.029 0.004 0.017 0.012 0.007 0.012 0.007 0.012 <td></td> <td>0.009</td> <td>0.075</td> <td>0.062</td> <td>0.031</td> <td>n.i.</td> <td>0.062</td>		0.009	0.075	0.062	0.031	n.i.	0.062
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α,α,4-Trimethyl-benzene methanol0.1180.1040.0880.1110.0720.078Benzyal alcoholn.i.n.i.n.i.n.i.n.i.0.026n.i.n.i.Phenyl ethyl alcohol0.0090.0070.0120.0360.015n.i.n.i.Benzothiazole0.0090.0040.0100.028n.i.n.i.n.i.Phenol0.0590.0060.0110.3910.007n.i.2-Ethyl phenol0.0250.0290.0040.0340.017n.i.Phenol, 2-methoxy-3-(2-propenyl)0.0240.0050.0080.0770.0120.007Totals0.5350.3280.3081.1340.1480.212Terpenes0.0120.0300.0070.0130.005n.i.3,7-Dimethyl-1,6-Octadien-3-ol0.1080.2790.2900.4660.1750.2471-Methyl-4-(1-methylethyl)-3-Cyclohexen-1-ol0.0610.0420.019n.i.0.0270.0304-Methyl-1-(1-methylethyl)-3-cyclohexen-1-ol0.2530.6220.1450.3080.1610.1741-Methyl-4-(1-methylethenyl)-cyclohexanol0.0820.1310.1170.1490.1040.091(Z)-3,7-Dimethyl-2,6-Octadien-1-oln.i.0.0140.0160.018n.i.n.i. p -Menth-1-en-8-ol0.8410.9280.6940.3080.4670.880Totals0.5961.1520.6131.0550.5110.567 <tr< td=""><td></td><td>0.089</td><td>0.009</td><td>0.053</td><td>0.021</td><td>n.i.</td><td>0.044</td></tr<>		0.089	0.009	0.053	0.021	n.i.	0.044
Benzyal alcohol n.i.		0.118	0.104	0.088	0.111	0.072	0.078
Benzothiazole 0.009 0.004 0.010 0.028 n.i. n.i. Phenol 0.059 0.006 0.011 0.391 0.007 n.i. 2-Ethyl phenol 0.025 0.029 0.004 0.034 0.017 n.i. Phenol, 2-methoxy-3-(2-propenyl) 0.024 0.005 0.008 0.077 0.012 0.007 Totals 0.535 0.328 0.308 1.134 0.148 0.212 Terpenes Eucalyptol 0.012 0.030 0.007 0.013 0.005 n.i. 3,7-Dimethyl-1,6-Octadien-3-ol 0.108 0.279 0.290 0.466 0.175 0.247 1-Methyl-4-(1-methylethyl)-3-Cyclohexen-1- ol 0.061 0.042 0.019 n.i. 0.027 0.030 1-Methyl-4-(1-methylethenyl)-cyclohexanol 0.080 0.034 0.019 0.101 0.039 0.025 p-Mentha-1,8-dien-6-ol 0.082 0.131 0.117 0.149 0.104 0.091 <td< td=""><td></td><td>n.i.</td><td>n.i.</td><td>n.i.</td><td>0.026</td><td>n.i.</td><td>n.i.</td></td<>		n.i.	n.i.	n.i.	0.026	n.i.	n.i.
Benzothiazole 0.009 0.004 0.010 0.028 n.i. n.i. Phenol 0.059 0.006 0.011 0.391 0.007 n.i. 2-Ethyl phenol 0.025 0.029 0.004 0.034 0.017 n.i. Phenol, 2-methoxy-3-(2-propenyl) 0.024 0.005 0.008 0.077 0.012 0.007 Totals 0.535 0.328 0.308 1.134 0.148 0.212 Terpenes Eucalyptol 0.012 0.030 0.007 0.013 0.005 n.i. 3,7-Dimethyl-1,6-Octadien-3-ol 0.108 0.279 0.290 0.466 0.175 0.247 1-Methyl-4-(1-methylethyl)-3-Cyclohexen-1-ol 0.061 0.042 0.019 n.i. 0.027 0.030 1-Methyl-4-(1-methylethenyl)-cyclohexanol 0.080 0.034 0.019 0.101 0.039 0.025 p-Mentha-1,8-dien-6-ol 0.082 0.131 0.117 0.149 0.104 0.091 (Z)-	Phenyl ethyl alcohol	0.009	0.007	0.012	0.036	0.015	n.i.
2-Ethyl phenol 0.025 0.029 0.004 0.034 0.017 n.i. Phenol, 2-methoxy-3-(2-propenyl) 0.024 0.005 0.008 0.077 0.012 0.007 Totals 0.535 0.328 0.308 1.134 0.148 0.212 Terpenes Eucalyptol 0.012 0.030 0.007 0.013 0.005 n.i. 3,7-Dimethyl-1,6-Octadien-3-ol 0.108 0.279 0.290 0.466 0.175 0.247 1-Methyl-4-(1-methylethyl)-3-Cyclohexen-1-ol 0.061 0.042 0.019 n.i. 0.027 0.030 1-Methyl-4-(1-methylethenyl)-cyclohexanol 0.080 0.034 0.019 0.101 0.039 0.025 p-Mentha-1,8-dien-6-ol 0.082 0.131 0.117 0.149 0.104 0.091 (Z)-3,7-Dimethyl-2,6-Octadien-1-ol n.i. 0.014 0.016 0.018 n.i. n.i. p-Menth-1-en-8-ol 0.841 0.928 0.694 0.308 0.467 0.880	Benzothiazole	0.009	0.004	0.010	0.028		n.i.
Phenol, 2-methoxy-3-(2-propenyl) 0.024 0.005 0.008 0.077 0.012 0.007 Totals 0.535 0.328 0.308 1.134 0.148 0.212 Terpenes Eucalyptol 0.012 0.030 0.007 0.013 0.005 n.i. 3,7-Dimethyl-1,6-Octadien-3-ol 0.108 0.279 0.290 0.466 0.175 0.247 1-Methyl-4-(1-methylethyl)-3-Cyclohexen-1-ol 0.061 0.042 0.019 n.i. 0.027 0.030 4-Methyl-1-(1-methyletheyl)-3-cyclohexen-1-ol 0.253 0.622 0.145 0.308 0.161 0.174 1-Methyl-4-(1-methylethenyl)-cyclohexanol 0.080 0.034 0.019 0.101 0.039 0.025 p-Mentha-1,8-dien-6-ol 0.082 0.131 0.117 0.149 0.104 0.091 (Z)-3,7-Dimethyl-2,6-Octadien-1-ol n.i. 0.014 0.016 0.018 n.i. n.i. p-Menth-1-en-8-ol 0.841 0.928 0.694 0.308 0.467	Phenol	0.059	0.006	0.011	0.391	0.007	n.i.
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	2-Ethyl phenol	0.025	0.029	0.004	0.034	0.017	n.i.
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Phenol, 2-methoxy-3-(2-propenyl)	0.024	0.005	0.008	0.077	0.012	0.007
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		0.535	0.328	0.308	1.134	0.148	0.212
3,7-Dimethyl-1,6-Octadien-3-ol 0.108 0.279 0.290 0.466 0.175 0.247 1-Methyl-4-(1-methylethyl)-3-Cyclohexen-1-ol 0.061 0.042 0.019 n.i. 0.027 0.030 4-Methyl-1-(1-methylethyl)- 3-cyclohexen-1-ol 0.253 0.622 0.145 0.308 0.161 0.174 1-Methyl-4-(1-methylethenyl)-cyclohexanol 0.080 0.034 0.019 0.101 0.039 0.025 p-Mentha-1,8-dien-6-ol 0.082 0.131 0.117 0.149 0.104 0.091 (Z)-3,7-Dimethyl-2,6-Octadien-1-ol n.i. 0.014 0.016 0.018 n.i. n.i. p-Menth-1-en-8-ol 0.841 0.928 0.694 0.308 0.467 0.880 Totals 0.596 1.152 0.613 1.055 0.511 0.567 Furans 5-Methyl-2(3H)-furanone 0.022 0.170 n.i. n.i. 0.076 0.128 5-Methyl-2-furancarboxaldehyde 0.138 0.009 0.004 0.632 n.i.	Terpenes						
1-Methyl-4-(1-methylethyl)-3-Cyclohexen-1- ol 0.061 0.042 0.019 n.i. 0.027 0.030 4-Methyl-1-(1-methylethyl)- 3-cyclohexen-1- ol 0.253 0.622 0.145 0.308 0.161 0.174 1-Methyl-4-(1-methylethenyl)-cyclohexanol 0.080 0.034 0.019 0.101 0.039 0.025 p-Mentha-1,8-dien-6-ol 0.082 0.131 0.117 0.149 0.104 0.091 (Z)-3,7-Dimethyl-2,6-Octadien-1-ol n.i. 0.014 0.016 0.018 n.i. n.i. p-Menth-1-en-8-ol 0.841 0.928 0.694 0.308 0.467 0.880 Totals 0.596 1.152 0.613 1.055 0.511 0.567 Furans 5-Methyl-2(3H)-furanone 0.012 n.i. 0.003 0.011 0.016 0.028 5-Methyl-2-furancarboxaldehyde 0.138 0.009 0.004 0.632 n.i. n.i.	Eucalyptol	0.012	0.030	0.007	0.013	0.005	n.i.
ol	3,7-Dimethyl-1,6-Octadien-3-ol	0.108	0.279	0.290	0.466	0.175	0.247
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		0.061	0.042	0.019	n.i.	0.027	0.030
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		0.253	0.622	0.145	0.308	0.161	0.174
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		0.080	0.034	0.019	0.101	0.039	0.025
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Furfural 0.627 0.170 n.i. n.i. 0.076 0.128 5-Methyl-2-furancarboxaldehyde 0.138 0.009 0.004 0.632 n.i. n.i.		0.012	n.i.	0.003	0.011	0.016	0.004
5-Methyl-2-furancarboxaldehyde 0.138 0.009 0.004 0.632 n.i. n.i.		_	_	_	+	_	1
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	2-Furanmethanol	0.061	n.i.	n.i.	0.155	n.i.	n.i.

540 Y. Zhao et al.

Table 55.1 (continued)

Compounds	Concentration (mg/L)						
	1	2	3	4	5	6	
5-Hydroxymethyl-2-furancarboxaldehyde	0.043	0.009	0.012	0.028	n.i.	n.i.	
Totals	0.881	0.188	0.019	0.826	0.092	0.132	
Acids				•			
Acetic acid	0.361	0.151	0.103	1.739	0.107	0.129	
Propanoic acid	0.045	0.078	0.005	0.059	n.i.	n.i.	
2-Methyl-butanoic acid	0.072	0.035	0.014	0.756	0.040	0.019	
Octanoic acid	0.006	n.i.	0.003	0.059	n.i.	n.i.	
Totals	0.484	0.264	0.125	2.613	0.147	0.148	
Sulfurs							
Dimethyl sulfide	0.814	0.855	0.610	0.686	1.571	0.784	
Totals	0.814	0.855	0.610	0.686	1.571	0.784	
Compounds	Concer	ntration	(mg/L)				
	7	8	9	10	11	12	
Aromatic compounds							
1-Methyl-3-(1-methylethyl)benzene	0.038	0.055	n.i.	0.147	0.067	0.090	
Acetic acid phenyl methyl ester	0.041	n.i.	n.i.	0.016	n.i.	0.021	
Methoxy-phenyl-oxime	0.026	0.015	n.i.	0.072	0.008	0.024	
2-Methyl-3-phenylpropanal	0.140	0.013	0.017	0.044	0.004	0.024	
Acetic acid 2-phenylethyl ester	0.108	0.003	0.019	0.051	0.012	0.020	
α,α,4-Trimethyl-benzene methanol	0.027	0.079	0.106	0.058	0.066	0.068	
Benzyal alcohol	n.i.	n.i.	n.i.	n.i.	n.i.	n.i.	
Phenyl ethyl alcohol	0.030	n.i.	n.i.	0.158	0.007	0.013	
Benzothiazole	0.060	0.004	n.i.	0.005	n.i.	0.020	
Phenol	0.037	0.007	0.006	0.027	n.i.	0.018	
2-Ethyl phenol	n.i.	n.i.	n.i.	0.016	n.i.	n.i.	
Phenol, 2-methoxy-3-(2-propenyl)	0.042	0.007	n.i.	0.011	0.005	0.008	
Totals	0.549	0.183	0.148	0.605	0.169	0.306	
Terpenes				•			
Eucalyptol	0.127	0.015	n.i.	0.543	0.003	0.018	
3,7-Dimethyl-1,6-Octadien-3-ol	0.239	0.195	0.220	0.241	0.232	0.287	
1-Methyl-4-(1-methylethyl)-3-Cyclohexen-1-	0.018	0.030	0.027	0.034	0.013	0.022	
ol							
4-Methyl-1-(1-methylethyl)- 3-cyclohexen-1-	0.109	0.162	0.187	0.148	0.122	0.141	
ol	0.025	0.622	0.621	0.022	0.01.5	0.025	
1-Methyl-4-(1-methylethenyl)-cyclohexanol	0.027	0.023	0.021	0.033	0.015	0.020	
p-Mentha-1,8-dien-6-ol	0.057	0.091	0.116	0.125	0.077	0.059	
(Z)-3,7-Dimethyl-2,6-Octadien-1-ol	n.i.	n.i.	n.i.	0.022	n.i.	0.024	
p-Menth-1-en-8-ol	0.336	0.581	0.854	0.345	0.617	0.577	
Totals	0.577	0.516	0.571	1.146	0.462	0.571	

Table 55.1 (continued)

Compounds	Concentration (mg/L)					
	7	8	9	10	11	12
Furans						
5-Methyl-2(3H)-furanone	0.021	n.i.	0.012	0.033	0.004	n.i.
Furfural	0.057	0.134	0.126	n.i.	0.086	0.352
5-Methyl-2-furancarboxaldehyde	0.077	0.020	n.i.	n.i.	n.i.	0.048
2-Furanmethanol	n.i.	n.i.	n.i.	n.i.	n.i.	n.i.
5-Hydroxymethyl-2-furancarboxaldehyde	0.038	n.i.	n.i.	n.i.	n.i.	0.010
Totals	0.193	0.154	0.138	0.033	0.09	0.41
Acids						
Acetic acid	0.022	0.221	n.i.	0.211	0.066	0.215
Propanoic acid	0.025	n.i.	0.036	0.035	n.i.	n.i.
2-Methyl-butanoic acid	n.i.	0.003	0.029	0.018	0.008	0.021
Octanoic acid	0.024	n.i.	n.i.	0.009	n.i.	0.008
Totals	0.071	0.224	0.065	0.273	0.074	0.244
Sulfurs						
Dimethyl sulfide	2.125	0.051	0.501	0.195	0.838	0.475
Totals	2.125	0.051	0.501	0.195	0.838	0.475

n.i. Not identified

55.3.1 Esters

Esters contribute to the "fruity" and "floral" sensory properties, along with sweet, pineapple, apple, and rose odors [4, 7]. The total concentration of the esters in all samples (100.6 mg/L) was greater than the other volatile compounds accounting for 55 % of the total volatile compounds, so esters were the most important volatile compounds in the hawthorn juices. According to the results reported by Gao [5], esters were also the most abundant volatiles in the hawthorn fruits accounting for 78.41 % of the total volatile compounds. As seen in Table 55.1, a total of 16 esters were detected in the hawthorn juices and ethyl acetate, butyl acetate, methyl benzoate, and 1,2-ethyl-3-hydroxyhexyl 2-methylpropanoate were the most common compounds.

The total ester content of the juice ranged from 4.879 to 18.26 mg/L. The highest ester contents were found in sample 7 (18.26 mg/L) and the sample with lowest ester contents was sample 6 (4.879 mg/L). In the current study, ethyl acetate was present at levels of 2.329–9.027 mg/L, the most abundant volatile ester in practically all samples. Methyl acetate was second most abundant ester in the samples accounting for 12.2 % of the total volatile esters; its largest concentration was found in sample 3, while the lowest was with sample 9. The other important esters detected in most samples included 2-methyl-1-butanyl acetate (0.098–3.278 mg/L) and 3-methyl-1-butanyl acetate (0.01–3.486 mg/L).

55.3.2 Alcohols

Alcohols were the second most abundant volatile group in the tested samples. A total of 16 alcohols were identified in the 12 hawthorn juices. Among these compounds, methyl alcohol, 2-methyl-3-buten-2-ol, 3-methyl-1-butanol, 1-pentanol, 1-hexanol, 3-hexen-1-ol, 2-ethyl-1-hexanol, 1-octanol, and 3-methyl-2-buten-1-ol were found in all samples. The total alcohol content of the hawthorn juices ranged widely from 1.886 to 11.589 mg/L. Sample 10 had the highest alcohol content, while, the lowest was detected in Sample 11. Two alcohols, i.e., 3-methyl-1-butanol, methyl alcohol were considered as predominant alcohol compounds (concentration ≥10 mg/L) in all hawthorn juices. Of these, 3-methylbutanol, imparting a fusel-like odor [11], was the most important volatile alcohol due to its highest concentration. Its largest amount (6.04 mg/L) was detected in sample 10.

55.3.3 **Sulfurs**

Dimethyl sulfide was the only sulfur detected in all tested samples which imparted juices unpleasant flavors, such as cooked onion, rubber, garlic, and asparagus. Although the low contents, sulfur plays an important role in the flavor profile of juices because of the low sensory threshold values [10]. The content of dimethyl sulfide varied from 0.051 to 2.125 mg/L. High sulfur contents were found in sample 7, while the sample with low sulfur contents was sample 8.

55.3.4 Terpenes

Terpenes own a very pleasant aroma and low olfactory threshold, so they are easy to perceive, even at low concentrations [16]. A total of eight terpenes were identified in all tested samples including eucalyptol, 3,7-dimethyl-1,6-Octadien-3-ol, p-menth-1-en-8-ol, 1-methyl-4-(1-methylethyl)-3-Cyclohexen-1-ol, 4-methyl-1-(1-methylethyl)-3-Cyclohexen-1-ol, 1-methyl-4-(1-methylethenyl)-cyclohexanol, p-mentha-1,8-dien-6-ol and (Z)-3,7-dimethyl-2,6-octadien-1-ol. Total values of this group showed higher levels in sample 2 (2.08 mg/L) and the lowest level in sample 7 (0.913 mg/L).

3,7-Dimethyl-1,6-octadien-3-ol, 4-methyl-1-(1-methylethyl)-3-cyclohexen-1-ol, p-mentha-1, 8-dien-6-ol, 1-methyl-4-(1-methylethenyl)-cyclohexanol and p-menth-1-en-8-ol were found in all tested samples, which accounted for 92.4 % of the total terpenes. p-Menth-1-en-8-ol was the key terpenes and detected in sample 2 with highest concentrations (0.928 mg/L) in this study.

55.3.5 Aromatic Compounds

A total of 12 aromatic compounds were found in all samples. The total aromatic compounds content of the samples ranged from 0.148 to 1.134 mg/L. Its largest concentration was found in sample 4, whereas, its lowest level was with sample 5. Aromatic compounds contribute to the honey aroma in juice [17]. Although accounting for 2.4 % of the total volatile compounds, aromatic compounds were very important to the sensory characteristics of juice because of their high perception thresholds. Among this group, $\alpha,\alpha,4$ -trimethyl-benzene methanol and 2-methyl-3-phenylpropanal were the predominate representatives according to their higher contents. Only one aromatic compound, i.e., $\alpha,\alpha,4$ -trimethyl-benzene methanol was detected in all samples.

55.3.6 Acids

The volatile acids which usually contribute to rancid and sweaty odors did not damage the flavor of beverages [18]. The total acid content varied between 0.065 and 2.613 mg/L. Sample 5 had the highest acid among all sample, whereas sample 15 had the lowest. In all 20 samples, only 4 carboxylic acids (acetic acid, propanoic acid, 2-methyl-butanoic acid and octanoic acid) were detected using SPME because of their poor affinity with the SPME fiber [11]. Acetic acid, which had the largest content in sample 5 (1.739 mg/L) was the major acids in tested samples. 2-Methyl-butanoic acid was detected in all tested samples except sample 13.

55.3.7 Furans

Fives furans were found by HS–SPME in the hawthorn juices, including 5-methyl-2 (3H)-furanone, furfural, 5-methyl-2-furancarboxaldehyde, 2-furanmethanol, 5-hydroxymethyl-2-furancarboxaldehyde. The total content of these compounds in sample 3 and sample 10 were far less than the other samples. 2-Furanmethanol was only found in sample 1 and 4 by our research. Furfural, responsible for sweet and almond-like aroma [18], could be the major furan compounds in all samples due to its largest content. The largest concentration of furfural was found with sample 1 (0.627 mg/L) and the lowest was in sample 7 (0.057 mg/L).

55.3.8 *Ketones*

The total ketone content of the samples ranged from 0.122 to 0.844 mg/L. Its largest concentration was found in sample 7, whereas, its lowest level was with sample 4.

544 Y. Zhao et al.

A total of four ketones were identified in the 12 hawthorn juices, including 3-octanone, 3-hydroxy-2-butanone, 6-ethyl-7-hydroxy-4-octen-3-one, and 6-methyl-5-hepten-2-one. 3-Octanone and 6-methyl-5-hepten-2-one were common in all samples. The most abundant ketone was 3-octanone which had a concentration of 0.013–0.814 mg/L in these samples. The highest concentration was detected in sample 7 and the lowest was in sample 1.

55.4 Conclusions

This research provides much information about the volatile compositions of 12 hawthorn juices. A total of 61 volatile compounds were identified by GC–MS in all tested samples, including 16 alcohols, 12 aromatic compounds, 11 esters, 8 terpenes, 5 furans, 4 carboxylic acids, 4 ketones, and 1 sulfur. 21 volatiles were detected in all samples.

Esters and alcohols made great contributions to the total volatiles comparing with ketones, acids, furans, sulfurs, terpenes, and aromatic compounds. Ethyl acetate, 3-methylbutanol, 3-octanone, acetic acid, furfural, dimethyl sulfide, and *p*-menth-1-en-8-ol were suggested to the predominant volatile compounds in all samples according to their higher concentrations. For most volatile compounds detected in the current study, their content in sample 7 was higher than other samples.

Generally, the volatile compositions of the 12 hawthorn juices tested in this work were rather different and the major reasons for the differences among these hawthorn juices were due to the hawthorn species. The hawthorn fruits of different species can be distinguished according to the valuable information provided by the study.

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Chapter 56 Optimization of Extraction Conditions for Crude Antibacterial Proteins/Peptides from *Clarias gariepinus* By-products

Yan Wang, Yunxia Xu, Junyan Mei, Chengxun Chen and Xiaomei Wang

Abstract In this paper, acetic acid was used for extracting crude antimicrobial proteins/peptides from by-products of healthy Clarias gariepinus reared at high stocking density. For the extraction process, three variables, namely the concentration of acetic acid, extraction time, and the ratio of liquid to solid were optimized by using orthogonal design $L_0(3^3)$ with the yield and antibacterial activity of the crude proteins/peptides extracts as test indices. The yields of extracted crude proteins were analyzed by range analysis. The results showed that the yield of crude proteins/peptides was different among nine groups, the highest was group 7, producing 1.5842 g (dialyzed in 1,000 MWCO and lyophilized) per 100 g raw material. All the nine extracted samples had similar inhibitory activity against Aeromonas hydrophila, Escherichia coli, Staphylococcus aureus, Citrobacter sp., and Microbacterium sp. except that groups 7 and 9 exhibited higher inhibitory activity against A. hydrophila and E. coli; moreover, the inhibitory activity of group 7 was higher than group 9 against the two bacteria strains. Nine groups did not display obvious different patterns on the SDS-PAGE gel. Based on these results above, the final conclusion was that group 7, namely 10 % acetic acid, 12 h extraction, and 5:1 liquid to solid ratio was the best combination for isolating the crude antimicrobial proteins/peptides from by-products of *C. gariepinus*.

Keyword *C. gariepinus* · By-product · Antimicrobial proteins/peptides · Yield · Acetic acid · Bacteriostatic test · Electrophoresis

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548 Y. Wang et al.

56.1 Introduction

The first antibacterial protein is reported and named cecropin by Steiner [1]. It is becoming clear that antimicrobial proteins or peptides are an abundant and diverse group of molecules produced by many tissues and cell types in various organisms and are an important component of the innate immune system of all species of life [2, 3]. Such proteins or peptides display stable physical and chemical properties, broad-spectrum antimicrobial activity, and do not induce bacterial resistance.

Fish are aquatic creatures, which are constantly exposed to relatively high concentrations of bacteria, virus, and fungi, many of them that may be harmful to the fish. Under normal conditions fish maintain a healthy state by especially heavily relying on innate defenses for initial protection against pathogen invasion [4]. Antimicrobial proteins or peptides are among the earliest developed molecular effectors of innate immunity and are significant in the first line of host defense response of diverse invaders [5]. It is generally acknowledged that raising fish at relatively high stocking densities means maximizing the usage of fish production infrastructure and is an effective measure for improving farm profitability [6]. However, this results in disease outbreaks in farmed fish and causes significant losses in aquaculture. To avoid economic loss caused by disease, chemicals and antibiotics are being used as prophylactic and therapeutic agents. The inevitable residues of the agent in the end product, in addition to being harmful to consumer's health, the use of antibiotics can also result in the development of resistant strains of bacterial pathogens [7]. The evolution of resistance of pathogenic bacteria to currently available antibiotics has stimulated enormous efforts to develop alternative antimicrobial agents from alternative natural sources to prevent and control disease [8]. Because antimicrobial proteins or peptides have some properties, especially, they kill bacteria very rapidly and their mode of action is considered unlikely to lead to development of resistance, they may be possibly exploited as promising candidates for new antibiotics [9]. However, because of extraction of natural antibacterial proteins or peptides from tissue with lower yield and higher processing cost, researchers need to make heroic efforts at better yields and cost reduction for the wide application of natural antibacterial proteins or peptides.

African catfish, *Clarias gariepinus* is native to the River Nile of Africa and was introduced to China in 1981. It has become an economically important aquaculture species in China owing to its fast growth and high disease resistance. The latter property of this fish might be attributed to its innate immunity system. In previous work published from our laboratory, it was mentioned that many bioactive antibacterial compounds, obtained with gradient ammonium sulfate fractionation, presented in skin mucus, skin, gill, suprabranchial organ, and intestine of *C. gariepinus* [10]. In contrast to other cultured fish, *C. gariepinus* has little intermuscular bones, processing products of muscles of the fish have good color and elasticity, so a certain proportion of commercial size individuals have been circulated to aquatic products processing factories and there are a large number of by-products during

the process. The purpose of this study was, by applying acetic acid as extract solvent, to gain a better protocol for isolating crude antibacterial proteins/peptides from by-products of *C. gariepinus*.

56.2 Materials and Methods

56.2.1 Fish and Sample Collection

Healthy *C. gariepinus* obtained from Deren aquaculture center, Tianjin, China, was maintained in static aerated tanks at a stocking density of 180 kg/m^3 at temperature of 25 ± 1 °C. Fish weight were 359.7–588.9 g at the time of sampling.

By-products including gill, suprabranchial organ, and viscera, such as hepatopancreas, spleen, kidney, digestive tract, and head kidney were collected and pooled for proteins/peptides isolation. Before sampling, experimental fish were anesthetized with a sub-lethal dose of tricaine methanesulphonate (MS222, sigma). The organs obtained above were flushed gently with cold sterilized physiological saline and intestinal contents were flushed out gently. The collected samples were stored at $-20~\rm ^{\circ}C$ for the subsequent protein extraction.

56.2.2 Extraction of Crude Antibacterial Proteins/Peptides

Acetic acid was used for extracting crude antibacterial proteins/peptides from *C. gariepinus* by-products. In this study, the concentration of acetic acid, extraction time, and ratio of liquid to solid levels were chosen for independent variables. Orthogonal design L9 (3⁴) [11] was used to optimize the extraction variables and the actual values of coded levels of each variable are shown in Table 56.1. Crude proteins/peptides were isolated from *C. gariepinus* by-products according to the procedure reported by Ji et al. [12] with some modifications. 100 g of *C. gariepinus* by-products were homogenized in a certain volume and concentration of acetic acid, the homogenates were placed at 4 °C for a few hours and stirred frequently at the same time. After that the homogenates were centrifuged at 8,000 rpm for 30 min at 4 °C. The pellets were weighed and treated with acetic acid using the same protocol mentioned above. The supernatants obtained twice were mixed, incubated

Table 56.1 Three independent variables and their coded and actual values in the orthogonal design

Independent variables	Symbol	Cod	els	
		1	2	3
Concentration of acetic acid (%)	A	1	5	10
Extraction time (h)	В	12	24	36
The ratio of liquid to solid (mL/g)	С	1	2	5

550 Y. Wang et al.

in a water bath at 80 °C for 20 min and centrifuged at 8,000 rpm for 30 min at 4 °C for removing high molecular weight proteins. The supernatants were adjusted pH with 1 % NaOH to 7 and then lyophilized at 5 mbar and -60 °C (Alpha 2-4, Marin Christ Company, Germany). The lyophilized crude protein/peptide extracts were redissolved in sterile deionized water and dialyzed in 1,000 MWCO (molecular weight cut-off) dialysis bag against sterile deionized water for at least 24 h at 4 °C with tenfold changes of dialysis water. Crude protein/peptide extracts obtained were lyophilized again, then weighed and kept at -20 °C. The data of the yields of the crude proteins/peptides were analyzed using range analysis method [11].

56.2.3 Antibacterial Activity Testing

Aeromonas hydrophila and Escherichia coli stored in Tianjin Aquatic Animal Infectious Disease Control and Prevention Center (Tianjin, China), Staphylococcus aureus purchased from Guangdong Microorganism Germplasm Resource Bank (Guangzhou, China), Citrobacter sp. and Microbacterium sp. provided by Prof. Liying Sui of Tianjin University of Science and Technology (Tianjin, China) were used as the main test bacteria. All strains were maintained in LB media with 30 % glycerol at -80 °C. Before the experiments, strains were sub-cultured in LB media at optimal temperature to logarithmic phase, harvested by centrifugation at 5,000 rpm for 15 min at 4 °C, washed twice with sterilized physiological saline and resuspended, and finally counted using a Neubauer hemocytometer.

The lyophilized crude proteins/peptides extracts were resuspended in sterilized physiological saline, respectively, to give a concentration of 200 mg/mL. The bacteriostasis tests were conducted according to the procedure described previously [10]. Briefly, 20 mL of 1.5 % (w/v) agar in LB medium were poured into sterile petri dishes and a layer was formed. The bacterial suspension was added aseptically to 5 mL agar medium at 45 °C to a final concentration of approximately 10⁶ cfu/ mL, then mixed well and poured immediately to form the second layer. After hardening, sterile Oxford cups (7.8 mm outer diameter and 6 mm inner diameter) were placed on this bilayer medium. 200 µL of resuspended samples were separately pipetted into Oxford cups and allowed to diffuse for 12 h at 4 °C. After incubation for 24 h at appropriate temperatures for bacteria growth, such as E. coli and S. aureus at 37 °C, others at 30 °C, the activity was evaluated based on the diameter of the inhibition zone. 200 µL of sterile water and 200 µL of 25 µg/mL oxytetracycline hydrochloride were used as negative and positive controls, respectively, for all assays. All inhibition assays and controls were carried out in triplicate.

56.2.4 Electrophoresis

SDS-polyacrylamide gel electrophoresis of Laemmli system was performed according to the method of Xia et al. [13] with some modifications. Acrylamide-bisacrylamide (30 % T, 2.67 % C) stock solution was used for making 15 % separating gel and 4 % stacking gel. 20 μg/μL crude proteins/peptides was mixed in equal volumes with protein loading buffer (4 % SDS, 8 % mercaptoethanol, 20 % glycerol, 0.05 % bromophenol blue, 100 mM Tris-HCl, pH 6.8), boiled for 10 min, and then loaded 10 μL supernatant to each well of the gel. Electrophoresis started with an initial current of 11 mA until the dye formed a line and maintained at this current for 20 min, and then ran at current of 24 mA until the dye front migrated to 1 cm from the bottom of the gel. After that, the gel was stained with 0.05 % Coomassie brilliant blue R-250 in 10 % acetic acid for 2 h and destained in 10 % acetic acid until clear bands were observed. The gel was transferred to deionized water for removing the acetic acid. The gel image was photographed with GeneSnap Version 6.0 software in the gel image system (GENEGENUSTM, Syngene Company, UK).

56.3 Results

Nine samples of crude proteins/peptides extracted from by-products of *C. gariepinus* with acetic acid were obtained, and termed as group 1 to group 9 in order. The yields of the 9 groups and the range analysis results on yields are shown in Table 56.2.

In this paper, all of nine samples at concentration of 200 mg/mL exhibited antibacterial activity against all of the five tested bacterial strains (*A. hydrophila*, *E. coli*, *S. aureus*, *Citrobacter* sp. and *Microbacterium* sp.) by producing clear zones of inhibition, but all of the diameters of inhibition zone were the same as outer diameter of Oxford cup (results not shown) except that groups 7 and 9 exhibited bigger zones of inhibition against *A. hydrophila* and *E. coli* as shown in Fig. 56.1. The diameters of zone inhibition against *A. hydrophila* and *E. coli* produced by group 7 were 12.11 \pm 0.27 and 10.10 \pm 0.59 mm, respectively, while those produced by group 9 were separately 9.40 \pm 0.43 and 9.35 \pm 0.29 mm.

The SDS-PAGE profiles of *C. gariepinus* by-products extracts showed a broad range of proteins/peptides, including those of small proteins with molecular weights from 4.1 to 6.5 kDa that were abundant in each sample (Fig. 56.2).

56.4 Discussion

The ability of fish to avoid infection depends on their mechanisms of innate immunity. Antibacterial proteins or peptides are one of the key innate immune factors. *Clarias gariepinu* possess high disease resistance and fast growth rate at high stocking density, and bioactive antibacterial proteins or peptides exist in

552 Y. Wang et al.

			*		
Group no.	Coded level	ls			Yield (dialyzed
	A	В	C	Blank	lyophilized weight, g)
1	1	1	1	1	0.5028
2	1	2	2	2	0.626
3	1	3	3	3	0.4931
4	2	1	2	3	0.7002
5	2	2	3	1	1.3543
6	2	3	1	2	0.6564
7	3	1	3	2	1.5842
8	3	2	1	3	0.7692
9	3	3	2	1	0.7970
K_1	1.6219	2.7872	1.9284	2.6541	
K_2	2.7109	2.7495	2.1232	2.8666	
K_3	3.1504	1.9465	3.4316	1.9625	
\overline{K}_1	0.5406	0.9291	0.6428	0.8847	
\overline{K}_2	0.9036	0.9165	0.7077	0.9555	
\overline{K}_3	1.0501	0.6488	1.1439	0.6542	
R	0.5095	0.2802	0.5011	0.3014	
Best levels	A3	B1	C3		

Table 56.2 The yields of crude proteins/peptides and range analysis results on yields

 K_1 , K_2 and K_3 mean the sum of yields at the same level of each variable. \overline{K}_1 , \overline{K}_2 and $\overline{K}_3 = K_1$, K_2 and K_s is divided respectively by n, here n = 3. R means range each variable, $R = \overline{K}_{max} - \overline{K}_{min}$

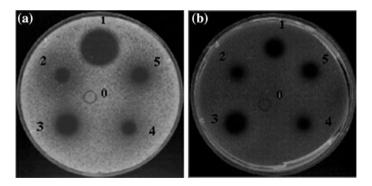


Fig. 56.1 Antimicrobial activity of crude proteins/peptides isolated from Clarias gariepinus byproducts against Aeromonas hydrophila (a) and Escherichia coli (b). Number 0 and 1 negative (0) and positive control (1), Number 2-5 crude proteins/peptides samples, group 6-9

various tissues of the healthy bodies [10, 14]. Such fish are very good sources of antimicrobial proteins or peptides.

It is clearly noted that acid soluble proteins/peptides of C. gariepinu by-products showed different yields among 9 groups. The highest yield of the crude proteins/

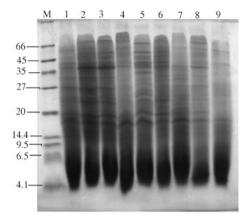


Fig. 56.2 Coomassie brilliant blue R-250 stained SDS-PAGE of soluble proteins extracted with acetic acid from *C. gariepinus* by-products. *Lane M* broad range protein markers. *Lane 1–9* crude proteins/peptides samples, group 1–9 extracted under the conditions shown in Table 56.2. Numbers on the *left side* of the gel correspond to the molecular weight of the markers in kDa

peptides extracted with group 7 condition (10 % acetic acid and 5:1 liquid to solid ratio for 12 h), was 1.5842 g (dialyzed in 1,000 MWCO and lyophilized) per 100 g raw material. According to the range analysis (Table 56.2), the values of R_A were greater than R_C and both were greater than R_D . This indicates that the concentration of acetic acid was the most influential factor, and the ratio of liquid to solid was the second one for yields of crude extracts. While the value of R_B was less than R_D , it means that the influence of extraction time on yields could not be proven in this study.

In this paper, groups 7 and 9 exhibited higher inhibitory activity against *A. hydrophila* and *E. coli*; moreover, the inhibitory activity of group 7 was higher than group 9 against the two bacterial strains. The observed variation in antimicrobial activity among 9 samples extracted under different conditions from same raw material in this paper was also observed in other organisms in previous research findings. For example, Jin et al. [15] reported that for *Rana chensinensis*, the highest antibacterial activity was from the sample extracted with pH 3.0 acetic acid at 0 °C and 22.5 of liquid to solid ratio for 24 h. Antibacterial peptides could be obtained from blue mussels (*Mytilus edulis*) by using 0.5 % acetic acid at 4 °C and 1 of liquid to solid ratio for 12 h [16]. These results indicate that different extraction protocols are suited for different organisms for extracting active antibacterial proteins or peptides.

From the image of the SDS-PAGE, it is obvious that there were a number of bands in each line, which were due to the extract samples not being subjected to any purification process. The homogeneous protein or peptide is obtained and the inhibitory activity against bacteria is improved through purification of the crude antibacterial proteins/peptides extracts [17, 18]. Further work is under progress in this laboratory for purification of active proteins or peptides and analysis of

554 Y. Wang et al.

physical and chemical stability and antimicrobial activity of the purified samples, and the results will be reported in the near future.

According to the results of the yield of crude extracted proteins/peptides, anti-bacterial activity testing as well as SDS-PAGE in the present study, group 7 (10 % acetic acid, 12 h extraction and 5:1 liquid to solid ratio) had the best combinations for isolating the crude antimicrobial proteins/peptides from the by-products of C. gariepinus.

To our knowledge, this study is the first report on optimization of extraction conditions and bacteriostatic analysis of crude antimicrobial protein/peptide from by-products of healthy *C. gariepinus* in China. This study provides a scientific evidence, in other words, *C. gariepinus* can serve as a valuable source of antimicrobial proteins or peptides.

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Chapter 57 Screening, Isolation, and Identification of *Bacillus coagulans* C2 in Pu'er Tea

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Abstract Pu'er Tea is a popular post-fermented tea and has a variety of health benefits. Many microorganisms, especially lactobacillus, exist and play an important role in the fermentation process of Pu'er tea. In this study, we try to screen some lactobacillus by plate method from the Pu'er tea, and strain C2 with significant cycle isolated from Pu'er tea. 16S rRNA gene sequencing results showed that strain C2 had 99 % sequence similarity to that of the type strain of three *Bacillus coagulans* (NR 102791.1, CP002472.1, NR 115727.1), and the results show that methyl red test and V&P test were positive. Sugar fermentation test suggested that strain C2 produced acid but not gas, which is identical to *B. coagulans*. This study confirmed that lactic acid bacteria play a role in Pu'er tea fermentation.

Keywords Pu'er tea · Bacillus coagulans · 16S rRNA · Lactic acid bacteria

57.1 Introduction

Pu'er tea is a distinctive post-fermented tea produced mainly in Yunnan province, China. Modern clinical medicine has shown that Pu'er tea has a variety of healthcare functions in the human body, like lowering blood pressure, [1] reducing weight, [2] moderating the risk of cancer, [3], and preventing cardiovascular disease [4]. The large leaves of *Camellia sinensis* (*C. sinensis assamica*) were used as raw material, and it was further processed by a manufacturing technology called 'pile fermentation'. Complex biological transformations were implemented in this important fermentation process, Pu'er tea is characterized by its unique color, savory flavors, and aroma.

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Previous studies have shown that many microorganisms play an important role in the fermentation process of Pu'er tea. Bacterial and fungal communities involved in Pu'er tea pile fermentation have also been reported, but the results obtained are different [5, 6]. A large number of studies show that fungi including yeast, Aspergillus sp., and Penicillium sp. were the dominant fungi in Pu'er tea samples. [7, 8]. Jeng [9] identified numerous Actinomyces such as Actinoplanes sp. and Streptomyces sp. in Pu'er tea, Candida sp. and Aspergillus sp. [10, 11] were also separately found to be dominant microbes in Pu'er tea. However, a metagenomic study [12] of Pu'er tea shows that the dominant microbes of Pu'er tea fermentation were bacteria and yeasts rather than molds accounting for the overwhelming majority of Eukaryota. These differences in microorganism communities are due to the different Pu'er tea samples [13]. In the previous study, several lactobacillus were isolated in the Pu'er tea which indicated that it played a role in the pile fermentation process. Lactobacillus is a major part of the lactic acid bacteria groups, which could convert lactose and other sugars into lactic acid. The production of lactic acid makes its environment acidic, which inhibits the growth of some harmful bacteria. In the human gastrointestinal tract, they could make up a small portion of the gut flora to regulate intestinal health [14]. In this study, we explored the useful lactic acid bacteria in Pu'er tea, Bacillus coagulans C2 was isolated, and the physiological and biochemical characters were identified.

57.2 Materials and Methods

57.2.1 Pu'er Tea Sample Collection

Pu'er tea samples of days 0, 15, 30, and 45 during pile fermentation were obtained from the courageous tea reprocessing factory in Yunnan, China. Each tea samples of different fermentation time were first soaked with sterile water for 30 min, and then centrifuged at 5,000 rpm for 10 min to remove the leaf residue. After filtration using six layers of gauze filter, the liquor samples containing the microbe, were used to isolate the probiotics.

57.2.2 Separation Probiotics

MRS medium, used to isolate the probiotics in the Pu'er tea, was purchased from Shanghai source Biotechnology Co. Ltd., which contains 2 % calcium carbonate and the final concentration of 10 ng/mL cycloheximide. Each tea liquor sample of 200 μ L was coated on the solid MRS culture medium, and cultivated under 30 °C in an anaerobic incubator (Yiwu Refrigeration Factory, Zhejiang, China). After 48 h, the isolated clones with dissolved calcium which might produce acid was selected,

and was cultured in MRS medium at 30 °C in an anaerobic incubator for further screening and identification. The isolated microorganisms were maintained in agar slants consisting of MRS medium supplemented with 2 % (w/v) agar.

The strain was by Gram staining test [15] and bacteria morphology was observed using a microscope.

57.2.3 DNA Extraction

Each isolated microbe with 5 mL medium was centrifuged at 12,000 rpm for 1 min to collect microbial cell. The cell residues were suspended by 750 μL CTAB-SDS lysis buffer at 55 °C for 1 h. The cell residues were suspended by 500 μL TE buffer (10 M Tris-HCl, 1 M ethylence diamine tetraacetic acid disodium salt), the cell lysis by 6 μL RNase enzymes (10 $\mu g/\mu L$), 30 μL lysozyme at 37 °C for 1 h, 30 μL 10 % SDS, 10 μL proteinase K (20 $\mu g/\mu L$) at 55 °C for 15 min, 100 μL 5 M NaCl, 80 μL CTAB/NaCl (10 % CTAB, 0.7 M NaCl) at 65 °C for 10 min. The DNA of the microbial cell was extracted by 750 μL phenol/chloroform/isoamyl alcohol (25:24:1), the supernatant of 300 μL was added to precooling of isopropyl alcohol and stored at -20 °C for 1 h. and centrifuged 12,000 rpm for 10 min. The residues was washed with 200 μL of 70 % ethanol twice, and then dissolved with 40 μL sterile water and stored at -20 °C. The extracted DNA was loaded on agarose and electrophoresed at 100 V for 30 min.

57.2.4 16S rRNA Sequencing of the Strain C2

Identification of the isolates was confirmed by 16S rRNA sequencing. DNA of the strain C2 was used as DNA template, the universal primer 27F (AGAGTTTGA TCCTGGCTCAG) and 1492R (TACGGCTACCTTGTTACGACTT) was used for amplification of 16S rRNA of these isolated microbes, and the product was approximately 1,600 bp. The PCR conditions were as follows: pre-heating at 94 °C for 5 min, 30 cycles of denaturation at 94 °C for 30 s, annealing at 55 °C for 30 s and extension at 72 °C for 2 min, and a final extension for 10 min at 72 °C. The PCR fragments were purified by agarose gel electrophoresis and PCR purification kit (TaKaRa Biotechnology Co. Ltd., China), and were ligated to the linearized vector pMD18-T (TaKaRa Biotechnology Co. Ltd., China). The ligated products were transformed into competent Escherichia coli DH5α cells and coated on the LB medime plate, which contained 100 μg/μL ampicillin. After culture at 37 °C overnight, the positive clone with correctly inserted PCR fragment was selected by PCR and was sequenced by Shanghai Invitrogen Technology Co. Ltd., China. The DNA sequences were analyzed using the BLAST program (http://blast.ncbi.nlm. nih.gov/Blast.cgi).

560 C. Feng et al.

57.2.5 Physiological and Biochemical Identification

The strain C2 was cultured for 12 h. Sugar fermentation test, [14] methyl red reaction test, [15] V&P test [16] detected the physiological and biochemical characters of the strain C2.

57.3 Result

57.3.1 Isolation of LAB in Pu'er Tea

After being cultured at 37 °C for 48 h, about 25 clones showed a significant dissolved calcium cycle. Clone C2 shows the largest dissolved calcium cycle at the first and second screenings as shown in Fig. 57.1a. Gram staining test shows that C2 is Gram-positive bacteria (Fig. 57.1b).

57.3.2 DNA Extraction and 16S rRNA Identification

The DNA of strain C2 was successfully extracted and the correct size is shown in Fig. 57.2a. With strain C2 genome DNA as template, 16s rRNA sequence was successfully amplificated, and the PCR products verified by electrophoresis detection with nonspecific amplification (Fig. 57.2b). The PCR products were purified and ligated into PMD18-T vector, the positive clone was selected by PCR method, and the rRNA sequence of strain C2 was approximately 1,600 bp. Based on BLAST analysis, strain C2 shared 99 % identity with 16s rRNA sequence of

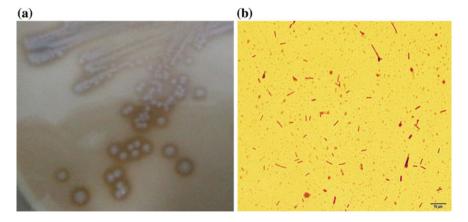
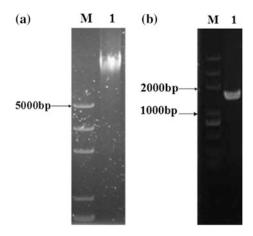


Fig. 57.1 Observed the morphology of strain C2. a Dissolved calcium *circle* of the culture medium. b The strain morphology was observed under a microscope

Fig. 57.2 DNA extraction and 16s rRNA identification. a DNA extracted from strain C2. b PCR product of 16s rRNA gene sequence of strain C2



B. Coagulans, (NR 102791.1, CP002472.1, NR 115727.1), which suggests strain C2 belongs to the *B. coagulans*. It was named *B. coagulan* C2 (KM980629).

57.3.3 Physiological and Biochemical Identification

Physiological and biochemical identification of strain C2 included sugar fermentation test, methyl red reaction test, V&P test, etc. The result shows methyl red test and V&P test was positive, sugar fermentation test suggested that the strain C2 produced acid but not gas. The experimental data are given in Fig. 57.3.

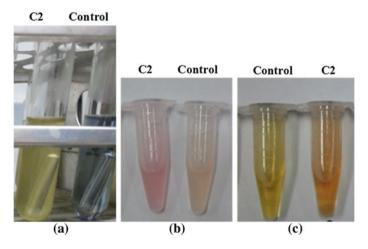


Fig. 57.3 The physiological and biochemical characteristics of strain C2. a Sugar fermentation test of strain C2. b Methyl red test of strain C2. c V&P test of strain C2

562 C. Feng et al.

57.4 Conclusion

A large number of microorganisms in Yunnan's Pu'er tea has a very important effect on the human body. Probiotics in Pu'er tea plays a very important role. In our previous work, we were separated from probiotics in Pu'er Tea. We successfully isolated a strain C2. Strain C2 16s rRNA sequence by comparing gene bank, the strain C2 was identified as *B. coagulans*. The similarity is 99 %.

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Chapter 58 Production and Identification of Antifungal Compounds Produced by *Bacillus subtilis* B579

Fang Chen, Yu Zheng, Jianmei Luo, Deduo Han and Min Wang

Abstract Biological control has become an important approach to suppress many pathogens. Bacillus subtilis is considered to be an excellent biocontrol agent not only due to its ability on inducing plant systematic resistance, but also on producing various hydrolytic enzymes and antibiotics. In this study, polymerase chain reaction (PCR) was used to detect the 12 genes related to the antifungal compounds biosynthesis. Six genes were detected that exist in the genome DNA of B579 by the sequence homology analysis. Five genes were related to biosynthesis of lipopeptide antifungal compounds, and one gene was related to biosynthesis of protein antifungal compounds. Lipopeptide antifungal compounds were obtained from the supernatant of B579 using the method of acid deposition and methanol extraction. Two homologies with the molecular weight of m/z 1043.59 and m/z 1057.35 were detected in the lipopeptide antifungal compounds, which had the similar molecular weight with iturin A. Three homologies with the molecular weight of m/z 1008.32, m/z 1022.06, and m/z 1036.13 were detected in the lipopeptide antifungal compounds, which had the similar molecular weight with surfactin. The antifungal compounds produced by B579 could be of a good prospect for being used as a new tool for biological control.

Keywords Bacillus subtilis · Production · Identification · Antifungal compounds

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58.1 Introduction

Soilborne disease was considered to be the major cause of yield loss on vegetable crops all over the world [1]. Pathogenic fungi, such as *Fusarium oxysporum* and *Rhizoctonia solani*, are soilborne and worldly distributed in many soil types [2]. They could infect plants through the roots, and eventually resulted in the wilt of entire plants. They could grow and survive in soil for long periods about 10–15 years as pathogenic saprophytes to many plant species, including cucumber, pepper, tomato and potato, throughout vegetable growing regions [3–6].

Synthetic chemical fungicides have long been used as active agents in reducing the incidence of soilborne diseases. However, they are costly, can cause environmental pollution, and may also induce pathogen resistance to chemical fungicides. Considering the limitations of chemical fungicides, it seems appropriate to search for a supplemental control strategy which was environmental friendly and safe to human being [7]. Biological control, the use of microorganisms or their secretions to prevent soilborne diseases, offers an attractive alternative or supplement to chemical fungicides for the management of soilborne plant diseases without the negative impact of chemical fungicides. It could also avoid the genetic resistance of pathogens to chemical fungicides. Therefore, biological control tactics have become an important approach to facilitating sustainable agriculture.

As Bacillus subtilis has the characteristics of omnipresence in soils, thermal tolerance, rapid growth, and ready formation of resistant spores, it is considered to be a good biological control agent. B. subtilis has a direct antagonistic activity not only by inducing plant systematic resistance [8], but also by producing various hydrolytic enzymes (for example, chitinase, β-1,3-glucanase) and antibiotics [9]. From the large amount of antimicrobials produced, lipopeptides stand among the most representative. Bacillus lipopeptides may be divided into three families iturin, fengycin, and surfactin. These antifungal lipopeptides are either linear or cyclic. They frequently contain amino acid residues, which are unique and not commonly found in proteins, with high stability to pH, heat, and protease [10]. The specific hydrophobous section of the antifungal peptides could insert to the cell wall of hyphae, and resulted in the leakage of cytoplasm. Antifungal peptide TasA produced by *B. subtilis* is a translocation-dependent antimicrobial spore component. It could inhibit the growth of some relative bacteria [11]. These antifungal peptides have been proved safe to people and no pollution to environment [12, 13]. So, they have high potential for being used in biological control, food antisepsis, medicine, and so on.

Bacillus subtilis B579 (deposition number CGMCC No. 2270) was isolated from rhizosphere of cucumber in Tianjin, China. It could effectively inhibit the growth of 13 species plant pathogenic fungi, and exhibited a broad spectrum antifungal ability [14]. The primary objective of this study was to identify the antifungal compounds produced by B579. Genes involved in the biosynthesis of four families of antibiotics, surfactin, iturin, fengycin and TasA, were amplificated by polymerase chain reaction (PCR) with the genomic DNA of B579 as template. Lipopeptide antifungal

compounds were obtained from the fermentation supernatant of B579, and were identified through HPLC and molecular weight.

58.2 Materials and Methods

58.2.1 Microorganisms and Growth Conditions

Bacillus subtilis B579 was grown in LB broth at 37 °C in a rotary shaker with 180 r/min for 24 h. They were maintained on LB agar plate at 4 °C. For antibiotic preparation, the synthetic medium was used with (g L⁻¹): dextrose, 10; DL-glutamic acid, 10; MgSO₄·7H₂O, 1.02; K₂HPO₄, 1.0; KCl, 1.0; and 1 mL trace elements (MnSO₄·H₂O, 0.25; CuSO₄·5H₂O, 8; and FeSO₄·7H₂O, 0.0075 in 50 mL medium). The pH was adjusted to 7.5 prior to autoclaving. *B. subtilis* B579 was grown at 37 °C for 48 h with continuous shaking with 180 r/min.

58.2.2 PCR Conditions for Detection Antibiotic Biosynthesis-Related Genes

Genomic DNA of B. subtilis B579 and B. subtilis 168 were extracted by standard protocols [14]. The primers used in PCR were either previously described or designed with sequences obtained from GenBank using Primer Premier 5.0 (Table 58.1). PCR amplifications were carried out in 50 µL reaction mixtures containing 1 × PCR buffer (TaKaRa Inc. China), 2.5 mM MgCl₂, 1.5 U rTaq DNA polymerase (TaKaRa Inc. China), 400 nM of each forward and reverse primers, 200 μM dNTP (dATP, dGTP, dCTP, and dTTP), and 3 μL of template DNA (~ 100 ng of bacterial genomic DNA). The PCR were performed in a thermocycler (Bio-Rad 2001) with an initial denaturation at 94 °C for 5 min, followed by 30 cycles of denaturation at 94 °C for 1 min, annealing at 55 °C for 1 min, and extension at 72 °C for 1.5 min. There was a final extension at 72 °C for 10 min. Each amplification reaction was analyzed by electrophoresis using a 1.2 % agarose gel. Expected PCR products were eluted using the DNA purification kit (DingGuo Inc. China). The amplicons were sequenced by Invitrogen Co. Ltd. (Shanghai, China). The sequences were compared using the BLAST (http://www.ncbi.nlm.nih. gov/BLAST/) for identification of the homology.

58.2.3 Purification and Identification of Lipopeptide Antifungal Compounds

Antifungal compounds were extracted using the method of acid deposition and methanol extraction basically according to Bie et al. [15]. The antifungal activity

Table 58.1 PCR detection of antibiotic biosynthesis genes

Antibiotic	Gene	Sequences	Expected size of PCR product/ detected in B579	References accession number
Fengycin	fenA	F1:AAGGACTCGGCAGAAATGAA R1:TGAATGTGAAGCGCAGGTAG	900 bp/no	CP000560, AJ576102
Iturin	ituB	F1:ATCCAGAATATCCGCAAGAC R1:GCTCGATATGCGGACTCAAT	1479 bp/yes	AB050629, EU263005
Surfactin	srfA	F1:GGAGCAAAGGTGCTTCTCAC R1:GAAACGTTGGCATCAGGAAT	1200 bp/yes	Z99105, BSU03480
TasA	tasA	F1: CACCATGGGTATGAAAAAGAAATT R1: TTAATTTTTATCCTCGCTATGCGCTT	790 bp/yes Z99116, BSU03480	
Bacillomy- cin D	bamC	F1: AGTAAATGAACGCGCCAATC R1: CCCTCTCCTGCCACATAGAG	957 bp/no	AY137375
Bacilysin	bacD	F1: AAAAACAGTATTGGTTATCGCTGA R1: CCATGATGCCTTCGATACTGAT		
Ericin	eriB	F1: GCACAGATGGAAAATCTGAAG R1: GAAAATTGCTCCCCAAATGA	688 bp/no AF233755, BSU09819	
Mersacidi- n	mrsA	F1: GGGTATATGCGGTATAAACTTATG R1: GTTTCCCCAATGATTTACCCTC	579 bp/no AJ250862	
Mycosubti- lin	тусС	F1: AATCAATTGGCACGAACCTT R1: ATCGCCCGTTTTGTACATTC	1026 bp/no	AF184956
Sublancin	sunT	F1: GCTTTGTTAGAAGGGGAGGAAT R1: CTTGTCCCAACCCATAGGATAA	974 bp/yes	AF014938
Subtilin	spaB	F1: GCACAGATGGAAAATCTGAAG R1: GAAAATTGCTCCCCAAATGA	688 bp/no AF233755	
Subtilosin	albF	F1: TCAACAGCTGGATGAACGAAC R1: AGGCGGTATGTTTGCTGTATCT	888 bp/yes	AJ430547

Note "yes" representation the genes were detected; "no" representation the genes was not detected in B. subtilis B579

was determined according to the bioassay method described earlier [16]. The crude antifungal compounds were further purified using reverse phase high-pressure liquid chromatography (HPLC) with a C_{18} column (Agilent 1100 C_{18} semi-preparative column, 21.4 mm \times 250 mm) at 30 °C. The aqueous methanol solution of the active fraction was injected into the column, and eluted with a linear gradient at 1 mL/min. Elution was monitored at 215–280 nm with a diode array UV monitor. Liquid chromatography–mass spectrometry (LC–MS) analysis were conducted for

each active peak, using reverse phase HPLC with a C_{18} column (Agilent 1100 C_{18} semi-preparative column, 4.6 mm \times 250 mm), and electrospray ionization mass spectrometry (ESI–MS) with quadrupole mass analyzer at 30 °C.

58.3 Results

58.3.1 PCR Detection of Antibiotic Biosynthesis-Related Genes

Specific primers were designed according to sequences related to biosynthesis of antagonistic compounds published on GenBank. Twelve genes related to the antifungal compounds biosynthesis were detected (Table 58.1). Six genes were detected that exist in the genome DNA of B579 by the sequence homology analysis [14]. Five genes were related to biosynthesis of lipopeptide antifungal compounds, and one gene was related to biosynthesis of protein antifungal compounds.

The DNA sequences obtained from these amplicons showed high homology to genes accessed in NCBI. Analysis of the 1022 bp PCR product with the ituB-F1/R1 primer pair showed 99 % identity with a region of the iturin biosynthesis operon. PCR products of sunT and albF (934 and 845 bp, respectively) also showed 99 % identity with the genes involved in the biosynthesis of sublancin and subtilosin. Bacilysin biosynthesis-related gene bacD showed 98 % identity. Analysis of the 1153 bp PCR product using the srfA-F1/R1 primer pair showed 98 % identity with a region of srfAA, a gene involved in the biosynthesis of surfactin. Finally, analysis of the 766 bp PCR product with the tasA-F1/R1 primer pair showed 99 % identity to gene involved in the biosynthesis of tasA.

58.3.2 HPLC Analysis of Antifungal Compounds

Lipopeptide antifungal compounds were obtained from the fermentation supernatant of B579 using the method of acid deposition and methanol extraction. Four major active peaks (named 24.05, 29.00, 67.54, and 69.75) were selected for further study (Fig. 58.1).

58.3.3 LC-MS Analysis of Antifungal Compounds

The active peaks of 24.05 and 29.00 were further analyzed, respectively, using LC–MS. Two homologies with the molecular weight of m/z 1043.59 (Fig. 58.2a) and m/z 1057.35 (Fig. 58.2b) were detected in the lipopeptide antifungal compounds, which

568 F. Chen et al.

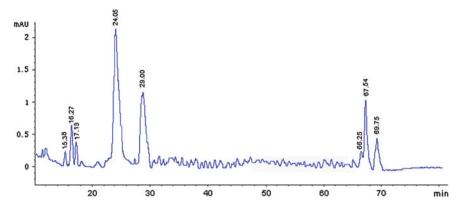


Fig. 58.1 HPLC analysis of lipopeptide antifungal compounds produced by *Bacillus subtilis* B579

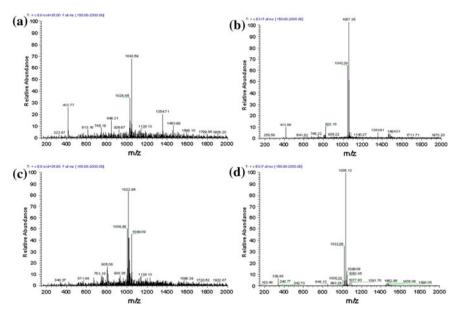


Fig. 58.2 Analysis of lipopeptide antifungal compounds produced by *B. subtilis* B579 using ESI-MS

had the similar molecular weight with the two homologies of iturin A. Two homologies with the molecular weight of m/z 1008.48 and m/z 1022.96 (Fig. 58.2c) were detected by analyzing the active peak 67.54, and the active peak 69.75 showed the molecular weight of m/z 1036.13 and m/z 1022.06 (Fig. 58.2d). The molecular weight of m/z 1008.48, m/z 1022.96, and m/z 1036.13 with one –CH₂ molecular

weight difference in turn, which had the similar molecular weight with three homologies of C_{13} – C_{15} surfactin.

58.4 Discussions

So far, numerous microorganisms with antagonistic ability have been identified, and many have been confirmed effective in field experiments [1–3]. The grampositive bacteria, like *B. subtilis*, have been studied intensively in recent years, because many *B. subtilis* strains are considered to be safe biocontrol agents [17]. So, *B. subtilis* is not only an organism known for protease production [18], but also a good bacteria used for fungicide production. Previous reports by other researchers have shown that antagonistic *B. subtilis* could produce β -1,3-glucanase, chitinase, and antibiotics such as iturin A and surfactin [19]. Iturin is a cyclolipopeptide containing seven residues of α -amino acids and one reside of a β -amino acid. It is a small molecule yet displays strong antifungal activity. The other lipopeptide, surfactin, in contrast has weak antibiotic activity. However, it exhibits a stronger antifungal ability when is synergism with iturin [10].

In the present study, genes involved in the biosynthesis of six antifungal compounds were detected in genomic DNA of B579. Two kinds of lipopeptide antifungal compounds were identified by HPLC and LC-MS. The separation of antifungal compounds and the effect of these antifungal compounds on the morphology of the pathogens are under extensive investigation in our laboratory. The related genes for biosynthesis of bacilysin, sublancin, subtilosin were detected by PCR, but they were not been detected by HPLC and LC-MS which need us for further study. They probably were produced too little to separation and identification, or need to find better conditions for their production and separation.

In our previous studies, *B. subtilis* B579 not only exhibited effective biological control ability against Fusarium wilt of cucumber under greenhouse conditions, but also could induce plant systematic resistance and promote plant growth in biological control of Fusarium wilt [16]. Formulation of powder preparation of B579 endospores had been developed (Chinese patent registration number: 200810153180.1). Some isolates of Bacillus have been successfully commercialized and marketed, such as the Gustafson product Kodiak is widely used for suppression of cotton disease in the US [20]. As the interest in biological control of soilborne plant pathogens has increased considerably in the last few decades, we believe that *B. subtilis* will be used more in production agriculture and horticulture based on the recent progress shown in implementing microbial inoculants.

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Chapter 59 Application of Molecularly Imprinted Polymers in Purification and Separation for Epothilones

Ruicheng Sun, Lin Zhao, Jikun Yang, Naiqiang Wang and Xinli Liu

Abstract A highly effective purification process combined with molecularly imprinted polymers (MIPs) was developed to selectively isolate Epothilones (Epos) from a complex milieu. The MIPs of Epos were successfully synthesized by precipitation polymerization, and the optimal ratio of the template molecule to functional monomer and cross-linker was 1:4:20 (with the molar ratio). Under the optimized condition, the resulted products were characterized by scanning electron microscope (SEM). The binding properties of the MIPs were evaluated by the adsorption kinetics and static adsorption. From the correlation coefficients (R^2) of the fitting models, the equilibrium data fitted well to Freundlich model ($R^2 = 0.9987$), indicating multilayer adsorption. Finally, we successfully applied the MIPs in the solid-phase extraction (SPE) process of Epos, and its purity was improved more than 40 % than the origin. It achieved an effectively selective Epos cleanup procedure, sequentially established a favorable foundation for later research on Epos.

Keywords Molecularly imprinted polymers • Epothilones • Static adsorption • Adsorption kinetics • Solid-phase extraction

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572 R. Sun et al.

59.1 Introduction

Epothilones (Epos, Fig. 59.1) belong to the macrocyclic lactone class of antibiotics, which consist of several structurally related groups: Epothilone A, B, C, D, E, F, and K. They were first described by G. Höfle et al. in 1993, as secondary metabolites synthesized by *Sorangium cellulosum* strain SMP44, which was collected at the banks of the Zambezi river in the Republic of South Africa [1–4]. Epothilones were confirmed to be able to kill dividing cells by stabilizing microtubules at nanomolar and subnanomolar concentrations, and considered as potential advantageous successors to cancer treatment, which can be effective against taxol-resistant cells [5, 6]. Unfortunately, the inefficiency in purification seriously restricted the industrialization of Epos. For the isolation of epothilones, macroporous adsorption resins were added into the medium during fermentation, but some impurities with similar polarities or structures just as antifoaming agents and other hydrophobic component were coinstantaneous adsorbed and hard to be eliminated by later traditional column chromatography. Therefore, developing a fast, convenient, and stable method for the isolation and purification of Epos is expected.

In recent years, molecular imprinting technology (MIT) has attracted considerable attentions and molecularly imprinted polymers (MIPs) have been widely applied in many fields, such as capillary electrochromatography, solid-phase extraction (SPE), artificial enzymes, and chemical sensors. That is because of their outstanding advantages, involving mechanical and chemical stabilities, desired selectivity, as well as relative ease and low cost of preparation [7–13].

In this study, we synthesized the MIPs of Epos by precipitation polymerization, and evaluated them on the morphological characterization and binding ability. Finally, we applied the MIPs in the SPE, achieving an effectively selective Epos cleanup procedure, which remarkably simplified the purification process.

59.2 Experimental

59.2.1 Reagents

Epothilones was obtained from Shandong Provincial Key Laboratory of Microbial Engineering, Qilu University of technology (Jinan, China). Methanol and

Fig. 59.1 Chemical structures of epothilones

chloroform were purchased from Tianjin Chemical Reagent Company (Tianjin, China). Methacrylic acid (MAA) and Ethylene glycol dimethacrylate (EGDMA) were purchased from Aladdin Industrial Corporation (Shanghai, China). 2,2-azoi-sobutyronitrile (AIBN) was supplied by Tianjin Damao Chemical Reagent Factory (Tianjin, China).

59.2.2 Preparation of the MIPs for Epos

The MIPs of Epos were synthesized by precipitation polymerization method. The general synthesis route was shown in Fig. 59.2. Using MAA as functional monomer, chloroform as porogen, EGDMA as cross-linker and AIBN as initiator, a traditional preparation procedure was carried out as follows: A certain amount of Epos and MAA were dissolved in 40 mL chloroform. After being placed for 3 h, EGDMA and AIBN were added. Then the solution was purged with nitrogen for 10 min, sealed, and polymerization took place at 60 °C for 24 h in a thermostat water bath. The obtained polymers were extracted in a Soxhlet apparatus with methanol/acetic acid (4:1, v/v) for 48 h, until no Epos could be detected. Then the polymers were washed with methanol for 24 h to remove the remained acetic acid and dried at 60 °C to reach a constant weight. The NIPs were prepared and treated under identical conditions without template present during polymerization.

Fig. 59.2 Scheme of the synthesis route of MIPs using epothilones as template molecule

574 R. Sun et al.

59.2.3 Morphological Characterization

The morphology of the polymers was observed by a scanning electron microscope (SEM) at 15.0 kV.

59.2.4 Binding Analysis

To evaluate the binding property of the MIPs and NIPs, 15 mg MIPs and NIPs were each added into a 20 mL vial and mixed with 10 mL Epos solution in chloroform. After being vertical shaken at room temperature for 3 h, the supernatant was centrifuged at 10,000 rpm for 10 min, then the concentration of Epos was measured. The amount of Epos which bound to the polymers was counted by subtracting the amount of Epos in the final solution from the initial amount. The adsorption property value Q could be evaluated by the following equation:

$$Q = (C_0 - C) \times V/W \tag{59.1}$$

 C_0 and C were the initial and final concentrations of Epos, V was the solution volume, and W was the mass of polymer.

59.2.5 SPE Experiments

Polypropylene 1 mL SPE cartridge secured by glass-fiber frits was packed with 25 mg of the MIPs or NIPs. The cartridge was conditioned consecutively with 1 mL methanol and 1 mL water. Then, 2 mL Epos dissolved in chloroform were passed through the cartridge at a rate of 0.1 mL min⁻¹, and rinsed with 1 mL methanol/ water (4:1, v/v) to wash off the interferences. Finally, for the elution step, the cartridge was eluted with mixture containing methanol and acetic acid (4:1, v/v) to release the bound analytes. All the fractions were collected; quantification of Epos was carried out by HPLC analysis.

59.3 Results and Discussions

59.3.1 Synthesis of the MIPs

The main factors affecting the adsorption behavior of the MIPs were estimated by the ratios of the template to functional monomer and cross-linker, as shown in Table 59.1.

No.	Epos (mmol)	MAA (mmol)	EGDMA (mmol)	$Q (mg g^{-1})$	IF
MIP1	0.5	1.0	10.0	57	1.78
NIP1	_	1.0	10.0	32	
MIP2	0.5	2.0	10.0	128	2.25
NIP2	_	2.0	10.0	57	
MIP3	0.5	3.0	10.0	112	1.49
NIP3	_	3.0	10.0	75	
MIP4	0.5	2.0	7.5	78	1.73
NIP4	_	2.0	7.5	45	
MIP5	0.5	2.0	12.5	98	1.19
NIP5	_	2.0	12.5	82	

Table 59.1 The MIPs and NIPs of Epos for preparation with different ratios and adsorption capacity

In this study, the imprinting factor IF (IF = $Q_{\rm MIPs}/Q_{\rm NIPs}$) was employed to confirm the optimal synthesis conditions. From the results in Table 59.1, we could see that insufficient amount of functional monomer was not be able to synthesis enough binding sites (MIP1), the overdosing functional monomer lead to association by themselves, forming little effective binding sites (MIP3). Furthermore, low cross-linker amount could not ensure the rigidity of the polymers (MIP4), and high cross-linker amount could promote the formation of nonspecific binding cavities (MIP5).

In conclusion, after survey in the main factors affecting the adsorption behavior to the MIPs, the optimal synthesis ratios were 1:4:20 (template molecule:functional monomer:cross-linker, with the molar ratio), and the imprinting factor IF reached maximum to 2.25.

59.3.2 Morphological Characterization of the MIPs

For both polymers, the particles possed a porous surface due to pores with irregular shapes and sizes, which could play an important role in the adsorption process. In the MIPs, the roughness of the particle surface itself causes the increase in the surface area than the NIPs, which possessed a uniform, compact, and smooth shape. The nonporous structure in the NIPs particles was due to the lack of specific binding sites which were created for Epos-MIPs and suggested that the MIPs had great potential applied for sorbents (SEM, as shown in Fig. 59.3).

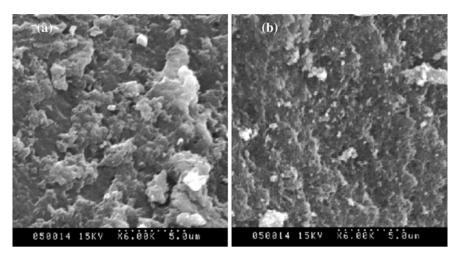


Fig. 59.3 Scanning electron micrographs of the MIPs (a) and NIPs (b)

59.3.3 Adsorption Kinetics of the MIPs

In order to explore the capacity of MIPs, the adsorption kinetics experiments for both MIPs and NIPs were carried out. 15 mg MIPs and NIPs were mixed with 10 mL Epos solution in chloroform, then incubated at room temperature with continuous vertical shaking for 5 h. The initial concentration of Epos was kept in 1.0 mg mL⁻¹, and the amount of Epos bound to the polymers was measured every 30 min. The results were shown in Fig. 59.4. It could be seen that the adsorption rate of MIPs for Epos increased rapidly in the first 30 min, which is mainly because many binding cavities existed on the MIPs surface. Furthermore, the adsorption saturation of MIPs was attained after 3 h, while NIPs was about 1 h, it showed that MIPs had many specificity binding cavities compared with NIPs.

59.3.4 Adsorption Isotherms of the MIPs

Furthermore, the static adsorption experiments of MIPs were also investigated. Compared with the adsorption kinetics experiments, the initial concentration of Epos was ranged from 0.2 to 1.0 mg mL⁻¹. The results were shown in Fig. 59.5. It could be seen that the amount of Epos bound to the polymers were raised with the initial concentration increased.

To explain how the adsorbate distributed on the adsorbent, the adsorption system were designed and established. Two classical isotherm models were used to analyze the equilibrium data: the Langmuir model (Eq. 59.2) and the Freundlich model (Eq. 59.3).

Fig. 59.4 Absorption kinetics of the MIPs and NIPs for Epos

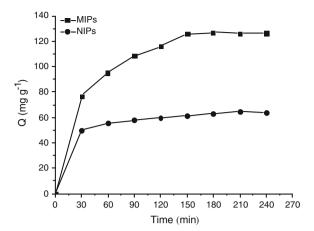
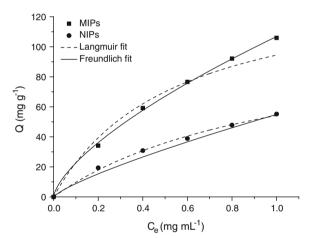


Fig. 59.5 Adsorption isothermic of the MIPs and NIPs for Epos



$$Q = K_{\rm L} Q_{\rm e} C_{\rm e} / (1 + K_{\rm L} C_{\rm e}) \tag{59.2}$$

where $C_{\rm e}$ was the initial concentration, Q was the amount of Epos bound at equilibrium, $Q_{\rm e}$ was the theoretical maximum monolayer capacity, and $K_{\rm L}$ was the Langmuir constant related to the affinity of the active sites.

$$Q = K_{\rm F} C_{\rm e}^{1/n} \tag{59.3}$$

where $C_{\rm e}$ was the initial concentration, Q was the amount of Epos bound at equilibrium, and n and $K_{\rm F}$ are Freundlich constants, which were related to the adsorption favorability and adsorption capacity, respectively.

A comparison of the isotherm models for Epos adsorption onto the MIPs and NIPs using nonlinear regression (Eqs. 59.2 and 59.3) were also presented in Fig. 59.5,

578 R. Sun et al.

Polymers	Langmuir			Freundl	Freundlich		
	$Q_{ m e}$	$K_{\rm L}$	R^2	n	K _F	R^2	
MIPs	145.15	1.86	0.9684	1.49	106.84	0.9987	
NIPs	110.13	0.97	0.9965	1.26	54.95	0.9736	

Table 59.2 Adsorption isothermic parameters for the adsorption of Epos onto MIPs and NIPs

and the parameters obtained from the isotherm models were also listed in Table 59.2. As shown in Fig. 59.5 and Table 59.2, the correlation coefficients (R^2) of 0.9684 and 0.9987 were obtained on the MIPs for the Langmuir and Freundlich models, respectively. While on the NIPs were 0.9965 and 0.9736, respectively. From that we could see the Freundlich model was better than the Langmuir model in describing the adsorption behavior of Epos on the MIPs in terms of the higher correlation coefficients (R^2) values, it demonstrated that the MIPs belongs to multilayer adsorption, they probably exhibited a logarithmic distribution of the binding sites in multilayers [14, 15]. However, for the NIPs, the Langmuir model was better than the Freundlich model in terms of the higher correlation coefficients (R^2) values, which indicated that the NIPs belong to homogeneous adsorption [16, 17].

59.3.5 SPE Analysis for the MIPs

For the Epos extraction, the offline SPE was employed with the MIPs as sorbents, the results were shown in Fig. 59.6. A quantity of 25 mg the MIPs was packed into 1 mL SPE cartridges with two polypropylene frits between the stuffing. The initial complex milieu (the metabolites containing Epos and impurities dissolved in chloroform, the purity of Epos were 40.62 % (in peak area), chromatogram A in

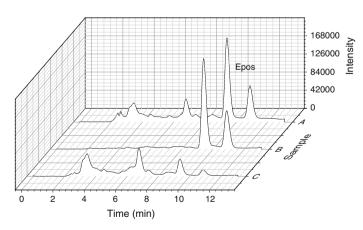


Fig. 59.6 Separation of Epos with the MIPs as a adsorbent in SPE procedure: **a** Initial complex milieu before SPE, **b** final milieu after SPE, **c** the solutions washing impurities

Fig. 59.6) were passed through at a rate of 0.1 mL min⁻¹. After all the steps were completed, collecting the elution fractions and HPLC analysis. The final milieu after SPE was eluted (chromatogram B in Fig. 59.6) successfully separating Epos from the solution. The purity of Epos in the solution was 83.83 %, improving more than 40 % compared with the initial solutions. Although most the interferences were washed off from the Epos solutions, as shown in Fig. 59.6 chromatogram C, there is still a small amount of Epos that was washed off together with impurities, and we will continue to follow up work. From the results we indicated that the MIPs of Epos could be successfully applied as an adsorbent into SPE process for separation of Epos.

59.4 Conclusions

In this paper, we successfully synthesized the MIPs of Epos by precipitation polymerization. The optimal ratio of the template to functional monomer and cross-linker is 1:4:20. Under the optimal composed conditions, a series of experimental analyses including SEM, FT-IR, adsorption kinetics, and adsorption isotherms were carried out to demonstrate the formation of the MIPs for Epos. The results showed that the equilibrium data fitted well to the Freundlich model, indicating multilayer adsorption. Moreover, the MIPs of Epos were successfully applied as sorbent into SPE process, brought about an effectively and simply Epos cleanup procedure.

Acknowledgments The work was financially supported by Research Projects of Science and Technology Division, Shandong (2012GSF12107), and Provincial Post-Doctoral Innovation Foundation (201203006).

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580 R. Sun et al.

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Chapter 60 Identification of Peanut Intersectional Hybrids with SSR Markers

Shuo Meng, Xiu Zhen Wang, Yue Yi Tang, Qi Wu, Quan Xi Sun, Chen Jiang, Chuan Tang Wang and Li Feng Liu

Abstract Wild peanut relatives constitute an invaluable gene source for broadening the narrow gene base of the cultivated peanut. However, severe pre-/post- fertilization obstacles impeded the utilization of incompatible species within the *Arachis* genus. In the present study, through SSR profiling, three true intersectional F_1 hybrids of Qunyu $101 \times Arachis pusilla$ and a true intersectional F_1 hybrid of Huayu $40 \times A$. *pusilla* resulting from post-pollination hormone treatment of flower bases were identified, providing strong evidence for the effectiveness of the in vivo hormone treatment method in overcoming cross-incompatibility in the genus *Arachis*.

Keywords Arachis pusilla · Hybrid · Intersectional cross · SSR · Wild species

60.1 Introduction

The cultivated peanut, *Arachis hypogaea* L., as an important cash crop in the world, provides over 30 essential nutrients and phytochemicals including oil and protein [1]. Plumpy'Nut, a ready-to-use peanut-based paste, was invented and used to treat severe acute child malnutrition in Africa [1, 2]. However, as pointed out by Mallikarjuna et al. [1], the peanut crop, an amphidiploid member of the *Arachis* section with *A. duranensis* and *A. ipaensis* as its progenitors [3–5], is the product of evolution after a series of six bottlenecks [1]. It was estimated that the cultivated peanut contains no more than 13 % of the genetic diversity present in its closest

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582 S. Meng et al.

wild relatives in the section [1, 6]. Narrow genetic base of the peanut crop accounts for its genetic vulnerability to various biotic/abiotic stresses and limited productivity potential [7].

The *Arachis* genus, now with 80 named species, is partitioned into 9 sections by Krapovickas and Gregory [8]. Reportedly, some of the genetically diversified wild *Arachis* species contained high protein/oil content [9]. It is believed that wild peanut also possesses high-yielding factors [10]. Moreover, high levels of stress resistances have been identified in wild peanut specie, some of which have never been found in the peanut cultigen [7]. Hence wild peanut relatives constitute an invaluable gene source for the genetic improvement of the cultivated peanut [7, 9]. Nevertheless, only species from section *Arachis* are cross-compatible with *A. hypogaea*. Owing to severe pre-/post- fertilization barriers, fertile hybrids between the cultivated peanut and species outside section *Arachis*, viz., incompatible species, are not readily available [7].

The aim of this study was to identify true peanut intersectional hybrids resulting from post-pollination hormone treatment of flower bases using simple sequence repeat markers.

60.2 Materials and Methods

60.2.1 Peanut Materials

Two intersectional crosses were made between peanut cultivars, Qunyu 101 and Huayu 40 (female parents) and wild peanut *Arachis pusilla* PI289628, a *Heteranthae* section species (male parent). Through post-pollination hormone treatment of flower bases, resultant seeds were harvested in fall and sown in the field under polythene film mulch in the next spring. Routine agronomic practices were followed as per the description by Wan et al. [11]. Genomic DNA was extracted from unexpanded leaves of field-grown peanut plants using EasyPure Plant Genomic DNA kit (TransGen Biotech, Beijing) according to manufacturer's instructions.

60.2.2 SSR Profiling

In primary evaluation, of the 12 SSR primer pairs evaluated, 3 primer pairs, PM201 [12], PM468 [13] and IPAHM123 [14], were found to be informative and capable of producing reproducible and distinguishable bands. Therefore, the three primer pairs were used to identify true hybrids in the study (Table 60.1). SSR analysis was conducted in three replications.

The PCR mixture (20 μ L total volume) consisted of 1 μ L of DNA template (30 ng), 0.5 μ L of upstream and downstream SSR primers (10 μ M) each, and 10 μ L

Primer ID	Forward primer (5′–3′)	Reverse primer (5′–3′)
PM201	CCTTTATAGAGGACCTTCCCTCTC	GCCTATTTGGTATCGGCTCA
PM468	TCAAGCCATAATATGTTCCACA	AAAACAACCCAAGCACCTCT
IPAHM123	CGGAGACAGAACAAACCA	TACCCTGAGCCTCTCTCTCG

Table 60.1 List of informative SSR primer pairs

of $2 \times Taq$ PCR Master Mix (DBI Bioscience, Catalog # DBI-2028). Thermal cycling profile was denaturing at 95 °C for 3 min, followed by 35 cycles of 95 °C for 45 s, 55 °C for 45 s, and 72 °C for 45 s, and a final extension of 72 °C for 5 min.

Bands were separated on a 6 % denaturing polyacrylamide gel. Gel was washed with distilled water twice (1 min). Fixation and silver staining were conducted in a single step. Gel was placed in solution containing 0.2 % AgNO₃, 1 % glacial acetic acid, and 10 % ethanol for 10 min. Then the gel was washed with distilled waterr twice (2 min). Later, gel was placed in developing solution (3 % NaOH, with 1 mL formaldehyde added in 200 mL total volume just prior to developing). Generally, 13 min was enough to develop clear bands [15].

60.3 Results and Discussion

For both crosses, no difference was detected among banding patterns of the 3 replications. Judged by ployacrylamide gel electrophoresis profile, true hybrids could be easily selected in both crosses.

60.3.1 Qunyu 101 × A. pusilla

In the cross, a total of 8 single plants were tested by PCR using the 2 SSR primer pairs, PM468 and PM201. Merely three single plants (37.5 %) were identified as true F₁ hybrids (Fig. 60.1a, b). PM468 and PM201 resulted in the same results. True hybrids were characterized by the presence of a band co-immigrating with the band from *A. pusilla*, whereas the female parent Qunyu 101 and selfs did not produce such a band (Fig. 60.1a, b). The results were supported by SSR profiling with a third primer pair, IPAHM123, where totally 16 single plants were utilized. Not surprisingly, with this new primer pair, all of the three "hybrid" plants previously identified using primer pairs PM468 and PM201 produced a characteristic band co-immigrating with the band from the male parent, *A. pusilla* (Fig. 60.2a).

584 S. Meng et al.

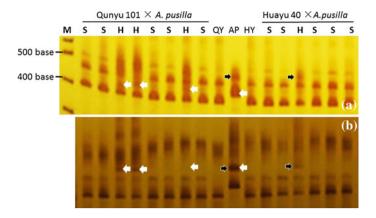


Fig. 60.1 PM468 (a) and PM201 (b) identified true F_1 hybrids (H) and selfs (S) resulting from Qunyu $101 \times A$. pusilla and Huayu $40 \times A$. pusilla, arrow indicating characteristic band present in A. pusilla (AP, male parent) but absent in female parents, Qunyu 101 (QY) and Huayu 40 (HY). M 50 bp DNA Ladder (Fermentas) S self. H hybrid

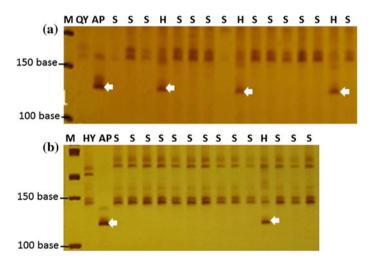


Fig. 60.2 IPAHM123 identified true F_1 hybrids (H) and selfs (S) resulting from Qunyu 101 \times *A. pusilla* (a) and Huayu 40 \times *A. pusilla* (b), *arrow* indicating characteristic band present in *A. pusilla* (AP, male parent) but absent in female parents, Qunyu 101 (QY) and Huayu 40 (HY). *M* 50 bp DNA Ladder (Fermentas) *S* self. *H* hybrid

60.3.2 Huayu $40 \times A$. pusilla

In this cross, totally six single plants were tested by PCR using primer pairs, PM468 and PM201, among which one plant (16.67 %) was identified as true F_1 hybrids only (Fig. 60.1a, b). Similar to the situation in Qunyu $101 \times A$. *pusilla*, both the true

hybrid and the male parent *A. pusilla* produced a 100–150 nt band absent in Huayu 40. When more single plants (a total of 14 single plants) were included in SSR profiling using primer pair IPAHM123, the true hybrid identified previously produced a characteristic band co-immigrating with the band from *A. pusilla* (Fig. 60.2b).

60.4 Conclusions

In the present study, through SSR profiling, three true intersectional F_1 hybrids of Qunyu $101 \times A$. pusilla and a true intersectional F_1 hybrid of Huayu $40 \times A$. pusilla were identified, providing evidence for the effectiveness of the in vivo hormone treatment method in overcoming cross-incompatibility in the genus Arachis. The easy-to-use in vivo hormone treatment method may facilitate the utilization of wild incompatible peanut species in the genetic improvement of the cultivated peanut.

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Chapter 61 Partial Purification and Chemical Characterization of a Bioemulsifier and Its Application in MEOR

Dayuan Dong, Xingbiao Wang, Mingyu Cai, Jingjing Wang, Yifan Han, Xiaoxia Zhang and Zhiyong Huang

Abstract The bioemulsifier secreted by strain XS2, which was isolated from oil-contaminated soil samples in Yumen oil field, China, was partially purified and analyzed. Protease hydrolysis and heat treatment results showed that the protein composition of the bioemulsifier played a significant role in emulsification activity. High-performance liquid chromatography (HPLC) study suggested that the protein composition may exist as glycoprotein. Monosaccharide analysis by gas chromatograph-mass spectrometer (GC-MS) showed that galactose, glucose, xylose, and mannose presented in the bioemulsifier. The emulsification activity of this bioemulsifier was relatively high due to different carbohydrates (n-hexadecane 61 %, liquid paraffin 60 %, Toluene 77 %). Furthermore, single well stimulation trial in Changging oil field, was done to explore the potential application in Microbial-enhanced oil recovery (MEOR). Up to September 2014, the cumulative increased crude oil production was about 88 T in total 240 days, and the water content of the well-produced liquid decreased from 85 to 25 %. These data illustrated that the bioemulsifier performed a high potential in applications and had important economic values.

Keyword Bioemulsifier \cdot XS2 \cdot Chemical characterization \cdot MEOR \cdot Single well stimulation

D. Dong and X. Wang are contributed equally to this work.

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61.1 Introduction

Bioemulsifiers were compounds with surface emulsifying activity, which were usually amphipathic polysaccharides, proteins, lipopolysaccharides, lipoproteins, or complex mixtures of these biopolymers [1]. The ability to form stabilized o/w or w/o emulsion and keep the condition was the typical feature of these substances, which ensured its potential application in microbial-enhanced oil recovery (MEOR) [2, 3]. Plentiful studies about bioemulsifiers nowadays were done which mainly focused on isolation, screening of active strains, and the characteristics of their products [4–8]. De Sousa and Bhosle [5] studied the isolation and character of a lipopeptide bioemulsifier, which the emulsification activity (E24) was varied from 40 to 60 % for various tested hydrocarbons [5], while it was a pity that no further application test was carried. The study of Colin [7] was more comprehensive with an applied research for bioremediation besides the basic character of the bioemulsifier, nevertheless a more wonderful work will be presented if added the exploration of the bioemulsifier field application [9]. Although extensive researches have been conducted on variety of aspects of different bioemulsifiers including production, emulsification activity and cell surface hydrophobic activities, however, few analysis of the field application of bioemulsifiers based on the emulsification activity and production were carried out further. The most likely application of bioemulsifier was in MEOR [2, 3], which of low requirements toward the emulsifier concentration and purity. Bioemulsifiers played a significant role in extracting the remaining oil from oil reservoirs, which was cost-effective, environmental-friendly, and efficient particularly in the exploitation of oil remained trapped in capillary pores of the formation rock or in areas not swept by the classical or modernenhanced oil recovery (EOR) methods [3]. Here, the partial purification and chemical characterization of the bioemulsifier produced by XS2 was reported and its application potential was explored in Changqing oil field by single well stimulation trial.

61.2 Materials and Methods

61.2.1 Microorganism and Media

Strain

The strain XS2, identified as *Geobacillus pallidus*, was isolated from oil contaminated soil samples in Yumen oil field, China. It could produce bioemulsifier at 60 °C with water-soluble carbon source glucose and the E24 of the fermentation liquid was about 60 %. Former chemical composition studies exhibited that the bioemulsifier consisted of carbohydrates (68.6 %), lipids (22.7 %), and proteins (8.7 %) [10].

Media

Luria-Bertani liquid medium (LB) (g/L), NaCl 10, tryptone 10, yeast extraction 5, pH 7.2, sterilized at 121 °C for 20 min.

Glucose–inorganic salt medium (GS) (g/L), glucose 20, NaNO₃ 3, KH₂PO₄ 0.42, K₂HPO₄ 1.2, FeSO₄ 0.05, MgSO₄ 0.5, CaCl₂ 0.05, pH 7.2, sterilized at 121 °C for 20 min [11].

The strain was cultured at 60 °C, 200 rpm with a 10 % inoculum size in 250 mL flasks. The XS2 cells as seed was incubated in LB medium for 10 h and then harvested and transferred into GS medium and cultured at 60 °C, 200 rpm for 24 h.

61.2.2 Purification of the Bioemulsifier

After culturing, the fermentation broth was centrifuged at 10,000 rpm for 20 min at 4 °C to remove the cells, and then the supernatant was obtained and precipitated with three volumes cold methanol for crude bioemulsifier isolation. The precipitate was redissolved in distilled water, dialyzed against distilled water for 48 h by dialysis bag (8,000–14,000 Da) to remove the excess inorganic salts and small molecular saccharides [10]. Then the free proteins were removed used sewage methods for three times until there was no protein layer existing. And the proteins were further dialyzed against with distilled water again for 24 h, and lyophilized.

61.2.3 Proteins Activity Identification of the Bioemulsifier

Two different treatments were carried to explore the protein role in the whole bioemulsifier. (1) Dealt with 20 mg/mL final concentration protease K at 37 $^{\circ}$ C for 30 min, (2) Set the bioemulsifier solution (5 mg/mL) at 100 $^{\circ}$ C in a water bath for 2 h, and tested the emulsification activity after treatment.

61.2.4 Gel Filtration Chromatography for the Protein Component in the Bioemulsifier

20 mg purified bioemulsifier was dissolved in 1 mL Tris-NaCl buffer (20 mmol/L Tris, 150 mmol/L NaCl) to separate the different protein components. The column used was Hiload 16/60 Superdex 200 with 0.15 Mpa column pressure and a 0.5 mL/min flow rate. The sample volume was 1.0 mL [12].

590 D. Dong et al.

61.2.5 HPLC Analysis for the Glycosyl of the Bioemulsifier

High-performance liquid chromatography (HPLC) was used to study the glycosyl of the bioemulsifier. The chromatography column was used Agilent PL aquagel-OH MIXED-H (300 \times 7.5 mm, 8 μm), and the column temperature was set at 35 °C. Double-distilled H_2O was used as the mobile phase with 0.8 mL/min flow velocity. And the injection volume was 10 μL . The Refractive index detector (RID) was set to detect the carbohydrates, at the same time an ultraviolet detector was set at 280 nm to detect proteins.

61.2.6 GC-MS Analysis for the Monosaccharide Components of the Bioemulsifier

For GC–MS analysis, the carbohydrate was first hydrolyzed by trifluoroacetic acid (TFA) and then subjected to trimethylsilylation (TMS) according to the references [13, 14]. The specific steps were done according to the method described by Peng [14]. The carrier gas was nitrogen at a flow rate of 30 mL/min. The oven temperature was set to 70 °C for 0.5 min. The temperature gradient was 130–280 °C at a rate of 10 °C/min.

61.2.7 Microbial Community Structure Analysis of the Test Well

Total DNA was extracted from the well production water sample of the test well Liu55-4 before the field trial by Ultra Clean TM Soil DNA Isolation Kit (Mo Bio Laboratories, Inc., USA) according to the operation manual. The microbial community was tested by 16S rDNA gene clone library, DNA sequencing, alignment, and phylogenetic analysis according to the reported method [15]. Aiming to compound a nutrient liquid to stimulate the indigenous bacteria, the community consisted was studied by the above method.

61.2.8 Field Study of MEOR

A pilot test in oil field was carried out to examine the applicability of the bioemulsifier produced by the strain XS2 after laboratory studies. To get a better effect, fermentation broth of XS2 and glycolipid were mixed by different ratios. The perfect ratio that the mixed liquid could make the solution keep a relative low surface tension and a high E24 activity at the same time was set as the final used ratio. Microbial inorganic salt nutrients were determined by the microbial community structure, which was learned from the oil well production water samples. Then the compounds mixture used as the composite oil displacement agent in MEOR including bioemulsifier of XS2, glycolipid, and inorganic salt nutrients was injected into Liu55-4 well, Changqing oilfield, China. The well had been closed for three months after composite oil displacement agent injection, and then begun to exploit crude oil. Two data mainly, crude oil production and water content of the well production liquid were recorded as the parameters to evaluate the bioemulsifier efficiency in MEOR.

61.3 Results and Discussions

61.3.1 Protein Activity Identification of the Bioemulsifier

For the protein activity identification of the bioemulsifier, an interesting phenomenon suggested that the protein component in bioemulsifier acted as a vital role for its emulsification activity. After treated by protease K for 30 min at 37 °C, the emulsification ability of the bioemulsifier was disappeared, but thermal treatment at 100 °C for 2 h acted slight influence on the emulsion ability of the bioemulsifier (Fig. 61.1). From this test, two conclusions were gotten. One conclusion was that the protein component performed a significant role in emulsification ability, the

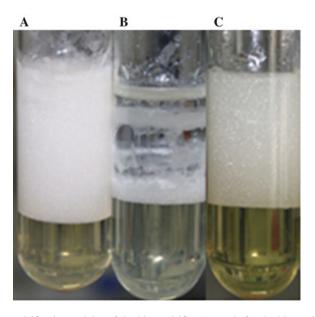


Fig. 61.1 The emulsification activity of the bioemulsifier, *A* stands for the bioemulsifier solution with no treatment, *B* stands for treated by protease K, and *C* stands for treated by high temperature

592 D. Dong et al.

other was that the bioemulsifier performed certain thermo stability. A reasonable interpretation may be that the optimum growth temperature of this strain was 60 °C, the thermal protein was a result of evolving to adapted the environment in the heat oil reservoir. This may strengthen the theoretical support of the bioemulsifier of this strain application in MEOR.

61.3.2 Gel Filtration Chromatography for the Protein Components of the Bioemulsifier

To further purify the bioemulsifier, gel filtration chromatography of the protein components was done. Figure 61.2 showed that the protein in bioemulsifier was consisted of 3 fractions, and then the fractions were collected, respectively, and carried out the emulsification activity test for each fraction. The results showed that the second fraction performed the central emulsion activity, while the first and third fractions had no emulsification activity. SDS-PAGE was done for fraction 2, and there were totally four bands about 72, 70, 42, 24 KDa, respectively (Fig. 61.3). From the analysis of the Gel filtration chromatography, fraction 2 was consisted of totally four protein fragments and played a key role in the emulsion activity of the bioemulsifier. And further components research was carried out by HPLC.

61.3.3 HPLC Analysis for Fraction 2 of the Bioemulsifier

To explore the existing form of the protein in fraction 2, HPLC analysis was performed. The HPLC data showed that when there was a peak in the RID, there appeared a peak in ultraviolet detector almost the same time, which mean that the saccharine component and the protein component were detected almost at the same time. There comes a hypothesis that the protein component may exist as glycoprotein in the bioemulsifier (Fig. 61.4).

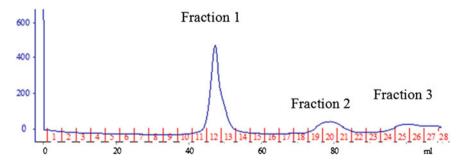


Fig. 61.2 Results of the bioemulsifier gel filtration chromatography

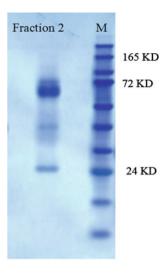


Fig. 61.3 SDS-PAGE for fraction 2

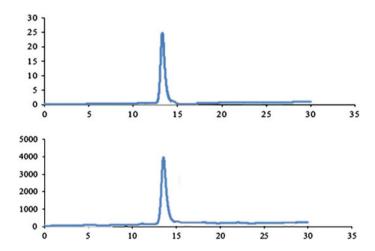


Fig. 61.4 HPLC result of fraction 2, the *above* peak was detected by RID detector, standing for the saccharide detected, and the *below* peak was detected by ultraviolet detector, standing for the protein detected

61.3.4 Monosaccharide Components Analysis of the Bioemulsifier

GC-MS analysis was done to analyze the monosaccharide composition of the bioemulsifier. Figure 61.5 showed there were four main peaks, and each peak

594 D. Dong et al.

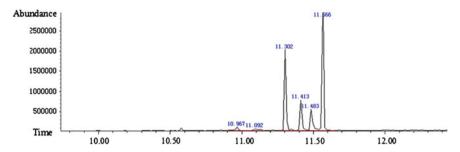


Fig. 61.5 GC-MS analysis of the monosaccharide of the bioemulsifier

stands for one silanized saccharide. Compared to the database, the monosaccharide components were likely to be galactose, glucose, xylose, and mannose (Table 61.1). A certain result will be obtained followed by further study. Research conducted by Llamas et al. [16] reported that the monosaccharide composition of the EPS was mannose, glucose, and rhamnose. Monteiro et al. [17] also mentioned the bioemulsifier produced by CLA2 had xylose, mannose, and glucose. Glucose and mannose were the shared monosaccharide in the high-molecular polymer. The unique monosaccharide of the bioemulsifier produced by XS2 was galactose compared to the other two researches. Although different products from various bacteria have diverse monosaccharide composition, there may be some usual monosaccharide. The different monosaccharides and the composition ratio of the high-molecular polymers may present on various activity and function.

61.3.5 Microbial Community Structure Analysis of the Test Well

Microbial community structure in the well Liu55-4 was analyzed, which could provide much more useful and abundant information for selecting appropriate inorganic salt as the injected nutrients for MEOR. Based on the analysis data, a total of 7 genera were discovered in Liu55-4, which were *Pelobacter*, *Petrotoga*, *Bacillus*, *Acinetobacter*, *Synergiste*, *Sedimentibacte*, and *Halabaerobiu*. The dominant species was *Palobacter*, about 38.3 % of total bacteria. *Sedimentibacte* and *Halanaerobiu* were infrequent species, about 4.26 %. Most microorganisms existed in the oil well could utilize the hydrocarbons in the oil reservoir, and according to

Table 61.1 The monosaccharide component of the bioemulsifier

Remain time	Compound	Structure formula
11.302	Galactopyranose	$C_{21}H_{52}O_{6}$
11.413	Glucopyranose	$C_{21}H_{52}O_{6}$
11.483	D-Xylose tetrakis	C ₁₇ H ₄₂ O ₅
11.566	Mannose	$C_{21}H_{52}O_{6}$

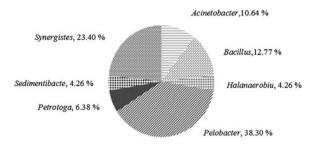


Fig. 61.6 The microbial community structure of 55-4 oil water

nutrients necessary for these bacteria and XS2, a nutrient solution including nitrates, phosphate, trace elements, and vitamins was determined for pilot research in oil field (Fig. 61.6).

61.3.6 Field Study of MEOR

XS2 fermentation by various scale fermenters (7, 30, 200, 2,000 L) was performed to obtain enough fermentation liquid, all the cell concentration maintained more than 5×108 /mL in the fermentation broth. Aimed at getting effective oil displace agent, the fermentation broth was compound with glycolipid solution (1 g/L). Finally, an 87 % emulsion activity on paraffin and the surface tension of the oil displace agent solution kept 33.8 mN/m. At the same time, 5 % (w/v) crude oil, 75 % inorganic salt medium (v/v) were mixed and added the following three liquids, respectively: (1) 20 % fermentation broth of XS2, (2) 20 % glycolipid solution (1 g/L), (3) 20 % compound agent of XS2 fermentation broth and glycolipid. The flasks were incubated at 60 °C and 180 rpm in shaker for 5 min, and then observed the emulsification state of crude oil (Fig. 61.7). The system added compound agent showed the crude oil dispersed as small droplets and the emulsification state could keep more than 15 days. While the other two systems added fermentation broth and glycolipid solution separately, both presented a weaker effect. This oil displacement agent compound consisted of fermentation broth and glycolipid gathered both the advantages of bioemulsifier and glycolipid which could decrease the surface/interfacial tension, meanwhile maintain a high emulsification state more than 15 days.

Liu55-4 well, Changqing oilfield, China, was chosen as the experimental well. Since November 2011, the water content of the well-produced liquid increased from 20 to 85.7 % and the yield of crude oil fall down to 0.23 m³/d (0.03 T). The oil displacement agent was injected into the Liu55-4 well in December, 2013, and then closed the well for three months before exploitation. Crude oil production and water content of the produced liquid were recorded as the main parameters. The oil yield and water content gradually tended to stable after 20 days extraction. The oil

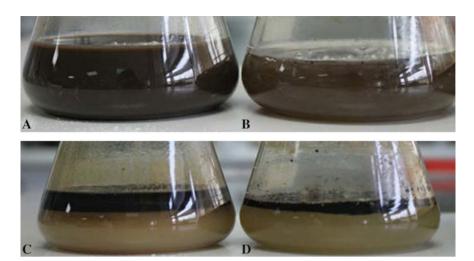


Fig. 61.7 The comparison of single liquid and compound liquid acting on crude oil, A and C were compound liquid of XS2 fermentation broth and glycolipid, B and D stands for the result added glycolipid only. A and B were uniformed by shaking, and C and D were the static condition

production was 0.65 m³ (0.36 T), with 25 % water content. To date, the well had produced crude oil for more than 8 months, and the average daily output was 0.45 T. Partial data after trial was displayed in Table 61.2 and Fig. 61.6, which showed the comparison of oil production and water content before and after the oil displacement agent injection. The data showed crude oil production was increased more than 12 times and the water content was decreased more than 70 %. Totally, 100 T crude oil was extracted after this trial, and compared to the data before, the increased production of the crude oil was 88 T. Oil displacement agent performed a sound effect for oil extraction, and the bioemulsifier played a significant role in the process, which give a robust proof that the strain XS2 has a huge application potential in MEOR (Fig. 61.8).

Table 61.2 The results of oil recovery

596

Time (d)	Liquid (m ³)	Water content (%)	Oil production (T)
1	6.72	100	0
4	3.41	95	0.13
6	2.19	23.4	1.43
12	2.19	25.6	1.38
17	0.63	26.6	0.39
24	0.6	26.6	0.36
40 55	0.6	26.6	0.36
55	0.6	26.6	0.36

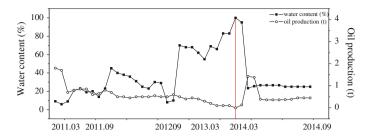


Fig. 61.8 Oil production and water content curve of the Liu55-4 well in Changqing oil field. The *red vertical line* was a mark. On the *left* and *right of the red line* were the oil production data before and after oil displacement agent injection, respectively

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Chapter 62 **Optimization of a Whey Containing** Medium for β-Galactosidase Production by Lactobacillus reuteri

Mengfei Li, Huiming Zhu, Huibin Qin, Yan Zhang and Hongjiang Yang

Abstract Lactobacillus reuteri MF2-2 was screened out for producing a high level of β-galactosidase from chicken fecal samples and identified with 16S rRNA gene sequence analysis. To reduce production cost, a new whey containing medium was formulated on the basis of MRS medium and optimized. Nutrients including whey, yeast extract, NH₄H₂PO₄, and vitamin B₁ were investigated at the individual and interactive level by the Taguchi orthogonal tests. The production of β -galactosidase was improved from 490.0 U/g in lactose-based MRS (L-MRS) medium to 814.6 U/g in the optimized whey-based MRS (W-MRS) medium. W-MRS medium did not comprise of lactose, peptone, and beef extract and was expected to cut the raw material cost approximately in half compared to L-MRS medium.

Keywords Lactobacillus reuteri · β-Galactosidase · Whey-based MRS medium (W-MRS)

62.1 Introduction

β-Galactosidases are mainly used to hydrolyze lactose in milk for producing low-lactose dairy products [1]. β-Galactosidases are widely present in plants, bacteria, fungi, yeast, and animal cells. The industrial β-galactosidases are mainly obtained from the probiotic microorganisms, such as Aspergillus sp. and Kluyveromyces sp. [2, 3].

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⁵⁹⁹

600 M. Li et al.

Lactic acid bacteria (LAB) are a taxonomically diverse group of gram-positive bacteria and usually categorized as "generally regarded as safe" (GRAS) organisms. All LAB strains produce lactic acid as the only or main end product of carbohydrate metabolism [4, 5]. Due to their GRAS status and ability to hydrolyze lactose, a number of strains may have the potential to be starters for industrial production of β-galactosidases [6]. Among them, Streptococcus thermophilus, which is the most important species in the manufacture of cheese and vogurt, has been well studied for β-galactosidases production [7]. Bifidobacteria are obligate anaerobes existing in the intestinal tract of humans and animals, and β-galactosidase productions have been investigated in strains of this genus [8–10]. Moreover, a wide variety of Lactobacillus species have also been assessed as potential source of β-galactosidases, such as L. acidophilus, L. bulgaricus, L. helviticus, L. kefiranofaciens, L. lactis, L. sporogenes, L. themophilus, L. delbrueckii, and L. fermentum [2, 11]. In this work, lactose-based MRS (L-MRS) medium was used in isolation of lactic acid bacteria (LAB) producing β-galactosidase. The highest β-galactosidase producer was selected and the lactase was characterized. In addition, a wheycontaining medium was formulated and optimized based on L-MRS medium to improve the enzyme productivity and reduce production cost in the β -galactosidase fermentation process.

62.2 Materials and Methods

62.2.1 Media, Growth Conditions, and Isolation of β-Galactosidase Producing Lactobacilli

MRS medium was composed of 20 g/L glucose, 10 g/L peptone, 8 g/L beef extract, 4 g/L yeast extract, 5 g/L sodium acetate, 2 g/L ammonium citrate, 0.2 g/L MgSO₄, 0.05 g/L MnSO₄, 2 g/L KH₂PO₄, and 0.1 % Tween-80 [12]. Lactose-MRS (L-MRS) agar (pH 5.0) was used in isolating β -galactosidase producing *lactobacilli* by replacing glucose in MRS with 20 g/L lactose. L-MRS broth (pH 6.4) was used for the cultivation of *lactobacilli* isolates and β -galactosidase production. Both plates and liquid cultures were statically incubated at 35 °C. Thirty fresh chicken fecal samples were collected from five henneries located in Beijing and Henan Province, respectively. To isolate β -galactosidase producing *lactobacilli*, the chicken fecal samples were suspended in 1× PBS buffer. Serial dilutions were spread on L-MRS agar containing 20 μ g/mL X-gal and incubated.

62.2.2 Determination of β-Galactosidase Activity

β-Galactosidase activity was tested by a modified method [13, 14]. In brief, 3 mL Z-buffer (pH 7.0) and 80 μL chloroform were added to 1 mL culture and mix thoroughly by vortex. The crude permeabilized cell suspension was added with 0.8 mL of 4 mg/mL ONPG (o-nitrophenol-β-D-Galactopyranoside) and incubated at 37 °C. The reaction was terminated by adding 1 mL of 1 M Na_2CO_3 when the mixture turned yellow. The mixture was centrifuged at 12,000 rpm for 3 min and the supernatant was measured spectrophotometrically at 420 nm. One unit of β-galactosidase activity was defined as the enzyme quantity that liberated 1 μmol o-nitrophenol per minute. Lactase concentration (U/g) was determined as the number of the enzyme units per gram of dry cell weight.

62.2.3 Identification of the Isolated Organism

Genomic DNA of the isolated *lactobacilli* was extracted and used as template for amplification of 16S rRNA gene with primers described previously [15]. The PCR reaction system was 50 μ L and the parameters include denaturation at 94 °C for 5 min, 1 cycle; denaturation at 94 °C for 45 s, annealing at 53 °C for 45 s, and extension at 72 °C for 1.5 min, 30 cycles; 72 °C for 10 min. Sequence similarity was analyzed by using the tool BLAST from the NCBI web site.

62.2.4 Time Course of β-Galactosidase Production

To study the β -galactosidase production of different periods, fermentations were performed in 250 mL shake flasks containing 100 mL L-MRS medium with 1 % inoculums of seed cultures. Samples were collected at 2 h time intervals for measuring cell density (OD₆₀₀) and β -galactosidase activity during fermentation process.

62.2.5 Formulation and Optimization of a Whey Containing Medium

Whey contained high content of lactose and it was often used as an alternative carbon source to lactose. On the basis of L-MRS medium, a whey medium was formulated by using dry whey (sweet type) powder as the carbon source. The effect of different nitrogen sources and vitamin B_1 on β -galactosidase production was investigated. Fermentation parameters were tested at individual and interactive

602 M. Li et al.

level. An L_{16} (4⁴) array used in the Taguchi orthogonal tests was designed to optimize the whey containing medium for improvement of β -galactosidase production. Lactose content in whey was determined by Lane-Eynon's method [16].

62.3 Results and Discussions

62.3.1 Isolation of β-Galactosidase Producing Lactobacilli

L-MRS agar containing 20 μ g/mL X-gal was used to isolate β -galactosidase producing lactobacilli. From 30 chicken fecal samples, 13 strains were obtained and purified. The β -galactosidase production was measured after fermentation in L-MRS for 24 h. Strain MF2-2 produced the highest level of β -galactosidase with the activity of 517.7 U/g.

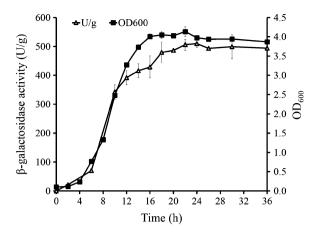
62.3.2 Identification of Strain MF2-2

16S rRNA gene of strain MF2-2 was amplified and sequenced. The obtained sequence was deposited in GenBank with the accession number KJ495694. 16S rRNA gene comparison was conducted and the result showed that strain MF2-2 exhibited a 99 % sequence homology with *L. reuteri* TD1 [17]. *L. reuteri* is the most abundant species of the heterofermentative *lactobacilli* in the gastrointestinal tract in humans and animals. The bacterium is responsible for hydrolyzing lactose inside the gastrointestinal tract with β -galactosidase. *L. reuteri* MF2-2 was deposited in China Center for Industrial Culture Collection and the number was CICC No. 10855.

62.3.3 Time Course of β-Galactosidase Production

Time course of cell growth and β -galactosidase production of *L. reuteri* MF2-2 were analyzed and shown in Fig. 62.1. As the bacterium began to grow, the bacterial cells began to produce β -galactosidase. The cultures entered stationary phase after 16 h incubation at an OD₆₀₀ of 4.0, while β -galactosidase production reached maximum level of 509.3 U/g at 24 h. Only a limit amount of β -galactosidase was accumulated after 20 h and the result indicated the enzyme could be harvested at the window of 20–24 h.

Fig. 62.1 Time course of the growth and β-galactosidase production by *Lactobacillus reuteri* MF2-2



62.3.4 Formulation of a Whey Containing Medium for β-Galactosidase Production

Several ingredients in L-MRS medium were expensive, including lactose, yeast extract, peptone, beef extract, and ammonium citrate. To reduce the high production cost, L-MRS medium was modified by testing different formulas. Sweet type whey powder was determined containing 71.1 % lactose by Lane-Eynon's method, and it was used as an alternative carbon source to lactose. In W-MRS-1 (whey-based MRS) medium with 20 g/L whey, the β -galactosidase yield was 957.5 U/g which was much higher than L-MRS (490.0 U/g) (Table 62.1). However, beef extract and peptone were present in both media which were still very expensive for β -galactosidase production.

In L-MRS-2 and L-MRS-3 media, whey was used as a nitrogen source to replace peptone or beef extract, the cultures produced higher levels of β -galactosidase, 594.0 and 622.7 U/g, respectively, compared to fermentations in L-MRS medium (Table 62.1). The result indicated that the whey could provide nitrogen sources for fermentation of the strain MF 2-2. To test this hypothesis, W-MRS-4 medium was further tested and the strain produced 517.1 U/g of β -galactosidae in the absence of both peptone and beef extract (Table 62.1). The production level was comparable to L-MRS medium (Table 62.1). Furthermore, NH₄H₂PO₄ was used to replace inorganic nitrogen source ammonium citrate and KH₂PO₄ in W-MRS-5 medium, and the β -galactosidase production was 520.2 U/g which was at the same level with L-MRS medium (490.0 U/g) (Table 62.1). The results demonstrated that whey and NH₄H₂PO₄ were able to provide enough nitrogen for the enzyme production.

Vitamin B_1 , also called thiamine or thiamin, is a cofactor of a variety of enzymes which are involved in carbohydrate catabolism. The effect of vitamin B_1 on β -galactosidase production was also investigated. The result showed that vitamin

Table 62.1 Investigation of the nutrients affecting β -galactosidase production

Medium	Compone	Components (g/L)								β-Galactosidase
	Whey	Lactose	Peptone	Beef	Yeast	Ammonium	KH ₂ PO ₄	NH ₄ H ₂ PO ₄	$VB_1 (\times 10^{-3})$	activity (U/g)
				extract	extract	citrate				
L-MRS	0	20	10	8	4	2	2	0	0	490.0 ± 35.0
L-MRS-1	0	20	10	8	8	2	2	0	0	453.4 ± 20.0
L-MRS-2	0	20	10	8	4	2	2	0	4.5	700.6 ± 15.1
W-MRS-1	20	0	10	8	4	2	2	0	0	957.5 ± 34.7
W-MRS-2	10	20	0	8	4	2	2	0	0	594.0 ± 35.9
W-MRS-3	8	20	10	0	4	2	2	0	0	622.7 ± 52.1
W-MRS-4	20	0	0	0	4	2	2	0	0	517.1 ± 11.3
W-MRS-5	20	0	0	0	4	0	0	4	0	520.2 ± 36.7
W-MRS-6	20	0	0	0	4	0	0	4	4.5	742.0 ± 23.9
W-MRS-7	20	0	0	0	2	0	0	4	4.5	349.4 ± 4.3
In addition to	the compo	onents listed	in the table,	, all media	also contain	In addition to the components listed in the table, all media also contain CH ₃ COONa 5 g/L, MgSO ₄ 0.2 g/L, MnSO ₄ 0.05 g/L, and Tween-80 0.1 %	,/L, MgSO ₄ (0.2 g/L, MnSO ₄	0.05 g/L, and T	ween-80 0.1 %

 B_1 could increase β -galactosidase production significantly in both L-MRS-2 and W-MRS-6 media, 700.6 and 742.0 U/g, respectively (Table 62.1).

Yeast extract content was also monitored. When its concentration was increased to 8 g/L, no increase of β -galactosidase production was obtained in L-MRS-1 compared L-MRS medium (Table 62.1). However, the decrease of yeast extract content brought a significant reduction of β -galactosidase production, only 349.4 U/g was obtained in W-MRS-7 medium (Table 62.1). Based on the obtained results, the final formula of W-MRS medium was determined and composed of whey 20 g/L, yeast extract 4 g/L, NH₄H₂PO₄ 4 g/L, CH₃COONa 5 g/L, MgSO₄ 0.2 g/L, MnSO₄ 0.05 g/L, 0.1 % Tween-80, and vitamin B₁ 4.5 mg/L.

62.3.5 The Optimization of W-MRS Medium

To further improve β -galactosidase production, W-MRS medium was optimized by orthogonal array. The effects of whey, yeast extract, NH₄H₂PO₄, and vitamin B₁ in W-MRS medium on fermentation performance were investigated at the individual level first. As shown in Fig. 62.2, each factor with five concentrations was evaluated by measurement of β -galactosidase production in the cultures. The optimum

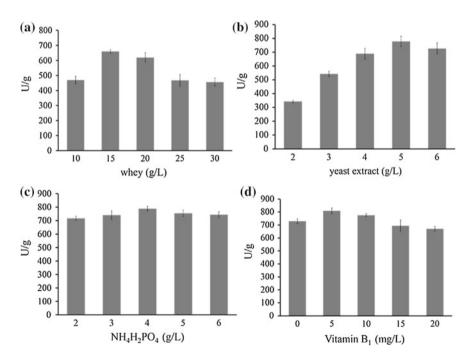


Fig. 62.2 Effect of different whey **a** yeast extract, **b** $NH_4H_2PO_4$, **c** and vitamin B_1 , **d** concentrations on β-galactosidase production

606 M. Li et al.

Table 62.2 Selected factors and their assigned levels for the orthogonal experimental design

Factors		Levels				
No.	Descriptions	1	2	3	4	
A	Whey (g/L)	12.5	15.0	17.5	20.0	
В	Yeast extract (g/L)	4.5	5.0	5.5	6.0	
С	NH ₄ H ₂ PO ₄ (g/L)	3.5	4.0	4.5	5.0	
D	Vitamin B ₁ (mg/L)	2.5	5.0	7.5	10.0	

concentration was 15 g/L whey, 5 g/L yeast extract, 4 g/L $NH_4H_2PO_4$, and 5 mg/L vitamin B_1 , respectively, which was indicated by the highest corresponding β -galactosidase activity in each set of fermentations (Fig. 62.2).

According to the obtained optimum concentration for each nutrient, an L_{16} (4⁴) array used in the Taguchi orthogonal tests was designed as described in Tables 62.2 and 62.3. Fermentations using 16 different media were conducted. Based on the R value obtained, factors affecting β -galactosidase activity were ranked from high to low as whey, vitamin B_1 , yeast extract, and $NH_4H_2PO_4$. The optimal conditions were determined as $A_3B_4C_3D_3$, which was whey powder 17.5 g/L, vitamin B_1

Table 62.3 Orthogonal experiment design for optimization of W-MRS medium

No.	Factor le	vels			β-Galactosidase activity (U/g)
	A	В	С	D	
1	1	1	1	1	614.1 ± 30.3
2	1	2	2	2	610.1 ± 29.7
3	1	3	3	3	720.6 ± 92.8
4	1	4	4	4	616.5 ± 59.3
5	2	1	2	3	680.6 ± 40.6
6	2	2	1	4	652.8 ± 61.7
7	2	3	4	1	658.4 ± 32.5
8	2	4	3	2	789.8 ± 52.1
9	3	1	3	4	679.5 ± 46.7
10	3	2	4	3	721.9 ± 52.8
11	3	3	1	2	751.1 ± 13.0
12	3	4	2	1	713.6 ± 76.6
13	4	1	4	2	664.3 ± 33.8
14	4	2	3	1	616.3 ± 43.4
15	4	3	2	4	651.0 ± 56.3
16	4	4	1	3	779.6 ± 51.7
T1/4	640.3	659.6	699.4	650.6	
T2/4	695.4	650.3	663.8	703.8	
T3/4	716.5	695.3	701.5	725.7	
T4/4	677.8	724.9	665.3	649.9	
R	76.2	74.6	37.7	75.7	

7.5 mg/L, yeast extract 6 g/L, and NH₄H₂PO₄ 4.5 g/L. Together with other components, including CH₃COONa 5 g/L, MgSO₄ 0.2 g/L, MnSO₄ 0.05 g/L, and 0.1 % Tween-80, the optimum parameters of W-MRS medium was defined. Under the same fermentation conditions, the β -galactosidase production was 814.6 U/g in the optimized medium, much higher than L-MRS medium.

62.4 Conclusion

Whey was a by-product generated in the cheese processing industry. More than 55 % of milk nutrients were remained in the whey liquid, including lactose (4.5–6.0 % w/v), protein (0.6–1.1 % w/v), and other nutrients. After dehydration, lactose and protein constituted approximately 70 and 10 % of whey powder, respectively [18]. Whey products provided carbon and nitrogen source to the microorganisms capable of metabolizing lactose. Currently, whey has been widely applied in a variety of fermentation processes to produce a number of bioproducts, such as ethanol, lactic acid, bacteriocins, and enzymes [19–21].

In this study, W-MRS medium was formulated and optimized. The medium did not contain lactose, peptone, and beef extract and the medium cost was reduced approximately to 50 % of L-MRS medium with a relatively high β -galactosidase activity 814.6 U/g. However, there is still room for further improvement of β -galactosidase production if more nutrients were supplemented. W-MRS-1 medium gave 957.5 U/g of β -galactosidase even without optimization of the related fermentation parameters, though the enzyme production cost went up significantly. Our results suggested strain MF2-2 might be used as a starter strain for β -galactosidase production and W-MRS medium had the potential to cut down the production cost during fermentation process.

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Chapter 63 Enzymatic Bioconversion for γ-Aminobutyric Acid by *Lactobacillus*brevis CGMCC No. 3414 Resting Cells

Xiu-feng Shi, Bo Zheng, Chuan-you Chang, Peng Cao, Hong-jin Yang and Qiang Gao

Abstract In this work, γ-aminobutyric acid (GABA) was prepared by the decarboxylation of L-glutamate via L-glutamate decarboxylase in the resting cells of *Lactobacillus brevis* CGMCC No. 3414. The influence of cell concentration, cell age, buffer system, reaction time, and substrate concentration were investigated. The optimal composition of bioconversion system was composed of 50 g/L resting cells, cell age at 48 h fermentation, 0.2 M disodium hydrogen phosphate–citric acid buffer, and 25 mM monosodium glutamate. When the bioconversion system was performed at pH 4.6, 30°C, and 180 r/min shaking for 4 h, GABA production in biotransformation solution was 23.29 mM and the molar yield rate of bioconversion reached 93.15 %.

Keywords γ-Aminobutyric acid (GABA) · *Lactobacillus brevis* · Resting cells · L-Glutamate decarboxylase · Bioconversion

63.1 Introduction

γ-Aminobutyric acid (GABA), an ubiquitous natural nonprotein amino acid, is widely distributed in plants and animals [1]. GABA, which acts as an important type of inhibitory neurotransmitter in mammal brain and spinal cord, could play an important role in hormone and nutritional factors in non-nerve tissues [2]; so GABA has an important regulating effect on the normal physiological functions of an organism [3]. GABA has several well-known physiological functions, such as antihypotension effects, improving cerebral function, delaying senility, tranquilizer

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K. Shi et al.

effects, activating renal and liver functions, and promoting the secretion of growth hormone [4–9]. GABA, therefore, has the potential as a bioactive component in foods and pharmaceuticals due to its bioactive effects. Glutamate decarboxylase (GAD, EC 4.1.1.15), which is widely distributed in prokaryotic and eukaryotic cells [10, 11], is the unique enzyme to catalyze the conversion of L-glutamate or its salts to GABA and CO_2 via the single-step α -decarboxylation [12–14].

With the increasing commercial demand for GABA, various chemical and biological synthesis methods for GABA have been studied [15–17]. Chemical synthesis of GABA has a high cost and a low production ratio. Furthermore, it cannot be used in the food industry due to dangerous or poisonous solvents used in the synthesis. Microbiological fermentation method is characterized by easy operation, moderate reaction conditions, safety, and low cost, whereas the downstream process and production cycle is long. Bioconversion method of GABA may be a much more promising method due to simple reaction procedure, high catalytic efficiency, saving production cost, shortening production cycle, less by-product, less pollution, and environmental compatibility [18]. In the current study, the process of enzymatic conversion for γ -aminobutyric acid by *Lactobacillus brevis* CGMCC No. 3414 resting cells and the influence of several factors on GABA production were investigated and the optimal composition of bioconversion system was determined in order to provide reference for using GAD of GRAS lactic acid bacterium to catalyze the conversion of L-glutamate or its salts to GABA.

63.2 Materials and Methods

63.2.1 Materials and Equipment

The *L. brevis* strain used in the present work was isolated from naturally pickled Chinese vegetables in our previous work, and it is stored at the China General Microbiological Culture Collection Center as *L. brevis* CGMCC No. 3414. The experimental equipments are high performance liquid chromatography (Agilent 1,260 infinity, Agilent Technologies, USA) with an Agilent ZORBAX StableBond C18 column $(4.6 \times 250 \text{ mm})$, Hitachi CR22GIII high-speed refrigerated centrifuge (Hitachi Koki Co., Ltd., Japan), and desk high speed centrifuge (Hunan Xiang Yi Laboratory Instrument Development Co., Ltd., China).

The glucose-yeast extract-peptone (GYP) medium for seed preparation contained (g/L): glucose, 10; yeast extract, 10; peptone, 5; sodium acetate, 2; MgSO₄·7H₂O, 0.02; MnSO₄·4H₂O, 0.001; FeSO₄·7H₂O, 0.001; NaCl, 0.001; and pH 6.8.

The fermentation medium consisted of the following (g/L): glucose, 30; yeast extract, 30; peptone, 5; sodium acetate, 3; FeSO₄·7H₂O, 0.001; MgSO₄·7H₂O, 0.03; NaCl, 0.001; MnSO₄·4H₂O, 0.019; L-sodium glutamate (L-MSG), 10; pH 6.8.

GABA standard was purchased from Sigma-Aldrich. L-MSG (purity ≥ 99 %) was provided by Tianjin Red Rose Food Co., Ltd, China. Chromatography reagents were obtained from Tianjin Kemiou Chemical Reagent Co., Ltd, China. Other reagents were of analytical reagent grade and biochemical reagents.

63.2.2 GABA Biotransformation by L. brevis CGMCC No. 3414 Resting Cells

One loop of slant culture of *L. brevis* CGMCC No. 3414 was inoculated into a 50-mL seed medium in a 500-mL Erlenmeyer flask and incubated without agitation at 30 °C for 24 h. The seed culture was then inoculated at 10 % (v/v) into 200 mL fermentation medium in a 500-mL Erlenmeyer flask and cultivated without agitation at 30 °C. After 16–72 h cultivation, the cells were harvested by centrifugation at 5,000×g for 10 min at 4 °C, washed twice with sterile water, and then resuspended in 0.2 M disodium hydrogen phosphate–citric acid buffer (containing 25 mM L-MSG, pH 4.6) containing 50 g wet cells per liter. The bioconversion was performed at 30 °C and 180 r/min shaking for 4 h, and then the bioconversion broth was detected after centrifugal filtration.

63.2.3 Analytical Methods

The volume of 1 mL bioconversion broth was centrifuged at 12,000 r/min for 2 min, then a 10-μL supernatant was pipetted into a 1.5-mL centrifuge tube, added with 100 μL derived buffer, 100 μL derivative agent (2,4-dinitro-1-fluorobenzene), and 790 μL potassium dihydrogen phosphate buffer, uniformly mixed, water bath at 60 °C for 1 h. After that, the derivation mixture was filtered by a 0.22 μm cellulose acetate membrane filter, and the filtrate was stored at 4 °C for HPLC assay. Gradient elution conditions: Agilent ZORBAX StableBond C18 column, 30 °C column temperature, 360 nm ultraviolet detection wavelength, 0.8 mL/min flow velocity, 10 μL sample amount, mobile phase A: water, mobile phase B: acetonitrile, mobile phase C: methyl alcohol, and mobile phase C: 0.05 M sodium acetate solution [19].

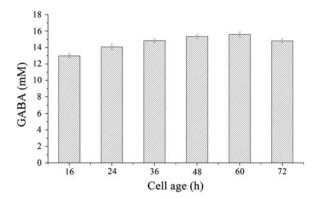
63.3 Results and Discussion

63.3.1 Effect of Cell Age on GABA Production by L. brevis CGMCC No. 3414 Resting Cells

The glutamate decarboxylase produced by lactic acid bacteria is an inducible enzyme [20]. The current study evaluated the resting cell of different cell age

K. Shi et al.

Fig. 63.1 The influence of cell age on GABA production



bioconversion of MSG to GABA. *L. brevis* CGMCC No. 3414 cells were harvested at culture time of 16, 24, 36, 48, 60, and 72 h, respectively, which is collected according to the above methods. The resting cells with the same concentration were resuspended in bioconversion system, and then the bioconversion broth was detected after a period of reaction time.

As shown in Fig. 63.1, the GABA yield was greatly improved with the increase of culture time. When the *L. brevis* CGMCC No. 3414 cells were harvested at culture time of 60 h, the cells had the strongest transformation ability, whereas at 72 h, the transformation ability of cells decreased. This is mainly because that a great amount of secondary metabolites produced by the cells began to accumulate after strains were developed to the stationary phase, especially the glutamate decarboxylase. When *L. brevis* CGMCC No. 3414 cells were harvested in this period for transformation, the yield of GABA could reach the highest. However, we conjecture that the content of GAD had little difference between the cells cultivated at 48 and 60 h. Therefore, the yield of GABA is of no significant difference between them. Considering the fermentation period and cost, culture time of 48 h was used for subsequent studies on GABA production by *L. brevis* CGMCC No. 3414 resting cells.

63.3.2 Effect of Cell Concentration on GABA Production by L. brevis CGMCC No. 3414 Resting Cells

Generally speaking, the yield of GABA will increase with the increased cell concentration. As shown in Fig. 63.2, the yield of GABA increased rapidly in the concentration range of 10–25 g/L. When the cell concentration increased further, the yield of GABA increased little. The specific conversion ratio was investigated in the current study. Figure 63.2 indicates that the specific conversion ratio decreases with the increased cell concentration, and the turning point appears at cell concentration of 50 g/L. Given the yield of GABA, training cost, and efficiency of the

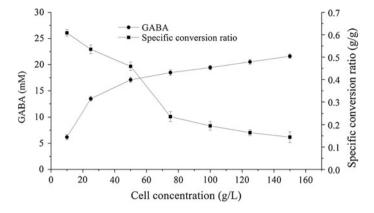


Fig. 63.2 The influence of cell concentration on GABA production

cell culture, 50 g/L cell concentration was used for subsequent studies on GABA production by *L. brevis* CGMCC No. 3414 resting cells.

63.3.3 Effect of Bioconversion Time on GABA Production by L. brevis CGMCC No. 3414 Resting Cells

When GABA is produced by bioconversion, glutamate reacts with GAD after penetrating the cell membrane and getting into the cell cytoplasm. The reaction ratios of GABA bioconversion were the highest at (0–3) h, slowed down at (4–5) h, and then became almost constant after 5 h (Fig. 63.3). This is mainly because that substrate consumption and product accumulation reach relative balance, and the

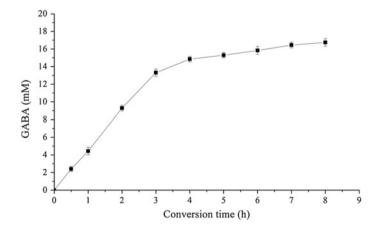


Fig. 63.3 The influence of bioconversion time on GABA production

K. Shi et al.

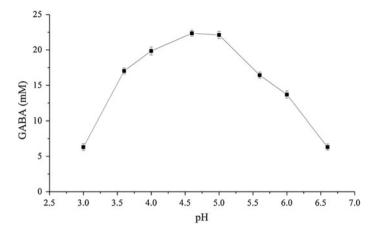


Fig. 63.4 The influence of pH on GABA production

decrease of substrate has an effect on reaction ratio. Thus, considering product accumulation and reaction ratio, a conversion time of 4 h was used for subsequent studies on GABA production by *L. brevis* CGMCC No. 3414 resting cells.

63.3.4 Effect of pH on GABA Production by L. brevis CGMCC No. 3414 Resting Cells

The reaction pH of GAD from different sources has a great difference. GADs of plant origin show the highest activity at pH 5.5–6.6 [21]. Nomura et al. [22] reported that the optimum pH of GAD produced by *Lactococcus lactis* subsp. *lactis* was 4.7. Huang and Mei [23] reported that the optimum pH of GAD produced by *L. brevis* CGMCC 1306 was 4.4. Figure 63.4 exhibits that the effect of pH has influence on bioconversion. The result showed that the suitable pH range of GAD was 4.6–5.0. The study proves that *L. brevis* GAD displays good pH stability in the pH 4.0–5.0 and the optimum pH is about 4.5. The optimum pH of bioconversion in the current study is 4.6 which is similar with the references. Thus, pH 4.6 was used for subsequent experiments.

63.3.5 Effect of MSG Substrate Concentration on GABA Production by L. brevis CGMCC No. 3414 Resting Cells

Substrate concentration has great influence on reaction ratio in enzymatic reaction. Substrate usually has double effects of accelerating the reaction speed and inhibiting

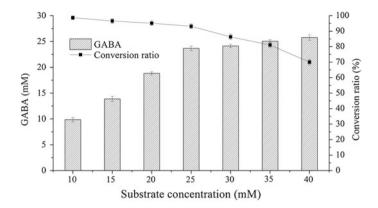


Fig. 63.5 The influence of MSG substrate concentration on GABA production

enzyme activity. The influence of initial MSG concentration on GABA production and the ratio of bioconversion by *L. brevis* CGMCC No. 3414 resting cells were studied in current research. In Fig. 63.5, the yield of GABA by resting cell transformation at lower ranges of MSG substrate concentration (10–25 mM) gradually increased with the increase of MSG concentration by 4-h biotransformation. The final GABA yield was almost constant while the substrate concentration increased to over 25 mM; however, the conversion ratio decreased significantly. This result indicates that low concentration of substrate has little effect on enzymatic reaction and the product inhibition is not found during this bioconversion. Thus, 25 mM MSG was used for GABA bioconversion by *L. brevis* CGMCC No. 3414 resting cells.

63.4 Discussion

Lactobacillus brevis, which has full GAD activity and is the preferable species to catalyze the conversion of L-glutamate or its salts to GABA, is a GRAS microorganism. Using bioconversion method for GABA production not only has advantages like high yield and short production cycle, but also obtains high purity product and overcomes the limitation of microbiological fermentation method. Since each influencing factor of GABA production by *L. brevis* CGMCC No. 3414 resting cells was analyzed and discussed, the optimal composition of bioconversion system contained 50 g/L resting cells, cell age at 48 h fermentation, 0.2 M disodium hydrogen phosphate—citric acid buffer, and 25 mM MSG. When the bioconversion system was performed at pH 4.6, 30 °C, and 180 r/min shaking for 4 h, GABA production in biotransformation mixture was 23.29 mM, and the molar yield rate of bioconversion reached 93.15 %. Overall, this study laid a foundation for catalyzing the conversion of L-glutamate or its salts to GABA by *L. brevis* resting cells.

K. Shi et al.

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Part IV Progress of Biotechnology

Chapter 64 [FeFe]-Hydrogenase: Catalytic Center and Modification by Genetic Engineering

Jiayi He and Chunfei Wu

Abstract In this review, we highlight the [FeFe]-hydrogenase, which is capable of catalyzing the splitting of molecular hydrogen to produce electrons and protons or catalyzing the reversible reaction: $2H^+ + 2e^- = H_2 \uparrow$ as a potential renewable fuel. We have focused on [FeFe]-hydrogenase because of structural studies have shed more light on the hydrogenase activity than the [NiFe]-hydrogenase. Our studies on the [FeFe]-hydrogenase from *Chlamydomonas reinhardtii* CC-503 ($HydA_1$) have also been highlighted. There are two factors influencing the multiplexed hydrogenase activity: a single hydrophobic channel of catalytic center, which is known as the H-cluster active site based on site-directed mutagenesis, moreover, some promising results have already been obtained. Modifications of [FeFe]-hydrogenase can improve its stabilization and activity in vitro, increase the efficiency of bioenergy utilization, and promote industrial amplification of biofuel production.

Keywords Renewable energy resource \cdot Biohydrogen synthesis \cdot [FeFe]-hydrogenase \cdot Gene mutation

64.1 Introduction

Driven by the growing appetite for energy consumption and increasing severity of global environment issue, the requirement for replacing traditional fossil fuel with alternative energy becomes urgent. Sustainable, secure energy supply plays a crucial role in stabilizing global economic development. Thus, to seek reproducible, clean and effective new energy source has become the primary task in the field of

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522 J. He and C. Wu

energy research [1]. Biohydrogen synthesis technology can be one preferred solution to these conflicts in future energy supply, using water as substrate and solar energy as power supply, possessing varieties of incomparable advantages, such as high transformation efficiency, environmental amity and mild reaction conditions, for which this technology has been enrolled in strategic agenda and environment protection issue around the world.

Currently, biohydrogen synthesis technology can be performed in three ways: by photosynthetic bacteria, by fermentation and by microalgae, among which microalgae hydrogen production technology is considered to be the most promising one, for its high energy conversion efficiency, unlimited resource of solar power and relatively low pollution [2]. The major subclasses of microalgae that can be used for hydrogen synthesis are *cyanophyta* and *chorophyta*. Reversible hydrogenase is the predominant hydrogen production source for these microalgae. The activity of reversible hydrogenase in *chlorophyta* is higher than that of nitrogen-based hydrogen production enzyme in *cyanophyta*, for mechanism of hydrogen production varies with organisms. Hydrogenase can transform proton into hydrogen directly, with no consumption of ATP (adenosine triphosphate), which in turn result in spontaneous accumulation of solar energy. Research on reversible hydrogenase in *chlorophyta* has become a hot topic in the field of biohydrogen synthesis recently [3].

64.2 Mechanism for Hydrogen Production in *Chorophyta* and Key Enzymes

64.2.1 Mechanism of Biohydrogen Synthesis

It was reported by Jackson that *Cyanobacterium Anabaena* has the ability to generate hydrogen as early as 1896 [4]. Gaffron and Rubin discovered that green algae Scenedesmus *obliquus* can produce hydrogen in the condition of hypoxia [5]. Researchers also reported that *Chlamydomonas reinhardtii* and *Chlorella* could generate hydrogen persistently under certain condition. The basic advantage of hydrogen synthesis by green algae is that solar energy can be fully used—water act as a direct raw material. Solar transformation efficiency is 10 times more than the other plants which live on photosynthesis, sometimes 10 % (lower limit of transformation efficiency in the field of hydrolysis of hydrogen production with applicable economic benefit). Increasing number of microorganisms and their hydrogen synthesis mechanisms have been studied in recent years and significant progress has been made.

Based on the metal ion that hydrogenase contains, hydrogenase fall into the following major classes: [FeFe]-hydrogenase, [NiFe]-hydrogenase, and metal-free hydrogenase. However, [FeFe]-hydrogenase is considered to be the most promising hydrogen synthesis catalyzed at its unique catalytic center and with higher efficiency than the other two classes [6, 7]. In green algae, light is captured by specialized Light

Harvest Complex (LHC), which confers the energy to photosynthetic reaction centers of photosystem I (PSI) and photosystem II (PSII). Hydrogen activation is mediated by a family of enzymes, termed hydrogenases, which either provide these organisms with reducing power from hydrogen oxidation or act as electron sinks, following the reaction: $2H^+ + 2e^- = H_2 \uparrow [8]$.

The electrons are passed down via chain reaction of plastoquinone (PQ), cytochrome b6f complex (Cytb6f), and plastocyanin (PC) to PSI, and then electrons are transmitted to Ferredoxin (Fd), which helps hydrogenase produce H₂ (Fig. 64.1) [9, 10]. Aerobic photosynthesis and mitochondrial respiration action can be changed in the absence of sulfur during algae culturing, because sulfur is an essential component of protein residue cysteine and methionine, and the repair cycle can be blocked without the presence of sulfur [6]. Eventually, respiration reaction can exceed photosynthesis action, as a result of the absence of sulfur, leading to the decrease of oxygen concentration, which can provide an ideal environment to maintain the activity of hydrogenase [7].

64.2.2 Key Enzyme Hydrogenase and Its Reaction Center

[FeFe]-hydrogenase is found in single cell green algae (e.g. *Chlamydomonas reinhardtii*), anaerobic fungi, and some procaryotic organisms so far. [FeFe] H₂ases house four iron–sulfur (FeS) clusters in addition to the active-site "H-cluster". Each H-cluster is coordinated by diatomic molecules CN- and CO ligand [11, 12]. In

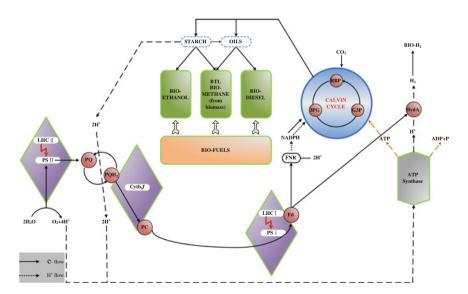


Fig. 64.1 Mechanism for biohydrogen synthesis in green algae

J. He and C. Wu

[FeFe]-Hydrogenase

Fig. 64.2 H-cluster active site of [FeFe]-hydrogenase [15]

addition, there is also a Fe bridge ligand exiting, apart from the structure that each Fe atom is bridged to each other by two S atoms in H-cluster active site. The known [FeFe]-hydrogenase atomic structures, respectively, show a common core, which contains a moiety, deeply buried inside the protein, with an Fe–Fe dinuclear center, nonproteic bridging, terminal CO and CN– ligands attached to each of the iron atoms, and a dithio moiety, which also bridges the two iron atoms and has been tentatively assigned as a di(thiomethyl)amine (Fig. 64.2) [8].

Hydrogenase has its own active channel to generate hydrogen. Current data indicate that HydF, HydE, and HydG would have the key role of a scaffold upon chemically modifying the [FeFe]-hydrogenase, a carrier that transfers the latter HydA, thus completing the maturation process, which means that if these proteins are expressed alone or two, the activity of hydrogenase will be very low with little hydrogen production ability [13, 14].

64.2.3 Tolerance to Oxygen

H-cluster active site, the activity site of hydrogenase, is highly sensitive to O_2 , which is produced by PSII and can make hydrogenase lose its activity. In order to improve its tolerance to oxygen, the mechanism of oxygen sensitivity of [FeFe]-hydrogenase was studied. It is reported that this sensitivity is related to the catalytic center [2Fe–2S]. When O_2 reaches its catalytic site, an irreversible oxidation reaction will take place, in which O_2 binds to Fe_2 of [2Fe–2S], causing H^+ to lose its ability to bind to Fe_2 of the H-cluster active site [15]. One plausible way to increase oxygen endurance ability of hydrogenase is to modify its structure, preventing O_2 from reaching the catalytic site of [FeFe]-hydrogenase by shrinking the diameter of gas channel, through which only H_2 can pass other than O_2 . It is proven that the oxygen sensitivity of [FeFe]-hydrogenase is related to its amino acid sequence.

64.3 Application of Biotechnology in the Study of Hydrogenase

Researchers have attempted to optimize the amino acid residue inside the hydrogen synthesis channel, in order to solve the problem that hydrogenase can be easily poisoned by O₂. Several plausible ways, including decreasing the diameter of gas channel, shrinking the cubage of catalytic site on this enzyme, was experimented to increase hydrogen production efficiency. Next, we discuss some of the genetically modified strains that led to improved hydrogen production.

64.3.1 Recombination of Gene In Vitro

DNA recombinant method is used to introduce extracellular DNA into cell, making it possible for exchanging genes between species. It includes homologous recombination, site specific recombination, transposon recombination, and abnormal recombination, which is a common way of mutation in nature.

Vanabendroth reported that the production of hydrogenase protein can doubly increase when the original gene $HydA_1$ of C. acetobutylicum was replaced by a homologous gene [16]. Many attempts have been made to generate O₂-tolerant hydrogenases through random mutagenesis in vivo and in vitro [17]; Yacoby used protein engineering method to fuse Fd to the N terminal of HydA, as a result, the expression levels of protein increased by about four times. The activity and purification of enzyme increased dramatically, which was proven by the following experiment [18, 19].

64.4 Gene Mutation Method

Two main types of gene mutation method systems exist: site specific mutation and random mutation. Cycle PCR and Quick-change Kit are used in the method of site-specific mutation to produce key mutated sites. Random mutation method runs in other way. The reaction condition, including concentration of Mg²⁺, Mn²⁺, dNTP, or low-cost polymerase was modified during amplification of DNA to optimize the frequency of mutation, in order to generate random mutation site in gene of interest.

64.5 Selection of Mutation Site

The key site of interest is the one that can affect the gas channel and can be modified by protein engineering method. It is reported that the hydrophobic amino acid residue, which is vital to the surface of protein, is essential for the diffusion of

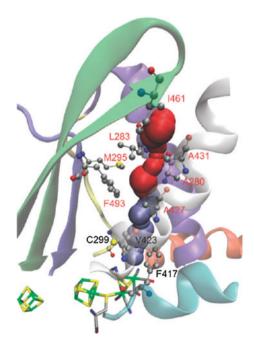
526 J. He and C. Wu

oxygen. One way to select mutation site is to choose the amino acid which is small in the channel in nature. If the small residue can be replaced by larger one, it can prevent oxygen from reaching the active site of hydrogenase. The structure of hydrogenase of *Clostridium pasteurianum* has been revealed by X-ray method. The arrangement of amino acid inside the hydrophobic environment can be acquired by computer simulation (Fig. 64.3). It was reported that the activity of hydrogenase can increase when Leu-238 was changed into trp, Val-423 was mutated into Ala-321 was replaced [20].

The other choice of mutation sites is that highly conserved amino acid residues. The H-cluster active site has its unique [Fe-s] diatomic center, in which each Fe has CN and CO ligand. The function of CN and CO is to stabilize Fe, making it easier to bond to hydrogen. The helper genes HydE, HydF, and HydG have its unique domain that can help synthesize these ligands, such as [Fe–S] binding site. In order to obtain high activity mutation of hydrogenase, the amino acid sequence of helper genes was chosen for mutation, The residues Cys 302, 354, 356 and residues His 304, 352 of HydF were mutated in new Apollo hot mantle bacteria by Berto in 2012, which has demonstrated that the three Cys residue help the binding of HydF with [4Fe-4S] and the activation of hydrogenase [22].

Modification of hydrogenase active site is done by genetic engineering based on the above theory in our experiments. The fact that co-expression of active [FeFe]hydrogenase requires strictly anaerobic conditions has always been interpreted as revealing the extreme oxygen sensitivity of these enzymes. Compared with wild

Fig. 64.3 X-ray analysis of gas channel [21]



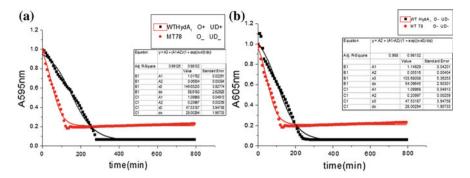


Fig. 64.4 a, b Specific activities of wile-type and mutant $HydA_1$. $Dark \ red$, mutant 78 (residues Ala to Leu); black, wild-type $HydA_1$; O^+ , aerobic treatment; O^- , anaerobic treatment; UD^+ , cell disruption with sonic oscillator in aerobic; UD^- , cell disruption with sonic oscillator in anaerobic. **a** Specific activities of MT 78(O^- , UD^-) and WT $HydA_1(O^+$, UD^+). **b** Specific activities of MT 78 (O^- , UD^-) and WT $HydA_1(O^-$, UD^-)

type, our data and quantitative analysis show that the rate of hydrogen synthesis by mutant is increased significantly. Some promising results have been obtained already by regulatory factor experiment, and regulation experiments showed that the control system of reliable, stable operation, high universality, moreover establishment of calculation and related model is under study, all of the further researches are in progress (Fig. 64.4a, b).

64.6 Outlook

As a promising ideal energy, hydrogen has relatively high efficiency without apparent pollution to the environment. However, much more need to be done in terms of expression, tolerance to oxygen, and the activity of the hydrogenase. Genetic engineering or protein engineering methods have been used major tools in the study of hydrogen biosynthesis. Moreover, modern molecular biology methods have been discussed in this paper, such as gene modification, site specific mutation method [23], and molecular dynamics calculations [21]. In addition, it is a very urgent problem to be solved, which will require further studies on the specific activities, dynamics calculations, and mechanistic modeling of different mutant strains.

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Chapter 65 Characteristics of *Staphylococcus aureus*Isolates from Raw Milk

Xiaomei Zhang, Qian Li and Hongjiang Yang

Abstract *Staphylococcus aureus* has raised great health concerns worldwide. In this work, with the traditional culture-based methods, totally 70 bacterial isolates on Barid–Parker medium from raw milk samples are collected from 181 healthy cows in a dairy farm located in Northern China. With comparative sequence analysis of 16S rDNA genes, 18 isolates were identified as *S. aureus*. One isolate was resistant to cefoxitin and tentatively considered as a methicillin-resistant *S. aureus* (MRSA) strain. *Spa* typing was performed to discriminate the strains. Four *spa* types were determined, including t3104 in 12 strains, t267 in 3 strains, t521 in 2 strains, and t528 in strain Z1584. The type t11557 was a new *spa* type discovered in this work and its repeats succession was 07-12-21-13-34-34-33-13. Our work suggested that the prevalence of MARA strains was quite low in fresh raw milk produced in the dairy farm located in Northern China.

Keywords Microbial communities • Methicillin-resistant *Staphylococcus aureus* (MRSA)

65.1 Introduction

Currently, a few molecular techniques have been developed that can identify bovine mastitis causing agents. Numeral bacterial genes were selected as targets and detected with such methods, including conventional PCR [1], multiplex PCR [2], real-time PCR [3]. Commercial kits were already available for this kind of assay with relatively high accuracy and sensitivity [4].

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K. Zhang et al.

In this work, we collected raw milk samples from 181 healthy cows in one dairy farm located in Togtoh County, a relatively remote area in northern rural China. Bacteria were isolated from raw milk via traditional culture-based methods. By comparative analysis of bacterial 16S rDNA genes, the isolates were further determined to species level. Finally, we investigated the phenotypic and genotypic traits of the *Staphylococcus aureus* isolates with various methods.

65.2 Materials and Methods

65.2.1 Cultivation of Bacteria in Raw Milk

Raw milk samples were collected from 181 healthy dairy cows without clinical signs of mastitis in one dairy farm in Togtoh County of Inner Mongolia Autonomous Region.

Prepare tenfold serial dilutions of milk samples and spread 100 μ L samples of each dilution on Baird–Parker agar plates. After incubation at 35 °C for 24–48 h, colony morphology was recorded to discriminate bacteria in all experiments.

65.2.2 Identification of the Isolates with Comparative Sequence Analysis of the 16S rDNA Genes

Genomic DNA of the isolates was extracted from cultures as described previously. Universal primers 27f (5'-AGAGTTTGATCCTGGCTCAG-3') and 1492r (5'-GGTTACCTTGTTACGACTT-3') were used for 16S rDNA gene amplification and sequencing of PCR products [5]. Sequence similarity was analyzed with BLAST program provided by NCBI for species identification.

65.2.3 Antibiotics Resistance Test of S. aureus Isolates

Antibiotic resistance in *S. aureus* strains was determined with disk diffusion test. Cefoxitin, oxacillin, tetracycline, ciprofloxacin, and gentamicin were tested and the results were recorded according to the CLSI documents (CLSI 2009). Gene *mecA* was detected with primers mecA1 and mecA2 [6].

65.2.4 Detection of Virulence Genes of S. aureus Isolates

Virulence genes were tested as described previously [7]. Briefly, *S. aureus* fibronectin binding protein A precursor gene *fnbA* was amplified with primers FNBA-1

and FNBA-2. *Staphylococcus aureus* fibronectin binding protein B precursor gene *fnbB* was amplified with primers FNBB-1 and FNBB-2. *Staphylococcus aureus* clumping factor A gene *clfA* was amplified with primers CLFA-1 and CLFA-2. *Staphylococcus aureus* clumping factor B gene *clfB* was amplified with primers CLFB-1 and CLFB-2.

65.2.5 Spa Typing

Staphylococcus aureus spa gene coding for protein A was amplified with primers 1113f (5'-AAAGACGATCCTTCGGTGAGC-3') and 1514r (5'-CAGCAGTAGT-GCCGTTTGCTT-3'). The sequence data was analyzed for spa typing with the Bionumerics software (http://www.ridom.de) [8].

65.3 Results

65.3.1 Isolation and Identification of S. aureus Isolates

In Baird–Parker medium, 70 isolates with colonies displaying gray-black color, opaque zone surround, and an outer clear halo were selected as presumptive *Staphylococcus* strains.

To identify the isolates, sequence analysis of 16S rDNA gene was performed in 70 presumptive *Staphylococcus* strains. Totally, 18 *S. aureus* isolates were identified.

65.3.2 Antibiotic Resistance Analysis of S. aureus Isolates

Strain Z1395 was resistant to cefoxitin (zone diameter = 17.7 mm), and it is tentatively considered as methicillin resistance according to the recommendation of CLSI (CLSI 2009). Strain Z1395 was also resistant to oxacillin, ciprofloxacin, and tetracycline showing its multiple drug resistance characteristics. Strain Z16004 was intermediate to cefoxitin (zone diameter = 21.2 mm) and ciprofloxacin, and susceptible to oxacillin, gentamicin, and tetracycline (Table 65.1). However, no *mecA* gene was detected by PCR in all 18 strains, suggesting non-*mecA*-mediated methicillin resistance mechanism might exist in strain Z1395.

K. Zhang et al.

Table 65.1 Analysis of antimicrobial drug resistance of *S. aureus* strains

Strains	Antibiotics resistance							
	Cefoxitin	Oxacillin	Ciprofloxacin	Gentamicin	Tetracycline			
Z160004	I	S	I	S	S			
Z164	S	S	S	S	I			
Z191	S	S	S	S	S			
Z221	S	S	S	S	S			
Z222	S	S	R	S	I			
Z37002	S	S	S	S	I			
Z1126	S	S	S	S	I			
Z1127	S	S	S	S	S			
Z1186	S	S	I	R	R			
Z1261	S	S	R	R	R			
Z1263	S	S	S	S	S			
Z1394	S	S	I	R	R			
Z1395	R	R	R	S	R			
Z1583	S	S	S	S	R			
Z1584	S	S	I	S	I			
Z1585	S	S	I	S	R			
Z1587	S	S	I	S	I			
Z1588	S	S	S	S	S			

S susceptible; I intermediate; R resistant

65.3.3 Characterization of Virulence Factors of the S. aureus Isolates

Clinical strains C123 and C126 were used as positive controls in coagulase and hemolytic activity tests. Eighteen strains were coagulase positive. Twelve strains displayed alpha hemolysis, five strains displayed beta hemolysis, and one strain displayed gamma hemolysis (Table 65.2). In virulence gene test, PCR results showed both *clfA* and *clfB* genes were positive in all 18 strains (Table 65.2). Their sequences were subjected to further analyses (accession no. KC212097–KC212118).

65.3.4 Spa Typing

Partial *spa* gene was amplified and sequenced (accession no. KC212119–KC212140). Four *spa* types were determined, including t3104 in 12 strains, t267 in 3 strains, t521 in 2 strains, and t528 in strain Z1584. These *spa* types were different from t11557 of strain C123 and t2310 of strain C126 (Table 65.2). The type t11557

Strains	Virulend	Virulence genes				Hemolysis	Ridom type
	fnbA	fnbB	clfA	clfB			
Z160004	-	_	+	+	+	γ	t3104
Z164	-	-	+	+	+	β	t3104
Z191	-	-	+	+	+	α	t3104
Z221	-	+	+	+	+	α	t267
Z222	_	+	+	+	+	β	t267
Z37002	_	+	+	+	+	β	t267
Z1126	-	-	+	+	+	α	t3104
Z1127	-	-	+	+	+	α	t3104
Z1186	-	-	+	+	+	β	t3104
Z1261	_	+	+	+	+	β	t521
Z1263	+	+	+	+	+	α	t521
Z1394	+	_	+	+	+	α	t3104
Z1395	+	-	+	+	+	α	t3104
Z1583	-	-	+	+	+	α	t3104
Z1584	_	_	+	+	+	α	t528
Z1585	-	-	+	+	+	α	t3104
Z1587	_	+	+	+	+	α	t3104
Z1588	+	-	+	+	+	α	t3104
C123			+	+	+	β	t11557
C126			+	+	+	β	t2310

Table 65.2 Characterization of virulence factors in the isolated S. aureus strains

was a new *spa* type discovered in this work and its repeats succession was 07-12-21-13-34-34-33-13.

65.4 Discussion

In this work, culture-based method was used for the first round isolation of microorganisms in raw milk samples. Colony morphology was the primary criterion for discrimination of different isolates. Other characteristics, including growth on Baird–Parker medium, hemolysis, and coagulase activity, were also taken into considerations for isolates screening and discriminating. Genus *Staphylococcus* was the predominant group isolated from bovine milk samples, including 18 *S. aureus* isolates and 13 coagulase-negative staphylococci. It had been shown that *S. aureus* was the major pathogen causing bovine mastitis [9], and the coagulase-negative staphylococci were minor pathogens isolated from clinical or subclinical mastitis cases [10].

A variety of gram-negative bacteria were isolated and identified in our work. Some species were opportunistic pathogens normally existing in bovine milk K. Zhang et al.

including Citrobacter freundii, Escherichia coli, Enterococcus cloacae, Enterococcus faecalis, and Serratia marcescens. These coliform bacteria could possibly be responsible for the infections in some mastitis cases [11].

MRSA strains can be isolated from clinical samples, communities, and livestock-associated sources across the world. The emergence of MRSA in dairy cows has raised a great public health concern. In this work, one isolate was considered as a tentative MRSA strain without *mecA* gene detected. However, it was possible that *mecA* homologs including *mecC* gene may be responsible for its MRSA phenotype [12].

Spa typing method has been used extensively in discrimination of *S. aureus* isolates from various environments. In this study, 18 *S. aureus* isolates were determined belonging to four *spa* types. Most of them (12/18) were of t3104, which was first assigned to strains isolated from intensive care unit in Netherlands [13]. Clinical isolates C123 and C126 had different *spa* types compared with the 18 isolates from raw bovine milk, especially C123 had a new *spa* type t11557 (Repeats Succession: 07-12-21-13-34-34-33-13) assigned by the database. The results indicated that the strains might originate from different lineages [14].

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Chapter 66 EST-SSR Marker-Based Assay for Purity Identification of Melon "Green Angle"

Ou-Jing Li, Xiao-Mu Chen, Pu-Xian Xia, Zhong-You Pei, Yong Wang, Qing-Kuo Lan and Ruo-Wei Zhang

Abstract In order to define the purity of melon (*Cucumis melo* L.) "Green Angle" variety F1 seed, PCR amplification, detection, and analysis were carried out with leaf DNA, using EST-SSR molecular marker technology. The results show that, among 57 SSR primers, PCR products with one pair of primer showed polymorphism from parents. For this marker, parents showed single allele whereas the hybrids showed both the parental alleles indicating the heterozygosity of the hybrids. 51 seeds were tested using the marker, and one seed was off-type. The same lot of seeds was also tested in the field and the purity was 98.4 %. It showed that the EST-SSR marker-based PCR assay is effective in determining the purity of F1-hybrid seeds of melon "Green Angle".

Keywords Melon (*Cucumis melo* L.) · SSR marker · Hybrid seed purity

66.1 Introduction

Melon is one of the most important horticultural crops in the world. At present, the production has been widely used in hybrids [1]. Seeds are the most basic means of production in agricultural production, and its purity is the main indicators of seed quality.

Innovation team project of Tianjin "crop quality and genetic improvement of resistance"; Innovative talent training plan between Young and middle-aged backbone in Tianjin.

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Reducing seed purity will decrease crop yield and product quality, so the peasants will suffer huge economic losses. Conventionally, hybrid seed purity assessment is done through grow out test (GOT) which is based on the morphological and floral characters of plants grown to maturity [2]. Being land and labor intensive, time consuming, and influenced by the environment, there is an immense need to replace GOT with a simple, rapid, unbiased, and cost-effective DNA-based assay for hybrid purity assessment.

With the continuous development of modern agriculture technology, the seed purity identification tag technology from the traditional form developed to a comprehensive appraisal technology system which set morphological marker identification, biochemical marker identification, and DNA molecular marker identification as a whole. DNA molecular marker is the direct reflection of DNA genetic variation at the molecular level [3], and they have stability genetic, large amount of information, not affected by internal and external environment, and has nothing to do with gene expression or not, rapid detection and such outstanding advantages as easy operation, therefore, DNA molecular markers become one of the ideal of rapid detection methods for the identification of seed purity, including RAPD (random amplified polymorphic DNA), RFLP (restriction fragment length polymorphism), SSR (simple sequence repeat), and ISSR (inter-simple sequence repeat) markers. Because of its codominance, the advantages of high polymorphism, and simple experimental procedure, SSR markers are one of the most commonly used tags in the study of crop variety and seed purity identifications.

The use of SSR markers for assessing seed purity has been reported in agricultural crops like rice [4–6], maize [7], sunflower [8] and horticultural crops like tomato [9], cabbage [10], and melon [11]. However, genomic SSR markers have been developed in melon only recently. The current development of SSR markers are mainly genomic SSR and EST-SSR. Relative to the SSR genome, EST-SSR makes full use of existing sequencing data, eliminates some steps in the process of development of SSR primers, such as library construction and sequencing, and reduces the cost. In recent years, with the continuous improvement of sequencing technologies and molecular biology research, a large number of melons have been EST sequenced; so far, the international cucurbitaceae genome project team recorded 7,856 ESTs markers in watermelon, 126,940 markers in melon, and 359,105 markers in cucumber [12].

Through DNA extraction, PCR amplification, and EST-SSR analysis, this report confirms the codominant complementary belt type of stable hybrids and their parents, thus identifying the melon hybrid purity.

66.2 Materials and Methods

66.2.1 Plant Material and DNA Extraction

F1 hybrid of melon "Green Angle" (LV) and their parents were analyzed in this study. These materials were provided by the Tianjin academy of agricultural sciences' institute of vegetables.

DNA was extracted from young leaves using the CTAB extraction protocol. About 100 mg of tissue was placed in a 2.0 mL microfuge tube and incubated in 800 mL of CTAB extraction buffer (2 % CTAB, 100 mM Tris-HCl, pH 8.0, 20 mM EDTA, pH 8.0, 1.4 M NaCl) at 65 °C for 1 h. Centrifuged for 10 min and the supernatant was collected. To each tube, 500 mL of chloroform was added, mixed, and centrifuged for 10 min. DNA was precipitated by mixing 300 μ L of supernatant with 240 μ L 95 % isopropanol. The precipitation was washed with 70 % ethanol, dried to remove alcohol, and dissolved in 100 mL TE (10 mM Tris, pH 8.0). The DNA quantity for each sample was assessed on 1 % agarose gel and the DNA concentrations were normalized at 50 ng/ μ L.

66.2.2 Primers

Primers used for SSR reactions were designed using CMD SSR Server online. A set of 57 SSR primer pairs were used to assess hybrid purity. The length of the primer is 18–25 bp, the content of GC is 40–60 %. The primers were synthesized by Shanghai biological engineering Co., Ltd.

66.2.3 PCR Conditions

PCRs were performed in a 20 μ L reaction volume [10× Buffer (Mg²+) 2.0 μ L, dNTP (2.5 mM each) 1.6 μ L, TagE (5 U/L) 0.1 μ L, Primer (10 μ mol/L) 0.5 μ L, DNA(50 ng/ μ L) 2.0 μ L] in 96-well microtiter plate using ABI veriti 96-well Thermal cycler. An initial denaturation for 2 min at 95 °C was used. This was followed by initial 35 cycles of denaturation for 45 s at 94 °C, annealing for 45 s at 55 °C, and extension for 45 s at 72 °C. Subsequently, followed by 7 min final extension at 72 °C.

The amplification products were analyzed by 8 % polyacrylamide gel and visualized by silver staining. Take photos with the "Typhoon" Multifunction scanner.

66.3 Results

66.3.1 EST-SSR Polymorphism in the Parental Lines

In a preliminary study, a set of 57 simple sequence repeat (SSR) primer pairs were used to assess hybrid purity. These 57 EST-SSR markers were used for the polymorphism survey among the parental lines of melon hybrids. Only one EST-SSR marker (Table 66.1) detected polymorphism between the parents of the hybrids.

640 O.-J. Li et al.

			<u> </u>
No.	Marker	Sequence (5'-3')	Expected product (bp)
Pri73	M548-F	AACAGGTAGAGGAAAGCATG	145
	M548-R	TGACCCACTAGTACATCTCTC	

Table 66.1 Details of the polymorphic EST-SSR markers used in the study

For the marker, parents showed single allele whereas the hybrids showed both the parental alleles indicating the heterozygosity of the hybrids (Fig. 66.1).

66.3.2 Genetic Purity Testing of Safflower Hybrids

This report chose the EST-SSR marker (M548) which shows polymorphism between LV-1 and LV-2 were used to amplify the genomic DNA of LV hybrid seedlings through PCR (Fig. 66.2). The result showed that the hybrid purity analysis of these 51 seedlings with EST-SSR marker M548 could identify the same number of off-types which are consistent with the field test results and the purity of F1 generation is 98.04 %. It indicates the reliability of genetic purity assessment of hybrid and it also indicated that a single polymorphic marker should be sufficient to ascertain the genetic purity as suggested by Yashitola et al. [4] and Nandakumar et al. [5].

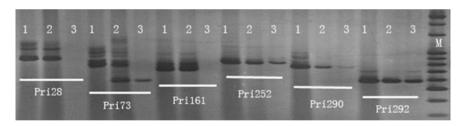


Fig. 66.1 EST-SSR markers polymorphism between parental lines of melon. I is female parent 'LV-1', 3 is male parent 'LV-2', 2 is 'LV' hybrid

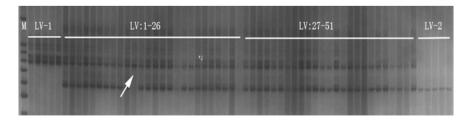


Fig. 66.2 Single seedling assay for detecting the genetic purity of the melon hybrid LV

66.4 Discussion

Though morphological and biochemical markers have been used for the characterization of genotypes, there are very few reports on the use of molecular markers for melon. In recent years, there are some reports on the use of molecular markers for cucurbitaceae [13, 14]. Though the utility of EST-SSR markers has been demonstrated in many species for applications discussed above, to the best of our knowledge, this is the first report of the EST-SSR marker-based genetic purity assessment of melon "Green Angle" hybrids.

In conclusion, molecular marker-based genetic purity assays will be highly useful in rapid and large-scale screening of hybrid seed lots. EST-SSR marker-based genetic purity testing would save the cost of hybrid seed storage for a whole season and the cost incurred on GOT. But, this pair of primer is only for melon "Green Angle" varieties; it is unknown for other varieties. The follow-up can be validated in other varieties and core primers of melon purity identification can be developed.

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642 O.-J. Li et al.

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Chapter 67 Exposure to Static Magnetic Fields Affects Insulin Secretion in INS Cells

Libin Mao, Zhixia Guo, Huiqin Wang, Qiongyao Wu, Nan Wang and Tong-Cun Zhang

Abstract The exposure to static magnetic fields (SMFs) has rapidly increased and is recently accompanied with the wide use of more and more electrical appliances. Although few researches have been reported on other types of magnetic fields that may have the effects on clinical therapy of diseases, it is rarely seen reports on the effects of moderate static magnetic fields on human health. In this study, in order to evaluate the influence of exposure to moderate static magnetic fields on biological systems, we cultured the INS-1 under exposure to moderate intensity of 400 mT and sham conditions for different times from 6 to 18 h, insulin secretion to medium and mRNA expression will be tested after exposure. Insulin secretion has significantly increased after exposed to SMFs with moderate intensity of 400 mT for 18 h compared with the sham condition. Furthermore, similar to the insulin secretion, insulin 1 and insulin 2 mRNA expression also increased after exposure. These results suggest that exposure to SMFs at 400 mT position for specific times might have positive stimulation in both insulin secretion and insulin mRNA expression, and these will be used hopefully in further investigations in vitro, even in the therapy clinically.

Keywords Static magnetic fields \cdot INS-1 cells \cdot Insulin secretion \cdot Clinical therapy

67.1 Introduction

There is no doubt that with more and more electrical appliances being used in our daily life [1], man has got more exposure to static magnetic fields (SMFs) than ever before. Previously, the beneficial effects of magnetic fields on inflammation [2],

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⁶⁴³

L. Mao et al.

osteoporosis [3], wound healing [4], and pain reduction [5] have been demonstrated by a plenty of intriguing studies using laboratory animals in vivo. Meanwhile, the frequency of exposure to SMFs has also increased rapidly, such as magnetic resonance imaging (MRI) and nuclear magnetic resonance (NMR) [6], which have been used in hospitals as noninvasive diagnostic tools. Therefore, it has become important to systematically elucidate the influence of SMFs on biological systems.

In an attempt to explain the biological effects of SMFs, it is useful to classify them as weak (<1 mT), moderate (1 mT-1 T), strong (1-5 T), and ultrastrong (>5 T). Previous studies have shown that SMF alone does not exhibit lethal effects on cell growth, survival, and DNA damage [7, 8], but are still controversial [9, 10]. It is urgent for us to provide a clear answer regarding the impact of SMFs on cell function, which is due to the use of SMFs of certain intensities and different cell types.

Pancreatic islets play a vital role in regulating the blood glucose levels of the body through the secretion of hormones such as glucagon and insulin, and the insufficient release of insulin is the basis of various forms of diabetes. Previously, Sakurai et al. reported that the effects of magnetic fields on insulin-secreting cells. Although the effects of strong static magnetic fields whose magnetic flux density is from 3 to 10 T has already been reported [11, 12], the effects of moderate intensity of SMFs were rarely seen before. In addition, insulin-secreting cells are appropriate for evaluating the effects of SMFs on biological systems, because insulin secretion is affected by calcium and its channel [13, 14], which was considered to be a candidate for interaction with SMFs.

Accordingly, in this study, we have investigated the effects of exposure to moderate SMFs at 400 mT on insulin mRNA expression by the insulinoma cell line, INS-1, under exposure to sham and SMFs conditions.

67.2 Materials and Methods

67.2.1 SMF Exposure System and SMF Exposure

SMF exposure (purchased from Tianjin Nibboh Magnets Co., Ltd), a static magnetic field at a intensity of 400 mT, was supplied by the xxx company. Briefly, the SMF exposure system was formed by a magnet with 10 mm thickness, 20 mm inner diameter, and 35 mm outer diameter into a CO_2 incubator where the atmosphere is maintained with 5 % CO_2 and 95 % humidified air and, the direction of the field is vertical. The temperature in the incubator was monitored by thermocouple sensor probes and is maintained at 37 \pm 0.2 °C. For the sham-exposure system, we utilized another magnet of the same shape, but without the magnetic field. The SMF in the sham-exposure space was less than 5 μ T.

67.2.2 Cell Culture

INS-1 cells were cultured in RPMI-1640 medium supplemented with 10 % fetal bovine serum (FBS), 10 mM 2-[4-(2-hydroxyethyl)-1-piperazinyl] ethanesulfonic acid (HEPES), 2 mM L-Glutamine, 1 mM Sodium Pyruvate, 50 μ M β -mercaptoethanol, 100 U/mL of penicillin, and 100 μ g/mL of streptomycin at 37 °C with 95 % humidified air and 5 % CO₂.

67.2.3 Insulin Secretion

Cells were seeded on the dishes of 35 mm with 2 mL culture medium at a density of 1.5×10^5 cells/cm². After cultured for 48 h, cells were washed once using the fresh medium, then the medium was changed to another 800 μ L/dish fresh culture medium. After INS-1 cells were exposed to SMF in 6, 12 or 18 h at 37 °C with 95 % humidified air and 5 % CO₂, The supernatant was collected and insulin secreted into the culture medium was measured by ELISA assay kits for rat insulin (WUHAN BEINGLAY BIOTECH CO.). Total protein was used to normalize the insulin secreted into the medium.

67.2.4 RNA Extraction, cDNA Synthesis, and PCR Reaction

Cells were seeded on the dishes of 35 mm in diameter at a density of 1.5×10^5 cells/ cm² and cultured as described in cell culture. After culture for 48 h, cells were washed once with PBS, the culture medium was then changed to another 1 mL fresh medium in each dish. After cells were exposed to SMF at 37 °C with 5 % CO₂ and 95 % humidified air for 6, 12 or 18 h, total RNA was extracted from cells with TRIzol reagent (Invitrogen). After that, the cDNA synthesis was performed using the reverse transcription with random primers. The thermal cycle profile was as follows: denaturation for 30 s at 95 °C, annealing for 30 s at 54–57 °C depending on the primers used, and extension for 30 s at 72 °C. PCR products were visualized on 2 % agarose gels stained with ethidium bromide under ultraviolet (UV) transillumination. The primers used in this study are summarized in Table 67.1.

Table 67.1 Primers used in the present study

Gene	Primer sequence
GAPDH	F: 5'-ATTCAACGGCACAGTCAAGG-3'
	R: 5'-GCAGAAGGGGCGGAGATGA-3'
Rat Insulin 1	F: 5'-CCGTCGTGAAGTGGAGGA-3'
	R: 5'-CAGTTGGTAGAGGGAGCAGAT-3'
Rat Insulin 2	F: 5'-TGGGGAGCGTGGATTCTTCTA-3'
	R: 5'-ACCTCCAGTGCCAAGGTCTGA-3'

¹ F Forward, R Reverse

L. Mao et al.

Glyceraldehyde-3-phosphate dehydrogenase (GAPDH) was used as an internal control to show equal loading of the cDNA samples.

67.3 Results

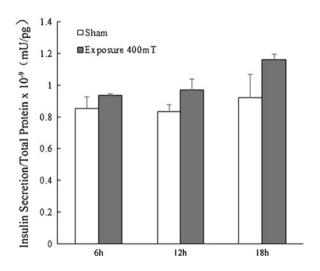
67.3.1 Insulin Secretion

INS-1 cells were exposed to SMF at 400 mT position for 6, 12, or 18 h, respectively, and insulin secretion were measured by rat ELISA assay Kit. Compared with the sham-exposed cells, insulin secretion was significantly increased under the exposure to SMF at 400 mT position for 18 h (Fig. 67.1). A slight increase in insulin secretion was observed following exposure to SMF for 6 and 12 h. The results show that 400 mT SMF can significantly enhance intracellular insulin concentration in INS cells in a time-dependent manner.

67.3.2 Effects of SMF on the Expression of Insulin mRNA

Next, the effect of SMF on the expression of insulin1 and insulin2 mRNA was evaluated by RT-PCR. When cells were exposed to SMF at 400 mT for 6, 12, and 18 h, the insulin secretion was significantly enhanced in a time-dependent manner, compared with sham-exposure conditions (Figs. 67.2 and 67.3).

Fig. 67.1 Alternations of insulin secretion levels during exposure to static magnetic fields (SMF). INS-1 cells were exposed to SMF at 400 mT position for 6, 12 or 18 h, respectively, compared with the sham-exposure cells. Insulin secretion levels in the culture medium were normalized by total protein



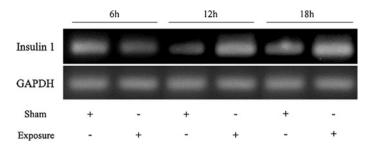


Fig. 67.2 Alternation of the expression of insulin 1 following exposure to static magnetic fields (SMF). INS-1 cells were exposed to SMF at 400 mT position for 6, 12 or 18 h, respectively, compared with the sham-exposure cells. Insulin 1 mRNA levels were normalized to GAPDH

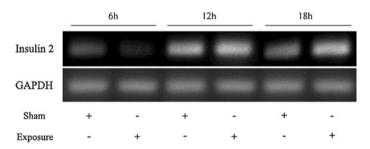


Fig. 67.3 Alternation of the expression of insulin 2 following exposure to static magnetic fields (SMF). INS-1 cells were exposed to SMF at 400 mT position for 6, 12 or 18 h, respectively, compared with the sham-exposure cells. Insulin 2 mRNA levels were normalized to GAPDH

67.4 Discussion

Diabetes mellitus has already become a deadly threat to human life. Its occurrence rate rapidly increased in recent years. The insufficient insulin secretion is recognized as one of the main causes of diabetes. There are a few studies that have already evaluated the influence of exposure to magnetic fields on pancreas islet function, including strong SMF [9] and ELFMF [15, 16], it is still difficult to compare the results which have been performed, not only because of the cells used are different from each other [13], but also the methods used were quite inconsistent in different researches. Therefore, the association between pancreas islet function and magnetic fields exposure is still controversial and uncertain [17]. In this study, we explored the effects of a medium SMF exposure on INS cells by measuring insulin secretion and the expression of mRNA. Exposure to SMF at 400 mT position for 18 h significantly increased insulin secretion. Similar to the results on insulin secretion, the expression of insulin 1 and insulin 2 mRNA were also obviously increased after exposure to SMF at 400 mT position for 18 h.

L. Mao et al.

In conclusion, SMF might be used clinically as a new medical therapy in diabetes. Of course, the precise underlying mechanisms remain unclear and need further studies.

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Chapter 68

The New Strategy of Breeding Cytidine Excessive Biosynthesis Mutants by pyr Operon Rearrangement of Bacillus amyloliquefaciens

Qing Wu, Huiyan Liu, Haitian Fang, Jianguo He, Xiaoguang He and Linan Yu

Abstract Cytidine is a good antitumor and antiviral intermediate which can also be used as a healthy food ingredients. With the market for cytidine increasing, the large-scale production of cytidine by microbial fermentation method has become a major way to solve this problem. While improving cytidine production, the regulation of pyr operon is essential to the excessive synthesis cytidine. However, its transcriptional regulation mechanism is unclear. The passage summarizes the regulation of de novo pyrimidine nucleotide biosynthesis (pyr genes) and the metabolic regulation mechanism of cytidine synthesis in *Bacillus amyloliquefaciens*. This paper makes the regulation protein pyrR as the research object, presents a series of *Bacillus amyloliquefaciens* pyrimidine operon transcriptional regulatory rearrangement strategy, effects the transcriptional regulation on cytidine biosynthesis, and aims to provide a theoretical basis for cytidine yielding strain.

Keywords Bacillus amyloliquefaciens \cdot Cytidine \cdot pyr operon \cdot pyrR

68.1 Introduction

Cytidine was known as cytidine, and $1-\beta$ -D-furan nucleoside cytosine. As a pyrimidine nucleoside, cytidine is a structural component of DNA in organisms. It is a nucleoside molecule that is widely used as a precursor for antitumor and antiviral drugs and it is the main raw material for the manufacture of cytarabine (Ara-CR), ring cytidine (CycloC), cytidine triphosphate (CTP), citicoline

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⁶⁴⁹

650 Q. Wu et al.

(CDP-Choline), and other drugs [1]. With the deep study of antiviral and antitumor, natural pyrimidine nucleoside demand become increasingly large, the demand for natural cytidine also increased. The present production methods to generate high cytidine production involves RNA hydrolysis, the combination of fermentation and synthetic, uracil precursor for the fermentation and direct fermentation method. However, the intracellular cytidine levels are very low and tightly controlled [2]. Compared to other methods, the use of mutant strains of microorganisms in direct fermentation for mass production of the nucleoside is an outstanding achievement of modern microbiology; such methods have the simplicity, short duration, easy control, high yield and low production cost as its advantages. The fermentation obtained strains using breeding techniques have capable of producing large-scale, single-cytidine, greatly reduced the production cost, accelerate research on fermentation for the production of cytidine.

68.2 Analysis of Pyrimidine Biosynthesis Pathway in *Bacillus amyloliquefaciens*

The genes encoding the enzymes of pyrimidine nucleotide biosynthesis de novo are the same in all bacteria. The regulation of cytidine biosynthesis in *B. amylolique-faciens* are summarized as follows (Fig. 68.1).

The first step of pyrimidine nucleotide biosynthesis de novo is to generate pyrimidine synthesis carbamoyl phosphate by carbamoyl phosphate synthetase catalyzed condensation reaction of CO₂ and glutamine. The next reaction catalyzed by aspartate transcarbamoylase (ATCase) with the condensation reaction of aspartate and carbamoylphosphate condensation to generate carbamoyl aspartate, which undergoes several reactions to produce UMP, the precursor for the synthesis of the cytidine. The de novo biosynthesis of pyrimidine nucleotides is regulated by intracellular concentrations of various nucleotides through feedback inhibition [3]. The intracellular CTP levels controls the intracellular UMP level through the control of ATCase and the feedback inhibition of CPS [4]. Further on that UDP is

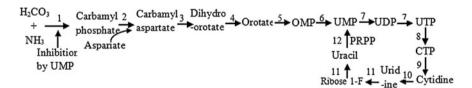


Fig. 68.1 Pyrimidine nucleotide biosynthetic pathway of *B. amyloliquefaciens*. The catalytic enzyme of each reactions are as follows: *1* Carbamylphosphate synthetase; *2* Aspartate transcarbamylase; *3* Dihydroorotase; *4* Dihydroorotate dehydrogenase; *5* Orotate phosphoribosyltransferase; *6* 5'-monophosphate decarboxylase; *7* UMP kinase; *8* CTP synthetase; *9* 5'-nucleotidase; *10* Cytidine deaminase; *11* Uridine phosphorylase; *12* Uracil phosphoribosyl transferase; *PRPP* Phosphoribosyl pyrophosphate

transformed by the UMP catalyzed by UMP kinase and CTP is transformed by the UTP catalyzed by the CTP synthase, which is regulated by CTP effectors through binding to specific allosteric sites on CTP synthase [5–7]. Moreover, the synthesis of CTP is located at a node in the cytidine biosynthesis pathway. Because of its importance, CTP synthase is also regulated by a feedback inhibition by CTP and cytidine [7]. Meanwhile, the CPS and ATCase enzyme reactions are key regulatory enzymes in the nucleotide metabolism and represent the major pathways for cytidine synthesis. It has been found that the CPS and ATCase generating UMP is a part of the overall regulation of CTP synthesis [8].

68.3 Research Progress of Pyrimidine High Yield Strains

Since 1980s to the 1990s, the Japanese scientists began to produce the pyrimidine nucleoside by fermentation [2]. High-level cytidine production depends on direct fermentation of carbohydrates by auxotrophic and regulatory mutant with *B. subtilis*, *B. pumilus*, *B. licheniformis*, *Escherichia coli*, and other organisms has been reported since the 1970s. Turnbough and Switzer [2, 9]. Among these species, *B. amyloliquefaciens*, *B. subtilis* and *E. coli* stains depending on their high growth rate and well-known physiological characteristics have been developed by mutagenesis or genetic manipulation and selected for the ability to grow on toxic cytidine analogues cytidine analogues [10–12]. The general strategy used for the development of a cytidine overproducing strain involves the alleviation of control mechanisms in key pathways [13–15]. Besides, these studies concentrated on the six operons encoding the enzymes required for the biosynthesis of UMP, the precursor of all pyrimidine nucleotides (Fig. 68.1). Nevertheless, we have never seen the report about the reconstruction of pyrimidine operon transcription regulatory system to optimize the cytidine metabolic network.

68.4 Pyrimidine Operon Regulatory Region Genes and Related to Pyrimidine Biosynthesis

Many diverse bacteria regulate the expression of genes of de novo pyrimidine nucleotide biosynthes by an transcriptional attenuation that is mediated by the PyrR, an mRNA-binding regulatory protein [2]. The genes of pyrimidine nucleotide biosynthesis in *B. amyloliquefaciens* are organized in an operon whose expression is governed by a single promoter and a regulatory protein encoded by *pyrR* [2]. The *pyr* operon of *B. amyloliquefaciens* contains 10 cistrons, which appears to be expressed as a single transcriptional unit. The first gene in the operon encodes pyrR, which has been shown to be the regulatory protein for the operon, and a uracil phosphoribosyltransferase (EC 2.4.2. 9) [2]. The second gene in the operon encodes pyrP, which is uracil permease. The remaining eight cistrons encode the six enzymes necessary

652 Q. Wu et al.

for de novo biosynthesis of UMP [2]. The map of *B. amyloliquefaciens pyr* operon are summarized as follows (Fig. 68.2).

The PyrR protein binds in a uridine nucleotide-dependent manner to three attenuation sites which lie upstream of the genes, called the binding loop and that is regulated by uridine and guanosine nucleotides, allowing a terminator hairpin to form and repressing the downstream genes. These have been seen in *Bacillus* [16], Lactobacillus [17, 18], Lactococcus [19], and Enterococcus [20] species. These attenuation regions are located in the 5' leader region of the operon upstream of pyrR (binding loop 1, BL1), the pvrR-pvrP intercistronic region (BL2), and the pvrR-pvrB intercistronic region (BL3) of pyr mRNA. Following transcription, the three attenuation sites within the nascent pyr mRNA are each predicted to fold into either an anti-antiterminator transcription terminator or a antiterminator structure. The key to regulation of the pyr operon lies in the ability of pyrR to favor formation of the terminator hairpins, which results in premature termination of transcription and reduces expression of the downstream genes. PyrR does this by binding to pyr mRNA when the protein is activated by binding of uridine nucleotides, PyrR binds to pyr mRNA at a site which lies upstream of the genes that it regulates. The binding of PyrR to the mRNA is dependent on the formation of a PyrR-UMP complex [2]. Previous research led to a model in which high levels of UMP stimulate PyrR to bind to a conserved sequence and secondary structure in the mRNA in each attenuation region called the anti-antiterminator or the binding loop; when UMP levels are low, the RNA-binding affinity of PyrR is reduced and the more stable antiterminator stem-loop is favored, leading to transcriptional read-through and expression of the downstream genes [2]. The pyrR protein of B. amyloliquefaciens also has been

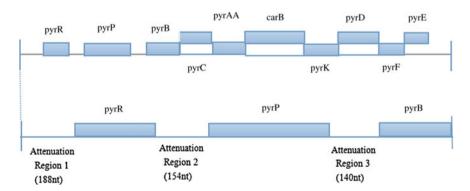


Fig. 68.2 The map of pyr operon in *B. amyloliquefaciens*. The genes shown in the figure and the encoded proteins are as follows: pyrR, pyr mRNA-binding attenuation regulatory protein; pyrP, uracil permease; pyrB, catalytic subunit of aspartate transcarbamylase; pyrC, dihydroorotase; pyrAA, glutamine-utilizing subunit of carbamylphosphate synthetase equivalent to carA; carB, catalytic subunit of carbamylphosphate synthetase; pyrK, electron-transferring accessory protein to dihdroorotate dehydrogenase; pyrD, dihydroorotate dehydrogenase; pyrF, OMP decarboxylase; pyrE, orotate phosphoribosyltransferase

shown to have the ability to catalyze the formation of UMP from uracil and PRPP, thus being a uracil phosphoribsyltransferase (UPRTase) as well as a regulatory protein [2].

68.5 Cytidine Excessive Biosynthesis Mutant Breeding Strategy-Pyrimidine Operon Regulatory Region Gene Modification

The prerequisite of cytidine fermentation is to design, build, and optimize the microbial pyrimidine nucleoside metabolic network, directional breeding highyielding strains that have capable of production cytidine. The pyrimidine metabolism of B. amyloliquefaciens is flux, whole genome sequencing has been completed, clear genetic background and suitable for breeding cytidine production of bacteria strains. Pyrimidine nucleotide operon (pyr operon) plays an important role in pyrimidine nucleotide synthesis of *B. amyloliquefaciens*. At present, a lot of studies about metabolism of cytidine focus on traditional mutagenesis breeding production bacteria, the research on molecular genetic breeding, and cytidine metabolic network reconstruction optimization report a little late. While there has little report on the impact of excess synthesis of cytidine about B. amyloliquefaciens and has never seen about the reconstruction of pyrimidine operon transcriptional regulatory system to optimize cytidine metabolic network. Through the modification of pyrimidine operon transcriptional regulatory region in B. amyloliquefaciens may be relieved the negative regulation of transcription, increase expression of gene dosage in the cells, and promote the rate-limiting step in the biochemical reactions, which may lead to increased production of cytidine.

68.5.1 pyrR Gene and Effect of PyrR Regulatory Protein Inactivation on Excessive Biosynthesis of Pyrimidine

The previous reports have indirectly showed that deletion of the gene leads to very high-level constitutive expression of *pyr* genes [2], and PyrR has been previously shown to act to regulate all three attenuators [2]. Our laboratories have analyzed carbamyl phosphate synthetase gene, aspartate aminotransferase, cytidine deaminase gene and homoserine dehydrogenase gene cytidine synthesis-related genes and cytidine accumulation at the early stage. It turned out that cytidine production has accumulated after metabolic block, indicating the metabolic pathways of cytidine and metabolic pathways of aspartic acid to homoserine block to the of cytidine is very necessary for the cytidine accumulation. Fixed point mutation cytidine synthesis from the de novo pathway of two key enzymes of cytidine accumulation is very effective, But accumulation level is low, which may be influenced by many

Q. Wu et al.

factors, series of cytidine synthesis enzyme gene transcription level lower the speed limit is the key factor. To solve this problem, we decided to try directed evolution and genetic recombination, focusing on the transformation of PyrR regulatory protein and attenuators of pyrimidine biosynthesis of *B. amyloliquefaciens*. We intend to adopt the homologous recombination to make the *pyrR* gene inactivation in *B. amyloliquefaciens*. RTFQ PCR is used to characterize the structural gene expression in transcription levels and then analyze the change in technology pyrimidine metabolism pathway protein levels by SDS-PAGE, analyzing the impact of pyrR gene inactivation pyrimidine operon transcriptional level.

68.5.2 Attenuations of pyr Operon Regulatory Region and Effects of Attenuator Gene Rearrangement on Excessive Biosynthesis of Pyrimidine

With further research that the interaction of the B. amyloliquefaciens regulatory protein PyrR with BL1, BL2 and BL3, which are located within the three different attenuation sites of pyr mRNA, has been analyzed in great detailing vitrousing gel mobility shift assays. Besides, It has been found that the apparent strength of the interaction between PyrR and the three different BLs was very different. PyrR bound significantly more tightly to the BLs of the second (BL2) and third (BL3) attenuation site than to the BL of the first (BL1) site. However, there has little reports about the influence of knocking out pyrR gene on expression of proteins in the bacterial cell about global responses of the bacterial cells affected by PyrR [21]. And still there has little reports about after knocking out pyrR gene and three attenuators like the following combinations (pyrR, pyrR, and BL in any one of pyrR and BL in any two arbitrary, pyrR and BLs), what it has impact on the level of transcription, and further effect on the cytidine synthesis. We intend to make the pyrR gene inactivation at first. After the pyrR gene inactivation and analysis of pyrimidine operon transcription level, we plan knockout attenuation in tandem to reconstruct attenuation regulatory region, even more monitor its impact on production of cytidine.

68.5.3 Blockade of Cytidine Catabolic Pathways

Aspartic acid is an important precursor in the de novo cytidine synthesis pathway of *B. amyloliquefaciens*. With carbamoyl phosphate in aspartate aminotransferase. Aspartate and carbamoyl phosphate catalyzed by aspartate aminotransferase produces carbamyl aspartic acid, and in the role of cytidine deaminase, cytidine further generate uridine. So we can delete cytidine deaminase gene to cut-off further metabolized to cytidine decomposition pathways uridine. There are studies indicating that knocking out of *cdd* gene can increase the production [12]. Therefore,

we plan to use homologous recombination to make the *cdd* gene inactivation of being transformation of the deficient strain and monitor the production of cytidine by fermentation.

68.6 Summary

After a series of ways to rearrange operon expression synthetic dose at the transcriptional level by monitoring the pyrimidine creatures, in order to get an impact on the cytidine through the gene modification of measures later. Doing that what we have imagined would increase or decrease the yield of cytidine and we do not know the results. Only through the experiment to draw conclusions and the process and results that is what we should strive to get. Through the modification of pyrimidine operon transcriptional regulatory regions in *B. amyloliquefaciens*, which may be relieve the negative regulation of transcription, increasing expression of gene dosage in the cells and promoting the rate-limiting step in the biochemical reactions, which may lead to increase production of cytidine. And, analyzing the change of the pyrimidine biosynthesis transcription level and structural gene expression can provide theoretical basis to enhance the cytidine production.

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656 Q. Wu et al.

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